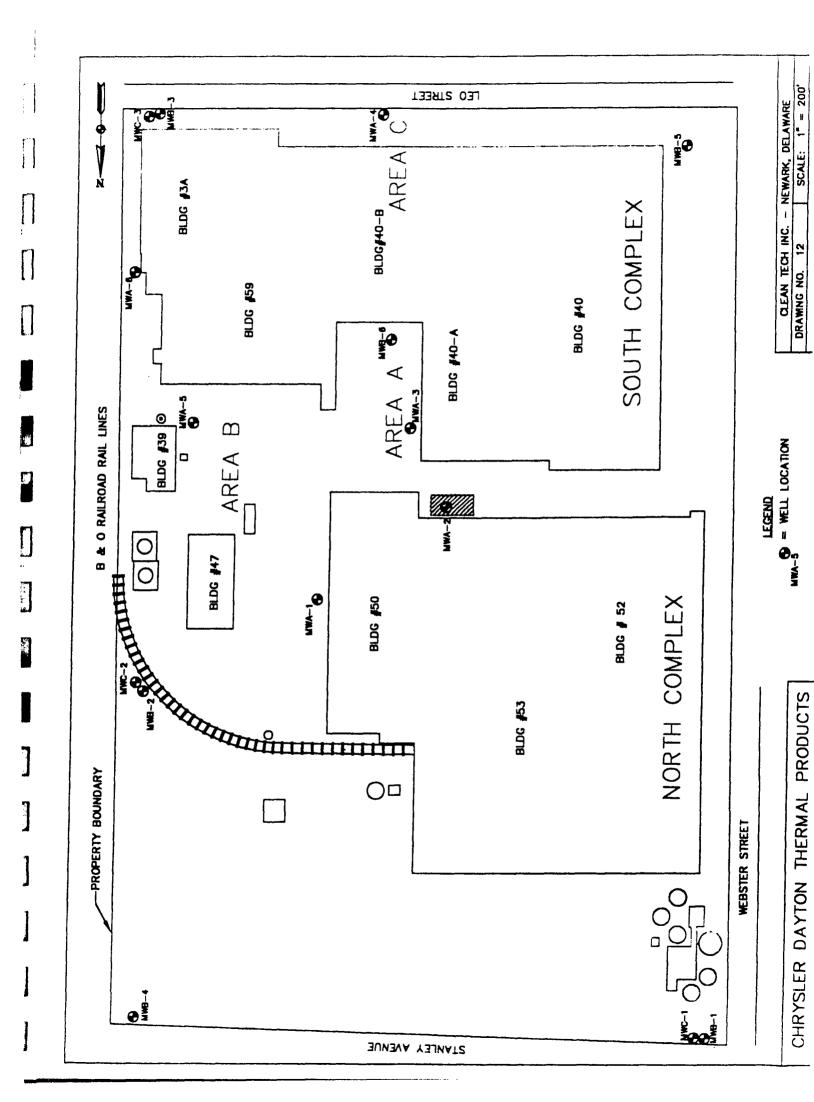
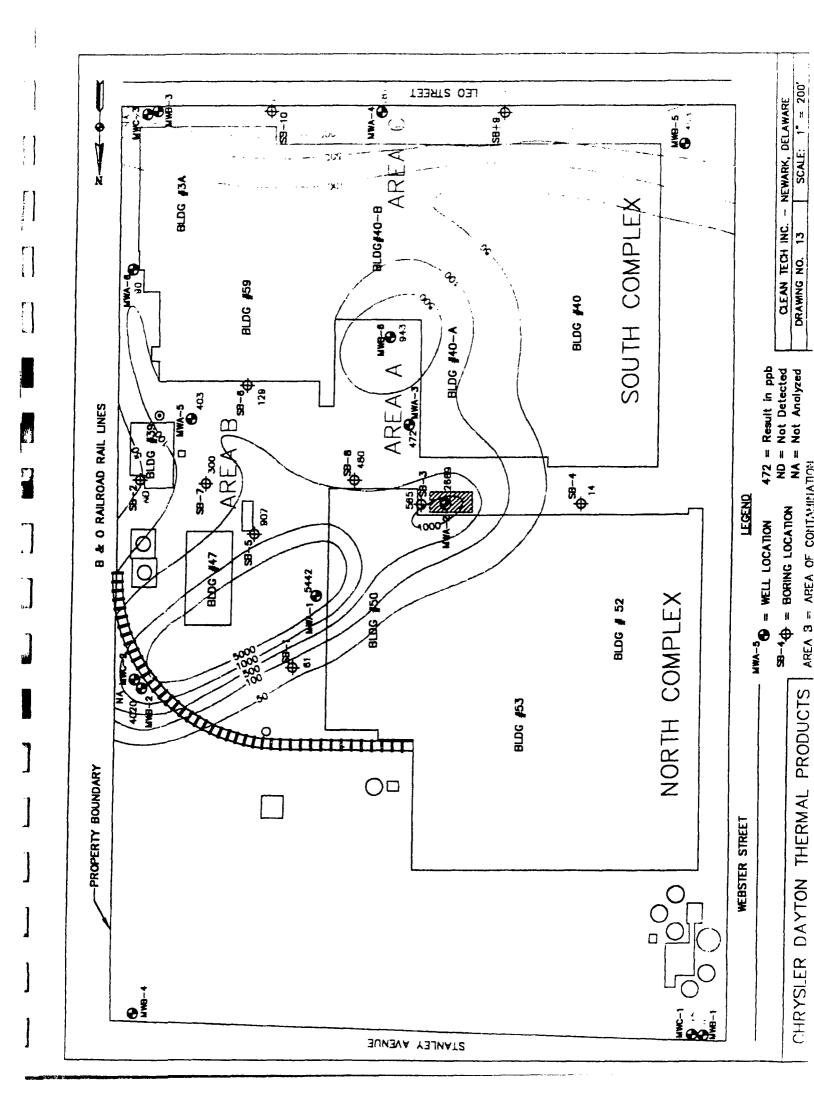


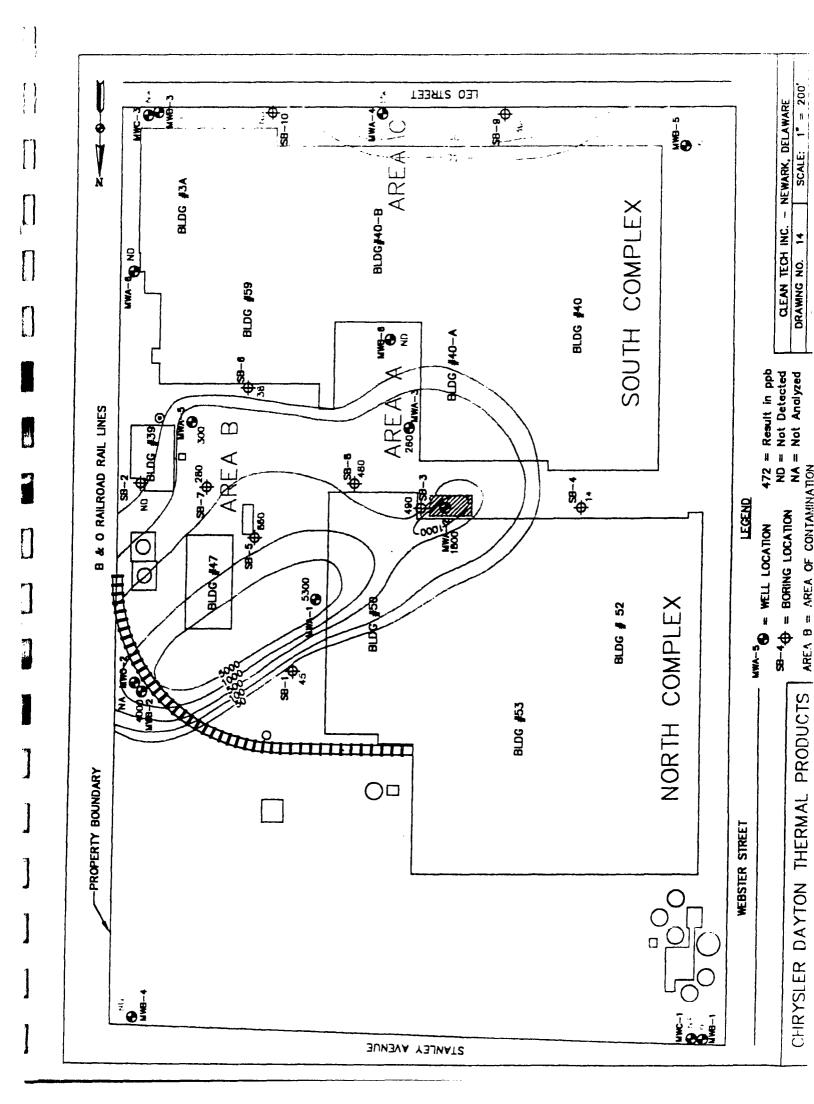
Groundwater Monitoring Well Locations
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



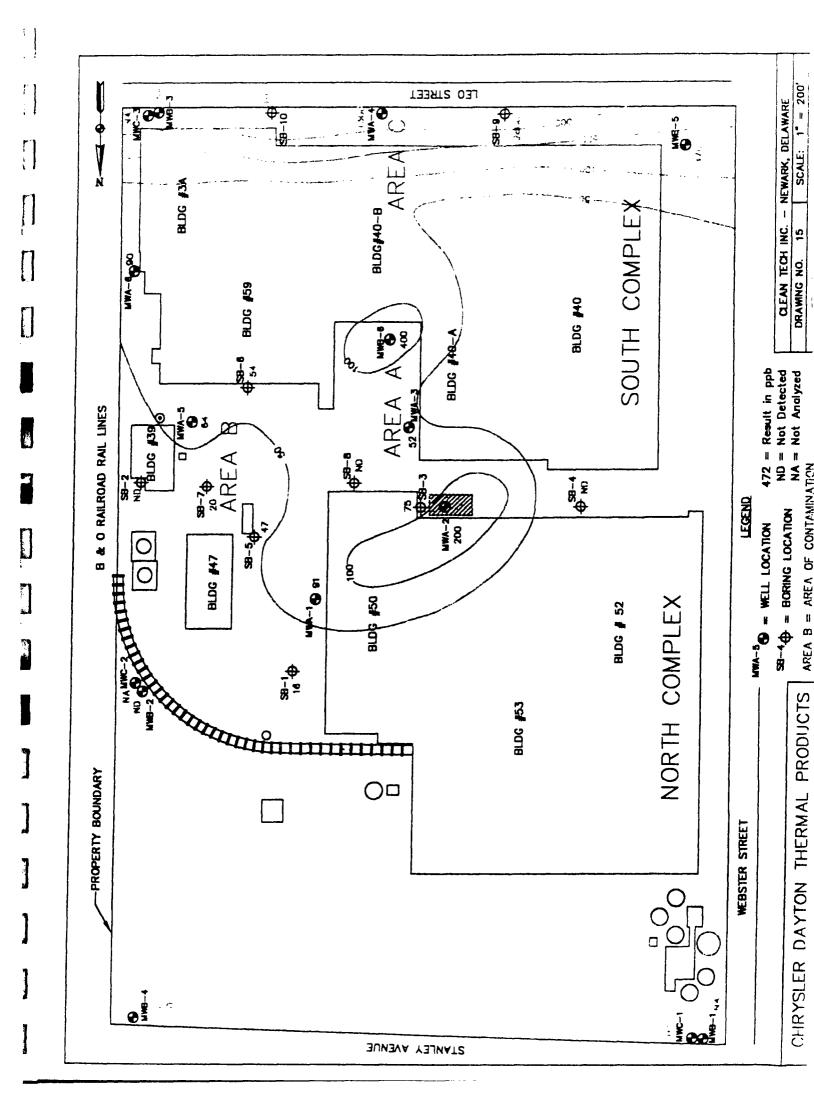
Soil Sample Results - Total VOCs
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



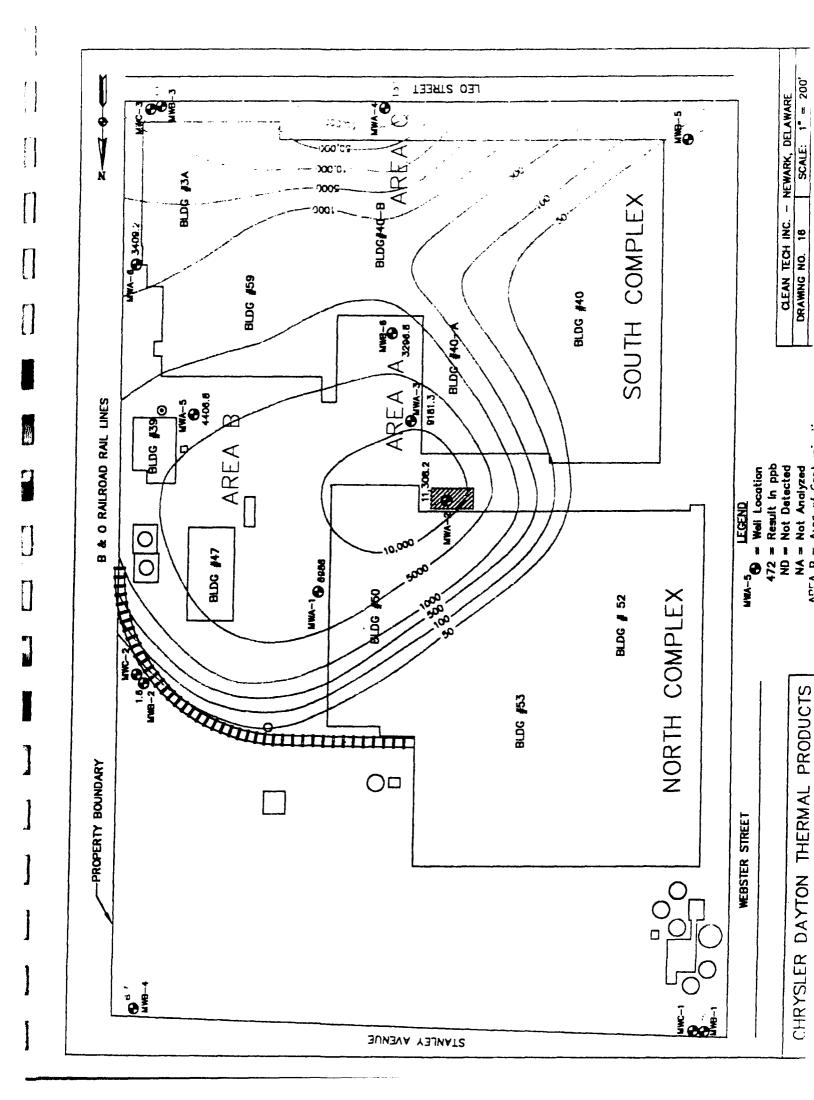
Soil Sample Results - Tetrachloroethylene
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



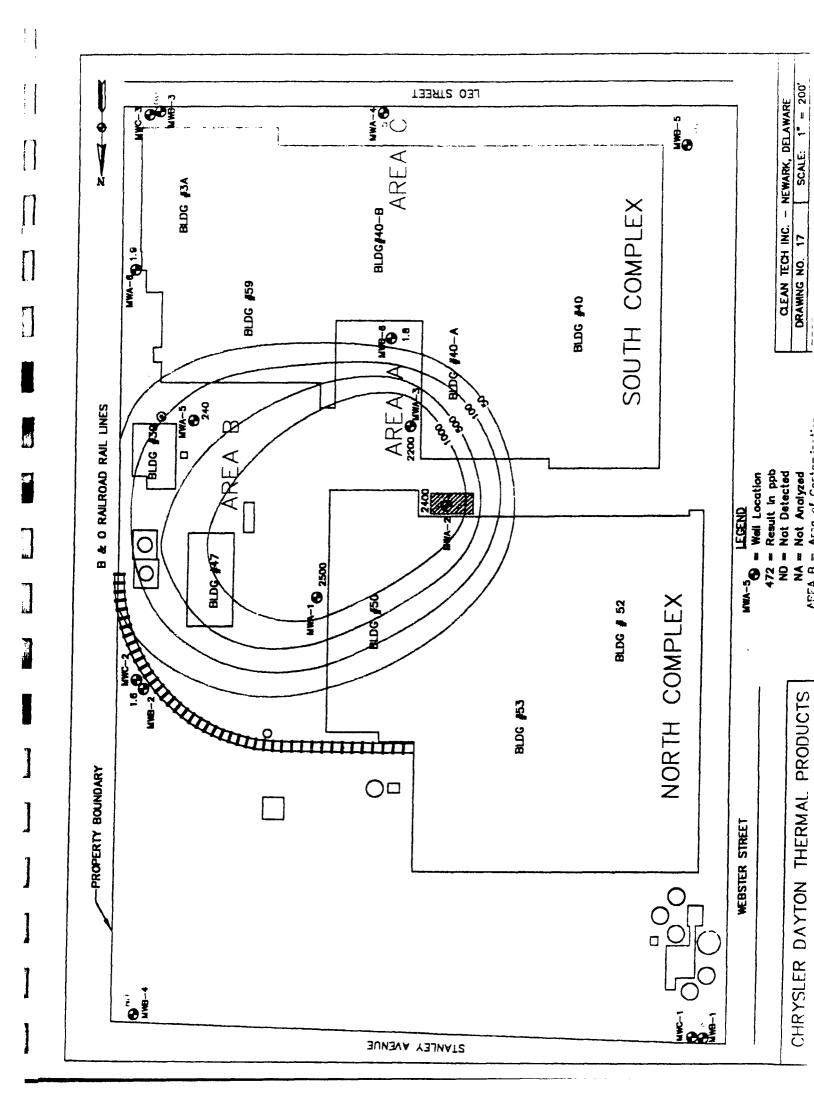
Soil Sample Results - Trichloroethene
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



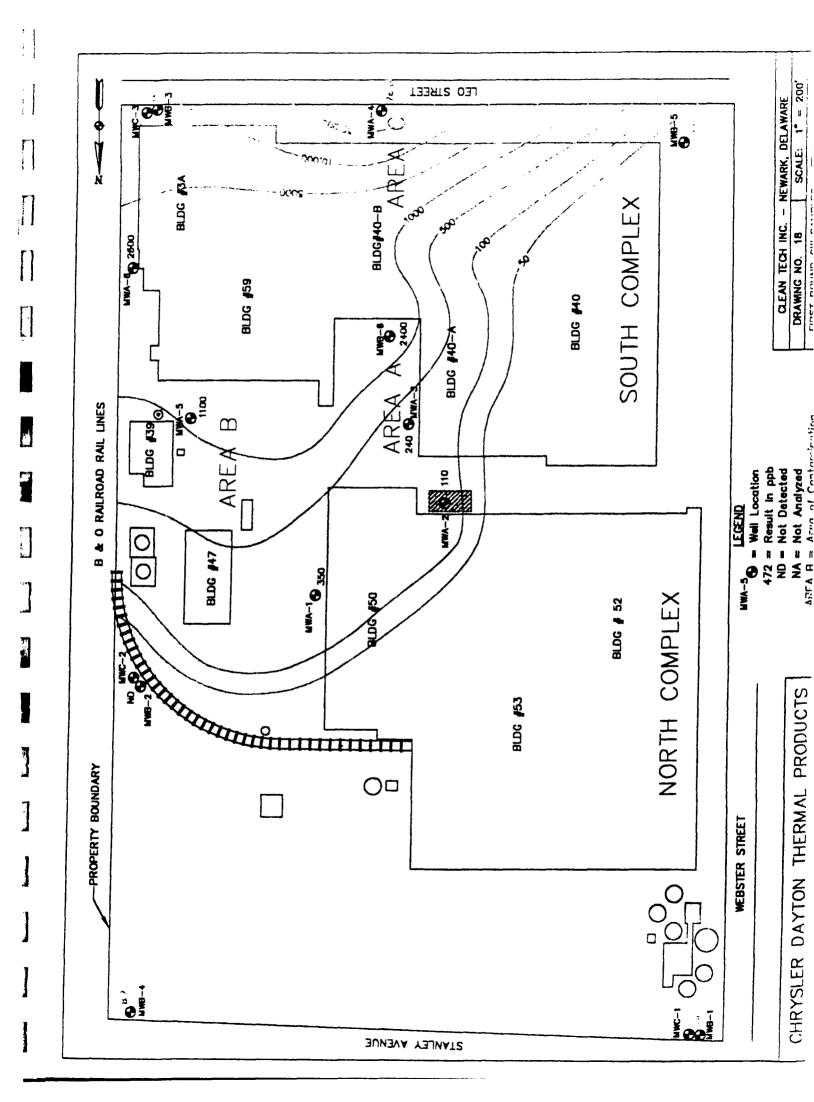
Round #1 Groundwater Results - Total VOCs
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



Round #1 Groundwater Results - Tetrachloroethylene
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



Round #1 Groundwater Results - Trichloroethene
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



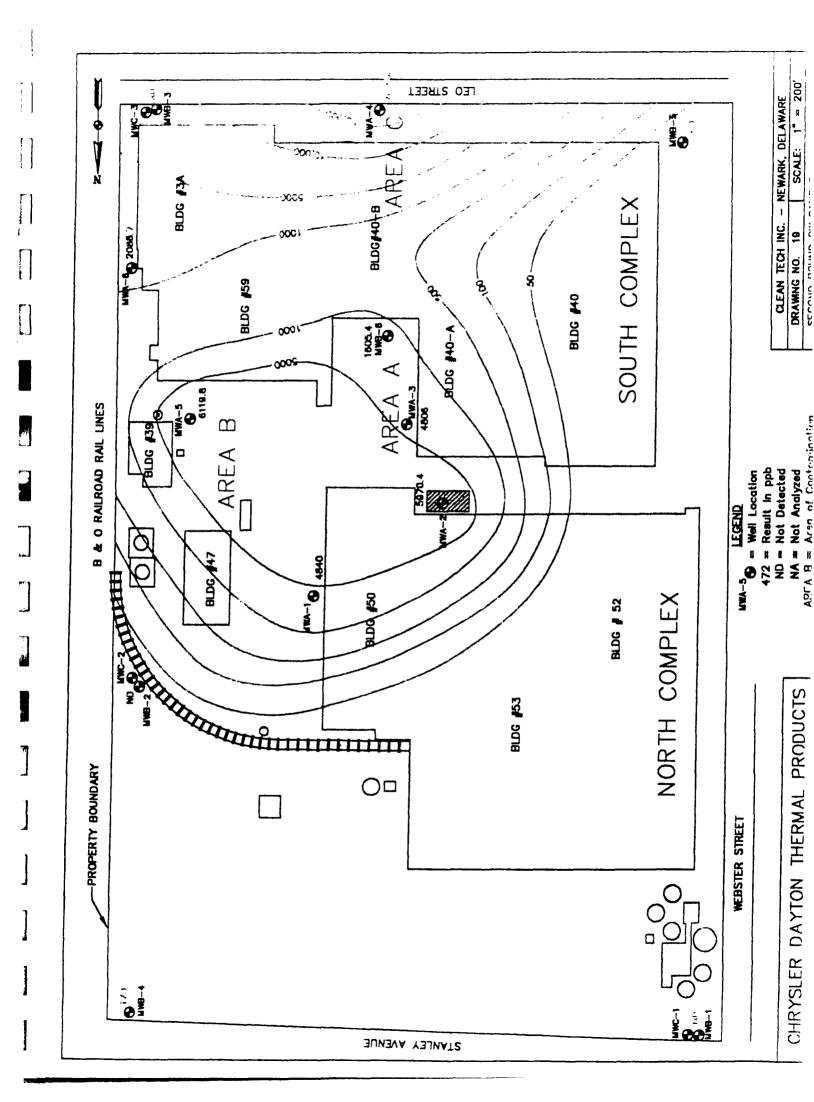
Round #2 Groundwater Results - Total VOCs

**Chrysler Corporation** 

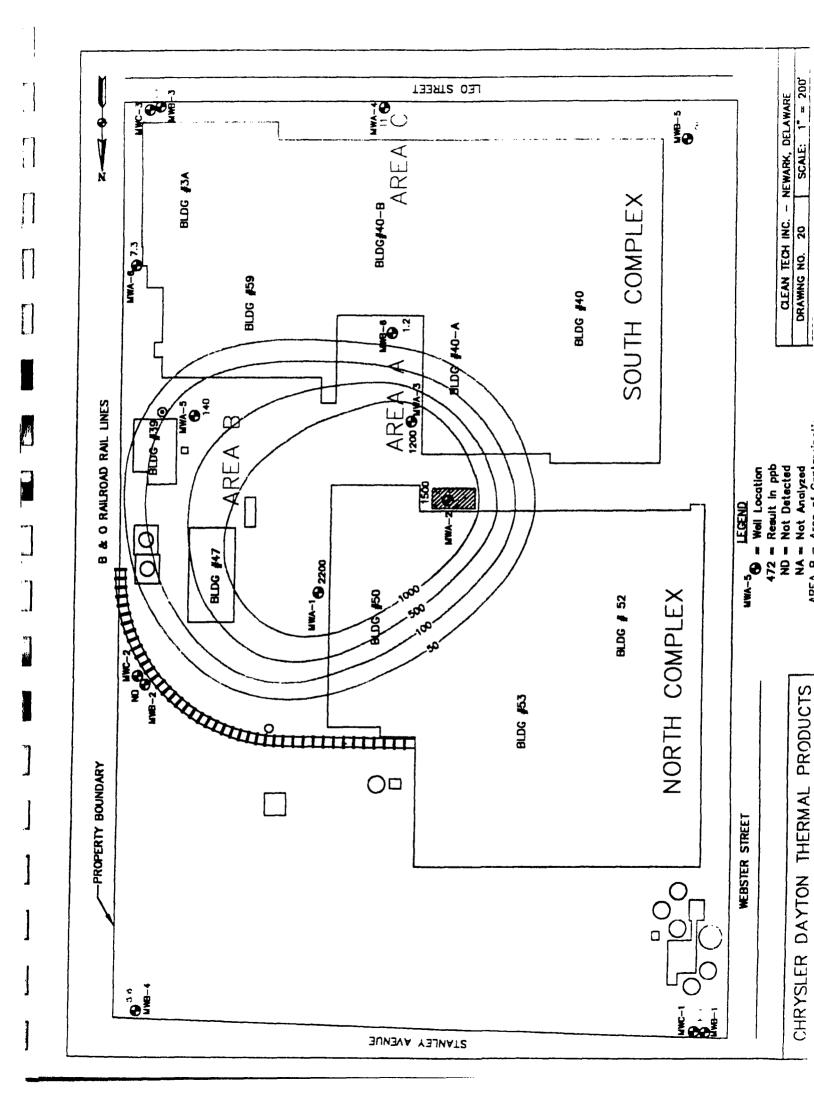
**Dayton Thermal Products Plant** 

1600 Webster Street

Dayton, Ohio 45404

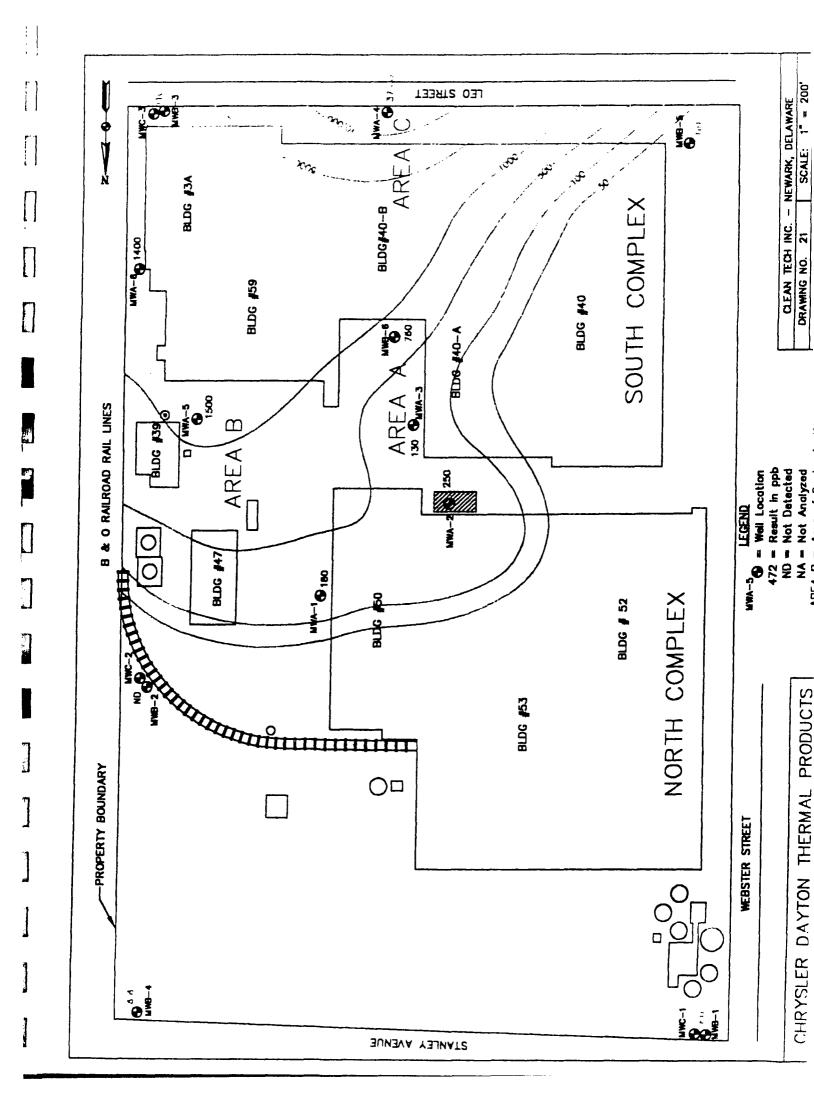


Round #2 Groundwater Results - Tetrachloroethylene
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



#### DRAWING 21.

Round 2 Groundwater Results - Trichloroethene
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



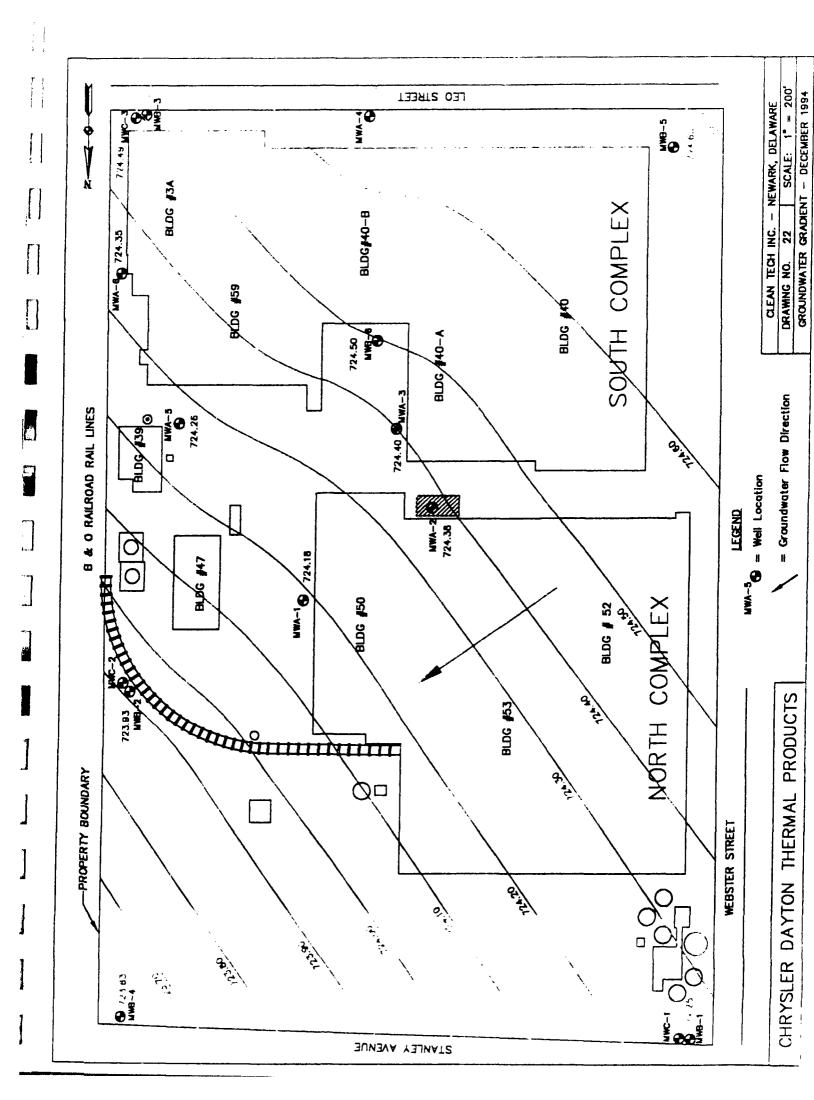
Groundwater Elevation Unconfined Aquifer - December 1994

Chrysler Corporation

Dayton Thermal Products Plant

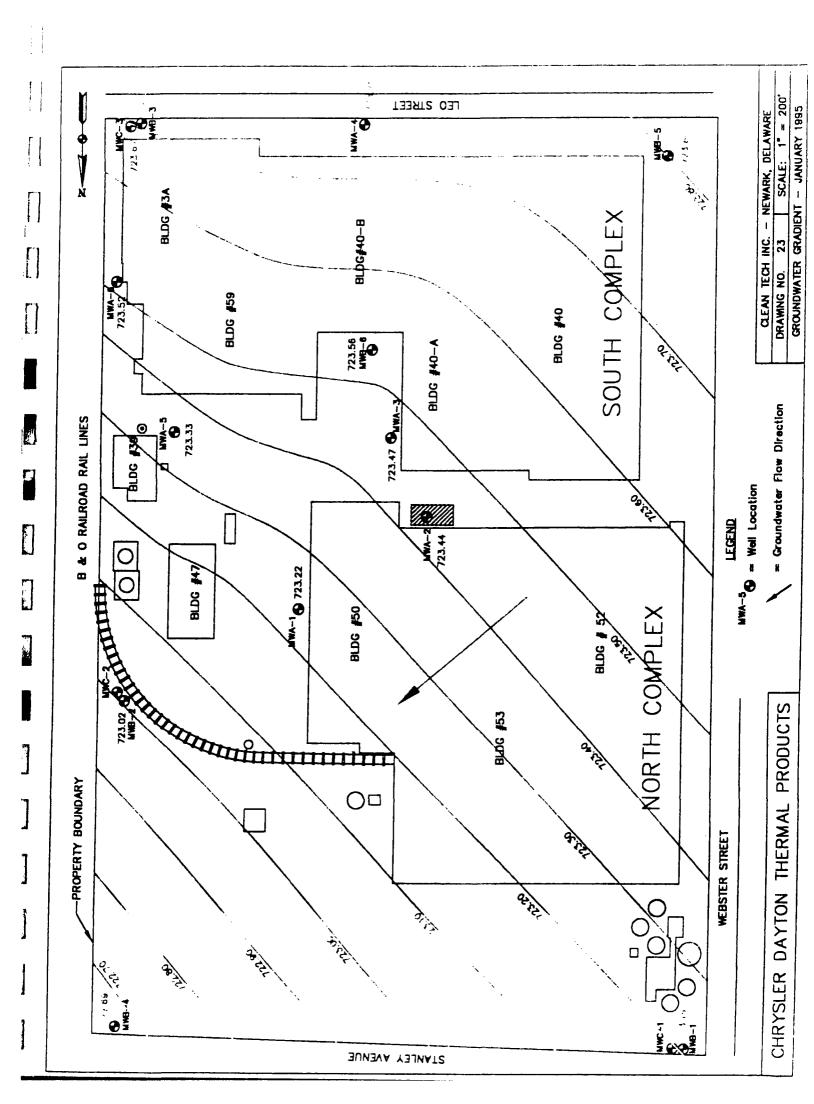
1600 Webster Street

Dayton, Ohio 45404

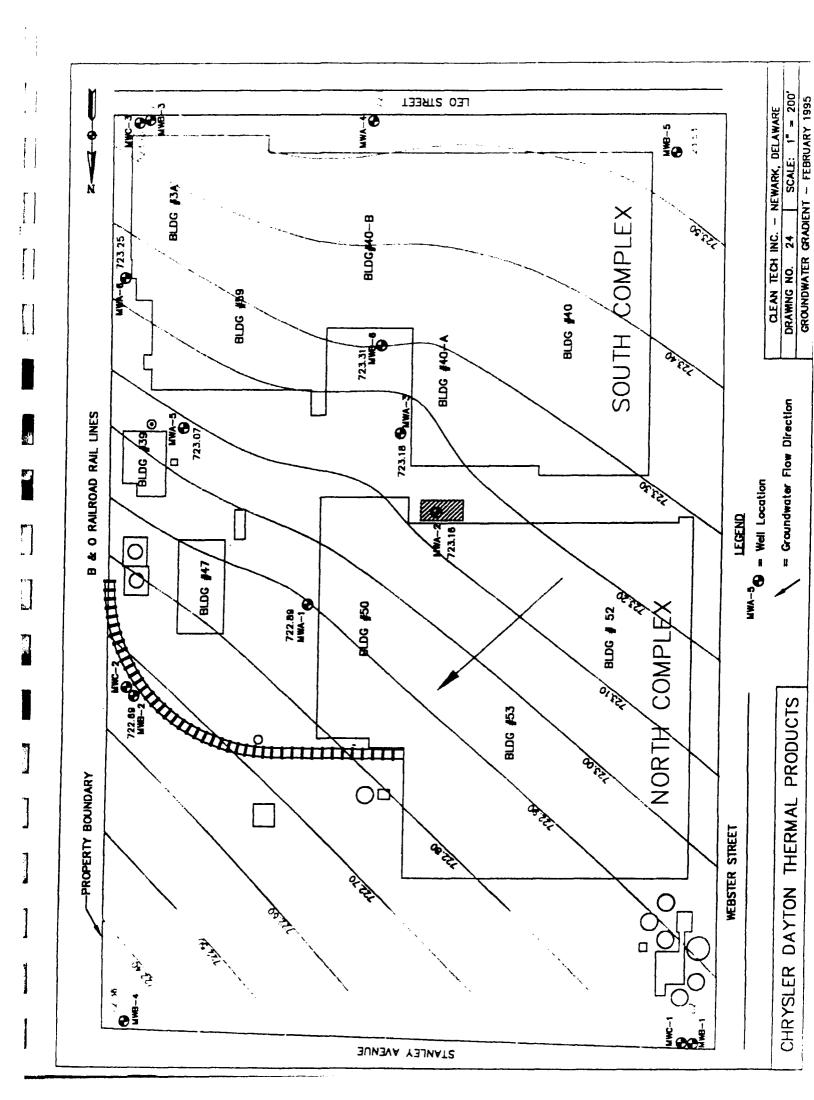


Groundwater Elevation Unconfined Aquifer - January 1995
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street

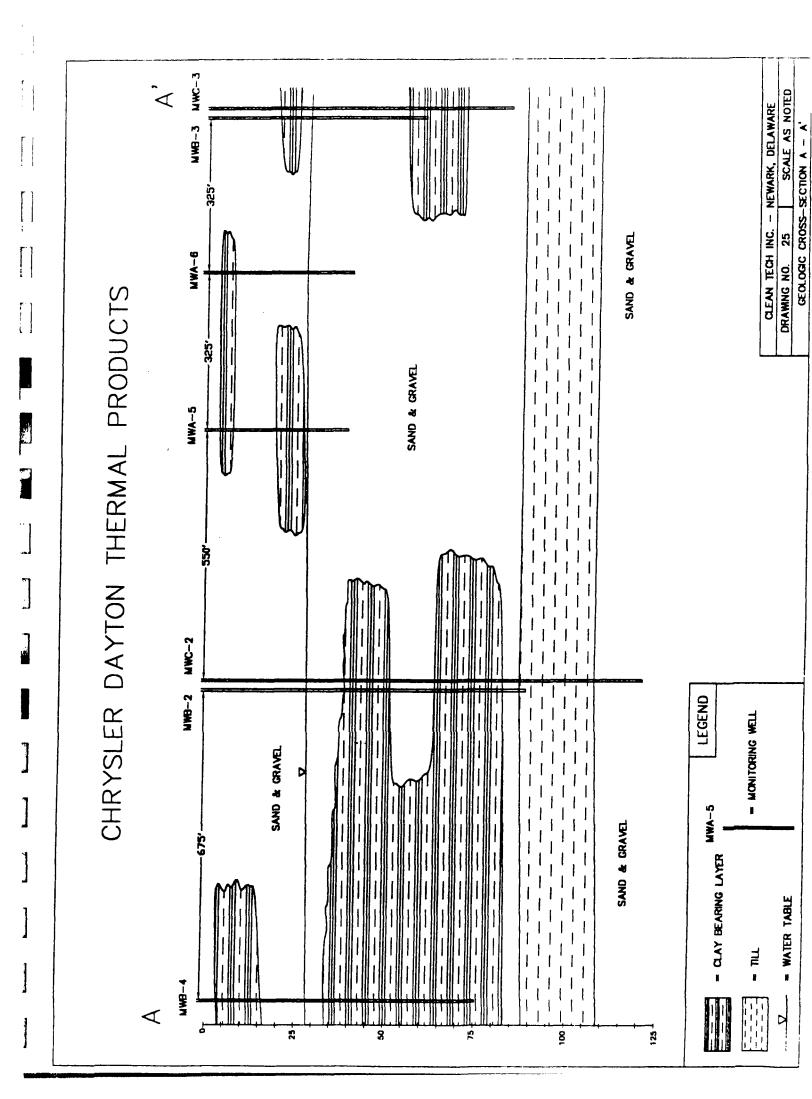
Dayton, Ohio 45404



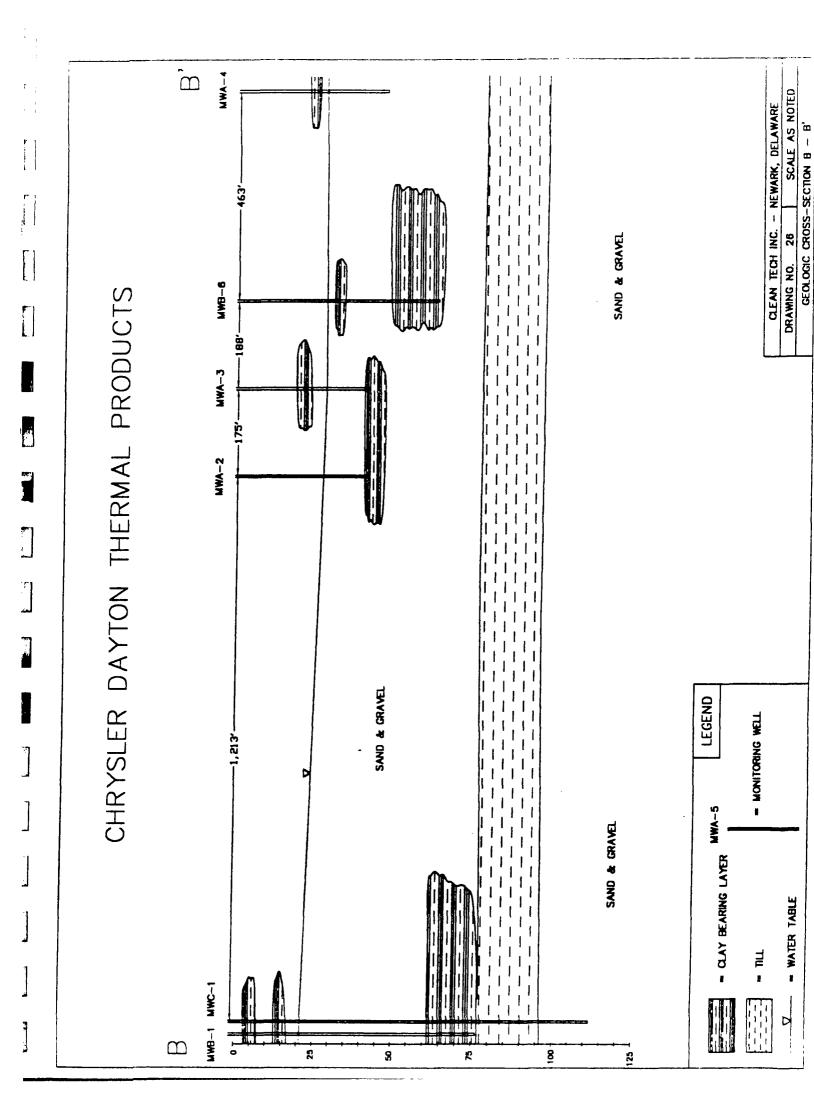
Groundwater Elevation Unconfined Aquifer - February 1995
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



Geologic Cross-Section A-A'
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404

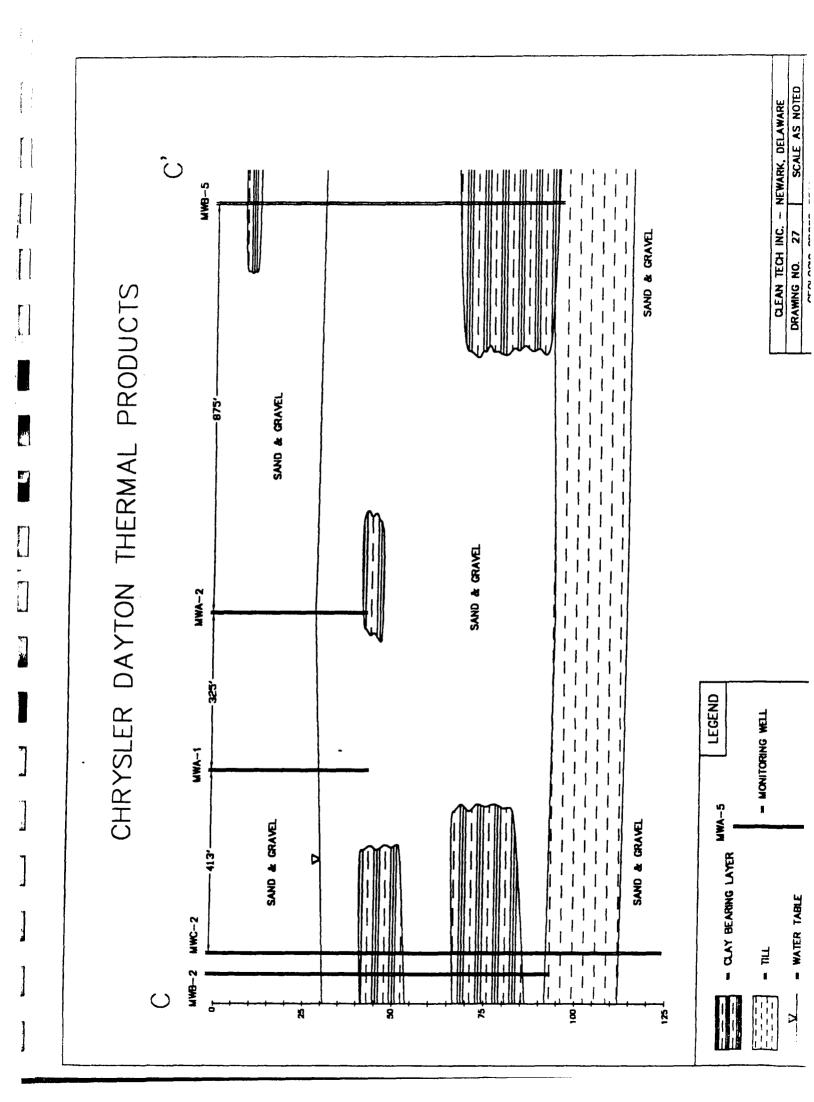


Geologic Cross-Section B-B'
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



DRAWING 27
Geologic Cross-Section C-C'
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street

Dayton, Ohio 45404





# SITE INVESTIGATION REPORT CHRYSLER CORPORATION DAYTON THERMAL PRODUCTS PLANT 1600 WEBSTER STREET DAYTON, OHIO 45404

Volume I of III
Report Text

#### **Prepared For**

Chrysler Corporation 800 Chrysler Drive CIMS 482-00-51 Auburn Hills, Michigan 48326-2757

#### Prepared By

Clean Tech 2700 Capitol Trail Newark, DE 19711 (302) 999-0924

September, 1995



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September 14, 1995

Mr. Curtis Chapman Chrysler Corporation 800 Chrysler Drive CIMS 482-00-51 Auburn Hills, MI 48326-2757

RE: Finalized Site Investigation Report

Chrysler Corporation Dayton Thermal Products Plant

Dayton, Ohio

Dear Mr. Chapman:

Enclosed please find the three volume finalized document <u>Site Investigation</u>, <u>Chrysler Corporation Dayton Thermal Products Plant</u>, <u>Dayton Ohio</u>. This submittal includes your review comments and requested report revisions. Comments received from Mr. Doug Orf are incorporated in this final submittal. This document has been forwarded to Mr. Orf per your request.

If you have any questions, please contact Clean Tech at (302) 999-0924.

Sincerely,

Steven W. Newsom, P.G.

Principal Geologist

**CLEAN TECH** 

Sincerely,

Deborah A. Buniski, P.E.

President

**CLEAN TECH** 

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CLEANTECH

September 14, 1995

Mr. Douglas J. Orf Chrysler Corporation Dayton Thermal Products Plant 1600 Webster Street Dayton, Ohio 45404

RE: Finalized Site Investigation Report

Chrysler Corporation Dayton Thermal P

Chrysler Corporation Dayton Thermal Products Plant Dayton, Ohio

Dear Mr. Orf:

Enclosed please find the three volume finalized document <u>Site Investigation</u>, <u>Chrysler Corporation Dayton Thermal Products Plant</u>, <u>Dayton Ohio</u>. This submittal includes comments and requested report revisions as received from you and Mr. Curtis Chapman. This document has been forwarded to Mr. Chapman.

If you have any questions, please contact Clean Tech at (302) 999-0924.

Sincerely,

Steven W. Newsom, P.G.

Principal Geologist

**CLEAN TECH** 

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Sincerely,

Deborah A. Buniski, P.E.

President

**CLEAN TECH** 

# Volume I of III Report Text Table of Contents

Section	Eva	cutive Summary	Page	
	LAC	cutive Summary		
1.0	Intro	oduction and Purpose	1	
2.0	Soil Vapor Survey			
	2.1	Sampling and Laboratory Methods	4	
	2.2	Sampling Locations	6	
	2.3	Findings	7	
		2.3.1 Contaminant Distribution Patterns	8	
	2.4	Discussion	11	
3.0	Soil	Borings	14	
	3.1	Soil Boring Locations	14	
	3.2	Installation Methods	15	
		3.2.1 Waste Disposal Methods	16	
	3.3	Soil Sampling and Analysis Methods	17	
		3.3.1 Chemical Analysis	18	
		3.3.2 Geotechnical Analysis	20	
4.0	Grou	Groundwater Monitoring Wells		
	4.1	Groundwater Monitoring Well Locations	23	
	4.2	Installation Methods	25	
		4.2.1 Waste Disposal Methods	30	
	4.3	Soil Sampling and Analysis Methods	32	
		4.3.1 Chemical Analysis	32	
		4.3.2 Geotechnical Analysis	36	
	4.4	Groundwater Sampling and Analysis Methods	37	
		4.4.1 Water Level Measurements	39	
5.0	Geology			
	5.1	Regional Geology	41	
	5.2	Site Geology	43	
6.0	Hydrogeology			
	6.1	Regional Hydrogeology	46	
	6.2	Site Hydrogeology	51	
		6.2.1 Unconfined Aquifer	52	
		6.2.2 Semi-Confined Aquifer	55	
		6.2.3 Vertical Flow Potential	56	

# Volume I of III Continued Report Text Table of Contents

Section		Page
7.0	Findings and Discussion for Soil Samples	58
	7.1 Chemical Analysis and Findings	58
	7.2 Geotechnical Analysis and Findings	69
8.0	Findings and Discussion for Groundwater Samples	75
	8.1 VOCs Analysis and Findings	76
	8.2 Metals Analysis and Findings	81
	8.3 Water Levels and Groundwater Flow	89
	8.4 Discussion of Findings	93
9.0	Interpretation of Contaminant Distribution Patterns	97
10.0	Targets for Soil and Groundwater Remediation	107

## **Executive Summary**

Clean Tech completed this site investigation at Chrysler Corporation's Dayton Thermal Products Plant located at 1600 Webster Street in Dayton, Ohio. The objectives were:

- Characterize the type and extent of contaminants in the unsaturated zone (above the water table) and saturated soil zones;
- Characterize the extent of dissolved phase contaminants in the groundwater;
- Assess the source of contaminants;
- Evaluate the potential for migration of contaminants off site;
- Obtain site data useful for evaluating remediation technologies;
- Evaluate potential for contamination due to dense non-aqueous phase liquids (DNAPL).

A review of existing information sources, a soil vapor survey, soil and groundwater sampling, and a hydrogeologic assessment permitted identification of three recognizable areas of the subject property having volatile organic compound (VOC) contamination. Groundwater and soil contamination by VOCs was documented with contaminant sources located in the central portion of the site near Building 53, and below Buildings 40A and 40B. Soil and groundwater contamination which originated from some off-site source to the south also appears to have impacted the subject site. VOC contamination appears to have entered the site from the south under the influence of an induced groundwater flow gradient originating at the Gem City Chemicals facility. Contaminants in groundwater have the potential to migrate off-site toward the Gem City facility.

No contamination by DNAPLs was observed. Groundwater contamination appears restricted to the shallow unconfined aquifer. The semi-confined aquifer does not appear to be affected by VOC contamination at this time. However, available information indicates the potential exists for groundwater to move downward from the unconfined aquifer to the semi-confined aquifer.

Solvents containing chlorinated organic compounds are interpreted to have entered the subsurface environment and penetrated to a depth near the base of the vadose (unsaturated soil) zone. Groundwater in the unconfined aquifer was brought in contact with the contaminated soil allowing contaminants to be released into the groundwater. Groundwater flow in the unconfined aquifer moved the groundwater toward the northeast under the influence of the steepening hydraulic gradient induced by the pumping well at Gem City Chemicals, Inc. The groundwater flowing past the contaminant sources acquired dissolved contaminants and carried the contaminants across the site toward the northeast forming the observed contamination plumes.

As groundwater moved toward the northeast carrying dissolved contaminants from the source locations, the soils in contact with the moving groundwater plumes absorbed some of the contaminants. This formed broad soil contamination plumes and may account for the similarity in location and pattern for both the soil contaminant and groundwater contaminant plumes. Seasonal fluctuations in water levels would be expected to exacerbate this situation over time. Under these conditions, the potential for off-site transport of contaminants is significant over time, first as dissolved groundwater contamination, and secondly as soil contamination near the base of the vadose zone.

The Ohio EPA currently seeks only to prevent significant contamination from reaching nearby public water supply wells through a Well Field Protection Program with Interim Action requirements. Interim Actions for groundwater are the only approved remedial actions which may be undertaken within the Well Field Protection Area. The need for groundwater gradient control is based Ohio EPA defined Interim Standards. If an Interim Standard for groundwater quality is exceeded, Ohio EPA will require the property owner control and remediate contaminated groundwater to prohibit it from leaving the effected property. Fourteen of the twenty-one VOCs detected at the subject site exceed the Interim Standards for those compounds.

# Section 1.0 - Introduction and Purpose

Clean Tech is pleased to present our report of findings for the Site Investigation completed at Chrysler Corporation's Dayton Thermal Products Plant (DTPP) in Dayton, Ohio. This plant is part of Chrysler Components, a division of Chrysler Corporation (Chrysler). The site is located at 1600 Webster Street in Dayton, Ohio as shown in Figure 1 (see Volume II of this report). The facility encompasses approximately 60 acres and contains over 1.3 million square feet under roof. Current operations at the facility include the manufacture, assembly, and finishing of heat exchangers and air conditioning components for motor vehicles. The facility consists of eight manufacturing buildings, a powerhouse, wastewater treatment plant, and associated storage buildings.

The facility is presently surrounded by the following industries: Brainerd Industries and Paint America Company on Webster Street, and American Lubricants and Gem City Chemical Company on Air City Avenue. There are several other industries and commercial operations in the near vicinity of the facility (DAP, Inc., Hohman Plating and Manufacturing, Gem City Stamping, Inc., RIS Paper Company, and Angell Manufacturing Company) in addition to private residences. A facility map which provides further details of the site including buildings and other operations is included as Figure 2 (see Volume II of this report).

Past operations at the site prior to Chrysler's acquisition in 1936 included the assembly of Maxwell automobiles from about 1907 through 1936, and other manufacturing processes such as furnaces, gun parts, aluminum and copper tube forming operations, light machining, plating, metal stamping, welding, soldering, degreasing, painting, plastic molding and assembly, as well as maintenance of these processes, equipment and structures. The Maxwell Complex, which was a group of twelve former buildings, was used by Chrysler until 1990 when it was demolished. The Maxwell Complex had been used primarily for storage purposes during the ten to twelve years prior to demolition. A

portion of the Maxwell Complex footprint was replaced by the new manufacturing Building 59 in 1991. Investigations completed during the demolition of the Maxwell Complex buildings (prior to the construction of Building 59) indicated that the site may have been impacted by historical manufacturing activities.

This site investigation was completed on behalf of Chrysler by Clean Tech of Newark, Delaware. Specifically, the objectives of the site investigation were as follows:

- Characterize the type and extent of contaminants in the vadose and saturated zones;
- Characterize the extent of dissolved phase contaminants in the groundwater;
- Assess the source of contaminants;
- Evaluate the potential for migration of contaminants off site;
- Obtain site data useful for evaluating remediation technologies; and
- Evaluate potential for contamination due to dense non-aqueous phase liquids (DNAPL).

The site investigation included the following:

- A review of existing information was conducted using sources which included aerial photographs (see Attachment A in Volume II), Sanborn maps, previous reports of limited investigations for the site, and documents describing investigations and remediation at nearby properties;
- A soil vapor survey was conducted to evaluate volatile organic compounds (VOCs) above the water table (vadose zone) for two distinct soil depth zones across the entire DTPP property using an on-site mobile laboratory;
- Soil borings were installed and subsurface soils were described and sampled for laboratory analysis of VOCs and selected metals. The installation of the soil

borings was completed using the results of the soil vapor survey as guidance for boring placement and selection of depth intervals for sampling;

- Monitoring wells were installed and the subsurface soils were described and sampled for laboratory analysis of VOCs and selected metals. The wells were screened at two distinct depths within the unconfined (water table) aquifer, and in the semi-confined aquifer. This was done to determine the nature and extent of groundwater contaminants. Two rounds of groundwater samples were collected for laboratory analysis of VOCs and selected metals;
- Three rounds of water level measurements were collected using the groundwater monitoring wells. This information was used to determine groundwater flow directions across the site with seasonal water level fluctuations noted.

The geophysical logging of an existing on-site water supply well was deleted from the scope of work. The geophysical log was proposed as a method to determine the depth to the confining clay layer separating the water table aquifer and the underlying semiconfined aquifer. This information was needed for the design and installation of the groundwater monitoring wells. The geophysical log became unnecessary since the needed information was obtained through additional research in existing data sources, and through discussions with the subcontracted well installation firm. The well installation firm used for the groundwater monitoring well installations, Moody's of Dayton, is a local business with many years of experience in the Dayton area and is familiar with drilling conditions in and near the site.

Slug testing of the groundwater monitoring wells was attempted, but provided minimal information. The aquifer conditions as encountered in both the unconfined and semi-confined aquifers (extreme permeability) made the slug testing of minimal use as a means of determining representative aquifer characteristics. The slug test findings are included in the report but will not be used in the overall analysis of site conditions.

# Section 2.0 - Soil Vapor Survey

The initial field task for the site investigation was the completion of a facility-wide soil vapor survey. The objectives of the soil vapor survey were to:

- Determine the areal extent of contamination by VOCs in the vadose zone;
- Confirm soil vapor survey results from a previous site investigation noting any correlation between elevated readings and potential sources of contaminants;
- Provide a basis for placement of soil borings and groundwater monitoring wells;
- Provide a qualitative evaluation of the potential of using soil venting to remediate vadose zone soils.

# 2.1 - Sampling and Laboratory Methods

On October 9, 1994 Clean Tech mobilized on-site to begin the soil vapor survey of the site. A Work Plan had been prepared (dated August, 1994) and submitted to Chrysler prior to mobilization. A total of thirty (30) soil vapor sample locations were proposed in the Work Plan. Soil vapor samples were to have been collected from three to eight feet below the ground surface at each location using a hydraulic hammer. The Clean Tech mobile laboratory was also mobilized on-site to provide accurate lab-quality data and rapid analysis of the collected samples.

Clean Tech altered the work plan through the use of a truck mounted Geoprobe subsurface sampling rig. The Geoprobe unit was utilized in order to collect more samples and to provide a better use of manpower and the mobile laboratory. The Geoprobe unit is capable of sampling at greater depths and more quickly than a hydraulic hammer.

The Geoprobe unit was mobilized to sample soil locations on October 10, 1994. The Geoprobe unit drives and withdraws a soil vapor sampling probe. By using Geoprobe, Clean Tech was able to collect soil vapor samples at multiple depths at each location. This

soil vapor survey method generated data which assisted in the understanding of the soil vapor contaminant distribution in the vadose zone.

Geoprobe equipment consists of a truck mounted hydraulic probe unit which drives four foot long sections of hollow metal pipe into the subsurface. The metal pipe is approximately one inch in diameter. A new drive point was used for each location. After the metal pipe was driven to the selected depth, Teflon tubing was inserted through the metal pipe and the surface connections were sealed to prevent the entry of atmospheric air. The probe was then extracted approximately one foot to create an open space at the bottom of the driven hole from which a soil vapor sample was collected.

Approximately fifteen to twenty liters of air were evacuated from the open space at the bottom of each driven hole prior to sample collection using a vacuum pump. A vacuum of approximately fifteen to twenty inches of mercury was applied through the Teflon tubing, and a soil vapor sample of approximately five ml was collected in a gas syringe or Tedlar bag. The sample was immediately brought to the Clean Tech mobile laboratory on-site and directly injected into the gas chromatograph (GC).

All samples were analyzed the same day they were collected. Analysis of the soil vapor samples was performed using modified EPA Method 601. The method detection level was 2.0 parts per billion (ppb).

Sample integrity was maintained through quality control procedures completed prior to, during and after sampling and analysis. New Teflon tubing was used for each sample then discarded. All soil vapor sampling equipment in contact with the sample or subsurface was decontaminated using an alconox solution (non-phosphate detergent) followed by a deionized water rinse. Decontamination protocol originally included a methanol rinse but this was found to cause interference in the GC. The use of methanol was therefore discontinued after the first day.

The GC calibration procedures included a system blank and performance of an external calibration curve using known analyte standards at the start of each day prior to analysis of any samples. Ambient air samples were periodically collected during each day through the soil vapor probe, and analyzed to ensure the effectiveness of decontamination procedures. One ambient air blank was analyzed following collection and analysis of every ten soil vapor samples.

Duplicate soil vapor samples were also analyzed as a quality assurance/quality control measure. One duplicate sample was analyzed following collection and analysis of every twenty soil vapor samples. GC calibration was performed at the end of each day following completion of the analysis of all soil vapor samples using a system blank.

# 2.2 - Sampling Locations

A total of 86 soil vapor samples were collected for analysis from 44 locations across the DTPP property. Daily quality checks of data allowed continuous quality control as the soil vapor survey progressed. Two soil vapor samples, one shallow sample and one deep sample, were collected (or attempted) at each sampling location. The shallow soil vapor sample was collected from approximately ten feet below local grade. The deep soil vapor sample was collected from approximately twenty feet below local grade. Groundwater was typically encountered approximately twenty-five feet below grade. The soil probe occasionally met refusal when advanced, or encountered groundwater shallower than twenty feet, causing an adjustment in sampling depth. The actual sampling depths with a description of each sample location are included in Attachment B (see Volume II of this report).

Soil vapor samples from locations 1 through 20 were collected and analyzed as a reconnaissance sampling effort focused in the eastern and central portions of the facility. The sample locations were relatively closely spaced in the eastern and central portions of the property as can be seen in Drawing 1 (Site Plan) and Drawing 2 (Sample Locations 1)

through 48). All drawings are contained in Volume II of this report. These initial soil vapor sample locations served to highlight those areas where focused soil vapor sample collection would provide the best data for accurate determination of contaminant distribution across the entire facility.

The initial samples from locations 1 through 20 were not directly incorporated as contoured data in the soil vapor isoconcentration maps (Drawings 3 through 10) because the data collected from locations 21 through 48 (a total of 49 samples collected from 25 locations) provided good areal coverage of the property. The reconnaissance data and the contoured data qualitatively agreed and in many instances were quite closely spaced in portions of the site. This situation provided both an internal quality check of the data used in the isoconcentration maps, and guidance in establishing the contoured pattern for the data. The soil vapor survey results including: sample numbers, sample locations, sample depths, and the amounts of detected compounds are presented in Attachment C (see Volume II of this report).

# 2.3 - Findings

The laboratory analysis determined that there were primarily eight volatile organic compounds present in the soil vapor samples, as had been identified in previous investigations. These compounds are listed below with the number of shallow and deep soil vapor samples found to contain the compound above the method detection level. Refer to Attachment C for a summary of the soil vapor results.

Target Compound	Shallow Samples (10')	Deep Samples (20')
1,1,1-trichloroethane (TCA)	16	17
tetrachloroethene (PCE)	22	23
vinyl chloride	14	16
1,1-dichloroethene	l	2
cis-1,2-dichloroethene	3	10
1,2-dichloroethane	5	7

Target Compound	Shallow Samples (10')	Deep Samples (20')
1,1,2-trichloroethane	5	6
1,1-dichloroethane	0	0

Eight isoconcentration contour maps were generated for the soil vapor survey findings based on the available data. Maps were prepared showing: Total VOCs (the sum of all eight compounds), TCA, PCE and vinyl chloride for the shallow soil vapor data, and for the deep soil vapor data. These maps are included as Drawings 3 through 10.

#### 2.3.1 - Contaminant Distribution Patterns

The distribution patterns were examined for each mapped contaminant as they were presented through the isoconcentration contour maps. Shallow and deep vadose zone distribution patterns were developed. These patterns of contaminant distribution were described as follows.

### Total VOCs - Shallow Vadose Zone - Drawing 3

The isoconcentration map for total VOCs in the shallow vadose zone was contoured using 50 ppb and 100 ppb contours. The isoconcentration map shows elevated levels of total VOCs were detected in the following areas:

- Within the northern portion of Buildings 40A and 40B, and in the paved area immediately northeast of those buildings;
- In the area of the former TCA tank south of Building 53;
- To the north and east of Building 59;
- In the area near Building 47 extending eastward toward the boiler house;
- In the area north of the boiler house and northeast of Building 47;
- In an isolated area south of Building 59 and west of Building 3A;
- In an isolated area northeast of the wastewater treatment plant.

### Total VOCs - Deep Vadose Zone - Drawing 4

The isoconcentration map for total VOCs in the deep vadose zone was contoured using 50 ppb, 100 ppb, and 1,000 ppb contours. The isoconcentration map shows elevated levels of total VOCs were detected in the following areas:

- Within the northern portion of Buildings 40A and 40B, and in the paved area immediately northeast of those buildings;
- In the area of the former TCA tank south of Buildings 50 and 53;
- To the north of Building 59 extending across the vicinity of Building 47;
- In an isolated area south of Building 59 and west of Building 3A;
- In an isolated area south of Building 40 and 40A along Leo Street.

### TCA - Shallow Vadose Zone - Drawing 5

The isoconcentration map for TCA in the shallow vadose zone was contoured using 50 ppb and 100 ppb contours. The isoconcentration map shows elevated levels of TCA were detected in the following areas:

- Within the northern portion of Building 40A and in the paved area immediately northeast of that building;
- In the area of the former TCA tank south of Buildings 50 and 53;
- To the south of Building 47 near the waste storage area.

### TCA - Deep Vadose Zone - Drawing 6

The isoconcentration map for TCA in the deep vadose zone was contoured using 50 ppb, 100 ppb, and 1,000 ppb contours. The isoconcentration map shows elevated levels of TCA were detected in the following areas:

 Within the northern portion of Building 40A and in the paved area immediately northeast of that building;

- In the area of the former TCA tank south of Buildings 50 and 53;
- In an area extending from north of Building 59 across the south of Building 47 near the waste storage area, and extending to the east near the boiler house.

### PCE - Shallow Vadose Zone - Drawing 7

The isoconcentration map for PCE in the shallow vadose zone was contoured using 50 ppb, and 100 ppb contours. The isoconcentration map shows elevated levels of PCE were detected in the following areas:

- In an isolated area south of Building 59 and west of Building 3A;
- Within the northern portion of Building 40A and in the paved area immediately northeast of that building and south of Building 50.

### PCE - Deep Vadose Zone - Drawing 8

The isoconcentration map for PCE in the deep vadose zone was contoured using 50 ppb, 100 ppb, and 1,000 ppb contours. The isoconcentration map shows elevated levels of PCE were detected in the following areas:

- In an isolated area south of Building 40 and 40A along Leo Street;
- In the area of the former TCA tank south of Buildings 50 and 53;
- In an isolated area south of Building 59 and west of Building 3A;
- In an area immediately to the north of Building 47;
- Within the northern portion of Building 40A and in the paved area immediately northeast of that building extending north of Building 59.

# Vinyl Chloride - Shallow Vadose Zone - Drawing 9

The isoconcentration map for vinyl chloride in the shallow vadose zone was contoured using 50 ppb and 100 ppb contours. The isoconcentration map shows elevated levels of vinyl chloride were detected in the following areas:

- In an isolated area south of Building 59 and west of Building 3A;
- In an isolated area immediately to the southeast of Building 47.

### Vinyl Chloride - Deep Vadose Zone - Drawing 10

The isoconcentration map for vinyl chloride in the deep vadose zone was contoured using 50 ppb and 100 ppb contours. The isoconcentration map shows elevated levels of vinyl chloride were detected in the area immediately west of Building 47 near the waste storage area.

#### 2.4 - Discussion

The soil vapor survey revealed the following patterns of contamination in the vadose zone across the DTPP property:

- VOC contamination levels in the vadose zone appear to be greatest in the central portion of the facility in the area to the north of Buildings 40A and 40B, and to the south of Buildings 50 and 53 (former TCA tank area).
   This pattern was found in both the shallow and deep portions of the vadose zone;
- VOC contamination in the vadose zone was noted at a lesser magnitude yet
  extends across a larger portion of the facility from north of Building 59 to
  the area of Building 47 and the associated waste storage area. This pattern
  was noted for both the shallow and deep portions of the vadose zone, but
  is much more pronounced in the deep vadose zone;
- Isolated areas of significantly elevated VOCs were noted in the southern portion of the site to the west of Building 3A and south of Building 59, and in the area to the south of Buildings 40 and 40A. This pattern was noted for both the shallow and deep portions of the vadose zone, but was found to be much more pronounced in the deep vadose zone;

 Larger amounts of VOC contamination with greater areal extent of VOC contamination were noted in the deep vadose zone as compared to the shallow vadose zone.

These findings are in close agreement with the work completed during previous soil investigations at the DTPP facility. The areas near Buildings 40A and 40B, the area to the south of Building 53 near the former TCA tanks, the area east of Building 50, and the western and southern portions of the former Maxwell Complex are identified as areas where elevated levels of VOCs may be expected in vadose zone soils.

The soil vapor survey permitted identification of recognizable areas of the DTPP property having a particular pattern of VOC contamination in the vadose zone. These areas were evaluated and are presented as reference areas for discussion of a working model of the site conditions. This model is presented for use in discussions of soil and groundwater contamination patterns, and identification of potential contamination sources. Refer to Figure 3 (see Volume II of this report) for a map of the facility showing these areas.

#### Area A

Area A was characterized as the central portion of the facility in the area to the north of Buildings 40A and 40B, and to the south of Buildings 50 and 53 (former TCA tank area). Area A exhibits a pattern of significantly elevated levels of total VOCs, TCA and PCE in both the shallow and deep portions of the vadose zone.

#### Area B

Area B was characterized as the east-central portion of the facility from the northern limit of Building 59 northward across the area of Building 47 and the associated waste storage area. Overall VOC contamination in the vadose zone appeared at a lesser magnitude in Area B than in Area A, but extended across a larger portion of the property in Area B. VOC contamination in Area B was noted for both the shallow and deep portions of the vadose zone, but was found to be more pronounced in the deep vadose zone. Area B

exhibits a pattern of significantly elevated levels of total VOCs in both the shallow and deep vadose zones, TCA in the deep vadose zone, and PCE in the deep vadose zone.

#### Area C

Area C was characterized as isolated areas of significantly elevated VOCs in the southern portion of the site to the west of Building 3A, south of Building 59, and south of Buildings 40 and 40A. Area C exhibited a pattern of significantly elevated levels of total VOCs and PCE in both the shallow and deep portions of the vadose zone, with elevated levels of vinyl chloride in the shallow vadose zone to the west of Building 3A. VOC contamination was noted for both the shallow and deep portions of the vadose zone, but was much more pronounced in the deep vadose zone.

Significantly elevated levels of VOCs in the deep vadose zone across the property in close proximity to the local water table clearly suggested groundwater contamination may have occurred. The soil vapor survey results provided a guide for placement of the soil borings and groundwater monitoring wells, and a qualitative evaluation showing the potential exists for using soil venting to remediate vadose zone soils.

# Section 3.0 - Soil Borings

The purpose of the soil borings was to observe and describe site stratigraphy, sample the vadose soil zone and analyze soils for the contaminants of concern, and determine the extent of contaminated soils in the vadose zone at the site. The selection of the soil boring locations, creation of a work plan for soil sample collection, and the selection of the target analytes for laboratory analysis was based on the review of available information, and information received from Chrysler Corporation environmental staff.

The results of previous investigations indicated vadose zone soils have been impacted by TCE, TCA, PCE and some heavy metal contamination (chromium and lead). The areas which may have been impacted include:

- Building 40B in the area of the former Freon-113 degreaser station;
- South side of Building 53 in the area of the former TCA storage tanks;
- Buildings 40A and 40B which contained former parts degreasers;
- West and southwest of the former Maxwell Complex or present Building 59;
- Storage areas located east of Building 50.

# 3.1 - Soil Boring Locations

The selection of the soil boring locations was based on the results of the previous investigations, existing information sources, and current soil vapor survey information which served as guidance for boring placement and selection of depth intervals for sampling. Drilling locations were also dependent on access restrictions due to operations of the facility, and underground and above ground utilities. Existing utilities were located and marked by Chrysler personnel. All drilling locations were pre-approved by appropriate DTPP personnel who assured each location was at least ten feet from underground utilities or structures, and that a twenty-foot minimum distance was maintained from above ground utilities.

The soil vapor survey proved a valuable guide to achieving optimum placement of the soil borings. Three areas were defined through the soil vapor survey which displayed patterns of shallow and deep vadose zone contamination by organic compounds. Refer to Figure 3 for a map of the facility showing these areas. Area A was characterized as the central portion of the facility which consists of an area to the north of Buildings 40A and 40B, and to the south of Buildings 50 and 53 (former TCA tank area). Area B was characterized as the east-central portion of the facility from the northern limit of Building 59 northward across the area of Building 47 and the associated waste storage area. Area C was characterized as isolated areas of significantly elevated VOCs in the southern portion of the site to the west of Building 3A, south of Building 59, and south of Buildings 40 and 40A.

On October 17, 1994 Clean Tech mobilized on-site to begin the soil boring installations. A Work Plan and site-specific Health and Safety Plan (HASP) were prepared and submitted to Chrysler Corporation's environmental activities staff prior to the start of drilling. Ten (10) soil borings were proposed and installed at the site.

Soil borings 3, 4, 6 and 8 were placed in <u>Area A</u>, soil borings 1, 2, 5 and 7 were placed in <u>Area B</u>, and soil borings 9 and 10 were placed in <u>Area C</u>. Drawing 11 shows the locations of the soil borings and the areas of contamination defined through the soil vapor survey.

#### 3.2 - Installation Methods

The soil borings were installed using the hollow stem auger drilling method. Drilling was performed by Moody's of Dayton, a local business with many years of drilling experience in the Dayton area, and familiar with subsurface conditions in and near the site. Each borehole was advanced using a CME 75 truck mounted hollow stem auger drilling rig. All soil borings were completed using 4.25" diameter I.D. augers. The soil borings were each advanced to the water table, which was typically encountered at approximately 25 to 30

feet below ground surface (BGS). Soil samples were collected at five foot intervals as each boring was advanced using a standard penetration test with split spoon sampler. The general procedures for drilling and soil sampling activities are presented in Attachment D (see Volume II of this report).

Each soil sample was tested using a hydrophobic dye for the presence of non-aqueous phase liquid. This was a qualitative screening test performed in the field at the time the sample was collected which could detect both light (LNAPL) and dense non-aqueous phase liquids (DNAPL) if present.

Field personnel maintained a field logbook with documentation of all pertinent information about field activities and samples, including sample identification information as included on the sample labels and chain of custody forms. Entries in the logbook were made in ink and included a description of field activities; names of project individuals; date, time, and any field measurement information.

A geologic log was generated for each soil boring. These logs included the depth of the boring, sampled intervals, sample identification and sample recovery, standard penetration test results (blow counts), descriptions of the samples, air monitoring measurements for the breathing zone, borehole and split spoon samples, and the results of the dye test for non-aqueous phase liquids. The geologic logs for the ten soil borings are included as Attachment E (see Volume II of this report).

# 3.2.1 - Waste Disposal Methods

The installation of the soil borings generated soil cuttings as waste materials. A total of ten soil borings were installed as per the Work Plan. The soil cuttings were placed in drums, labeled and staged on-site. A total of 143 drums of soil cuttings were generated during the combined soil boring and well installation activities.

As described in the Work Plan, the drilling cuttings generated during the soil boring installations were screened for organic vapor emissions using a PID. No free phase product was observed in any of the drilling cuttings. No organic vapor readings were measured which exceeded the action level of 50 ppm described in the HASP for any of the borings. The laboratory analytical results for the soil samples collected from the soil borings do not show significantly elevated levels of VOCs for any of the soil samples. Based on these findings, Chrysler Corporation will move the drummed soil cuttings to an area of the facility near the existing soil stock piles, and spread and grade the soil level on the ground surface as soon as possible.

# 3.3 - Soil Sampling and Analysis Methods

The soil samples collected from the soil borings were examined in the field, and laboratory analyzed for targeted chemical analysis, and geotechnical evaluation of the subsurface materials. A generalized guide to soil sample depth selection was based on the soil vapor survey.

For soil borings located near the center of areas of elevated soil vapor measurements, a sample was collected for chemical analysis from the split spoon sample having the highest observed PID readings. For soil borings located near the edge of areas of elevated soil vapor measurements, a sample was collected for chemical analysis from the split spoon sample below any elevated PID readings, or at the top of the water table, whichever was encountered first as the boring was advanced.

Geotechnical analysis was performed as an aid to identify applicable remedial technologies for the vadose zone. The geotechnical samples were selected as representative samples of the subsurface materials encountered, and at depths in the borings where an engineered remedial technology might be applied to the vadose zone.

### 3.3.1 - Chemical Analysis

Chemical analysis of soil samples from the soil borings consisted of quantitative field analysis using the photoionization detector, qualitative field analysis using hydrophobic dye, and laboratory analysis for volatile organic compounds, total organic carbon and metals.

#### Photoionization Detector

The soil samples were analyzed immediately upon opening the split spoon sampler using an HNu photoionization detector. The PID measured the levels of total volatile organic compounds and reported those measurements as parts per million (ppm) equivalent of the calibration gas, isobutylene. Results typically ranged from background (BG as reported in the geologic logs) for ambient air levels, to under 10 ppm total volatile organic compounds for the majority of the samples. Those samples with measured PID values of 10 ppm or greater are listed below:

Boring	Depth	PID	Comment
SB-3	14-16 ft	10 ppm	Sample approximately 10 ft above water table.
SB-5	19-21 ft	10 ppm	Sample approximately 5 ft above water table.
SB-5	29-31 ft	10 ppm	Sample approximately 5 ft below water table.
SB-7	14-16 ft	10 ppm	Sample approximately 10 ft above water table.
SB-9	19-21 ft	15 ppm	Sample approximately 5 ft above water table.
SB-10	29-31 ft	15 ppm	Sample approximately 5 ft below water table.

#### Field Analysis Dye Test

The hydrophobic dye test was performed using each sample collected. The results are summarized below for those samples yielding positive dye test results. All other soil boring samples yielded negative results for the dye test.

Boring	Depth	Comment	
SB-5	29-31 ft	Sample approximately 5 feet below water table.	
		PID slightly elevated at 10 ppm in sample.	
SB-9	19-21 ft	Sample approximately 5 feet above water table.	
		PID slightly elevated at 15 ppm in sample.	

#### Laboratory Analysis

One soil sample from each soil boring was collected and analyzed for the Target Compound List (TCL) volatile organic compounds (VOCs), and the Target Analyte List (TAL) metals. The TCL VOCs list includes 69 targeted organic compounds. Analysis was performed using EPA Method 8260. The TAL metals list includes 18 targeted metals. Analysis was performed using EPA Methods 6010/7000 and 7421 (lead by furnace). Metals analysis was performed as follows:

- ICP analysis for aluminum, barium, beryllium, cadmium, calcium, chromium (total), cobalt, copper, silver, sodium, vanadium, and zinc;
- Furnace analysis for antimony, arsenic lead, selenium, and thallium;
- Mercury analysis by cold vapor.

These lists provide a selection of targeted analytes which might be present based on the available information. One soil sample was collected and analyzed for total organic carbon (TOC) using EPA Method 9060. This was done as a preliminary design step to assist with the determination of possible remedial technologies. The sample collected for TOC analysis was selected as representative of the subsurface materials encountered. Soil samples collected for laboratory analysis are listed below:

Boring	Depth	Analysis Performed
SB-1	9-11 ft	TCL VOCs, TAL Metals
SB-2	19-21 ft	TCL VOCs, TAL Metals
SB-3	14-16 ft	TCL VOCs, TAL Metals
SB-4	14-16 ft	TCL VOCs, TAL Metals, TOC
SB-5	29-31 ft	TCL VOCs, TAL Metals
SB-6	14-16 ft	TCL VOCs, TAL Metals
SB-7	24-26 ft	TCL VOCs, TAL Metals
SB-8	24-26 ft	TCL VOCs, TAL Metals
SB-9	19-21 ft	TCL VOCs, TAL Metals
SB-10	29-31 ft	TCL VOCs, TAL Metals

### Quality Assurance and Quality Control

The Quality Assurance/Quality Control (QA/QC) program for chemical analysis of soil samples (both for the soil borings and groundwater monitoring wells) consisted of the collection and analysis of duplicate samples, spiked samples, and equipment blanks. The purpose of this program was to ensure the analyses performed by the analytical laboratory are reproducible. The chain of custody documentation, any QA/QC sample analytical results and the laboratory results for the soil boring samples are included as Attachment F (see Volume III of this report). The QA/QC program for chemical analysis of soil samples is included as Attachment G (see Volume II of this report).

# 3.3.2 - Geotechnical Analysis

Geotechnical analysis of selected soil samples was performed as an aid in determining applicable remedial technologies. Samples were selected for geotechnical analysis based on their representativeness of the subsurface materials encountered, and at a depth in the boring where an applicable remedial technology might be applied.

The Work Plan submitted to Chrysler proposed a total of six soil samples to be collected from the soil borings for geotechnical analysis. These samples were to have been collected

from the vadose zone and analyzed for particle-size distribution, porosity, permeability, and percent moisture.

The Work Plan was altered in response to site conditions as encountered during the initial phase of drilling. The coarse granular nature of the subsurface materials precluded the planned use of large (three-inch O.D.) split spoons for collection of the geotechnical samples. Minimal sample could be retained in the large split spoon sampler. Additionally, any soil samples collected using a split spoon were so disturbed as to make porosity and permeability measurements less than reliable, regardless of the size or type of sampler used.

Geotechnical samples were collected from the soil borings using a two-inch O.D. split spoon sampler. The soil samples were collected in clean glassware and submitted to Tetra Tech Richardson of Newark, Delaware for textural gradation analysis and percent moisture content. Soil samples collected from the soil borings for laboratory analysis are listed below:

Boring	Depth	Analysis Performed
SB-1	14-16 ft	% Moisture
SB-2	14-16 ft	% Moisture
SB-3	19-21 ft	% Moisture
SB-5	14-16 ft	Textural Gradation, % Moisture
SB-6	19-21 ft	Textural Gradation, % Moisture
SB-10	14-16 ft	Textural Gradation, % Moisture

### Quality Assurance and Quality Control

The Quality Assurance/Quality Control (QA/QC) program for geotechnical analysis of soil samples (both the soil borings and the groundwater monitoring wells) specified laboratory test procedures which followed ASTM procedures or approved equivalent methods for analysis of textural gradation and percent moisture. The QA/QC program for geotechnical analysis of soil samples is included as Attachment G (see Volume II of this report). The

results of the geotechnical analysis for the samples collected from the soil borings are included as Attachment H (see Volume III of this report).

# Section 4.0 - Groundwater Monitoring Wells

The groundwater monitoring wells were installed to satisfy two objectives. One objective of the groundwater monitoring wells, similar to the soil borings, was to observe and describe site stratigraphy, sample the vadose soil zone for laboratory analysis, and determine the extent of contaminated soils in the vadose zone at the site. The selection of the well locations and soil sampling depths was augmented by the soil boring information as well as previous site investigations, existing information sources and the soil vapor survey findings.

The other objective of the groundwater monitoring wells was to install a total of fifteen wells, twelve in the unconfined aquifer and three in the semi-confined aquifer, which would allow groundwater samples and water-level measurements to be obtained at points across the site in the unconfined and semi-confined aquifers. A total of three well pairs, each pair having one well screened in the unconfined aquifer and one well screened in the semi-confined aquifer, were installed to assess vertical hydraulic gradients at the site.

# 4.1 - Groundwater Monitoring Well Locations

The selection of the monitoring well locations was completed using the results of the previous site investigations, existing information sources, and the findings of the soil vapor survey and soil borings. This information guided well placement and selection of depth intervals for soil sampling.

Drilling locations were also dependent on access restrictions due to operations of the facility, and underground and above ground utilities. Existing utilities were located and marked by Chrysler personnel. All drilling locations were pre-approved by appropriate DTPP personnel who confirmed each location was at least ten feet from underground utilities or structures, and that a twenty-foot minimum distance was maintained from above ground utilities.

On October 17, 1994 Clean Tech mobilized on-site to begin installation of the groundwater monitoring wells. A Work Plan and site-specific Health and Safety Plan (HASP) were prepared and submitted to Chrysler prior to the start of drilling. A total of fifteen monitoring wells were proposed and installed at the site. Twelve wells were installed in the unconfined aquifer and three wells were installed in the semi-confined aquifer. The wells installed in the unconfined aquifer were installed in two depth ranges within the aquifer. This was done to assess the unconfined aquifer for the possible presence of DNAPLs.

Wells designated MWA were installed in the upper portion of the saturated zone. Total depths for these wells range from 39 to 45 feet BGS, approximately 15 feet below the top of the saturated zone (water table). Wells designated MWB were installed in the lower portion of the saturated zone. Total depths range from 54 to 90 feet BGS. Wells designated MWC were installed in the semi-confined aquifer as paired wells with MWB wells. The total depths for the MWC wells range from 84 to 122 feet BGS. The MWC wells were installed in the upper portion of the semi-confined aquifer.

Three areas were identified and delineated through the soil vapor survey which displayed patterns of shallow and deep vadose zone contamination by organic compounds. Refer to Figure 3 for the locations of these areas. Area A was characterized as the central portion of the facility in the area to the north of Buildings 40A and 40B, and to the south of Buildings 50 and 53 (former TCA tank area). Area B was characterized as the east-central portion of the facility from the northern limit of Building 59 northward across the area of Building 47 and the associated waste storage area. Area C was characterized as isolated areas of significantly elevated VOCs in the southern portion of the site to the west of Building 3A, south of Building 59, and south of Buildings 40 and 40A.

Groundwater monitoring wells MWA-2, MWA-3 and MWB-6 were placed in <u>Area A</u>, wells MWA-1, MWA-5, MWB-2 and MWC-2 were placed in <u>Area B</u>, and wells MWA-4,

MWB-3 and MWC-3 were placed in <u>Area C</u>. Groundwater monitoring wells MWA-6, MWB-1, MWB-4, MWB-5 and MWC-1 were located near the property boundary corners and separated from the identified contaminant areas. This provided coverage of the site as a whole, allowed determination of background water quality for groundwater flowing on to the site, and made data collection possible in additional areas of potential contamination. Drawing 12 shows the locations of the groundwater monitoring wells and the areas of contamination defined through the soil vapor survey.

### 4.2 - Installation Methods

The shallow groundwater monitoring wells in the unconfined aquifer were installed using the hollow stem auger drilling method. The deeper wells in the semi-confined aquifer were installed using the cable tool drilling method. Drilling was performed by Moody's of Dayton, a local business with many years of drilling experience in the Dayton area.

Soil samples were collected at five-foot intervals as each hollow stem auger boring was advanced using a standard penetration test with split spoon sampler. Soil samples were examined from the cable tool rig when the boring was bailed, and split spoon samples were collected from the confining unit (till layer) and from the portion of the semi-confined aquifer where the well screen was set. The general procedures for drilling and soil sampling activities are presented in Attachment D (see Volume II of this report).

Each soil sample was tested using a hydrophobic dye for the presence of non-aqueous phase liquid. This was a qualitative screening test performed in the field at the time the sample was collected which could detect both light (LNAPL) and dense non-aqueous phase liquids (DNAPL) if present.

Field personnel maintained a field logbook with documentation of all pertinent information about field activities and samples, including sample identification information as included on the sample labels and chain of custody forms. Entries in the logbook were made in ink

and included a description of field activities; names of individuals involved; date, time, and any field measurement information.

A geologic log was generated for each groundwater monitoring well. These logs include: the depth of the boring, sampled intervals, sample identification, sample recovery, standard penetration test results (blow counts), descriptions of the samples, air monitoring measurements for the breathing zone, borehole and split spoon samples, the results of the dye test for non-aqueous phase liquids, and well construction details. The geologic logs for the groundwater monitoring wells are included as Attachment I (see Volume II of this report).

#### **Unconfined Aquifer Wells**

A total of twelve wells were installed in the unconfined aquifer in two depth ranges. This was done to assess the shallow and deeper portions of the unconfined aquifer for both dissolved phase contaminants and the presence of DNAPLs. Analytical results at the Gem City, Inc. site immediately east of DTPP indicated higher VOC concentrations were present in the shallow portion of the unconfined aquifer.

Wells designated MWA were installed in the shallow portion of the saturated zone. Total depths for these wells range from 39 to 45 feet BGS, approximately 15 feet below the top of the saturated zone (water table). The water table is anticipated to fluctuate between ten and fifteen feet during the year. Wells designated MWB were installed in the deeper portion of the saturated zone immediately above the confining layer (till layer). Total depths range from 54 to 90 feet BGS. MWB-4 was completed at a more shallow depth (35 feet) due to conditions encountered during well installation.

Each shallow well boring was advanced using a CME 75 truck mounted hollow stem auger drilling rig. The borings were completed using 4.25" or 6.25" diameter I.D. augers.

The larger augers were used whenever site conditions allowed. Soil sampling was performed using procedures as presented in Attachment D.

Each well was constructed inside the hollow stem augers using two-inch diameter PVC casing and ten feet of 10 slot two-inch diameter PVC well screen. After insertion of the casing and screen, sand pack (Global Filter Pack #5) was poured to approximately two feet above the top of the well screen as the augers were withdrawn. An approximately three-foot thick bentonite seal was installed above the sand pack. The bentonite seal was allowed to hydrate and expand prior to placement of the grout. The remaining annular space was grouted using a positive pressure tremmie pipe. Care was taken to avoid disturbing the bentonite seal during grout placement. The grout mixture was allowed to cure before installation of flush mounted locking well covers.

#### Semi-Confined Aquifer Wells

Three groundwater monitoring wells were installed targeting the semi-confined aquifer below the confining layer (till layer). These wells were designated MWC as presented in the geologic logs (see Attachment I). MWC-1 was double-cased, and MWC-2 and MWC-3 were triple-cased to prevent groundwater migration between the unconfined aquifer and the semi-confined aquifer. The borings for the semi-confined aquifer wells were advanced using a BE22-W cable tool drilling rig.

The boring for MWC-1 was advanced and eight-inch diameter steel casing was driven through the unconfined aquifer to approximately five feet below the top of the till layer. The till consisted of clay with variable amounts of sand and gravel. The boring was then advanced through the till layer and a split spoon sample of the till was collected from two feet below the top of the till layer for lithologic description. Split spoon samples of the semi-confined aquifer were collected from the interval to be screened for lithologic description, and six-inch diameter steel casing was installed to the total depth of 112 feet BGS.

The MWC-1 well was constructed inside the six-inch casing using two-inch diameter PVC casing and ten feet of 10 slot two-inch diameter PVC well screen. The well screen was set from 102 to 112 feet BGS. The top of the well screen was positioned approximately six feet below the base of the till layer. After insertion of the casing and screen, sand pack (Global Filter Pack #5) was poured to approximately two feet above the top of the well screen as the six-inch casing was withdrawn from the boring. An approximately three-foot thick bentonite seal was installed above the sand pack. The bentonite seal was allowed to hydrate and expand prior to placement of the grout. The remaining annular space was grouted using a positive pressure tremmie pipe. Care was taken to avoid disturbing the bentonite seal during grout placement. The grout mixture was allowed to cure before installation of a flush mounted locking well cover.

The boring for MWC-2 was advanced and twelve-inch diameter steel casing was driven through the unconfined aquifer to approximately two feet below the top of the till layer where it was grouted in place. A split spoon sample of the till was collected from two to four feet below the top of the till layer for lithologic description. The boring was then advanced into the till layer and eight-inch diameter steel casing was driven to approximately eight feet below the top of the till layer where it was grouted in place. The boring was advanced through the till layer, and split spoon samples of the semi-confined aquifer were collected from the interval to be screened for lithologic description. Six-inch diameter steel casing was installed to the total depth of 122 feet BGS.

The MWC-2 well was constructed inside the six-inch casing using two-inch diameter PVC casing and ten feet of 10 slot two-inch diameter PVC well screen. The well screen was set from 112 to 122 feet BGS. The top of the well screen was positioned approximately ten feet below the base of the till layer. After insertion of the casing and screen, sand pack (Global Filter Pack #5) was poured to approximately two feet above the top of the well screen as the six-inch casing was withdrawn. The six-inch casing was withdrawn to nineteen feet above the top of the screen (into the till layer) where it jammed and was grouted in place. An approximately three-foot thick bentonite seal was installed above the

sand pack. The bentonite seal was allowed to hydrate and expand prior to placement of the grout. The remaining annular space was grouted using a positive pressure tremmie pipe. Care was taken to avoid disturbing the bentonite seal during grout placement. The grout mixture was allowed to cure before installation of a flush mounted locking well cover.

The boring for MWC-3 was advanced and twelve-inch diameter steel casing was driven through the unconfined aquifer to approximately two feet below the top of the till layer where it was grouted in place. A split spoon sample of the till was collected from two to four feet below the top of the till layer for lithologic description. The boring was then advanced into the till layer and eight-inch diameter steel casing was driven to approximately four feet below the top of the till layer where it was grouted in place. The boring was advanced through the till layer, and split spoon samples of the semi-confined aquifer were collected in the interval to be screened. Six-inch diameter steel casing was installed to the total depth of 84 feet BGS.

The MWC-3 well was constructed inside the six-inch casing using two-inch diameter PVC casing and ten feet of 10 slot two-inch diameter PVC well screen. The well screen was set from 74 to 84 feet BGS. The top of the well screen was positioned approximately five feet below the base of the till layer. After insertion of the casing and screen, sand pack (Global Filter Pack #5) was poured to approximately two feet above the top of the well screen as the six-inch casing was withdrawn from the boring. An approximately three-foot thick bentonite seal was installed above the sand pack. The bentonite seal was allowed to hydrate and expand prior to placement of the grout. The remaining annular space was grouted using a positive pressure tremmie pipe. Care was taken to avoid disturbing the bentonite seal during grout placement. The grout mixture was allowed to cure before installation of a flush mounted locking well cover.

Soil sampling for the semi-confined aquifer wells was performed following the procedures presented in Attachment D.

### Well Development

Each monitoring well was developed after installation to restore the natural hydraulic properties of the aquifer and facilitate free hydraulic connection between the aquifer and the well. Well development was performed by surging the screened interval and pumping the well. Field measurements were collected including conductivity, pH, and temperature. Water turbidity was monitored.

Each well was developed until the measured parameters stabilized, and the water pumped from the well was relatively turbidity-free. The wells were each developed for a period of approximately thirty to sixty minutes. Approximately 200 to 350 gallons of water were pumped from each well during the development process.

### 4.2.1 - Waste Disposal Methods

The installation and sampling of the groundwater monitoring wells generated soil cuttings and groundwater as waste materials. A total of fifteen groundwater monitoring wells were installed and sampled as per the Work Plan. The soil cuttings were placed in drums, labeled and staged on site. A total of 143 drums of soil cuttings were generated during the combined soil boring and well installation activities.

A total of approximately 3,630 gallons of water were pumped from the wells during the well development process. Additionally, the wells were each purged of at least three wellbore volumes of water prior to the collection of each round of groundwater samples producing a total of approximately 525 gallons of purge water.

#### Soil Cuttings

As described in the Work Plan, the drilling cuttings were screened for organic vapor emissions using a PID. No free phase product was observed in any of the drilling cuttings. No organic vapor readings were measured for the breathing zone or borehole which

exceeded the action level of 50 ppm as described in the HASP for any of the wells. The laboratory analytical results for the soil samples do not show significantly elevated levels of VOCs for any of the soil samples. Based on these findings, Chrysler Corporation will move the drummed soil cuttings to an area of the facility near the existing soil stock piles, and spread and grade the soil level on the ground surface as soon as possible.

#### Water

As described in the Work Plan, drilling fluids, well development water, purge water, and decontamination fluids generated by field investigation activities were screened for organic vapor emissions using the PID. No free phase product was observed in any of the produced water, however, organic vapor readings from these fluids were occasionally found to exceed the action level of 50 ppm as described in the HASP. Therefore, in accordance with the Work Plan, all drilling fluids, well development water, and decontamination fluids were temporarily containerized at the well head and transferred to the on-site wastewater treatment plant for final disposal at the time of well installation and development.

Purge water generated during purging of the wells prior to collection of both rounds of groundwater samples was drummed and staged at the well head. The laboratory analytical results for the groundwater samples do not show significantly elevated levels of VOCs for any of the groundwater samples. No free phase product was observed in any of the produced water, however, organic vapor readings from the purge water as measured during the first round of groundwater sampling were occasionally found to exceed the action level of 50 ppm as described in the HASP. Therefore, in accordance with the Work Plan, Chrysler Corporation will move all of the drummed purge water to the on-site wastewater treatment plant for final disposal as soon as possible.

### 4.3 - Soil Sampling and Analysis Methods

The soil samples collected from the groundwater monitoring wells were examined in the field, and laboratory analyzed for targeted chemical analysis and geotechnical evaluation of the subsurface materials. A generalized guide to soil sample depth selection was formed based on the soil vapor survey, and the site conditions encountered during the soil boring installations.

For wells located near the center of areas of elevated soil vapor measurements, a soil sample was collected from the split spoon sample having the highest observed PID readings. For wells located near the edge of areas of elevated soil vapor measurements, a soil sample was collected from the split spoon sample below any elevated PID readings or at the top of the water table, whichever was encountered first as the boring was advanced.

### 4.3.1 - Chemical Analysis

Chemical analysis of the soil samples from the groundwater monitoring wells consisted of quantitative field analysis using the photoionization detector, qualitative field analysis using hydrophobic dye, and laboratory analysis for volatile organic compounds, total organic carbon and metals.

#### Photoionization Detector

The soil samples were analyzed immediately upon opening the split spoon sampler using an HNu photoionization detector. The PID measured the levels of total volatile organic compounds and reported those measurements as parts per million equivalent of the calibration gas, isobutylene. Results typically ranged from background (BG as reported in the geologic logs) for ambient air levels to under 10 ppm total volatile organic compounds for the majority of the samples. Those samples with measured PID values of 10 ppm or greater are listed below:

Well	Depth	PID	Comment
MWA-1	19-21 ft	10 ppm	Sample approximately 5 ft above water table.
MWA-1	24-26 ft	10 ppm	Sample approximately 1 ft above water table.
MWA-1	34-36 ft	15 ppm	Sample approximately 10 ft below water table.
MWA-2	14-16 ft	10 ppm	Sample approximately 10 ft above water table.
MWA-2	19-21 ft	12 ppm	Sample approximately 5 ft above water table.
MWA-3	24-26 ft	12 ppm	Sample approximately 5 ft below water table.
MWA-3	29-31 ft	70 ppm	Sample approximately 3 ft below water table.
MWA-3	34-36 ft	70 ppm	Sample approximately 8 ft below water table.
MWA-4	19-21 ft	50 ppm	Sample approximately 5 ft above water table.
MWA-4	24-26 ft	75 ppm	Sample approximately at water table.
MWA-4	29-31 ft	80 ppm	Sample approximately 5 ft below water table.
MWA-4	34-36 ft	80 ppm	Sample approximately 10 ft below water table.
MWA-4	39-41 ft	100 ppm	Sample approximately 15 ft below water table.
MWA-4	44-46 ft	60 ppm	Sample approximately 20 ft below water table.
MWB-3	34-36 ft	15 ppm	Sample approximately 8 ft below water table.
MWB-3	44-46 ft	20 ppm	Sample approximately 18 ft below water table.
MWB-6	24-26 ft	14 ppm	Sample approximately at water table.

# Field Analysis Dye Test

The hydrophobic dye test was performed using each sample collected. The results are summarized below for those samples yielding positive dye test results, suggesting the presence of non-aqueous phase liquids. All other soil samples from the groundwater monitoring wells yielded negative results for the dye test.

Well	Depth	Comment
MWA-4	24-26 ft	Sample approximately at water table.
		PID elevated at 75 ppm in sample.
MWB-1	69-71 ft	Sample approximately 50 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-2	64-66 ft	Sample approximately 40 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-2	69-71 ft	Sample approximately 45 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-2	74-76 ft	Sample approximately 50 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-2	79-81 ft	Sample approximately 55 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-2	84-86 ft	Sample approximately 60 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-2	89-91 ft	Sample approximately 65 feet below water table.
		Possible natural oil from overlying clay unit.
MWB-3	39-41 ft	Sample approximately 15 feet below water table.
		Possible natural oil from underlying clay unit.
MWB-3	54-56 ft	Sample approximately 30 feet below water table.
		Possible natural oil from underlying clay unit.
MWB-5	39-41 ft	Sample approximately 10 feet below water table.
		Possible natural oil from clay unit.
MWB-5	49-51 ft	Sample approximately 20 feet below water table.
		Possible natural oil from clay unit.
MWB-5	54-56 ft	Sample approximately 25 feet below water table.
		Possible natural oil from clay unit.
MWB-5	59-61 ft	Sample approximately 30 feet below water table.
		Possible natural oil from clay unit.
MWB-5	64 <b>-</b> 66 ft	Sample approximately 35 feet below water table.
		Possible natural oil from clay unit.
MWB-5	67-71 ft	Sample approximately 40 feet below water table.
		Possible natural oil from clay unit.
MWB-5	74-76 ft	Sample approximately 45 feet below water table.
		Possible natural oil from clay unit.
MWB-5	79-81 ft	Sample approximately 50 feet below water table.
		Possible natural oil from clay unit.
MWB-5	84-86 ft	Sample approximately 55 feet below water table.
) (UD 5	80 01 6	Possible natural oil from clay unit.
MWB-5	89-91 ft	Sample approximately 60 feet below water table.
ACUTO	20.41.6	Possible natural oil from clay unit.
MWB-6	39-41 ft	Sample approximately 15 feet below water table.
NOVO (	44.46.5	Possible natural oil from underlying clay unit.
MWB-6	44-46 ft	Sample approximately 20 feet below water table.
		Possible natural oil this clay unit.

### **Laboratory Analysis**

Soil samples were collected from the monitoring well borings and analyzed for TCL VOCs, and TAL metals. The TCL VOC list analysis was performed using EPA Method

8260. The TAL Metals list analysis was performed using EPA Methods 6010/7000 and 7421 (lead by furnace). Metals analysis was performed as follows:

- ICP analysis for aluminum, barium, beryllium, cadmium, calcium, chromium (total), cobalt, copper, silver, sodium, vanadium, and zinc;
- Furnace analysis for antimony, arsenic lead, selenium, and thallium;
- Mercury analysis by cold vapor.

Four soil samples were collected and analyzed for total organic carbon (TOC) using EPA Method 9060. The samples collected for TOC analysis were selected as representative of the subsurface materials encountered. Soil samples collected for laboratory analysis are listed below:

Well	Depth	Analysis Performed
MWA-1	24-26 ft	TCL VOCs, TAL Metals
MWA-2	19-21 ft	TCL VOCs, TAL Metals
MWA-3	24-26 ft	TCL VOCs, TAL Metals
MWA-4	24-26 ft	TCL VOCs, TAL Metals
MWA-5	24-26 ft	TCL VOCs, TAL Metals
MWA-6	24-26 ft	TCL VOCs, TAL Metals
MWB-1	49-51 ft	TOC
MWB-2	24-26 ft	TCL VOCs, TAL Metals
MWB-3	24-26 ft	TCL VOCs, TAL Metals, TOC
MWB-4	19-21 ft	TCL VOCs, TAL Metals, TOC
MWB-5	24-26 ft	TCL VOCs, TAL Metals, TOC
MWB-6	24-26 ft	TCL VOCs, TAL Metals

#### Quality Assurance and Quality Control

The Quality Assurance/Quality Control (QA/QC) program for chemical analysis of soil samples (both for the soil borings and the groundwater monitoring wells) consisted of the collection and analysis of duplicate samples, spiked samples, and equipment blanks. The purpose of this program was to ensure the analyses performed by the analytical laboratory

are reproducible. The chain of custody documentation, any QA/QC sample analytical results and the laboratory results for the soil samples collected from the groundwater monitoring wells are included as Attachment J (see Volume III of this report). The QA/QC program for chemical analysis of soil samples is included as Attachment G (see Volume II of this report).

## 4.3.2 - Geotechnical Analysis

Geotechnical analysis of selected soil samples was performed to assist with the identification of possible remedial technologies. Samples were selected for geotechnical analysis based on their representativeness of the subsurface materials encountered, and at a depth in the boring where a possible remedial technology might be applied.

The Work Plan submitted to Chrysler proposed a total of six soil samples to be collected from the soil borings only. None were to have been collected from the monitoring well borings. Samples were to have been collected from the vadose zone and analyzed for particle-size distribution, porosity, permeability, and percent moisture. Three Shelby tube samples were to have been collected from the confining layer (till layer) during installation of the wells in the semi-confined aquifer. The Shelby tube samples were to have been analyzed for permeability using a constant head permeability test for granular soils.

The Work Plan was altered in response to site conditions encountered during drilling. The coarse granular nature of the subsurface materials precluded the planned use of large (three-inch O.D.) split spoons for collection of the geotechnical samples. Any soil samples collected using a split spoon were found to be so disturbed as to make porosity and permeability measurements less than reliable. Representatives of the well drilling firm, Moody's of Dayton, reported that their previous experiences attempting Shelby tube samples in the till layer were unsuccessful. Shelby tubes typically crush when pushed into the stiff clay and gravel of the till. If a Shelby tube was crushed in a boring advanced using the cable tool drilling method, there is no reliable way to recover it. The boring

would then need to be grouted and abandoned, and the well installation started again.

Geotechnical samples were collected from the monitoring well borings using either a two-inch or three-inch O.D. split spoon sampler. The soil samples were collected in clean glassware and submitted to Tetra Tech Richardson of Newark, Delaware for textural gradation analysis. Soil samples collected from the groundwater monitoring well borings are listed below:

Well	Depth	Analysis Performed
MWA-4	39-41 ft	Textural Gradation
MWA-5	34-36 ft	Textural Gradation
MWB-2	74-76 ft	Textural Gradation
MWC-1	104-106 ft	Textural Gradation
MWC-2	114-116 ft	Textural Gradation
MWC-3	76-78 ft	Textural Gradation

#### Quality Assurance and Quality Control

The Quality Assurance/Quality Control (QA/QC) program for geotechnical analysis of soil samples (both the soil borings and the groundwater monitoring wells) specified laboratory test procedures which followed ASTM procedures or approved equivalent methods for textural gradation analysis. The results of the geotechnical analysis are included as Attachment K (see Volume III of this report). The QA/QC program for geotechnical analysis of soil samples is included as Attachment G (see Volume II of this report).

# 4.4 - Groundwater Sampling and Analysis Methods

Groundwater samples were collected from each of the fifteen groundwater monitoring wells during two sampling events. The wells were sampled twice to determine if there are any effects on water quality due to seasonal water level fluctuations. The first sampling event was completed in December, 1994 and the second was completed in February,

1995. Both groundwater sampling events were performed using Clean Tech's standard sampling procedures.

Groundwater samples were collected and analyzed for TCL VOCs and TAL metals. Chromium analysis was performed using total chromium as the target analyte. Chromium VI analysis was not performed for the groundwater samples because the sample holding times were in excess of 24 hours.

Each well remained static for approximately two weeks following well development to allow the portion of the aquifer disturbed during the well installations to equilibrate. All wells were screened for evidence of organic vapors prior to collection of the first round of groundwater samples in December 1994 using a PID. The PID was inserted into the open well top immediately upon opening the well. The PID measurement was recorded in the field logbook. The wells were not screened for evidence of organic vapors at the time of the second round of groundwater samples in February 1995 because the ambient air temperature was so low during that time as to render the PID ineffective as an air monitoring tool. Water levels were measured from the top of the PVC casing prior to well purging. An interface probe was used to measure water levels and the thickness of any non-aqueous phase product (LNAPL or DNAPL) prior to purging the well in preparation for groundwater sampling. The water levels and notes regarding any non-aqueous phase product were recorded for both sampling rounds.

All wells were purged prior to sample collection to obtain a representative groundwater sample. Each well was purged of at least three wellbore volumes of water prior to sample collection. The groundwater sample collection procedures are presented in Attachment L (see Volume II of this report).

### Quality Assurance and Quality Control

The Quality Assurance/Quality Control (QA/QC) program for chemical analysis of groundwater consisted of the collection and analysis of duplicate samples, and equipment

and trip blanks. The purpose of this program was to ensure the analyses performed by the analytical laboratory are reproducible. The chain of custody documentation, any QA/QC sample analytical results and the laboratory results for the groundwater samples are included as Attachment M (Round #1) and Attachment N (Round #2), and are presented in Volume III of this report. The QA/QC program for chemical analysis of groundwater samples is included as Attachment O (see Volume II of this report).

#### 4.4.1 - Water Level Measurements

Water level measurements were collected at each groundwater monitoring well on a monthly basis for a period of three months after completion of the wells. Three rounds of water level measurements were collected to observe temporal variations in groundwater levels during the reporting period. The depth to water in each monitoring well was measured from the top of the PVC well casing. The PVC well casing tops were surveyed for each monitoring well and referenced to a standard elevation, thereby allowing computation of the reference elevation for the water level in each well.

All soil borings and monitoring well tops (top of PVC casing) were surveyed for vertical and horizontal control by a State of Ohio licensed surveyor. The survey was performed using a site specific local coordinate system for horizontal control which tied surveyed traverses to the on site buildings, and a United States Geologic Survey benchmark located at the intersection of Webster and Leo Street for vertical control referenced to feet above mean sea level. Elevations were measured to the nearest 0.01 foot.

Water-level measurements were obtained at the time of the first groundwater sampling event on December 13-14, 1994, on January 24, 1995, and at the time of the second groundwater sampling event on February 20, 1995. Water level measurements were collected from each monitoring well prior to the beginning of purging and groundwater sample collection. Piezometric surface maps for the unconfined aquifer are included as Drawings 22, 23, and 24. All Drawings are included in Volume II of this report.

The water level measurements were collected using the following procedure. All sampling team members wore new and clean disposable gloves during measurements at each well to protect team members from exposure to potentially contaminated groundwater, and to minimize the potential for cross-contamination between wells. The lock was removed from the locking well cap just prior to measuring. While standing upwind, the well cap was removed and the well was allowed to vent. The permanent measurement reference point was then located on the PVC well casing.

The decontaminated two-phase interface probe was lowered into the well to the static water level. The well was inspected for the presence of any LNAPL which might be presence as a layer on top of the static water level. No LNAPL layers were detected in any wells during any of the three water level measurement events. The monitoring well number and the distance from the permanent reference point to the static water level were recorded to the nearest 0.01 foot in the bound log book. The static water elevation was computed and recorded in the bound log book. This value is the elevation of the measured reference point minus the depth to the water in the well.

The interface probe was then lowered to the bottom of the well to inspect the well for the possible presence of a DNAPL layer (a different audible signal will sound if the interface probe contacts a non-water fluid). No DNAPL was detected in any well during any of the three water level measurement events. The elevation of the bottom of the well was computed and recorded in the bound log book. This value is the elevation of the measured reference point minus the depth to the bottom of the well. The two-phase interface probe was then removed from the well, and decontaminated. The measurement tape and probe were wiped along its entire length, discarding and replacing the towels as they became soiled. Field decontamination of the interface probe was accomplished by washing the instrument using a phosphate-free detergent followed by a potable water rinse. The equipment was then rinsed using deionized water and allowed to air dry.

### Section 5.0 - Geology

The geology of the Chrysler DTPP facility is discussed as a means of understanding the nature of the subsurface and its influence on potential contaminant migration pathways. The discussion of the regional geology of the Dayton area provides a reference framework. It was prepared from published reports and available site investigations obtained for nearby properties. The site geology was further examined through information gathered during installation of the soil borings and groundwater monitoring wells, with additional information obtained from the records of the water supply wells located on the property.

### 5.1 - Regional Geology

The regional geology of the Dayton, Ohio area has been examined and discussed by several authors. Original publications by Norris, (1959), and Walton and Scudder, (1960) were reviewed. Site investigations by QSource Engineering, Inc., (1993) for the Gem City Chemicals, Inc. facility, and by Mathes & Associates, (1991) for the DTPP property incorporate these and several additional previous studies. The regional geology of the area has been summarized here from these information sources.

The regional geologic setting of the Dayton, Ohio area consists of glacial and glacial-fluvial (outwash stream) sediments deposited over an irregular bedrock surface. Highly permeable calcareous sands and gravel fill pre-glacial or glacial valleys eroded into the underlying bedrock. These permeable glacial deposits are believed to be outwash deposits originating from retreating glaciers. The permeable deposits have formed shallow and deeper aquifers separated by low permeability confining layers. The confining layers are till layers composed primarily of clay with mixtures of gravel, sand, and silt.

The bedrock underlying the glacial sediments is believed to consist of relatively impermeable materials. It is mapped as the Ordovician Richmond Group, and is thought to

be composed of soft, light gray, calcareous shale with interbedded layers of limestone. Few wells in the region have reached the bedrock surface, which is estimated to be 250 to 300 feet BGS in most areas. The bedrock yields little to no water, provides little recharge to the overlying aquifers, and acts as an impervious lower and lateral boundary to the overlying aquifers.

Regional studies of the glacial and glacial-fluvial deposits have shown the uppermost recognizable geologic unit is a sand and gravel outwash deposit approximately 80 feet thick. This unit is typically recognized as the unconfined aquifer. Discontinuous till layers have been encountered within this unit at depths between 40 and 50 feet BGS.

The unconfined aquifer is generally underlain by a till layer present at approximately 80 feet BGS. This till layer appears to be laterally persistent, but may absent from some locations in the region due either to non-deposition or erosion. Till layers have been reported as massive clay units, or as zones of alternating clay with stratified sand and gravel. Till layers act as confining layers which control aquifer recharge and regional groundwater flow.

Till layers are known to contain significant amounts of natural hydrocarbons. The well drillers were quick to recognize the natural hydrocarbon as it was encountered in clays within the unconfined aquifer, and from the till underlying the unconfined aquifer. The hydrocarbon was a dark brown liquid found as non-aqueous phase product. It was noted in drilling cuttings, when bailing during cable tool drilling, and during examination of soil samples. The State of Ohio Geological Survey was consulted regarding the hydrocarbon and confirmed the hydrocarbon is a natural material found throughout the region. The Survey noted a major oil company had recently studied the viability of hydrocarbon production from till layers in the region.

Regional studies indicate that a second recognizable sand and gravel outwash deposit underlies the till layer found at approximately 80 feet BGS. This lower aquifer behaves as

a confined or semi-confined aquifer. However, if the till layer is thin or absent the hydraulically connected sand and gravel units act as a single unconfined aquifer.

Deep wells in the region suggest discontinuous till layers may exist within the second glacial outwash unit (the semi-confined aquifer), and additional semi-confined or confined aquifers exist at greater depths. These deeper aquifers are believed to be separated by till layers in much the same way as the shallower geologic units. Deeper aquifers were not examined in this investigation.

# 5.2 - Site Geology

The site geology was examined through information gathered during installation of the soil borings and groundwater monitoring wells, with additional information obtained from the records of the production wells located on the property. Refer to Drawings 11 and 12 for locations of the soil borings and monitoring wells in Volume II of this report. Refer to Attachments E and I (see Volume II of this report) for geologic logs containing descriptions of the materials found in the soil borings and monitoring wells. Geologic cross-sections were prepared using the information contained in the boring logs, specifically differentiating the subsurface units containing significant amounts of clay from the more permeable gravel and sand units. A map showing the locations of the geologic cross-sections is included as Figure 4, and the three geologic cross-sections are included as Drawings 25, 26, and 27 in Volume II of this report.

The site geology as determined from information obtained from the borings and wells consists primarily of sand and gravel with minor amounts of silt and clay. These are the glacial and glacial-fluvial sediments typical of the region. The sand and gravel is interbedded with till and clay layers composed of massive clay units, or zones of clay with sand and gravel. The uppermost two to four feet is typically a disturbed clay-bearing material which is absent in many places, probably due to site development activities. None of the borings or wells reached the bedrock surface.

The uppermost geologic unit at the site is a sand and gravel outwash deposit approximately 75 to 90 feet thick. This is the unconfined aquifer. Clay units, and units composed of clay, sand and gravel mixtures were encountered within the unconfined aquifer. Several of these units are laterally persistent suggesting they might exert some local control over potential contaminant migration pathways. Additional clay-bearing units were noted in the unconfined aquifer, but were restricted to certain small areas of the site. The more laterally persistent clay-bearing units within the unconfined aquifer are summarized as follow:

Clay Units Within the Unconfined Aquifer

Depth of Clay-Bearing Unit	Well or Boring Encountering Clay-Bearing Unit
4-6 ft	SB-1, SB-6 (to 11' BGS), SB-9,
	MWA-5, MWA-6,
	MWB-1 (to 16' BGS), MWB-4 (to 16' BGS), MWB-5
19-21 ft	SB-2 (to 26' BGS), SB-8, SB-10 (to 26' BGS),
	MWA-3, MWA-4, MWA-5 (to 26' BGS),
	MWB-3,
	MWC-3
29-31 ft	SB-4, SB-6, SB-10,
	MWB-6,
	Production Well #4
34-75 ft	MWB-4
39- 41 ft	MWA-2, MWA-3,
	MWB-2 (to 51')
64 ft -	MWB-1 (to 76' BGS), MWB-2 (to 81' BGS), MWB-5 (to 90' BGS)

A persistent till layer was encountered which was interpreted as forming the confining layer between the unconfined aquifer and the underlying semi-confined aquifer. The thickness of the till was 20 feet in MWC-1, 23 feet in MWC-2, 14 feet in MWC-3, 15 feet in Production Well #3, and 25 feet in Production Well #4. The approximate depth to the top of the till layer as seen in the deeper wells is listed below:

Depth to Confining Layer at Base of Unconfined Aquifer

Well	Depth to Till at Base Unconfined Aquifer
MWB-1, MWC-1	76 feet BGS
MWB-2, MWC-2	85 to 89 feet BGS
MWB-3, MWC-3	56 to 57 feet BGS
MWB-5	90 feet BGS
MWB-6	44 feet BGS
Production Well #3	85 feet BGS
Production Well #4	80 feet BGS

The semi-confined aquifer was encountered below the till layer in MWC-1, MWC-2, and in Production Wells #3 and #4. It consists of sand and gravel with minor amounts of fine-grained material, much like the unconfined aquifer. These materials are glacial and glacial-fluvial sediments typical of the region. The monitoring wells penetrated approximately 20 feet of the uppermost portion of the unit. No clay-bearing units were noted in the portion of the semi-confined aquifer examined. Drillers logs for Production Wells #3 and #4 describe the unit as coarse grained sand and gravel. A till layer was encountered in Production Well #3 between 128 and 129 feet BGS.

# Section 6.0 - Hydrogeology

The hydrogeology of the Chrysler DTPP facility is presented to gain an understanding of potential subsurface contaminant transport mechanisms. Groundwater flow behavior and aquifer properties form the hydraulic framework for an understanding of the documented pattern of contamination at the site. Regional and site hydrogeology are discussed.

### 6.1 - Regional Hydrogeology

The regional hydrogeology of the Dayton area has been discussed by several authors. Original publications by Norris, (1959), and Walton and Scudder, (1960) were reviewed, along with site investigations by QSource Engineering, Inc., (1993) for the Gem City Chemicals, Inc. facility, and by Mathes & Associates, (1991) and Clean Tech, (1994) for the subject property.

The regional geologic setting of the Dayton, Ohio area consists of highly permeable calcareous sands and gravel deposited in pre-glacial or glacial valleys eroded into the underlying bedrock. These glacial deposits form shallow and deeper aquifers separated by low permeability confining layers (glacial till) composed primarily of clay with mixtures of gravel, sand, and silt. The bedrock materials are of low permeability and act to form lateral and lower boundaries to the flow of groundwater through the permeable materials.

Regional studies of the permeable deposits have shown the uppermost recognizable hydrogeologic unit is a sand and gravel deposit approximately 80 feet thick which is recognized as the unconfined aquifer. Discontinuous till layers have been encountered within this unit which act as local confining layers. The hydraulic conductivity of the shallow aquifer is approximately 200 feet per day with a transmissivity reported to be approximately 15,000 to 40,000 square feet per day (QSource Engineering, Inc.).

Studies completed by Dames & Moore in 1991, and reviewed by Clean Tech, 1994 for the DAP site which is located about four miles north of this site, included an aquifer recovery test which monitored drawdown in the monitoring wells and piezometers surrounding the pumping well. Transmissivity values in the range of 249,000 gallons per day per foot to 747,000 gallons per day per foot were reported. The transmissivity appears to generally be lower in the shallow part of the aquifer and increases with depth.

The lithology of the deeper aquifer is very similar to the shallow aquifer. Based on reports prepared for Gem City Chemicals, the saturated thickness of the deep aquifer is approximately 60 feet thick. Hydraulic conductivity values range from 140 to 200 feet per day. Reported transmissivity ranges from 1,200 to 12,000 square feet per day. The reported storage coefficient of 0.001 is within the expected range for a confined aquifer.

Values for the aquifer parameters developed by CH<sub>2</sub>M Hill for the development of the Miami South Well Field were reviewed by Clean Tech, 1994:

### Upper Aquifer

Hydraulic Conductivity - 0.003 ft/sec (260 ft/day, 2021 GPD/ft²)

Storativity - 0.2 ft/ft

#### Till Layers

Hydraulic Conductivity - 0.44 x 10<sup>-6</sup> ft/sec (0.04 ft/day, 0.3 GPD/ft<sup>2</sup>)

Storativity - 0 ft/ft

#### Lower Aquifer

Hydraulic Conductivity - 0.001 ft/sec (87 ft/day, 710 GPD/ft²)

Storativity - 0.00001 ft/ft

The analysis assumed a 50 foot thick saturated zone in the upper aquifer, and variable thickness for the till and lower aquifer. The transmissivity values were not calculated

directly. All values were calculated assuming that each of the layers were homogeneous and isotropic. Due to the directions of flow that were calculated, the calculated hydraulic conductivities are likely to reflect the horizontal conductivity in the "upper" and "lower" aquifers, and the vertical conductivity through the till. Considerable local variability from these values is likely across the region.

During the pump test conducted at Gem City Chemicals, Inc. on February 21, 1990, the recovery well was pumped at a rate of 340 gpm and the water level in the piezometer installed 3.5 feet away from the pumping well was monitored. The drawdown was 0.75 feet after 450 minutes of pumping. This gives a reported value for transmissivity of 52,900 square feet per day or 395,000 gallons per day per foot, and a hydraulic conductivity of 0.226 centimeters per second (755 ft/day). This value is about three times the average value calculated from the regional studies.

Based on these values, the pre-pumping groundwater flow velocity was estimated to be about 1.2 feet per day. The current flow velocity in the area surrounding the pumping well is estimated to be 6.4 feet per day. The potentiometric surface elevations have been measured in the two well clusters located at the northeastern and southwestern limits of Gem City Chemicals, Inc. The levels measured in the three wells in each cluster are similar, which indicated that the groundwater flow is nearly level at both locations.

Due to the presence of the till layer separating the valley fill deposits into "upper" and "lower" aquifer systems, the direction of groundwater flow was evaluated separately at Gem City Chemicals for each of the two layers. As described previously, a low-permeability till layer is present beneath Gem City Chemicals, Inc. and for at least one-half mile surrounding the site. This till layer effectively isolates the uppermost, unconfined aquifer at Gem City Chemicals, Inc. from deeper confined aquifers.

Ground-water flow directions in the lower aquifer have changed considerably during the past thirty years, due to changes in water usage in the surrounding areas. Clean Tech,

1994 reported that potentiometric maps compiled by Norris & Spiker for 1959 and 1960 (prior to the time when the Miami South Well Field began operations) show groundwater flow to the southwest, toward a wide cone of depression developed beneath the central business district of Dayton, and also toward industrial facility water supply wells to the southwest. A major cone of depression developed beneath the South Miami Well Field following the beginning of water production from the well field in the early 1960's. Maps compiled by CH<sub>2</sub>M Hill for 1972 show this cone of depression. The location of Gem City Chemicals, Inc. appeared to be on or near a groundwater divide between these two cones of depression, and the direction of groundwater flow at the DTPP site was thought to be either to the north or to the south, or it could fluctuate depending on recharge variations and variability in the pumping rates at the city's well field.

In August of 1988, the City of Dayton adopted a Well Field Protection Program to protect its drinking water supplies. The southern limit of the Miami Well Field Protection Overly District is Stanley Avenue, which borders the DTPP property. Well yields for wells within the area as published in Norris & Spiker (1966) range from 20 gallons per minute (No. 209) to a maximum of 1,000 gallons per minute (No. 212), as reported by Clean Tech, 1994. A test well in the South Miami Well Field pumped at a rate of 2,283 gallons per minute. The City's Mad River Well Field is approximately two miles to the east of the site and does not receive any recharge from this area as reported by QSource Engineering.

The unconfined aquifer has been widely utilized as a water source throughout the region. The main source of groundwater recharge to the unconfined aquifer is infiltration from local rivers. Direct recharge by precipitation, and recharge by subsurface flow from the edges of buried valleys provide lesser amounts of recharge to the aquifer. Available annual precipitation is higher during the months of March through June in the Dayton region.

Wells constructed in portions of the aquifer having a substantial saturated thickness may yield up to 1,000 gallons per minute (gpm) for a short period of time, although yields of

100 gpm to 500 gpm are more common. The presence of thick layers of till within the aquifer has been shown to decrease these short-term yields up to 50%. Areas having thin deposits of sand and gravel, such as locations near the edges of the buried valleys, have been shown to yield substantially less water.

The unconfined aquifer is generally underlain by a till layer present at approximately 80 feet BGS. This till layer appears to be laterally persistent across areas on the order of a mile, but evidence suggests it may be discontinuous on a larger regional scale across the entire buried valley in the Dayton region. The till has been found to be absent from some locations in the region due either to non-deposition or erosion.

Till layers have been reported as massive clay units, or as zones of alternating clay with stratified sand and gravel. Till has been shown to have low permeability and yields little water to wells. It has been used as a local aquifer for domestic or farm water supply wells (up to 12 gpm) near the edges of the buried valley deposits when sand and gravel content are high within the till.

Till layers generally act as confining layers, controlling aquifer recharge and creating barriers to groundwater flow. Norris examined recharge to the aquifer underlying a regional till layer and found that leakage through the confining layer was responsible for the majority of the groundwater recharge to the lower aquifer. This leakage was not assumed to represent a breach in the till layer, but rather uniform transmission of water through the till under a hydraulic gradient. A leakage coefficient was computed for the till of 0.003 gallons per day per cubic foot.

Regional studies indicate that a second recognizable sand and gravel outwash deposit underlies the till layer found at approximately 80 feet BGS. This lower aquifer behaves as a confined or semi-confined aquifer. However, if the till layer is thin or absent the hydraulically connected sand and gravel units will act as a single unconfined aquifer. This

second recognizable sand and gravel deposit is the semi-confined aquifer examined during this investigation.

The semi-confined aquifer is utilized as a major water source throughout the region for municipal supplies. The main source of groundwater recharge to the semi-confined aquifer is the overlying unconfined aquifer. Wells constructed in the semi-confined aquifer have routinely yielded 2,000 gpm for extended periods of time. Chemical quality of the water from the unconfined and semi-confined aquifers has been shown to be similar by Walton and Scudder, 1960.

Deep wells in the region suggest discontinuous till layers may exist within the second glacial outwash unit (the semi-confined aquifer), and additional semi-confined or confined aquifers exist at greater depths. These deeper aquifers are believed to be separated by till layers in much the same way as the shallower geologic units.

# 6.2 - Site Hydrogeology

The hydrogeology of the site was examined through information gathered from groundwater monitoring wells, with additional information obtained from the records of the production wells located on the property. The hydrogeology of the site is typical of the region. The permeable subsurface materials consist of glacial and glacial-fluvial sediments made up of sand and gravel with minor amounts of silt. The sand and gravel is interbedded with till and other clay layers composed of massive clay units, or zones of clay with sand and gravel. Two aquifers were examined in this investigation: the shallow unconfined aquifer and the deeper semi-confined aquifer.

#### Slug Testing

Hydraulic conductivity testing was attempted following collection of the first round of groundwater samples on December 15 and 16, 1994. The planned hydraulic conductivity testing employed slug testing techniques. A slug test consists of causing a water-level

change within a well and measuring the rate at which the water level returns to its initial level. This rate of recovery can be related to the hydraulic conductivity of the surrounding aquifer material.

Approximately one gallon of deionized water was added to the wells. The volume of water added was chosen to displace an equal volume of water equivalent to approximately five feet of standing water in the well. The slug tests were conducted using a digital data logger and pressure transducer. All equipment in contact with the well or groundwater was decontaminated using a solution of ten percent methanol in potable water. The length of the transducer cable and the probe were wiped clean using the methanol solution, discarding and replacing the towel as it became soiled, and rinsed using deionized water.

Slug testing of the groundwater monitoring wells was attempted, but provided minimal information. The aquifer materials encountered in both the unconfined and semi-confined aquifers were recognized to have extreme permeability and porosity when first described during the well and boring installations. Published reports indicate these aquifers can have porosity values up to approximately 35%, and transmissibility of approximately 280,000 gallons per day per foot in the Dayton area. This extreme permeability as encountered at the site made the slug testing method minimally effective as a means of determining representative aquifer characteristics.

# 6.2.1 - Unconfined Aquifer

The hydrogeology of the unconfined aquifer was investigated through analysis of water level elevations, interpretation of groundwater flow direction and gradients, examination of the relationship between the water level measurements and confining units as encountered in the wells, and the pattern of contamination as observed in the groundwater and soil across the site.

#### Trichloroethene - Drawing 15

The isoconcentration map for trichloroethene was contoured using 50 ppb, 100 ppb, 500 ppb, 1,000 ppb and 5,000 ppb contours. The isoconcentration map shows elevated levels of trichloroethene were detected in the soils in the following areas:

- Area A: within the northern portion of Buildings 40A and 40B; in the
  paved area immediately north of those buildings; and in the area of the
  former TCA tank south of Building 53;
- Area B: in the area to the west of Building 47 with contamination noted to a lesser degree than seen in Area A or Area C;
- Area C: along the southern portion of the site adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40, 40A and 40B.

The soil sample results and interpreted distribution of contaminants revealed the following patterns of contamination in the soil across the Chrysler Corporation property:

- The levels of tetrachloroethylene in the soil appear to be greatest in the central portion of the facility within Area A and Area B. The distribution of the tetrachloroethylene controls the total VOCs distribution pattern in this portion of the site. The affected areas are within the northern portion of Buildings 40A and 40B; in the paved area immediately north and east of those buildings; in the area of the former TCA tank south of Building 53; to the north of Building 59; in the area near Building 47 extending northward and toward the east; and in the area north of the boiler house and northeast of Building 47;
- The levels or trichloroethene in the soil appear to be greatest along the southern portion of the site within Area C. The distribution of the trichloroethene controls the total VOCs distribution pattern in this portion of the site. The affected area is adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40, 40A and 40B.

These findings are in agreement with the work completed during previous soil investigations at the facility, and with the soil vapor survey completed as a part of this investigation. The soil vapor survey permitted identification of recognizable areas of the site having a particular pattern of VOC contamination in the vadose zone. These areas (Area A, B, C) were presented as a working model of the site conditions useful in the discussion of soil contamination patterns, and identification of potential contamination sources.

The areas near Buildings 40A and 40B, the area to the south of Building 53 near the former TCA tanks, the area east of Building 50, and the western and southern portions of the former Maxwell Complex are identified as areas where elevated levels of VOCs are present in the soil. Significantly elevated levels of VOCs have been identified in the soil in close proximity to the local water table.

#### Chemical Analysis: Metals

The laboratory analysis detected the following metals in the soil samples from the soil borings and the monitoring well borings. The results are listed in the following tables.

Soil Metals for Soil Boring Samples

Metals	SB-1	SB-2	SB-3	5B-4	SB-5	SB-6	SB-7	SB-8	SB-9	SB-10
	@ 9'	@ 19'	@ 14	@ 14"	@ 29'	@ 14	(a) 24'	@ 24'	@ 19"	@ 291
Aluminum	1600	1400	1900	1900	1700	1800	1500	1900	1600	1400
Barium	7.2	12	11	11	12	11	12	11	8	9.6
Beryllium	ND	ND	ND	0.22	ND	ND	ND	ND	ND	ND
Cadmium	0.65	0.63	0.58	0.67	0.53	0.72	0.73	0.47	0.72	0.46
Calcium	140,000	110,000	130,000	260,000	130,000	100,000	140,000	110,000	77,000	70,000
Chromium	7.1	6.9	6.9	5.6	8.7	5	4.5	6.8	5.2	4.8
Cobalt	ND	ND	ND	4.6	ND	ND	ND	ND	ND	ND
Silver	3.9	3.1	3.6	2.7	3.3	3.2	2.8	2.9	3	2.4
Copper	2.2	8.9	8.7	8.7	8	9.7	10	7.2	8.1	8.1
Sodium	280	140	190	200	210	210	180	160	290	140
Vanadium	ND	ND	ND	8.2	ND	ND	ND	ND	ND	ND
Zinc	16	18	20	16	14	17	24	16	19	22
Antimony	ND	ND	ND							
Arsenic	4.8	4.3	5.6	3	16	5.4	6.4	6.5	5.2	3
Lead	2	2	2	2	3	2.4	2.5	4.1	1.6	2.8
Selenium	ND	ND	ND							
Thallium	1.2	1.3	0.8	1.7	1.4	1.8	1.2	1.1	1.3	0.94
Mercury	ND	ND	ND							

Metals in Soil Samples from Soil Borings.

All Results in Parts per Million (mg/kg).

ND denotes analyte was not detected at the laboratory detection levels.

### Soil Metals for Monitoring Well Boring Samples

Metals	MWAI	MWA2	MWA3	MWA4	MWA5	MWA6
	@ 24'	@ 19'	@ 24'	@ 24'	@ 24"	@ 24"
Aluminum	2300	2300	3300	1900	1800	2300
Barium	12	13	11	15	9	14
Beryllium	0.099	0.21	0.26	ND	0.1	ND
Cadmium	0.3	0.71	0.78	0.66	0.68	0.71
Calcium	93,000	220,000	90,000	100,000	150,000	100,000
Chromium	7.4	9	7.1	6.4	7.9	5
Cobalt	5.7	4.9	4.1	2.6	5.4	2.8
Silver	3.7	2.7	3.3	2.1	4.2	2
Соррег	9	9.6	7.3	11	8.7	11
Sodium	140	150	120	140	190	120
Vanadium	9.7	9.6	8.7	5.6	8.7	5.4
Zinc	17	19	20	23	15	19
Antimony	ND	ND	ND	ND	ND	ND
Arsenic	6.5	4.4	5.6	2.6	2.5	4.2
Lead	4	3.2	3.3	3.2	3.5	3
Selenium	ND	ND	ND	ND	ND	ND
Thallium	2	1.9	0.9	0.82	2.6	0.61
Мегсигу	ND	ND	ND	ND	ND	ND

Metals in Soil Samples from Monitoring Wells. All Results in Parts per Million (mg/kg). ND denotes analyte was not detected at the laboratory detection levels.

Soil Metals for Monitoring Well Boring Samples

Metals	MWB2 @ 24'	MWB3 @ 24'	MWB4 (a) 24'	MWB4 @ 19'	MWB5 @ 24'	MWB6 @ 24'
Aluminum	2100	1500	2000	1900	2100	2900
Barium	12	12	14	13	10	10
Beryllium	0.088	0.3	0.23	0.21	0.27	0.28
Cadmium	0.62	0.68	0.62	0.67	0.55	0.77
Calcium	83,000	140,000	190,000	260,000	73,000	100,000
Chromium	7.2	6.8	6.5	6.2	6.3	7
Cobalt	5.1	8.2	4.4	4.8	8.3	4.4
Silver	3.7	3.1	2.5	2.5	2.7	3.1
Copper	10	12	8.8	9.1	9.1	8.1
Sodium	140	190	160	130	170	130
Vanadium	9.5	9.5	7.4	8.6	11	8
Zinc	22	19	18	18	22	18
Antimony	ND	ND	ND	ND	ND	ND
Arsenic	7.1	6	4.1	4.1	8.4	3.5
Lead	3.5	3	4.6	4.5	3.3	2.6
Selenium	ND	ND	ND	ND	ND	ND
Thallium	2	4.6	1.6	1.6	2.1	0.75
Mercury	ND	ND	ND	ND	ND	ND

Metals in Soil Samples from Monitoring Wells. All Results in Parts per Million (mg/kg). ND denotes analyte was not detected at the laboratory detection levels.

The detected metal species were compared to the mean concentrations of those metals occurring naturally in soils of the Eastern United States (includes Ohio). This information was compiled by the United States Geological Survey and presented in the 1984 publication Element Concentrations in Soils and Other Surficial Materials of the Conterminous United States, Professional Paper 1270, by Shacklette and Boerngen. A summary of this comparison is presented in the following table.

Comparison of Mean Metals Concentration and Sample Concentrations

Metals Species	Concentration Range (# Samples)	Eastern US Mean (Includes Ohio)
Aluminum	1400-2900 ppm (all)	5.7 %
Barium	7.2 - 15.0 ppm (all)	420 ppm
Beryllium	0.099-0.3 ppm (10 of 21)	0.85 ppm
Cadmium	0.3-0.78 ppm (all)	None Given
Calcium	70,000-260,000 ppm (all)	0.65 %
Chromium	4.5-9.0 ppm (all)	52 ppm
Cobalt	2.6-8.3 ppm (12 of 21)	9.2 ppm

Metals Species	Concentration Range (# Samples)	Eastern US Mean (Includes Ohio)	
Silver	2.0-4.2 ppm (all)	None Given	
Copper	2.2-12.0 ppm (all)	22 ppm	
Sodium	120-290 ppm (all)	0.78 %	
Vanadium	5.4-11.0 ppm (12 of 21)	66 ppm	
Zinc	14-24 ppm (all)	52 ppm	
Antimony	All Not Detected		
Arsenic	2.5-16.0 ppm (all)	7.4 ppm	
Lead	1.6-4.6 ppm (all)	17 ppm	
Selenium	All Not Detected		
Thallium	0.61-4.6 ppm (all)	None Given	
Mercury	All Not Detected		

A total of eighteen metals species were targeted for analysis. Three of the targeted metals were below detectable levels in the samples. Thirteen metals were found to be within the Eastern United States mean values for those metals. Two metals were found to be present at levels significantly above the Eastern United States mean values. Calcium was measured in the range between 7 % and 26 % for all the soil samples. This range is well above the Eastern United States mean value of 0.65 %, but within the expected range for calcium in the Dayton, Ohio area because the sands and gravels underlying the area are known to be calcareous sediments, that is, made up of calcium-bearing minerals. Arsenic was measured in the range between 2.5 ppm and 16.0 ppm. The Eastern United States mean value is 7.4 ppm, however, the observed range of naturally occurring arsenic was reported up to 73 ppm for the Eastern United States.

Three metals, cadmium, silver, and thallium, were not assigned mean values in the referenced report. Common ranges for cadmium and silver were compiled by Dragun, 1988 and were compared to the analyzed samples. Cadmium concentrations were between 0.3 and 0.78 ppm for all samples which is within the common range reported for cadmium of 0.01 to 7.0 ppm. Silver concentrations were within the range of 2.0 to 4.2 ppm for all samples which is within the common range reported for silver of 0.1 to 5.0 ppm. Thallium was detected in all samples at concentrations in the range of 0.61 to 4.6 ppm.

#### Chemical Analysis: TOC

Soil samples were collected and analyzed for TOC. The samples collected for TOC analysis were selected as representative of the subsurface materials encountered. Laboratory results for the TOC analysis are listed below:

**Total Organic Carbon in Soil Samples** 

SB-4	MWB1	MWB3	MWB4	MWB5
@ 30'	@ 49'	@ 49'	@ 49'	@ 49'
35,000	17,000	100	18,000	21,000

TOC in Soil Samples.

All Results in Parts per Million (mg/kg).

Organic carbon typically exists in a variety of oxidation states. Some of these carbon compounds can be oxidized further by biological or chemical processes. TOC provides a direct expression of the total organic content of the sample independent of the oxidation state of the organic matter. TOC does not measure other organically bound elements that can contribute to the total oxygen demand during the oxidation process.

The TOC concentrations in the range of 17,000 to 35,000 ppm indicate significant organic carbon is available to be oxidized further by biological or chemical processes. The relatively low concentration of TOC found in MWB-3 @ 49' indicates relatively little organic carbon is available for biological or chemical oxidation in that sample.

#### QA/QC

The QA/QC program for chemical analysis of the soil samples consisted of the collection and analysis of duplicate samples, spiked samples, and equipment blank. Two duplicate soil samples were collected and analyzed as was done for all soil samples. MWA-7 is a duplicate sample of MWB-2 from 24'. MWB-7 is a duplicate of MWB-4 from 19'. Soil samples were spiked in the laboratory, analyzed and the results were retained at the laboratory in their records retention system. One equipment blank was collected by pouring deionized water over the decontaminated sampling equipment and collecting the

rinsate for laboratory analysis. The rinsate water sample was identified as MWA-5-24. No VOCs were detected in the equipment blank sample indicating the decontamination of the equipment was thorough and no cross-contaminants were introduced to the samples.

The detected VOCs and targeted metals in the duplicate sample pairs are presented in the following table. Overall reproducibility of the laboratory results between the duplicate samples showed that sample handling did not appear to introduce any significant variability in the results.

**Duplicate Soil Sample Pairs** 

	MWB2		%	MWB4		%
Duplicate Samples	@ 24'	MWA7	Difference	@ 19'	MWB7	Difference
tetrachloroethylene	4000	2600	35	ND	ND	0
1,1,1-trichloroethane	ND	21	>100	ND	ND	0
trichloroethene	ND	38	>100	ND	ND	0
dichloromethane	20	ND	>100	ND	ND	0
Aluminum	2100000	2200000	4	1900000	1800000	5
Barium	12000	13000	7	13000	11000	15
Beryllium	88	95	7	210	190	9
Cadmium	620	750	17	670	630	6
Calcium	83000000	120000000	31	260000000	230000000	11
Chromium	7200	8400	14	6200	5900	5
Cobalt	5100	5800	12	4800	4200	12
Silver	3700	3400	8	2500	3600	31
Соррег	10000	11000	9	9100	8400	8
Sodium	140000	150000	7	130000	130000	0
Vanadium	9500	9200	3	8600	7800	9
Zinc	22000	25000	12	18000	16000	11
Antimony	ND	ND	0	ND	ND	0
Arsenic	7100	4000	44	4100	2500	39
Lead	3500	3600	3	4500	3000	33
Selenium	ND	ND	0	ND	ND	0
Thallium	2000	1800	10	1600	1900	16
Mercury	ND	ND	0	ND	ND	0

VOCs detected in Duplicate Soil Samples shown in Parts per Billion (ug/kg), Metals Targeted in Duplicate Soil Samples shown in Parts per Million (mg/kg). ND denotes analyte was not detected at the laboratory detection levels.

68

# 7.2 - Geotechnical Analysis and Findings

Geotechnical analysis of selected soil samples was completed to evaluate potential remedial technologies. Samples were selected for geotechnical analysis based on them being representative of the subsurface materials encountered, and at a depth where a remedial technology might be applied. The QA/QC program for geotechnical analysis of the soil samples specified laboratory test procedures which followed ASTM procedures or approved equivalent methods for analysis of textural gradation and percent moisture.

The results of the geotechnical analysis are included as Attachment H for the samples collected from the soil borings, and as Attachment K for the samples collected from the groundwater monitoring well borings, both of which are presented in Volume III of this report. Soil samples collected for geotechnical analysis are listed below:

#### Geotechnical Samples from Soil Borings

Boring	Depth	Analysis Performed
SB-1	14-16 ft	% Moisture
SB-2	14-16 ft	% Moisture
SB-3	19-21 ft	% Moisture
SB-5	14-16 ft	Textural Gradation, % Moisture
SB-6	19-21 ft	Textural Gradation, % Moisture
SB-10	14-16 ft	Textural Gradation, % Moisture

#### Geotechnical Samples from Monitoring Well Borings

Well	Depth	Analysis Performed
MWA-4	39-41 ft	Textural Gradation
MWA-5	34-36 ft	Textural Gradation
MWB-2	74-76 ft	Textural Gradation
MWC-1	104-106 ft	Textural Gradation
MWC-2	114-116 ft	Textural Gradation
MWC-3	76-78 ft	Textural Gradation

The laboratory results textural gradation and % moisture are presented in the following tables. Textural gradation was found by sieve analysis with the percent fines (or percent passing the sieve) reported. The % moisture test was conducted using ASTM D-2216.

Boring SB-5 Sieve Size	Textural Gradation Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	100
0.5"	81.8
0.375"	78.7
#4	59.6
#10	40.0
#20	27.9
#40	19.1
#60	13.5
#100	10.7
#140	9.5
#200	8.5



The SB-5 soil sample is described as a brown fine gravelly fine to coarse grained sand.

Boring SB-6 Sieve Size	Textural Gradation Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	92.6
0.5"	87.5
0.375"	81.5
#4	63.0
#10	42.9
#20	28.7
#40	17.1
#60	12.3
#100	9.9
#140	8.9
#200	8.0



The SB-6 soil sample is described as a brown fine gravelly fine to coarse grained sand with a trace of silt.

Boring SB-10 Sieve Size	Textural Gradation Percent Finer
2.5"	
2"	
1.5"	
1"	100
0.75"	88.8
0.5"	70.9
0.375"	62.0
#4	45.7
#10	31.1
#20	22.3
#40	15.9
#60	11.6
#100	9.1
#140	8.1
#200	7.4

The SB-10 soil sample is described as a brown sandy fine gravel with a trace of silt.

Well MWA-4	Textural Gradation
Sieve Size	Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	100
0.5"	100
0.375"	100
#4	97.3
#10	93.5
#20	82,0
#40	41.9
#60	13.1
#100	5.6
#140	4.2
#200	3.3

The MWA-4 soil sample is described as dark brown fine to medium sand with a trace of silt & fine gravel.

Well MWA-5 Sieve Size	Textural Gradation Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	100
0.5"	89.0
0.375"	84.3
#4	75.3
#10	60.6
#20	40.0
#40	21.8
#60	11.5
#100	6.7
#140	5.4
#200	4.6

The MWA-5 soil sample is described as dark brown fine to medium sand with fine gravel & a trace of silt.

Well MWB-2	Textural Gradation
Sieve Size	Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	92.0
0.5"	83.2
0.375"	81.8
#4	74.0
#10	59.0
#20	39.7
#40	24.1
#60	16.4
#100	11.6
#140	• 9.8
#200	8.5

The MWB-2 soil sample is described as gray gravelly fine to coarse sand with a trace of silt.

Well MWC-1 Sieve Size	Textural Gradation Percent Finer
2.5"	100
2"	100
1.5"	100
1"	86.6
0.75"	77.3
0.5"	70.6
0.375"	58.9
#4	41.1
#10	24.2
#20	13.7
#40	7.6
#60	4.7
#100	3.5
#140	2.8
#200	2.3

The MWC-1 soil sample is described as gray sandy fine to coarse grained gravel with a trace of silt.

Well MWC-2	Textural Gradation
Sieve Size	Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	100
0.5"	96.5
0.375"	96.5
#4	92.4
#10	83.1
#20	56.1
#40	20.3
#60	8.6
#100	6.1
#140	5.5
#200	5.2

The MWC-2 soil sample is described as gray fine to medium sand with a trace of silt & fine gravel.

Well MWC-3 Sieve Size	Textural Gradation Percent Finer
2.5"	100
2"	100
1.5"	100
1"	100
0.75"	100
0.5"	97.9
0.375"	95.3
#4	89.7
#10	80.6
#20	64.8
#40	41.1
#60	25.1
#100	18.4
#140	15.8
#200	13.5

The MWC-3 soil sample is described as dark brown fine to medium sand with a little silt & fine gravel.

The % moisture content analysis results are presented in the following table. Note that all soil samples were collected above the water table.

Geotechnical Analysis for % Moisture

Boring	Depth	% Moisture		
SB-1	14-16 ft	3.7 %		
SB-2	14-16 ft	6.2 %		
SB-3	19-21 ft	4.3 %		
SB-5	14-16 ft	4.9 %		
SB-6	19-21 ft	4.7 %		
SB-10	14-16 ft	4.5 %		

# Section 8.0 - Findings and Discussion for Groundwater Samples

Groundwater samples were collected from each of the fifteen groundwater monitoring wells during two sampling events. The wells were sampled twice to determine if there were any effects on water quality due to seasonal water level fluctuations. The first sampling event was completed in December, 1994, and the second was completed in February, 1995. Both groundwater sampling events were performed using EPA approved standard sampling procedures.

Groundwater samples were collected and analyzed for TCL VOCs, and TAL metals. Each well had remained static for approximately two weeks following well development prior to collection of the first round of samples. All wells were screened for evidence of organic vapors prior to collection of the first round of groundwater samples in December 1994 using a PID. The PID was inserted into each open well top immediately upon opening the well. The maximum instantaneous PID measurements were recorded in the field logbook as measured immediately upon opening the well top.

No significantly elevated levels of volatile organic compounds were detected using the PID, and none were measured in excess of levels of concern as described in the health and safety plan. The PID levels in the open wells were immediately reduced to ambient background level as each well vented following opening. The wells were not screened for evidence of volatile organic compounds at the time of the second round of groundwater samples in February 1995 because the ambient air temperature was so low as to render PID measurements unreliable, and the first round PID measurements did not encounter any significantly elevated levels of volatile organic compounds.

Water levels were measured from the top of the PVC casing prior to well purging. An interface probe was used to measure water levels and the thickness of any non-aqueous phase product (LNAPLs or DNAPLs) prior to purging the well in preparation for groundwater sampling. The water levels and notes regarding the possible presence of

non-aqueous phase product were recorded for both sampling rounds. No evidence of LNAPLs or DNAPLs was found during either of the two groundwater sampling rounds, or during collection of additional water level measurements in January 1995. The results of the laboratory analysis for VOCs and metals are presented with a discussion of those findings.

The QA/QC program for chemical analysis of the groundwater samples consisted of the collection and analysis of duplicate samples, spiked samples, equipment blanks and trip blanks. The purpose of this program was to ensure the analyses performed by the analytical laboratory are reproducible. The chain of custody documentation, any QA/QC sample analytical results and the laboratory results for the groundwater samples are included as Attachment M (First Round) and Attachment N (Second Round), both of which are presented in Volume III of this report.

The findings of the laboratory analysis of both rounds of the groundwater samples are presented separately for the VOCs and metals. The water level measurements are presented with an interpretation of groundwater flow directions for the three sets of water level measurements collected.

The contaminant distribution patterns as seen in the groundwater samples are discussed for all the findings including both rounds of groundwater samples and all water level measurements. The information presented for the MWA and MWB wells is applicable to the unconfined aquifer.

# 8.1 - VOCs Analysis and Findings

# First Round Groundwater Samples

The laboratory analysis detected several volatile organic compounds in the first round groundwater samples. The samples containing these detected compounds are listed in the following tables.

**VOCs in First Round Groundwater Samples** 

VOCs	MWA-1	MWA-2	MWA-3	MWA4	MWA-5	MWA-6
tetrachloroethylene	2500	2400	2200	15	240	1.9
trichloroethene	350	110	240	76000	1100	2600
benzene	ND	ND	ND	ND	ND	ND
1,2-dichloroethene (total)	110	8.3	160	30000	61	74
cis-1,2 dichloroethylene	110	8.3	160	30000	59	73
trans-1,2 dichloroethylene	ND	· ND	4.3	110	2.1	1.3
1,1,1-trichloroethane	3300	8600	5500	28	2600	640
1,1,2-trichloroethane	ND	ND	ND	ND	2.1	ND
chloroform	ND	ND	ND	ND	ND	ND
1,1-dichloroethane	65	3.6	590	ND	74	ND
1,2 -dichloroethane	ND	ND	5	ND	5	ND
1,1-dichloroethene	81	84	240	42	260	19
trichlorofluoromethane	ND	ND	ND	ND	2.1	ND
dichlorodifluoromethane	470	94	79	ND	1.5	ND
vinyl chloride	ND	ND	1.7	1100	ND	ND
chloroethane	ND	ND	1.3	ND	ND	ND

VOCs in Round #1 Groundwater Samples. All Results in Parts per Billion (ug/l). ND denotes analyte was not detected at the laboratory detection levels.

**VOCs in First Round Groundwater Samples** 

VOC	MWB-1	MWB-2	MWB-3	MWB-4	MWB-5	MWB-6
tetrachloroethylene	ND	1.6	ND	ND	ND	1.8
trichloroethene	ND	ND	9900	8.7	1	2400
benzene	ND	ND	2.4	ND	ND	ND
1,2-dichloroethene (total)	ND	ND	1700	ND	ND	310
cis-1,2 dichloroethylene	ND	ND	1700	ND	ND	290
trans-1,2 dichloroethylene	ND	ND	24	ND	ND	17
1,1,1-trichloroethane	ND	ND	320	ND	ND	81
1,1,2-trichloroethane	ND	ND	2.5	ND	ND	ND
chloroform	ND	ND	1.9	ND	ND	ND
1,1-dichloroethane	ND	ND	14	ND	ND	85
1,2 -dichloroethane	ND	ND	11	ND	ND	ND
1,1-dichloroethene	ND	ND	27	ND	ND	32
trichlorofluoromethane	ND	ND	3.8	ND	ND	ND
dichlorodifluoromethane	ND	ND	ND	ND	ND	34
vinyl chloride	ND	ND	21	ND_	ND	46
chloroethane	ND	ND	ND	ND	ND	ND

VOCs in Round #1 Groundwater Samples. All Results in Parts per Billion (ug/l). ND denotes analyte was not detected at the laboratory detection levels.

No volatile organic compounds were detected in the first round of groundwater samples collected from MWC-1, MWC-2, or MWC-3.

# Second Round Groundwater Samples

The laboratory analysis detected several volatile organic compounds in the second round groundwater samples. The samples containing these detected compounds are listed in the following tables.

**VOCs in Second Round Groundwater Samples** 

VOCs	MWA-1	MWA-2	MWA-3	MWA-4	MWA-5	MWA-6
tetrachloroethylene	2200	1500	1200	11	140	7.3
trichloroethene	180	250	130	37000	1500	1400
1,2-dichlorobenzene	ND	1.8	ND	ND	ND	ND
1,1-dichloropropene	ND	ND	2.2	ND	ND	ND
1,2-dichloroethene (total)	110	8.9	140	16000	190	110
cis-1,2 dichloroethylene	110	8.5	120	16000	190	110
trans-1,2 dichloroethylene	1.3	3.6	17	75	3.3	3.1
1,1,1-trichloroethane	2000	3800	2300	48	3600	420
1,1,2-trichloroethane	ND	ND	ND	1.8	2.6	ND
1,1,1,2-tetrachloroethane	ND	ND	ND	ND	1.4	ND
chloroform	ND	5	2.2	1.2	2	ND
1,1-dichloroethane	69	3.6	710	12	93	4.3
1,2 -dichloroethane	ND	ND	9.4	ND	7.5	ND
1,1-dichloroethene	100	270	100	37	390	34
dichlorodifluoromethane	70	42	17	ND	ND	ND
vinyl chloride	ND	77	53	1400	ND	ND
chloromethane	ND	ND	2.6	ND	ND	ND
chloroethane	ND	ND	2.6	ND	ND	ND

VOCs in Round #2 Groundwater Samples. All Results in Parts per Billion (ug/l). ND denotes analyte was not detected at the laboratory detection levels.

**VOCs in Second Round Groundwater Samples** 

VOCs	MWB-1	MWB-2	MWB-3	MWB-4	MWB-5	MWB-6
tetrachloroethylene	ND	ND	ND	5.6	ND	1.2
trichloroethene	ND	ND	1100	6.8	ND	760
benzene	ND	ND	2.1	ND	ND	ND
1,2-dichloroethene (total)	ND	ND	3500	1.7	ND	230
cis-1,2 dichloroethylene	ND	ND	3500	1.7	ND	210
trans-1,2 dichloroethylene	ND	ND	28	ND	ND	14
1,1,1-trichloroethane	ND	ND	350	1.3	ND	84
1,1,2-trichloroethane	ND	ND	3.9	ND	ND	ND
chloroform	ND	ND	4.8	ND	ND	ND
1,1-dichloroethane	ND	ND	28	ND	ND	130
1,2 -dichloroethane	ND	ND	42	ND	ND	1
1,1-dichloroethene	ND	ND	28	ND	ND	27
trichlorofluoromethane	ND	ND	ND	ND	ND	ND
dichlorodifluoromethane	ND	ND	ND	ND	ND	4.2
vinyl chloride	ND	ND	26	ND	ND	130
chloromethane	ND	ND	ND	ND	ND	14

VOCs in Round #2 Groundwater Samples. All Results in Parts per Billion (ug/l). ND denotes analyte was not detected at the laboratory detection levels.

No volatile organic compounds were detected in the second round of groundwater samples collected from MWC-1, MWC-2, or MWC-3.

# **Findings**

Three isoconcentration contour maps were generated which show the interpreted distribution of volatile organic compounds in groundwater within the unconfined aquifer. Maps were prepared showing: total VOCs (sum of the detected compounds in each well), tetrachloroethylene, and trichloroethene. These maps are included as Drawings 16, 17, and 18 for the first round results, and as Drawings 19, 20, and 21 for the second round results. All Drawings are presented in Volume II of this report.

No VOCs were detected in any of the three deeper wells during analysis of the first round or the second round groundwater samples. Recognizable areas of the site having a particular pattern of VOC contamination in the vadose zone (Area A, B, C) were presented as a working model. This model continues to be useful in the discussion of

contamination patterns, and identification of potential contamination sources. Refer to Figure 3 (see Volume II of this report) for a map of the facility showing these areas.

# Total VOCs - Drawing 16 (First Round) and Drawing 19 (Second Round)

The isoconcentration map for total VOCs was contoured using 50 ppb, 100 ppb, 500 ppb, 1,000 ppb, 5,000 ppb, 10,000 ppb, 50,000 ppb, and 100,000 ppb contours. The two isoconcentration maps show only minor variations in the contaminant distribution patterns between the two sampling rounds. The isoconcentration map shows elevated levels of total VOCs were detected in the groundwater within the unconfined aquifer in the following areas:

- Area A: within the northern portion of Buildings 40A and 40B; in the paved area immediately north and east of those buildings; in the southern portion of Building 50, and in the area of the former TCA tank south of Building 53;
- Area B: to the north of Building 59; in the area near Building 47 extending northward and toward the west; and in the area north of the boiler house;
- Area C: along the southern portion of the site adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40A and 40B.

# Tetrachloroethylene - Drawing 17 (First Round) and Drawing 20 (Second Round)

The isoconcentration map for tetrachloroethylene was contoured using 50 ppb, 100 ppb, 500 ppb, and 1,000 ppb contours. The two isoconcentration maps show only minor variations in the contaminant distribution patterns between the two sampling rounds. The isoconcentration map shows elevated levels of tetrachloroethylene were detected in the groundwater within the unconfined aquifer in the following areas:

• Area A: within the northern portion of Buildings 40A and 40B; in the paved area immediately north of those buildings; in the southern portion of Building 50, and in the area of the former TCA tank south of Building 53;

• Area B: to the north of Building 59; in the area near Building 47 extending toward the west; and in the area north and west of the boiler house.

# Trichloroethene - Drawing 18 (First Round) and Drawing 21 (Second Round)

The isoconcentration map for trichloroethene was contoured using 50 ppb, 100 ppb, 500 ppb, 1,000 ppb, 5,000 ppb, 10,000 ppb and 50,000 ppb contours. The two isoconcentration maps show only minor variations in the contaminant distribution patterns between the two sampling rounds. The isoconcentration map shows elevated levels of trichloroethene were detected in the groundwater within the unconfined aquifer in the following areas:

- Area A: within the northern portion of Buildings 40A and 40B and in the
  paved area immediately north of those buildings; this contaminant was
  found in the area of the former TCA tank south of Building 53 but at
  reduced levels;
- Area B: in the area to the north of Building 59 and west of Building 47
  with contamination noted across a broad area of the site but to a lesser
  degree than seen in Area A or Area C;
- Area C: significantly elevated levels of this contaminant were noted along the southern portion of the site adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40A and 40B.

# 8.2 - Metals Analysis and Findings

# First Round Groundwater Samples

Dissolved metals analysis was performed using field filtered groundwater samples. Eighteen metals were targeted in TAL metals analysis of the groundwater samples. The laboratory analysis detected the following metals in the groundwater samples. The first round groundwater samples containing these detected metals are listed in the following tables.

# Metals in First Round Groundwater Samples

Metals	MWA-1	MWA-2	E-AWM	MWA-4	MWA-5	MWA-6
Barium (ug/l)	210	160	250	320	290	150
Calcium (mg/l)	130	100	150	150	140	140
Cobalt (ug/l)	ND	ND	ND	ND	ND	61
Silver (ug/l)	45	34	38	ND	50	ND
Sodium (mg/l)	73	72	96	85	100	81
Vanadium (ug/l)	ND	ND	ND	29	ND	ND
Antimony (ug/l)	6.6	5	4.5	ND	4.9	4.3
Thallium (ug/l)	13	15	21	17	11	8.4

Dissolved Metals in Round #1 Groundwater Samples.

All Results Units as Indicated.

Metals in First Round Groundwater Samples

Metals	MWB-1	MWB-2	MWB-3	MWB-4	MWB-5	MWB-6
Barium (ug/l)	110	190	110	180	150	180
Calcium (mg/l)	150	160	170	100	150	140
Chromium (ug/l)	ND	ND	ND	ND	23	ND
Cobalt (ug/l)	ND	ND	ND	43	ND	ND
Silver (ug/l)	ND	ND	ND	33	36	45
Sodium (mg/l)	56	180	120	78	150	53
Vanadium (ug/l)	ND	36	42	30	41	ND
Zinc (ug/l)	ND	20	ND	ND	ND	ND
Antimony (ug/l)	ND	5.2	5.1	3	4.7	5.9
Arsenic (ug/l)	2.4	ND	ND	ND	ND	ND
Lead (ug/l)	ND	1.1	1.4	ND	1.1	ND
Thallium (ug/l)	13	42	23	15	26	4.7
Mercury (ug/l)	0.28	ND	ND	ND	ND	ND

Dissolved Metals in Round #1 Groundwater Samples.

All Results Units as Indicated.

ND denotes analyte was not detected at the laboratory detection levels.

Metals in First Round Groundwater Samples

Metals	MWC-1	MWC-2	MWC-3
Barium (ug/l)	210	220	130
Calcium (mg/l)	100	97	130
Chromium (ug/l)	23	23	ND
Silver (ug/l)	47	44	ND
Sodium (mg/l)	21	32	57
Vanadium (ug/l)	ND	75	31
Antimony (ug/l)	ND	5.7	5.9
Arsenic (ug/l)	7.5	6.9	ND
Lead (ug/l)	ND	ND	1
Thallium (ug/l)	4.8	5.7	11
Mercury (ug/l)	ND	0.58	ND

Dissolved Metals in Round #1 Groundwater Samples.

All Results Units as Indicated.

The concentrations of the detected metal species were similar when comparing the groundwater samples collected from the unconfined and the semi-confined aquifers. Naturally occurring calcium concentrations were examined by Walton and Scudder, 1960 in glacial aquifers like those found at the subject site. The calcium concentrations reported by Walton and Scudder were in the range of 104 to 144 ppm. The calcium concentrations in the first round groundwater samples were comparable to the published values.

## Second Round Groundwater Samples

Dissolved metals analysis was performed using field filtered groundwater samples. Eighteen metals were targeted in TAL metals analysis of the groundwater samples. The laboratory analysis detected the following metals in the groundwater samples. The results were reported by the laboratory in units of mg/l (parts per million), but were converted to the units as reported in the first round results for ease of comparison between the two sample rounds. The second round groundwater samples containing these detected metals are listed in the following tables.

Metals in Second Round Groundwater Samples

Metals	MWA-1	MWA-2	MWA-3	MWA-4	MWA-5	MWA-6
Barium (ug/l)	200	200	240	220	320	200
Calcium (mg/l)	110	110	140	130	120	140
Sodium (mg/l)	59	63	80	83	120	68
Vanadium ug/l)	ND	ND	20	ND	ND	ND
Thallium (ug/l)	11	9	14	12	21	14
Mercury (ug/l)	ND	ND	ND	0.22	ND	ND

Dissolved Metals in Round #2 Groundwater Samples.

All Results Units as Indicated.

Metals in Second Round Groundwater Samples

Metals	MWB-1	MWB-2	MWB-3	MWB-4	MWB-5	MWB-6
Aluminum (ug/l)	ND	150	ND	ND	180	ND
Barium (ug/l)	100	160	110	180	140	160
Calcium (mg/l)	110	130	150	96	140	130
Cobalt (ug/l)	34	32	ND	ND	29	20
Silver (ug/l)	ND	37	ND	ND	36	ND
Sodium (mg/l)	40	140	100	59	130	80
Antimony (ug/l)	4.4	ND	ND	ND	ND	ND
Arsenic (ug/l)	2.6	ND	ND	ND	ND	ND
Thallium (ug/l)	11	29	15	13	25	10

Dissolved Metals in Round #2 Groundwater Samples.

All Results Units as Indicated.

ND denotes analyte was not detected at the laboratory detection levels.

Metals in Second Round Groundwater Samples

Metals	MWC-1	MWC-2	MWC-3
Aluminum (ug/l)	140	ND	ND
Barium (ug/l)	210	250	140
Calcium (mg/l)	44	84	120
Cobalt (ug/l)	33	30	ND
Silver (ug/l)	46	40	ND
Sodium (mg/l)	7.9	20	43
Antimony (ug/l)	5.6	4.3	ND
Arsenic (ug/l)	10	6.3	ND
Thallium (ug/l)	4.5	5.4	7

Dissolved Metals in Round #2 Groundwater Samples.

All Results Units as Indicated.

ND denotes analyte was not detected at the laboratory detection levels.

As had been seen in the laboratory results from the first round groundwater samples, the concentrations of the detected metal species were similar in the second round groundwater samples when comparing the groundwater samples collected from the unconfined and the semi-confined aquifers.

# **QA/QC** for First Round Groundwater Samples

The QA/QC program for chemical analysis of the groundwater samples consisted of the collection and analysis of duplicate samples, spiked samples, and equipment and trip blanks. Duplicate samples were collected and analyzed for metals and VOCs as was done

for all the groundwater samples. Sample 94-2-MWC3-12 is a duplicate sample of 94-1-MWC3-12 for VOC analysis. Sample 94-2-MWA5-12 is a duplicate of 94-1-MWA5-12 for metals analysis. Samples were spiked in the laboratory, analyzed and the results were retained at the laboratory in their records retention system.

One equipment blank was collected by pouring deionized water over the decontaminated sampling equipment and collecting the rinsate for laboratory analysis. The rinsate water sample was identified as 94-3-RINS-12. No VOCs were detected in the equipment blank indicating no cross contamination of the samples took place associated with equipment decontamination procedures. One trip blank, identified as 94-4-TB-12, accompanied the samples during transit from the site to the laboratory. No VOCs were detected in the trip blank sample indicating cross-contamination did not take place during sample handling.

The detected VOCs and targeted metals in the duplicate sample pairs are presented in the following table. Overall reproducibility of the laboratory results between the duplicate samples showed that sample handling does not appear to have introduced any significant variability in the results.

**Duplicate First Round Groundwater Sample Pairs** 

Duplicate Samples		94-2- MWC3-12 (Duplicate)			94-2- MWA5-12 (Duplicate)	% Difference
All Targeted VOCs Not Detected	ND	ND	0			
Aluminum				ND	ND	0
Barium		•••		290	310	6
Beryllium				ND	ND	0
Cadmium			•••	ND	ND	0
Calcium				140000	140000	0
Chromium				ND	ND	0
Cobalt				ND	ND	0
Silver				50	34	32
Copper				ND	ND	0
Sodium				100000	110000	9
Vanadium				ND	49	>100
Zinc				ND	ND	0
Antimony	•••			4.9	3.3	33

Duplicate Samples	94-1- MWC3-12 (Sample)	The state of the s		94-2- MWA5-12 (Duplicate)	
Arsenic			 ND	ND	0
Lead			 ND	ND	0
Selenium			 · ND	ND	0
Thallium			 11	13	15
Mercury			 ND	ND	0

VOCs and Metals in Round #1 Duplicate Groundwater Samples, Results shown in Parts per Billion (ug/l).

ND denotes analyte was not detected at the laboratory detection levels.

# QA/QC for Second Round Groundwater Samples

The QA/QC program for chemical analysis of the groundwater samples consisted of the collection and analysis of duplicate samples, spiked samples, and equipment and trip blanks. Duplicate samples were collected and analyzed for metals and VOCs as was done for all the groundwater samples. Sample 95-2-MWB3-2 is a duplicate sample of 95-1-MWB3-2 for VOC analysis. Sample 95-2-MWB2-2 is a duplicate of 95-1-MWB2-2 for metals analysis. Samples were spiked in the laboratory, analyzed and the results were retained at the laboratory in their records retention system.

One equipment blank was collected by pouring deionized water over the decontaminated sampling equipment and collecting the rinsate for laboratory analysis. The rinsate water sample was identified as 95-4-EB-2. One VOC compound was detected in the equipment blank. Trichloroethene was detected at 1.2 ug/l (parts per billion). The low level of this compound as detected indicates significant cross-contamination of the samples did not take place due to equipment decontamination procedures.

One trip blank, identified as 95-4-TB-2, accompanied the samples during transit from the site to the laboratory. No VOCs were detected in the trip blank sample indicating cross-contamination did not take place during sample handling.

Overall reproducibility of the laboratory results between the duplicate samples showed that sample handling does not appear to have introduced any significant variability in the

results. The detected VOCs and targeted metals in the duplicate sample pairs are presented in the following table.

# **Duplicate Second Round Groundwater Sample Pairs**

Duplicate Samples	95-1- MWB3-2 (Sample)	95-2- MWB3-2 (Duplicate)	% Difference	95-1- MWB2-2 (Sample)	95-2- MWB2-2 (Duplicate)	% Difference
benzene	2.1	2.4	12			
chloroform	4.8	8.1	41			
1,1-dichloroethane	28	49	43			
1,2-dichloroethane	42	39	7			
1,1-dichloroethene	28	61	54			
cis-1,2 dichloroethylene	3500	1500	57			
trans-1,2 dichloroethylene	28	48	42			
1,2-dichloroethene (total)	3500	1500	57			
1,1,1-trichloroethane	350	230	34			
1,1,2-trichloroethane	3.9	5.1	24			
trichloroethene	1100	5700	81			
trichlorotrifluoromethane	ND	2.9	>100	***	***	
vinyl chloride	26	110	76			***
Aluminum				150	210	29
Barium				160	160	0
Beryllium	***			ND	ND	0
Cadmium				ND	ND	0
Calcium				130000	140000	7
Chromium				ND_	ND	0
Cobalt				32	29	9
Silver				37	36	3
Copper				ND	ND	0
Sodium				140000	150000	7
Vanadium				ND	ND	0
Zinc				ND	ND	0
Antimony				ND	ND	0
Arsenic				ND	ND	0
Lead		•••		ND	ND	0
Selenium	***			ND	ND	0
Thallium	***			29	27	7
Mercury				ND	ND	0

VOCs and Metals in Round #2 Duplicate Groundwater Samples, Results shown in Parts per Billion (ug/l).

# 8.3 - Water Levels and Groundwater Flow

Water level measurements were collected at each groundwater monitoring well on a monthly basis for a period of three months after completion of the wells. Measurements were obtained prior to the beginning of purging and groundwater sample collection at the time of the first groundwater sampling event on December 13-14, 1994, on January 24, 1995, and at the time of the second groundwater sampling event on February 20, 1995.

Three rounds of water level measurements were collected to observe temporal variations in groundwater levels during the reporting period. The depth to water in each monitoring well was measured and referenced to a standard elevation above mean sea level, thereby allowing computation of the reference elevation for the water level in each well.

# First Round Water Level Measurements

Water level measurements collected at the time of the first round groundwater sampling event on December 13-14, 1994 as presented in the following table. No LNAPLs or DNAPLs were detected in any well during the water level measurements.

First Round Water Level Measurements

Well	Depth to Water (Feet)	Well Top Elevation (Fert Above Mean Sea Level)	Water Elevation (Feet Above Mean Sea Level)
MWA-1	27.25	751.43	724.18
MWA-2	25.07	749.45	724.38
MWA-3	27.79	752.19	724.40
MWA-4	26.63	751.27	724.64
MWA-5	26.99	751.25	724.26
MWA-6	27.40	751.75	724.35
MWB-I	20.68	744.93	724.25
MWB-2	27.69	751.62	723.93
MWB-3	27.59	752.13	724.54
MWB-4	28.01	751.64	723.63
MWB-5	26.04	750.73	724.69

Well	Depth to Water (Feet)	Well Top Elevation (Feet Above Mean Sea Level)	Water Elevation (Feet Above Mean Sea Level)
MWB-6	26.87	751.37	724.50
MWC-1	25.53	745.00	719.47
MWC-2	31.47	751.60	720.13
MWC-3	27.66	752.15	724.49

The interpreted direction of groundwater flow was computed for the unconfined aquifer and is shown in Drawing 22. Wells MWA and MWB are completed in the unconfined aquifer. The water elevations were contoured to show lines of equal groundwater elevation above mean sea level. The map shows a generally non-uniform groundwater elevation change, producing a variable gradient, across the site from the southwest toward the northeast. The gradient near the southwestern portion of the site was approximately 0.0003 foot/foot. As can be seen on the map, the gradient becomes steeper in the northeastern portion of the site. There the gradient was approximately 0.001 foot/foot. The interpreted direction of groundwater flow in the unconfined aquifer across the subject site was toward the northeast.

Water level measurements at the MWC-3 well (completed in the semi-confined aquifer below a till layer encountered at approximately 57 feet) and the MWB-3 well installed immediately adjacent to it (completed in the lower portion of the unconfined aquifer) showed only 0.05 foot difference in water elevation. The water levels in the other two semi-confined aquifer wells, MWC-1 and MWC-2, were approximately 4 to 5 feet lower than the water elevation in the unconfined aquifer wells.

## Second Round Water Level Measurements

Water level measurements collected January 24, 1995 are presented in the following table. No LNAPLs or DNAPLs were detected in any well during the water level measurements.

**Second Round Water Level Measurements** 

	Depth to Water	Well Top Elevation (Feet Above Mean	Water Elevation (Feet Above Mean Sea Level)		
Well	(Feet)	Sea Level)			
MWA-1	28.21	751.43	723.22		
MWA-2	26.01	749.45	723.44		
MWA-3	28.72	752.19	723.47		
MWA-4	27.52	751.27	723.75		
MWA-5	27.92	751.25	723.33		
MWA-6	28.23	751.75	723.52		
MWB-1	21.74	744.93	723.19		
MWB-2	28.60	751.62	723.02		
MWB-3	28.41	752.13	723.72		
MWB-4	28.95	751.64	722.69		
MWB-5	26.93	750.73	723.80		
MWB-6	27.81	751.37	723.56		
MWC-1	25.81	745.00	719.19		
MWC-2	31.57	751.60	720.03		
MWC-3	28.48	752.15	723.67		

The interpreted direction of groundwater flow was computed for the unconfined aquifer and is shown in Drawing 23. All water elevations fell approximately 0.5 to 1.0 foot in the unconfined aquifer from the December 1994 levels. The map shows a generally non-uniform groundwater elevation change, producing a variable gradient, across the site from the southwest toward the northeast. The gradient near the southwestern portion of the site was approximately 0.0003 foot/foot. The gradient became steeper in the northeastern portion of the site where it was approximately 0.001 foot/foot. The interpreted direction of groundwater flow in the unconfined aquifer across the subject site remained toward the northeast.

The water level measurements at the MWC-3 well and the MWB-3 well installed immediately adjacent to it again showed only 0.05 foot difference in water elevation. The water levels in the other two semi-confined aquifer wells, MWC-1 and MWC-2, were

approximately 3.5 to 4.5 feet lower than the water elevation of the unconfined aquifer wells, and had fallen approximately 0.1 to 0.5 foot from their December 1994 levels.

## Third Round Water Level Measurements

Water level measurements collected at the time of the second round of groundwater sampling on February 20, 1995 as presented in the following table. No LNAPLs or DNAPLs were detected in any well during collection of the water level measurements.

Third Round Water Level Measurements

Well	Depth to Water	Well Top Elevation (Feet Above Mean Sea Level)	Water Elevation (Feet Above Mean Sea	
MWA-1	(Feet) 28.54	751.43	Level) 722.89	
MWA-2	26.29	749.45	723.16	
MWA-3	29.01	752.19	723.18	
MWA-4	27.72	751.27	723.55	
MWA-5	28.18	751.25	723.07	
MWA-6	28.50	751.75	723.25	
MWB-1	22.11	744.93	722.82	
MWB-2	28.93	751.62	722.69	
MWB-3	28.64	752.13	723.49	
MWB-4	29.28	751.64	722.36	
MWB-5	27.20	750.73	723.53	
MWB-6	28.06	751.37	723.31	
MWC-1	26.18	745.00	718.82	
MWC-2	31.97	751.60	719.63	
MWC-3	28.73	752.15	723.42	

The interpreted direction of groundwater flow was computed for the unconfined aquifer and is shown in Drawing 24. All water elevations fell approximately 0.25 to 0.5 foot in the unconfined aquifer from the January 1995 levels. The map shows a generally non-uniform groundwater elevation change, producing a variable gradient, across the site from the southwest toward the northeast. The gradient near the southwestern portion of the

site was approximately 0.0003 foot/foot. The gradient became steeper in the northeastern portion of the site where it was approximately 0.001 foot/foot. The interpreted direction of groundwater flow in the unconfined aquifer across the subject site remained toward the northeast.

The water level measurements at the MWC-3 well and the MWB-3 well showed only 0.07 foot difference in water elevation. The water levels in the other two semi-confined aquifer wells, MWC-1 and MWC-2, were approximately 3.75 to 4.5 feet lower than the water elevation of the unconfined aquifer wells, and had fallen approximately 0.3 to 0.4 foot from their January 1995 levels. The water elevation in the MWC-3 well fell 0.23 foot from the January 1995 level, much like what had occurred at the unconfined aquifer wells.

The technique of generating a graphical solution for establishing groundwater flow direction requires a minimum of three measurement points distributed across the area of interest in order to generate a credible flow direction. Three wells were installed in the semi-confined aquifer, however, the water elevation in the MWC-3 well suggests that well appears more similar to the unconfined aquifer wells, that is, the MWC-3 well appears to be hydraulically connected to the unconfined aquifer. With only two wells yielding water elevations which might be reliably associated with the semi-confined aquifer, its flow direction and gradient cannot be well understood at this time.

# 8.4 - Discussion of Findings

The groundwater sample results, groundwater flow patterns, and interpreted distribution of contaminants revealed the following patterns of contamination in the unconfined aquifer across the property:

The levels of tetrachloroethylene in the unconfined aquifer appear to be greatest in the central portion of the facility within Area A and Area B.

The distribution of the tetrachloroethylene controls the total VOCs

- distribution pattern in that portion of the site, indicating tetrachloroethylene is the principal contaminant in that portion of the site.
- The areas affected by tetrachloroethylene contamination are the northern portion of Buildings 40A and 40B and the paved area immediately north and east of those buildings; the southern portion of Building 50, the area of the former TCA tank south of Building 53; the area to the north of Building 59; the area near Building 47 extending northward and toward the west; and the area north and west of the boiler house;
- The concentrations of trichloroethene in the unconfined aquifer appear to be greatest along the southern portion of the site within Area C. The distribution of the trichloroethene controls the total VOCs distribution pattern in that portion of the site, indicating trichloroethene is the principal contaminant in that portion of the site.
- The areas affected by trichloroethene contamination are adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40A and 40B. Area A and Area B are affected by this contaminant but to a much lesser degree.

These findings agree well with the work completed during previous investigations at the facility, the soil vapor survey and the soil sample analysis completed as a part of this investigation. The areas near Buildings 40A and 40B, the area to the south of Building 53 near the former TCA tanks, the area east of Building 50, and the western and southern portions of the former Maxwell Complex are areas where elevated levels of VOC contamination have been documented in soil and groundwater.

The patterns of contamination by the principal mapped contaminants in the soil and groundwater, tetrachloroethylene and trichloroethene, reveal a contaminant transport relationship between the soil and groundwater. These two compounds account for the overall pattern of VOC contamination observed in the soil and groundwater.

Groundwater contamination plumes appear to originate from locations within the plant, or as plumes entering the site from off-site sources in areas where soil contamination is also present, particularly in the deeper portion of the vadose soil zone. Soil and groundwater contamination by tetrachloroethylene is greatest in Areas A and B. Soil and groundwater contamination by trichloroethene occurs mainly in Area C.

Additional relationships were observed. Several organic compounds having greater concentration levels were detected in the unconfined aquifer in the MWA wells than the MWB wells, that is, contamination is greater in the shallow portion of the unconfined aquifer than the deeper portions of the same aquifer. This pattern suggests groundwater in contact with the base of the unsaturated soil zone is in contact with a contamination source, possibly only during certain periods of time in response to seasonal water level fluctuations.

Contaminants may enter groundwater as they become dissolved into the water from contaminant sources in the unsaturated soil zone. Seasonal water level fluctuations, as observed during this investigation, appear to remove groundwater from contacting contaminated soil. As the water levels fell during the three month period of this investigation, the overall level of groundwater contamination by VOCs decreased. This contaminant migration model is supported by the pattern of greater contamination having been documented in the deep vadose soil and greater groundwater contamination having been documented in the wells completed in the upper portion of the unconfined aquifer.

The three groundwater elevation maps show the groundwater flow direction remained constant during the investigation. The groundwater flow gradient appeared to vary across the site for each set of water level measurements, but the amount of variance in the gradient appeared to remain constant between measurement sets. The groundwater contamination plumes as shown on the isoconcentration maps conform well to the interpreted groundwater flow direction. The contaminant plumes in Areas A and B appear

to originate from locations within the plant. The plume shown in Area C appears to originate either from the portion of the site directly adjacent to Leo Street, or from some off-site source.

The findings for both groundwater sampling rounds show the semi-confined aquifer does not appear to be affected by VOC contamination at this time. No VOCs were detected in any well completed in the semi-confined aquifer in either groundwater sampling round.

The metals concentrations were similar in both the unconfined and semi-confined aquifers. This suggests that since the absence of VOCs in the semi-confined aquifer can be interpreted as an indication of no impact to that aquifer. The similarity of the metals concentrations in the aquifers can be interpreted as evidence of no significant impact to groundwater in either aquifer by the targeted metals at this time.

# Section 9.0 - Interpretations of Contaminant Distribution Patterns

Contaminant distribution patterns were established using information generated from several investigative methods. A review of previous limited investigations and existing information sources was conducted. This was followed by a soil vapor survey which established contamination patterns in portions of the site designated as Areas A, B, and C. Soil samples were collected from soil borings and during the installation of the groundwater monitoring wells for TCL VOCs and TAL metals. Two rounds of groundwater samples were collected and analyzed for TCL VOCs and TAL metals. Groundwater flow directions and water level fluctuations were examined for the unconfined aquifer using three sets of water level measurements. This information was compiled as an integrated package and contaminant distribution patterns were developed for soil and groundwater. The affected areas as documented through the different investigative techniques have been related to the reference Areas A, B, and C.

Previous investigations indicated soils and groundwater had been impacted by volatile organic compounds with limited heavy metal contamination documented in the area of the former Maxwell Complex. Aerial photographs showed areas of possible concern within the northern areas of the plant prior to its development, unpaved areas within the plant boundaries which formerly existed near the manufacturing buildings, and an area where easy access to the undeveloped northern portion of the site was possible prior to its development. Additionally, several areas were identified where material storage took place within the plant boundaries over many years, particularly in the area adjacent to Leo Street in the extreme southern area.

Sanborn maps showed areas of possible concern associated with historical manufacturing processes and materials handling areas within the plant boundaries in association with the former Maxwell Complex. Off-site areas of concern were focused along the southern portion of the site including the former paint and varnish facility across Leo Street, the former service stations located across Leo Street and at the intersection of Leo and

Webster Streets, and the group of light industries located approximately 300 to 400 feet south of the DTPP property.

A review of potential contamination sources conducted by Clean Tech revealed several potential on-site sources of contaminants which may have impacted the soil or groundwater. These potential sources include: underground and above ground storage tanks, chemical handling or storage areas, hazardous waste generation and accumulation storage areas, sumps for waste oil or process wastewater, past spills, and various processes or operations of the plant.

Significant potential on-site sources of contamination include tanks which stored TCA and TCA sludge which were located on the south side of Building 53 and the north side of Building 40; a TCE degreaser station formerly located near the southern end of Building 53; a TCA degreaser formerly located in the northeast area of Building 40A; and a CFC-113 degreaser formerly located in the middle of Building 40A. Spill records revealed potential contamination from spills of approximately 500 gallons of chrome-containing paint sludge in Building 47; an overfill of TCA storage tank (quantity unspecified); a release of approximately 35 gallons of untreated wastewater containing flux rinse water near Building 50; and a release of an unspecified quantity of TCA from a tank next to Building 53.

The soil vapor survey revealed contamination in the vadose zone across the property. VOC contamination in the vadose zone appeared to be greatest in Area A in both the shallow and deep portions of the vadose zone. VOC contamination in the vadose zone was noted at a lesser magnitude across Area B in both the shallow and deep portions of the vadose zone, but was much more pronounced in the deep vadose zone. Isolated areas of significantly elevated VOCs were noted in Area C for both the shallow and deep portions of the vadose zone, but were much more pronounced in the deep vadose zone.

Soil sample analysis confirmed these patterns of soil contamination across the property. Tetrachloroethylene in the soil was greatest in the central portion of the facility within Area A and Area B. The distribution of the tetrachloroethylene controls the total VOCs distribution pattern in these areas. Trichloroethene in the soil was greatest along the southern portion of the site within Area C. The distribution of the trichloroethene controls the total VOCs distribution pattern in the southern area.

Groundwater sample analysis and groundwater flow patterns established a contamination pattern which relates vadose soil and groundwater contaminants in the unconfined aquifer across the property. No detectable levels of VOCs were found in the semi-confined aquifer for first round or second round groundwater samples. No LNAPLs or DNAPLs were detected in any well for any of the three sets of water level measurements.

The levels of tetrachloroethylene in the unconfined aquifer were greatest in the central portion of the facility within Area A and Area B in the same area and pattern similar to that seen for the soil contamination. The distribution of the tetrachloroethylene controls the total VOCs distribution pattern in those areas. The levels of trichloroethene in the unconfined aquifer were greatest along the southern portion of the site within Area C in the same area and pattern similar to that seen for the soil contamination. The distribution of the trichloroethene controls the total VOCs distribution pattern in that portion of the site, indicating trichloroethene is the principal contaminant in Area C.

The information gathered through each investigative technique may be summarized as follows for each contamination area:

#### Area A

Previous Investigations

Soil contaminants at Building 40B in the area of former CFC-113 degreaser station, and soil and groundwater contaminants on south side of Building 53 in

the area of the former TCA storage tanks, and at Buildings 40A and 40B which contained former TCA degreasers.

## Soil Vapor Survey

Areas with elevated VOCs in the area to the north of Buildings 40A and 40B, and to the south of Buildings 50 and 53 in the area of the former TCA storage tanks with VOCs elevated in both shallow and deep vadose soil zones with greater contamination in the deep vadose zone.

## Soil Samples

Affected areas are northern portions of Buildings 40A and 40B and the paved area immediately north and east of those buildings, and the southern portion of Building 50 in the area of the former TCA tank south of Building 53.

#### Groundwater

Areas affected by contamination are the northern portion of Buildings 40A and 40B and the paved area immediately north and east of those buildings, and the southern portion of Building 50 in the area of the former TCA tank south of Building 53.

#### Area B

#### Previous Investigations

Soil and groundwater contaminants noted in the west and southwest sections of the former Maxwell Complex, and in the storage areas located east of Building 50. Well 2, the production well located in the boiler house and completed in the unconfined aquifer, was found to contain organic compounds. Testing suggested a large volume of the aquifer may be affected by the contaminants.

## Soil Vapor Survey

Areas with elevated VOCs in the area to the north of Building 59 to the area of Building 47 and the associated waste storage area with VOCs elevated in both

the shallow and deep vadose zone and greater contamination noted in the deep vadose zone.

## Soil Samples

Affected areas are to the north of Building 59, in the area near Building 47 extending northward and toward the east; and in the area north of the boiler house and northeast of Building 47.

#### Groundwater

Areas affected by contamination are the area to the north of Building 59; the area near Building 47 extending northward and toward the west; and the area north and west of the boiler house.

#### Area C

## Soil Vapor Survey

Isolated areas of elevated VOCs were noted in the southern portion of the site to the west of Building 3A and south of Building 59, and in the area to the south of Buildings 40 and 40A. This pattern was noted for both the shallow and deep portions of the vadose zone, with greater contamination noted in the deep vadose zone.

## Soil Samples

Areas affected by contamination are adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40, 40A and 40B.

#### Groundwater

Areas affected by contamination are the area adjacent to Leo Street south of Building 59, Building 3A, and Buildings 40A and 40B.

The patterns of contamination by the principal mapped contaminants in the soil and groundwater, tetrachloroethylene and trichloroethene, suggest a contaminant transport mechanism involving interactions between the soil and groundwater. The patterns of soil

and groundwater contamination may be explained using the available information in the following conceptual model for Areas A and B, and for Area C of the site.

## Area A and Area B

The soil contamination as seen across Area A and Area B in the soil vapor and soil sample analysis is typically greater in the deeper portions of the vadose zone soils, with the clear exception of the area of the former TCA tanks where significant vadose zone soil contamination both shallow and deep suggests this is a primary source location of solvent contamination to the soils. Other possible secondary contaminant source areas include the area to the east of Building 50 and the waste storage area near Building 47 north of Building 59. Previous studies and the pattern of contamination found in this investigation identified the subsurface beneath Buildings 40A and 40B as potential contaminant source areas.

Solvents containing chlorinated organic compounds are interpreted to have entered the subsurface environment at these source locations. After soil contaminants penetrated to a depth near the base of the vadose zone, groundwater in the unconfined aquifer was brought in contact with the contaminated soil allowing contaminants to be released into the groundwater. Groundwater flow in the unconfined aquifer moved the groundwater toward the northeast under the influence of the steepening hydraulic gradient induced by the pumping well at Gem City Chemicals, Inc. The groundwater flowing past the contaminant sources acquired dissolved contaminants and carried the contaminants across the site toward the northeast forming contamination plumes.

The steepening hydraulic gradient which was persistently seen in all water elevation maps is interpreted to represent the influence of the groundwater recovery well system at the Gem City Chemical facility as it draws groundwater toward the pumping well. The recovery well pumps continuously in the unconfined aquifer at approximately 300 gpm. The pumping well was required by the Ohio EPA to prevent introduction of contaminants from Gem City Chemical into the South Miami Well Field. The groundwater elevation

maps show a generally non-uniform groundwater elevation change, producing a variable gradient, across the subject site from the southwest toward the northeast in all three measurement sets. The uniformity of the hydraulic conditions were maintained as water levels fell in the unconfined aquifer, indicating the pumping well exerted primary hydraulic control over the groundwater flow pattern. This interpretation is in agreement with the potentiometric measurements for the Gem City site as reported by QSource Engineering, Inc. QSource Engineering reported that changes in water levels and recharge do not appear to affect the general direction of groundwater flow. QSource Engineering estimated the groundwater flow direction was to the northeast in the portion of the Gem City site adjacent to the DTPP site. This flow direction and the reported configuration of the groundwater potentiometric surface agree closely between the two sites, further demonstrating the interconnection between pumping at Gem City and the behavior of the groundwater across the DTPP property.

As groundwater moved toward the northeast carrying dissolved contaminants from the source locations, the soils in contact with the moving groundwater plumes absorbed some of the contaminants. This formed broad soil contamination plumes and may account for the similarity in location and pattern for both the soil contaminant and groundwater contaminant plumes as mapped. The areas of the greatest soil contamination and greatest groundwater contamination are coincident. Seasonal fluctuations in water levels (reported annual total of 10 to 15 feet) would be expected to exacerbate this situation over time, causing an additional vertical thickness of soil near the base of the vadose zone to be exposed to the dissolved contaminants in the groundwater. Under these conditions, the potential for off-site transport of contaminants is significant over time, first as dissolved groundwater contamination, and secondly as soil contamination near the base of the vadose zone. As shown on the isoconcentration maps, the potential exists at present for some off-site transport of contaminants in groundwater to have occurred.

Additional supporting evidence for this mechanism comes in two forms. First, larger numbers of organic compounds having greater concentration levels are present in the

groundwater samples from the unconfined aquifer in the MWA (shallower) wells than the MWB (deeper) wells. Overall VOC contamination is greater in the shallow portion of the unconfined aquifer in contact with the deep vadose zone than the deeper portions of the same aquifer. Second, seasonal water level fluctuations as observed during this investigation appeared to remove groundwater from contacting the zone of contaminated soil. As the water levels fell during the three month period of this investigation, the level of groundwater contamination by VOCs decreased.

## Area C

Groundwater flows to the north onto the site from across Leo Street. The soil and groundwater contamination plume shown in Area C appears to originate from some off-site source, particularly from the area across Leo Street from the site boundary as groundwater flows on the site from adjacent properties. Certain properties in this area have been identified as potential sources of contaminants. The soil contamination as seen across Area C in both the soil vapor survey and the soil sample analysis is typically greater in the deeper portions of the vadose zone soils, and has trichloroethene as the primary compound present. The available information for the soil in the shallow vadose zone provided no clear indication of a source area for chlorinated solvents on the property. The soil vapor survey did not identify significantly elevated levels of VOCs in the shallow vadose zone in Area C. This evidence supports the idea of an off-site source for the contaminants.

A soil sample from SB-10 was collected from below the water table at 31 feet and was found to contain significantly elevated levels of VOCs, particularly trichloroethene, in the saturated soil approximately four feet below the water table. It may be interpreted that this contaminant is flowing on the property from an off-site source to the south. Additional supporting evidence for an off-site contaminant source is the presence of benzene in both rounds of groundwater samples collected from MWB-3 when that compound is not a contaminant of concern at any other location on the site. Benzene is a contaminant typically associated with petroleum and may have entered the subsurface

environment from a leaking storage tank, or possibly from some operation at the former paint and varnish operation located across Leo Street from the site boundary.

A hydraulic connection between the MWC-3 well and the unconfined aquifer apparently occurs because the till layer seen in the southeastern portion of the site becomes substantially thinner and ceases to be a hydraulic barrier in the east central portion of the site. The till layer has apparently protected the portion of the unconfined aquifer below the till layer in the extreme southeastern portion of Area C from encountering significant contamination by organic compounds. The MWB-3 well, completed in the unconfined aquifer above the till layer, was found to have up to 13,727.6 ppb total VOCs in a groundwater sample. Both groundwater samples collected from MWC-3, completed in the unconfined aquifer below the till layer, were found to have less than detectable levels of total VOCs.

Groundwater flowing in the unconfined aquifer moves northward toward the cone of influence developed by the Gem City recovery well. The groundwater within the unconfined aquifer is interpreted to pass this point both above and below the clay layer within the unconfined aquifer. This pattern of groundwater movement apparently prevents contaminants entering the aquifer from the ground surface from impacting the portion of the unconfined aquifer below the clay layer in which MWC-3 is screened. The clay layer is expected to extend intact to the south (on the opposite side of Leo Street) a distance at least as great as the distance to the proposed contamination source.

The findings for both groundwater sampling rounds show the semi-confined aquifer does not appear to be affected by VOC contamination at this time. No VOCs were detected in any well completed in the semi-confined aquifer in either groundwater sampling round. The vertical flow potential provides an assessment of the potential for movement of groundwater, and potential contaminants, from one aquifer to another. The available information as compiled during regional hydrogeological analysis, and for the site specific information generated as a part of this investigation indicate the potential exists for

groundwater to move downward from the unconfined aquifer to the semi-confined aquifer. The confining layer between the aquifers has apparently prevented contamination of the semi-confined aquifer based on the available information.

The metals concentrations in groundwater were similar in both the unconfined and semi-confined aquifers. Since the absence of VOC contamination in the semi-confined aquifer can be interpreted as an indication of no impact to that aquifer, the similarity of the dissolved metals concentrations in the aquifers may be interpreted as evidence of no significant impact to groundwater in either aquifer by the targeted metals at this time. The metals contamination in soils as encountered in the area of the former Maxwell Complex was apparently confined to the soils present in that limited area.

# Section 10.0 - Targets for Soil and Groundwater Remediation

The following discussion of targets for soil and groundwater remediation are presented as guidance for development of goals for remediation of soil and groundwater contamination at the Chrysler DTPP facility.

The Ohio EPA Division of Emergency and Remedial Response has developed guidance for site investigations and remediation programs. Ohio EPA evaluates every site independently and will not provide generic clean-up guidance or criteria. The policy was originally developed for unregulated hazardous waste sites but has been extended to the entire Ohio EPA Remedial Response Program.

The selection of soil and groundwater remediation targets typically begins with a determination of the levels of site contamination through a site investigation. A site is considered to be hazardous by the Ohio EPA if a contaminant is present on-site at concentrations significantly above background levels, or the contaminant is present on-site and is not detected in representative background samples. The DTPP facility appears to fall in the hazardous site category based on the findings of this investigation.

Ohio EPA guidance stipulates a determination of whether contamination poses a threat to public health or the environment. Normally this involves preparation of a site-specific health-based risk assessment for most locations within Ohio. Following preparation of the risk assessment, a review of applicable ARARs (applicable or relevant and appropriate standards and/or criteria) is usually undertaken. The selection of remedial alternatives and design goals normally results from these activities. However, an overriding issue in Dayton is the delineation of a portion of Dayton in the area of the DTPP facility as a Well Field Protection Area.

Mr. Joe Smindak of the Ohio EPA was contacted regarding this issue on April 25, 1995. Mr. Smindak indicated that the Well Field Protection Program instituted in Dayton is a nationally recognized program which seeks to monitor the public water supply source aquifer, and has as a future goal the development of a comprehensive Well Field Management Plan. This management plan has not yet been initiated but will be based on the findings of the Ohio EPA groundwater quality monitoring program now underway. The Ohio EPA routinely monitors groundwater using a network of wells installed throughout Dayton in public lands and right-of-ways. Mr. Smindak noted that the management plan is needed primarily because groundwater quality monitoring has shown organic contaminants are commonly dispersed across large areas in the subsurface throughout Dayton. The contaminants are commonly organic solvents which have found their way into the aquifer due to a long history of manufacturing land uses, and the effects of pumping from numerous water supply wells in Dayton. The contaminants are known to enter the aquifer near certain properties, and pass under other properties where they may be detected at monitoring points and in water supply wells. Mr. Smindak stated that at some point in the future a program to address these large areas of contaminants will be brought forward by the Ohio EPA.

The Ohio EPA currently seeks only to prevent significant contamination from reaching the public water supply wells through a program of Interim Action requirements. Interim Actions for groundwater are the only approved remedial actions which may be undertaken within the Well Field Protection Area. Groundwater gradient control is the most commonly required Interim Action. The need for groundwater gradient control is based on what Ohio EPA has defined as Interim Standards. If an Interim Standard for groundwater quality is exceeded at a site, then Ohio EPA seeks to have the property owner control and remediate contaminated groundwater to prohibit it from leaving the effected property. The process of calculating the Interim Standard for a site involves review of the following standards:

- Carcinogenic standards for each detected compound which will produce a frequency of 10<sup>-6</sup> cancer cases (one cancer case per million population exposed) in the effected population;
- Non-carcinogenic standards for each detected compound which will produce a
   Hazard Index of 1 for the effected population;
- Maximum Contaminant Levels (MCLs) derived from US EPA drinking water regulations and health advisories.

The lowest value resulting from the review of the three standards listed above becomes the Interim Standard for the detected compound, provided the standard is not less than 1 ppb. In those cases 1 ppb becomes the Interim Standard. Under the Interim Actions process only the groundwater pathway is assessed, and no cumulative or synergistic effects are incorporated into the risk analysis. Achieving protection of the groundwater within the Well Field Protection Area is done by meeting the Interim Standard set for each detected contaminant. If the detected contaminant levels are below the Interim Standards, then there is no requirement to achieve groundwater gradient control and perform any treatment of extracted groundwater.

Interim Standards were applied to the remedial actions which have been undertaken at the DAP, Inc. and the Gem City Chemicals, Inc. facilities. The DAP facility employed a soil vapor extraction system to remove the contaminant source within the soil as a method of preventing groundwater contamination. DAP achieved groundwater gradient control using a system of four groundwater recovery wells plus an air stripping tower to remove a variety of solvents from the groundwater. Routine monitoring of wells at the site demonstrated groundwater quality has improved to within the Interim Standards, and the recovery wells are expected to be shut down in the near future.

Gem City Chemicals, Inc. achieved groundwater gradient control using a groundwater recovery well and an associated air stripper system. A soil vapor extraction system was

also employed to remediate the soil which was contributing contaminants to the groundwater. The soil vapor extraction system was discontinued when no significant concentration of VOCs (≤5 ppm) were detected in the exhausted air. Groundwater gradient control and remediation continues on an ongoing basis using a single recovery well.

Interim Standards were computed using the laboratory results for the organic compounds detected in both the first and second round groundwater samples from the subject site. The detected concentrations were averaged for the MWA and MWB wells for each compound detected, using both the first and second round samples. The carcinogenic standard or hazard index was taken from the Risk Based Concentration (RBC) tables prepared by US EPA Region III, as recommended by the Ohio EPA, for each detected compound. The reported RBC value corresponds to a contaminant concentration exposure concentration through a certain pathway which produces a fixed level of risk, either the hazard index of 1 or lifetime cancer risk of 10<sup>-6</sup> whichever occurs at a lower concentration. The RBC value for water (tap water) as a pathway was used. The MCLs were taken from the November 1994 US EPA Drinking Water Regulations and Health Advisory tables.

## Interim Standards for Detected VOCs in Groundwater

VOCs	Averaged Concentration	RBC Value	MCLa	Ohio EPA Interim Standard	Interim Standard Exceeded?
tetrachloroethylene	777	1.1	5	1.1	Yes
trichloroethene	7107	1.6	None	1.6	Yes
benzene	2.3	0.36	5	1.0	Yes
1,2-dichloroethene (total)	3101	55	None	55	Yes
cis-1,2 dichloroethylene	3097	61	70	61	Yes
trans-1,2 dichloroethylene	21.7	120	100	100	No
1,1,1-trichloroethane	1981	1300	200	200	Yes
1,1,2-trichloroethane	2.6	0.19	5	1.0	Yes
chloroform	2.9	0.15	100	1.0	Yes
1,1-dichloroethane	134	810	None	810	No
1,2 -dichloroethane	11.6	0.12	5	1.0	Yes
1,1-dichloroethene	111	0.044	7	1.0	Yes
trichlorofluoromethane	3.0	1300	None	1300	No
dichlorodifluoromethane	90.2	390	None	390	No
vinyl chloride	317	0.019	2	1.0	Yes
1,2-dichlorobenzene	1.8	270	75	75	No
1,1-dichloropropene	2.2	None	None	None	
1,1,2-trichloroethane	2.2	0.19	5	1.0	Yes
1,1,1,2-tetrachloroethane	1.4	0.41	None	1.0	Yes
chloromethane	8.3	1.4	None	1.4	Yes
chloroethane	2.0	8600	None	8600	No

All Results in Parts per Billion (ug/l).

Fourteen of the twenty-one detected volatile organic compounds exceed the Interim Standards for those compounds. The Interim Standards may be viewed as goals for groundwater remediation at the Chrysler DTPP facility.

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- 2. Mathes & Associates, Soil Gas Investigation, May 1991.
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- 4. Clean Tech, Work Plan for Investigation, Chrysler Corporation DTPP, August 1994.
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# SITE INVESTIGATION REPORT CHRYSLER CORPORATION DAYTON THERMAL PRODUCTS PLANT 1600 WEBSTER STREET DAYTON, OHIO 45404

# Volume III of III Laboratory Results

# **Prepared For**

Chrysler Corporation 800 Chrysler Drive CIMS 482-00-51 Auburn Hills, Michigan 48326-2757

# Prepared By

Clean Tech 2700 Capitol Trail Newark, DE 19711 (302) 999-0924

September, 1995



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September 14, 1995

Mr. Curtis Chapman Chrysler Corporation 800 Chrysler Drive CIMS 482-00-51 Auburn Hills, MI 48326-2757

RE: Finalized Site Investigation Report

**Chrysler Corporation Dayton Thermal Products Plant** 

Dayton, Ohio

Dear Mr. Chapman:

Enclosed please find the three volume finalized document <u>Site Investigation</u>, <u>Chrysler Corporation Dayton Thermal Products Plant</u>, <u>Dayton Ohio</u>. This submittal includes your review comments and requested report revisions. Comments received from Mr. Doug Orf are incorporated in this final submittal. This document has been forwarded to Mr. Orf per your request.

If you have any questions, please contact Clean Tech at (302) 999-0924.

Sincerely,

Steven W. Newsom, P.G.

Principal Geologist

**CLEAN TECH** 

Sincerely,

Deborah A. Buniski, P.E.

President

**CLEAN TECH** 

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### CLEANTECH

September 14, 1995

Mr. Douglas J. Orf Chrysler Corporation Dayton Thermal Products Plant 1600 Webster Street Dayton, Ohio 45404

RE: Finalized Site Investigation Report

**Chrysler Corporation Dayton Thermal Products Plant** 

Dayton, Ohio

Dear Mr. Orf:

Enclosed please find the three volume finalized document <u>Site Investigation</u>, <u>Chrysler Corporation Dayton Thermal Products Plant</u>, <u>Dayton Ohio</u>. This submittal includes comments and requested report revisions as received from you and Mr. Curtis Chapman. This document has been forwarded to Mr. Chapman.

If you have any questions, please contact Clean Tech at (302) 999-0924.

Sincerely,

Steven W. Newsom, P.G.

Principal Geologist

**CLEAN TECH** 

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Sincerely,

Deborah A. Buniski, P.E.

President

**CLEAN TECH** 

# Volume III of III Laboratory Results Table of Contents

### Attachments

Г	Chemical, Soil Samples from Soil Borings
н	Laboratory Results:

- H Laboratory Results:
  Geotechnical, Soil Samples from Soil Borings
- J Laboratory Results: Chemical, Soil Samples from Well Borings
- K Laboratory Results:Geotechnical, Soil Samples from Well Borings
- M Laboratory Results: Chemical, First Round Groundwater Samples
- N Laboratory Results: Chemical, Second Round Groundwater Samples

### **ATTACHMENT F**

Laboratory Results: Chemical
Soil Samples from Soil Borings
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404

CLIENT: CLEAN TECH

REPORT NUMBER: 35848

SAMPLE ID: 4101683

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-1-9

DATE SAMPLED....: 10/17/94 TIME SAMPLED....: 14:20

DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
}		]		ĺ		[	
l	TAL METALS		Į.	l .		t	
•	Aluminum, Total	1600000	ug/kg	}	EPA 6010	10/31/94	SD
	Barium, Total	7200	ug/kg		EPA 6010	10/31/94	SD
t	Beryllium, Total	< 500	ug/kg	ł	EPA 6010	10/26/94	SD
	Cadmium, Total	650	ug/kg	}	EPA 6010	10/21/94	SD
)	Calcium, Total	140000000	ug/kg		EPA 6010	10/31/94	SD
ł	Chromium, Total	7100	ug/kg	i	EPA 6010	10/21/94	SD
.	Cobalt, Total	< 1000	ug/kg	)	EPA 6010	10/31/94	SD
	Silver, Total	3900	ug/kg		EPA 6010	10/21/94	SD
H	Copper, Total	2200	ug/kg	}	EPA 6010	10/21/94	SD
1	Sodium, Total	280000	ug/kg		EPA 6010	10/31/94	SD
- 1	Vanadium, Total	< 1000	ug/kg		EPA 6010	10/31/94	SD
H	Zinc, Total	16000	ug/kg	i	EPA 6010	10/21/94	SD
	Antimony Total, GFAA	< 1500	ug/kg	1	EPA 7041	10/25/94	NS
• 1	Arsenic Total, GFAA	4800 2000	ug/kg	ļ	EPA 7060 EPA 7421	10/25/94	NS
	Lead, Total, GFAA Selenium Total, GFAA	< 1000	lug/kg lug/kg		EPA 7421 EPA 7740	10/26/94 10/25/94	NS NS
	Thallium Total, GFAA	1200		1	EPA 7740	10/23/94	NS NS
	Mercury, Cold Vapor	< 40	ug/kg ug/kg	1	EPA 7470	10/27/94	SD SD
	TCL VOLATILE ORGANICS, GC/MS	<b>4</b> 0	ug/kg	ł	EPA 8260	10/25/94	VS
	Acetone	< 100	ug/kg	1	EPA 0200	10/29/94	<b>V3</b>
- 1	Acrolein	< 50	ug/kg				- 1
<b>T</b> I	Acrylonitrile	< 10	ug/kg				
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	Benzene	< 10	ug/kg	ļ			
	Bromobenzene	< 10	ug/kg	!			1
- 1	Bromochloromethane	< 10	ug/kg	j			
ı	Bromodichloromethane	< 10	ug/kg	i l			j
П	Bromoform	< 10	ug/kg	i l	1		Į.
Ŋ	Bromomethane	< 10	ug/kg	Ì	]		ļ
1	2-Butanone	< 100	ug/kg		İ		ļ
	n-Butyl benzene	< 10	ug/kg		{		Į.
	Sec-Butylbenzene	< 10	ug/kg	ĺ	ļ		Ī
•	tert-Butylbenzene	< 10	ug/kg	ľ			}
	Carbon Disulfide	< 100	ug/kg		ļ		
	Carbon Tetrachloride	< 10	ug/kg	]			į
	Chlorobenzene	< 10	ug/kg				ļ
	Chloroform	< 10	ug/kg				į
	Chloromethane	< 10	ug/kg				1
	2-Chlorotoluene	< 10	ug/kg				ł
BI.	- Chlorotoluene	< 10	ug/kg		Ì		1
	ibromochloromethane	< 10	ug/kg				1
"	ibromomethane	< 10 < 10	ug/kg				l
	,2-Dibromo-3-chloropropane	< 10	ug/kg	}			
	, = 5.5. Gilo 5 circol opi opane	` ''	ug/kg				1
I	1	İ		ļ			ţ
1				<u> </u>			

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 1

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 35848

SAMPLE ID: 4101683

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-1-9

DATE SAMPLED.....: 10/17/94 TIME SAMPLED ..... 14:20

DATE RECEIVED....: 10/20/94

TIME RECEIVED....: 16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	< 10	ug/kg	EPA 8260	10/29/94	VS
	i				

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 3

Canton Analytical Laboratory, Inc.

CLIENT: CLEAN TECH

REPORT NUMBER: 35848 SAMPLE ID: 4101683 DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-1-9

DATE SAMPLED.....: 10/17/94 TIME SAMPLED .....: 14:20

DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

Ĭ		T	T		T	Lau	8.11.11.19
}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN	TECHN
	ANALISES PERFORMED	KESULI	HEASURE	<u></u>	HE THOO	DAIL	TECHN.
1							<del></del>
I	1,2-Dichlorobenzene	< 10	ug/kg	[		,	
J	1,3-Dichlorobenzene	< 10	ug/kg	1		Ì	
	1,4-Dichlorobenzene	< 10	ug/kg			ļ	
	Dichlorodifluoromethane	< 10	ug/kg	1		}	
	1,1-Dichloroethane	< 10	ug/kg				
В	1,2-Dichloroethane	< 10	ug/kg	i		ł	
•	1,1-Dichloroethene	< 10	ug/kg			ſ	
	Cis-1,2 Dichloroethylene	< 10	ug/kg				
	Trans-1,2 Dichloroethylene	<b>&lt; 10</b>	ug/kg	ļ			
1	1,2-Dichloroethene (total)	< 10	ug/kg			!	
į,	Dichloromethane	< 30	ug/kg				
	1,2-Dichloropropane	< 10	ug/kg			f	
_	1,3-Dichloropropane	< 10	ug/kg		•		
ľ	2,2-Dichloropropane	< 10	ug/kg	ļ			
Ų	1,1-Dichloropropene	< 10	ug/kg	İ			
•	Cis-1,3-Dichloropropene	< 10	ug/kg				
	Trans-1,3-Dichloropropene	< 10	ug/kg				
ד	Ethylbenzene	< 10	ug/kg				
١	Hexachlorobutadiene	< 10	ug/kg				
J	2-Hexanone	< 100	ug/kg		i		
	Isopropyltoluene	< 50 *	ug/kg				
_	4-Isopropyltoluene	< 50 *	ug/kg				
1	4-Methyl-2-Pentanone	< 100	ug/kg	<u>'</u>			
H	Naphthalene	< 10	ug/kg				
9	n-Propylbenzene	< 10	ug/kg		Į.		
1	Styrene	< 10	ug/kg		ì		
	1,1,1,2-Tetrachloroethane	< 10	ug/kg				
	1,1,2,2-Tetrachloroethane	< 10	ug/kg				
5	Tetrachloroethylene	45	ug/kg		ļ		
	Toluene	< 10	ug/kg				
	1,2,3-Trichlorobenzene	< 10	ug/kg	}			
	1,2,4-Trichlorobenzene	< 10	ug/kg				
U	1,1,1-Trichloroethane	< 10	ug/kg				
7	1,1,2-Trichloroethane	< 10	ug/kg				
ı	Trichloroethene	16	ug/kg	ļ			
e ni	Trichlorofluoromethane	< 10	ug/kg	4	Ī		
	1,2,3-Trichloropropane	< 10	ug/kg				
4)	1,2,4-Trimethylbenzene	< 10	ug/kg	]			
	1,3,5-Trimethlybenzene	< 10	ug/kg	}	j		
	Vinyl Acetate	< 5.0	ug/kg	ì	Ì		ì
	Vinyl Chloride	< 10	ug/kg	Ì	1		
	MP-Xylene	< 20	ug/kg	J	1		j
	0-Xylene	< 10	ug/kg	į	Į.		i
	Xylenes, Total	< 30	ug/kg	ì	1		l
4				[	ļ		j
1				ì	1		i

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 2

Canton Analytical Laboratory, Inc.

QA/QC\_

CLIENT: CLEAN TECH

REPORT NUMBER: 35848 SAMPLE ID: 4101684 DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-2-19

DATE SAMPLED....: 10/18/94 TIME SAMPLED....: 11:00 DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOO	RUN DATE	TECHN
TAL METALS	İ	1		1	
Aluminum, Total	1400000	ug/kg	EPA 6010	10/31/94	SD
Barium, Total	12000	ug/kg	EPA 6010	10/31/94	SD
Beryllium, Total	< 500	ug/kg	EPA 6010	10/26/94	SD
Cadmium, Total	630	ug/kg	EPA 6010	10/21/94	SD
Calcium, Total	110000000	ug/kg	EPA 6010	10/31/94	SD
Chromium, Total	6900	ug/kg	EPA 6010	10/21/94	SD
Cobalt, Total	< 1000	ug/kg	EPA 6010	10/31/94	SD
Silver, Total	3100	ug/kg	EPA 6010	10/21/94	SD
Copper, Total	8900	ug/kg	EPA 6010	10/21/94	SD
Sodium, Total	140000	ug/kg	EPA 6010	10/31/94	SD
Vanadium, Total	< 1000	ug/kg	EPA 6010	10/31/94	SD
Zinc, Total	18000	ug/kg	EPA 6010	10/21/94	SD
Antimony Total, GFAA	< 1500	ug/kg	EPA 7041	10/25/94	NS
Arsenic Total, GFAA	4300	ug/kg	EPA 7060	10/25/94	NS
Lead, Total, GFAA	2000	ug/kg	EPA 7421	10/26/94	NS NS
Selenium Total, GFAA	< 1000	ug/kg	EPA 7740	10/25/94	NS
Thallium Total, GFAA	1300	ug/kg	EPA 7841	10/27/94	NS
Mercury, Cold Vapor	1300 < 40	ug/kg	EPA 7470	10/25/94	SD
	T 40	Ug/kg	EPA 8260	10/29/94	VS
TCL VOLATILE ORGANICS, GC/MS	< 100	ug/kg	EPA 0200	10/29/94	٧5
Acrolein	< 50	ug/kg ug/kg	i	1	
Acrylonitrile	< 10	ug/kg ug/kg			
Benzene	< 10	ug/kg ug/kg		1	
Bromobenzene	< 10	ug/kg ug/kg	J		
Bromochloromethane	< 10	1 1	j	<b>†</b>	
Bromodichloromethane	< 10	ug/kg	i	1	
Bromoform	< 10	ug/kg			
Bromomethane	< 10	ug/kg		ļ	
12-Butanone	< 100	ug/kg	(	[	
1		lug/kg	1	i	
n-Butylbenzene	< 10	ug/kg			
Sec-Butyl benzene	< 10	ug/kg	İ		
tert-Butylbenzene	< 10	ug/kg	1		
Carbon Disulfide	< 100	ug/kg	}	]	
Carbon Tetrachloride	< 10	ug/kg			
Chlorobenzene	< 10	ug/kg	ì	1	
Chloroform	< 10	ug/kg	[		
Chloromethane	< 10	ug/kg	ľ	1	
2-Chlorotoluene	< 10	ug/kg	1	]	
4-Chlorotoluene	< 10	ug/kg	l l		
Dibromochloromethane	< 10	ug/kg			
1,2-Dibromoethane	< 10	ug/kg			
Dibromomethane	< 10	ug/kg		1	
1,2-Dibromo-3-chloropropane	< 10	ug/kg			
			1	1	

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 4

DA/DC KI

CLIENT: CLEAN TECH

REPORT NUMBER: 35848

SAMPLE ID: 4101684

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-2-19

10/18/94 TII

AIF	SAMPLED	10/18/94	DAIL	KECEIVED	10/20/94
IME	SAMPLED:	11:00	TIME	RECEIVED:	16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
						yer or vegit 11 yer (11 ye
1,2-Dichlorobenzene	< 10	ug/kg			!	
1,3-Dichlorobenzene	< 10	ug/kg				
1,4-Dichlorobenzene	< 10	ug/kg		ļ.	1	
Dichlorodifluoromethane	<b>  &lt; 10</b>	ug/kg		1	1	
1,1-Dichloroethane	< 10	ug/kg			ļ	
1,2-Dichloroethane	< 10	ug/kg		}	İ	
1,1-Dichloroethene	< 10	ug/kg				
Cis-1,2 Dichloroethylene	< 10	ug/kg		1		
Trans-1,2 Dichloroethylene	< 10	ug/kg		<u> </u>	ł	
1,2-Dichloroethene (total)	< 10	ug/kg		1		
Dichloromethane	< 30	ug/kg			ł	
1,2-Dichloropropane	< 10	ug/kg		<b>\</b>	ł	
1,3-Dichloropropane	< 10	ug/kg			ł	
2,2-Dichloropropane	< 10	ug/kg		ļ		
1,1-Dichloropropene	< 10	ug/kg		ľ	}	
Cis-1,3-Dichloropropene	<b>  &lt; 10</b>	ug/kg			ļ	
Trans-1,3-Dichloropropene	< 10	ug/kg			ĺ	
Ethylbenzene	< 10	ug/kg			ł	
fexach lorobutadiene	< 10	ug/kg			ļ	
2-Hexanone	< 100	ug/kg				
sopropyltoluene	< 50 *	ug/kg				
-Isopropyltoluene	< 50 *	ug/kg				
-Methyl-2-Pentanone	< 100	ug/kg				
laphthal ene	<b>  &lt; 10</b>	ug/kg				
n-Propylbenzene	< 10	ug/kg				
Styrene	<b>∤ &lt; 10</b>	ug/kg				
,1,1,2-Tetrachloroethane	<b>∤ &lt; 1</b> 0	ug/kg			ļ	
,1,2,2-Tetrachloroethane	< 10	ug/kg				
etrachloroethylene	< 10	ug/kg				
oluene	<b>  &lt; 10</b>	ug/kg				
,2,3-Trichlorobenzene	< 10	ug/kg				
,2,4-Trichlorobenzene	< 10	ug/kg		ľ		
,1,1-Trichloroethane	< 10	ug/kg				
,1,2-Trichloroethane	< 10	ug/kg	İ	ļ		
richloroethene	< 10	ug/kg	1	ĺ		
richlorofluoromethane	<b>  &lt; 10</b>	ug/kg	1	Ì		
,2,3-Trichloropropane	< 10	ug/kg	ľ			
,2,4-Trimethylbenzene	< 10	ug/kg	i			
,3,5-Trimethlybenzene	< 10	ug/kg				
inyl Acetate	< 5.0	ug/kg				
inyl Chloride	< 10	ug/kg	J			
P-Xylene	< 20	ug/kg				
-Xylene	< 10	ug/kg				
ylenes, Total	< 30	ug/kg				
	1	) i	1	)		

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 5

QA/QC	Ka/
	,

REPORT NUMBER: 35848

SAMPLE ID: 4101684

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-2-19

DATE SAMPLED....: 10/18/94 TIME SAMPLED....: 11:00 DATE RECEIVED....: 10/20/94

TIME RECEIVED....: 16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	< 10	ug/kg	EPA 8260	10/29/94	vs
			·		
·					

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

The results above apply only to the sample analyzed. This report may only be represented in full, and may only be submitted to a bird party with the written permission of CLEAN TECH

PAGE 6

Canton Analytical Laboratory, Inc.

QA/QC K)

CLIENT: CLEAN TECH

REPORT NUMBER: 35848 SAMPLE ID: 4101685 DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-3-14

DATE SAMPLED.....: 10/19/94 TIME SAMPLED.....: 06:05

DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
7							***************************************
	TAL METALS Aluminum, Total Barium, Total Beryllium, Total Cadmium, Total Calcium, Total Chromium, Total Cobalt, Total	1900000 11000 < 500 580 130000000 6900 < 1000	ug/kg ug/kg ug/kg ug/kg ug/kg ug/kg ug/kg		EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010	10/31/94 10/31/94 10/26/94 10/21/94 10/31/94 10/31/94	SD SD SD SD SD SD SD
]	Silver, Total Copper, Total Sodium, Total Vanadium, Total Zinc, Total	3600 8700 190000 < 1000 20000	ug/kg ug/kg ug/kg ug/kg ug/kg	•	EPA 6010 EPA 6010 EPA 6010 EPA 6010	10/21/94 10/21/94 10/31/94 10/31/94 10/21/94	SD SD SD SD SD
•	Antimony Total, GFAA Arsenic Total, GFAA Lead, Total, GFAA Selenium Total, GFAA	< 1500 5600 2000 < 1000	ug/kg ug/kg ug/kg ug/kg		EPA 7041 EPA 7060 EPA 7421 EPA 7740 EPA 7841	10/25/94 10/25/94 10/26/94 10/25/94	NS NS NS NS
	Thallium Total, GFAA Mercury, Cold Vapor TCL VOLATILE ORGANICS, GC/MS Acetone Acrolein	800 < 40 < 100 < 50	ug/kg ug/kg ug/kg ug/kg		EPA 7841 EPA 7470 EPA 8260	10/27/94 10/25/94 10/29/94	SD VS
	Acrylonitrile Benzene Bromobenzene Bromochloromethane	< 10 < 10 < 10 < 10	ug/kg ug/kg ug/kg ug/kg				
AP.	Bromodichloromethane Bromoform Bromomethane 2-Butanone n-Butylbenzene	< 10 < 10 < 10 < 100 < 100	ug/kg ug/kg ug/kg ug/kg ug/kg				
	Sec-Butylbenzene tert-Butylbenzene Carbon Disulfide Carbon Tetrachloride	< 10 < 10 < 100	ug/kg ug/kg ug/kg ug/kg ug/kg				
]	Chlorobenzene Chloroform Chloromethane 2-Chlorotoluene	< 10 < 10 < 10 < 10	ug/kg ug/kg ug/kg ug/kg				
]	4-Chlorotoluene Dibromochloromethane 1,2-Dibromoethane Dibromomethane 1,2-Dibromo-3-chloropropane	< 10 < 10	ug/kg ug/kg ug/kg ug/kg ug/kg				

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 7

Canton Analytical Laboratory, Inc.

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-3-14 REPORT NUMBER: 35848 SAMPLE ID: 4101685

DATE SAMPLED.....: 10/19/94 TIME SAMPLED ..... 06:05

DATE RECEIVED ....: 10/20/94 TIME RECEIVED....: 16:50

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
7						l	
l	1,2-Dichlorobenzene	< 10	ug/kg			l	
J.	1,3-Dichlorobenzene	< 10	ug/kg	1			
	1,4-Dichlorobenzene	< 10	ug/kg			Į.	
_	Dichlorodifluoromethane	< 10	ug/kg			]	
	1,1-Dichloroethane	< 10	ug/kg				
	1,2-Dichloroethane	< 10	ug/kg	1			
_	1,1-Dichloroethene	< 10	ug/kg			i	
	Cis-1,2 Dichloroethylene	< 10	ug/kg			ł	
	Trans-1,2 Dichloroethylene	< 10	ug/kg				
7	1,2-Dichloroethene (total)	< 10	ug/kg	]		Ì	
	Dichloromethane	< 30	ug/kg				
	1,2-Dichloropropane	< 10	ug/kg			]	
	1,3-Dichloropropane	< 10	ug/kg				
	2,2-Dichloropropane	< 10	ug/kg	j			1
٧,	1,1-Dichloropropene	< 10	ug/kg	]			
_	Cis-1,3-Dichloropropene	< 10	ug/kg			ı	}
	Trans-1,3-Dichloropropene	< 10	ug/kg	1			
ገ	Ethylbenzene	< 10	ug/kg	}			}
1	Hexachlorobutadiene	< 10	ug/kg				
اد	2-Hexanone	< 100	ug/kg				j
	Isopropyltoluene	< 50 *	ug/kg				1
_	4-Isopropyltoluene	< 50 *	ug/kg	ļ			
4	4-Methyl-2-Pentanone	< 100	ug/kg	i			
П	Naphthalene	< 10	ug/kg	l i			Į.
٦	n-Propylbenzene	< 10	ug/kg	}	ļ		ŀ
- 1	Styrene	< 10	ug/kg	į į	Į.		ŀ
П	1,1,1,2-Tetrachloroethane 1,1,2,2-Tetrachloroethane	< 10 < 10	ug/kg  ug/kg		Ì		ı
	Tetrachioroethylene	490	"-"   -		{		
	Toluene	< 10	ug/kg ug/kg				ļ
Į	1,2,3-Trichlorobenzene	< 10	ug/kg		l		ļ
	1,2,4-Trichlorobenzene	< 10	ug/kg	ŀ			ļ
	1,1,1-Trichloroethane	< 10	ug/kg				
	1,1,2-Trichloroethane	< 10	ug/kg	]			
	Trichloroethene	75	ug/kg	]			
- 1	Trichlorofluoromethane	< 10	ug/kg	]	]		
2.2	1,2,3-Trichloropropene	< 10	ug/kg	]			ĺ
	1,2,4-Trimethylbenzene	< 10	ug/kg		Ì		ì
	1,3,5-Trimethlybenzene	< 10	ug/kg	İ			ľ
	Vinyl Acetate	< 5.0	ug/kg	İ	i		1
1	Vinyl Chloride	< 10	ug/kg		1		I
	MP-Xylene	< 20	ug/kg		}		j
	0-Xylene		ug/kg		İ		1
1	Xylenes, Total	< 30	ug/kg				į.
4							ı
Ħ			ì	j			1

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

The results above apply only to the sample analyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of CLEAN TECH

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Canton Analytical Laboratory, Inc.

QA/QC KI

REPORT NUMBER: 35848 SAMPLE ID: 4101685 DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-3-14

DATE SAMPLED....: 10/19/94 TIME SAMPLED....: 06:05 DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

ANALYTICAL	UNITS OF	ANALYTICAL	RUN

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
]	Chloroethane	< 10	ug/kg	EPA 8260	10/29/94	Vs
}						
1						
}						

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 9

Canton Analytical Laboratory, Inc.

GA/GC FI

REPORT NUMBER: 36123

SAMPLE ID: 4110327

DESCRIPTION: PROJECT-CHRYSLER ACUSTAR S84-14

DATE SAMPLED.....: 10/31/94 TIME SAMPLED.....: 13:30

DATE RECEIVED....: 11/03/94

TIME RECEIVED ....: 09:45

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
						100000000000000000000000000000000000000
AL METALS						
luminum, Total	1900000	ug/kg		EPA 6010	11/15/94	SD
arium, Total	11000	ug/kg	1	EPA 6010	11/08/94	SD
eryllium, Total	220	ug/kg	1	EPA 6010	11/15/94	SD
admium, Total	670	ug/kg		EPA 6010	11/15/94	SD
alcium, Total	260000000	ug/kg	ļ .	EPA 6010	11/15/94	SD
hromium, Total	5600	ug/kg	İ	EPA 6010	11/08/94	SD
obalt, Total	4600	ug/kg	į i	EPA 6010	11/15/94	SD
ilver, Total	2700	ug/kg	1	EPA 6010	11/15/94	SD
opper, Total	8700	ug/kg		EPA 6010	11/08/94	SD
odium, Total	200000	ug/kg		EPA 6010	11/15/94	SD
anadium, Total	8200	ug/kg		EPA 6010	11/15/94	SD
inc, Total	16000	ug/kg		EPA 6010	11/08/94	SD
ntimony Total, GFAA	< 1500	ug/kg	[	EPA 7041	11/10/94	NS
rsenic Total, GFAA	3000	ug/kg	1	EPA 7060	11/09/94	NS
ead, Total, GFAA	2000	ug/kg		EPA 7421	11/09/94	NS
elenium Total, GFAA	< 1000	ug/kg		EPA 7740	11/10/94	NS
hallium Total, GFAA	1700	ug/kg		EPA 7841	11/09/94	NS
	1700   < 40		<b>(</b>	EPA 7471	11/08/94	SD
ercury, Cold Vapor	1 40	ug/kg		_	1 .	VS
CL VOLATILE ORGANICS, GC/MS	. 400			EPA 8260	11/13/94	V5
cetone	< 100	ug/kg				
crolein	< 50	ug/kg			Į .	
crylonitrile	< 10	ug/kg			ĺ	
enzene	< 10	ug/kg			ĺ	
romobenzene	< 10	ug/kg				
romochloromethane	< 10	ug/kg			i	
romodichloromethane -	< 10	ug/kg			Ì	
romoform	< 10	ug/kg				
romomethane	< 10	ug/kg				
Butanone	< 100	ug/kg				
Butylbenzene	< 10	ug/kg			1	
ec-Butylbenzene	< 10	ug/kg		ı		
ert-Butylbenzene	< 10	ug/kg				
rbon Disulfide	< 100	ug/kg	`			
rbon Tetrachloride	< 10	ug/kg	1			
lorobenzene	< 10	ug/kg	. 1		1	
loroform	< 10	ug/kg	ł			
loromethane	< 10	ug/kg	I			
Chlorotoluene	< 10	ug/kg	ł			
Chlorotoluene	< 10	ug/kg	l			
bromochloromethane	<b>√ &lt; 10</b> -	ug/kg	Į			
2-Dibromoethane	< 10	ug/kg	Į			
	< 10			l l		
bromomethane 2-Dibromo-3-chloropropane	( < 10	lug/kg	Į.	· ·		

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 1

Canton Analytical Laboratory, Inc.

CLIENT: CLEAN TECH

REPORT NUMBER: 36123 SAMPLE ID: 4110327

DESCRIPTION: PROJECT-CHRYSLER ACUSTAR SB4-14

DATE SAMPLED....: 10/31/94 TIME SAMPLED .....: 13:30

DATE RECEIVED....: 11/03/94

TIME RECEIVED....: 09:45

	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
]1,2-Dichlorobenzene	< 10	ug/kg	1			
1,3-Dichlorobenzene	< 10	ug/kg	Ì			
1,4-Dichlorobenzene	< 10	ug/kg	t		1	
Dichlorodifluoromethane	< 10	ug/kg	1			
1,1-Dichloroethane	< 10	ug/kg	-	Į.		
1,2-Dichloroethane	< 10	ug/kg		1	1	
1,1-Dichloroethene	< 10	ug/kg	- }	<b>,</b>	1	
Cis-1,2 Dichloroethylene	< 10	ug/kg	l.		1	
Trans-1,2 Dichloroethylene	j < 10	ug/kg		1	1	
11,2-Dichloroethene (total)	< 20	ug/kg	[	1	İ	
Dichloromethane	< 10	ug/kg		1	1	
1,2-Dichloropropane	< 10	ug/kg		ļ	1	
1,3-Dichloropropane	< 10	ug/kg	1	j .	1	
2,2-Dichloropropane	< 10	ug/kg		İ	i	
1,1-Dichloropropene	) < 10	ug/kg	1	)	1	
Cis-1,3-Dichloropropene	< 10	ug/kg	ı	]	ł	
Trans-1,3-Dichloropropene	< 10	ug/kg		]	}	
Ethylbenzene	< 10	ug/kg		ļ	ļ	
Hexachlorobutadiene	< 10	ug/kg				
2-Hexanone	< 100	ug/kg		ļ	1	
Isopropyltoluene	< 50 *	ug/kg			1	
K-Isopropyltoluene	< 50 *	ug/kg	į.	Į.	Ţ	
K-Methyl-2-Pentanone	< 100	ug/kg			]	
Naphthalene	< 10	ug/kg		}	}	
n-Propylbenzene	< 10	ug/kg			1	
Styrene	< 10	ug/kg			<b>S</b>	
1,1,1,2-Tetrachloroethane	< 10	ug/kg			l	
1,1,2,2-Tetrachloroethane	< 10	ug/kg	ľ		1	
Tetrachloroethylene Toluene	14	ug/kg	j		[	
	< 10	ug/kg	i i		Ì	
1,2,3-Trichlorobenzene	< 10	ug/kg				
1,2,4-Trichlorobenzene	< 10	ug/kg				
M,1,1-Trichloroethane	< 10	ug/kg			ł	
1,1,2-Trichloroethane	< 10	ug/kg			ļ	
Trichloroethene Trichlorofluoromethane	< 10	ug/kg	į į		ţ	
,2,3-Trichloropropane	< 10 < 10	ug/kg	1		1	
1,2,4-Trimethylbenzene	< 10	ug/kg	1		[	
1,2,4-rrimethytbenzene 1,3,5-Trimethlybenzene	< 10	ug/kg	1		1	
Vinyl Acetate	< 10	ug/kg	{		ļ.	
Vinyl Chloride	< 10	ug/kg			]	
MP-Xylene	< 20	lug/kg	{		1	
0-Xylene	< 10	ug/kg ug/kg			!	
Xylenes, Total	< 30	,	1		1	
1,	` 30	ug/kg	]		1	
}	<b>\</b>		1		1	
1	ſ	ı	1		1	

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 2

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Canton Analytical Laboratory, Inc.

1								
1		LABO	RATORY R CLIENT: CLEAN T					
إ	REPORT NUMBER: 36123 SAMPLE ID: 4110	327 DESCRIPTION	ON: PROJECT-CHRYS	LER ACUSTAR SB4-1	14			
	DATE SAMPLED: 10/31/94 TIME SAMPLED: 13:30			DA T I	DATE RECEIVED: 11/03/94 TIME RECEIVED: 09:45			
	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN	
	Chloroethane	< 10	ug/kg					
	-							
1						!		
]						1		
1								
L								

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 3

Canton Analytical Laboratory, Inc.

ga/gc KW

CLIENT: CLEAN TECH

REPORT NUMBER: 36270

SAMPLE ID: 4111216

DESCRIPTION: PROJECT-CHRYSLER ACUSTAR \$84-30

DATE SAMPLED....: 10/29/94 TIME SAMPLED....: 14:15

DATE RECEIVED....: 11/09/94 TIME RECEIVED....: 09:45

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
]	Total Organic Carbon	35000	mg/kg	EPA 9060M	11/16/94	AF
ţ						
]						
7						
1						
}						

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PAGE 7

Canton Analytical Laboratory, Inc.

QA/QC KN

SAMPLE ID: 4101687

REPORT NUMBER: 35848 DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-5-29

DATE SAMPLED.....: 10/19/94 TIME SAMPLED.....: 15:30

DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
ANALISES PERFORMED	RESULT	HEASURE	METRO	VAIE	TECHA
TAL METALS					
Aluminum, Total	1700000	ug/kg	EPA 6010	10/31/94	SD
Barium, Total	12000	ug/kg	EPA 6010	10/21/94	SD
Beryllium, Total	< 500	ug/kg	EPA 6010	10/26/94	SD
Cadmium, Total	530	ug/kg	EPA 6010	10/26/94	SD
Calcium, Total	130000000	ug/kg	EPA 6010	10/20/94	SD
Chromium, Total	8700	, • •	EPA 6010	10/26/94	SD
	< 1000	ug/kg	EPA 6010		SD
Cobalt, Total	3300	ug/kg		10/31/94	SD
Silver, Total	8000	ug/kg	EPA 6010	10/26/94	SD
Copper, Total		ug/kg	EPA 6010	10/21/94	SD
Sodium, Total	210000	ug/kg	EPA 6010	10/31/94	
Vanadium, Total	< 1000	ug/kg	EPA 6010	10/31/94	SD SD
Zinc, Total	14000	ug/kg	EPA 6010	10/26/94	
Antimony Total, GFAA	< 1500	ug/kg	EPA 7041	10/25/94	NS
Arsenic Total, GFAA	16000	ug/kg	EPA 7060	10/25/94	NS
Lead, Total, GFAA	3000	ug/kg	EPA 7421	10/26/94	NS
Selenium Total, GFAA	< 1000	ug/kg	EPA 7740	10/25/94	NS
Thallium Total, GFAA	1400	ug/kg	EPA 7841	10/27/94	NS
Mercury, Cold Vapor	< 40	ug/kg	EPA 7470	10/25/94	SD
TCL VOLATILE ORGANICS, GC/MS	ł		EPA 8260	10/29/94	VS
Acetone	< 100	ug/kg		ĺ	
Acrolein	< 50	ug/kg	İ	ł	
Acrylonitrile	< 10	ug/kg	ł	1	
Benzene	< 10	ug/kg		ì	
Bromobenzene	< 10	ug/kg	ĺ	]	
Bromochloromethane	< 10	ug/kg	ĺ	ļ	
Bromodichloromethane	<b>∤ &lt; 10</b>	ug/kg		ļ	
Bromoform	< 10	ug/kg			
Bromomethane	< 10	ug/kg			
2-Butanone	< 100	ug/kg	1		
n-Butylbenzene	< 10	ug/kg			
Sec-Butylbenzene	<b>  &lt; 10</b>	ug/kg			
tert-Butylbenzene	< 10	ug/kg			
Carbon Disulfide	< 100	ug/kg			
Carbon Tetrachloride	<b>  &lt; 10</b>	ug/kg			
Chlorobenzene	< 10	ug/kg	]		
Chloroform	< 10	ug/kg			
Chloromethane	< 10	ug/kg			
2-Chlorotaluene	< 10	ug/kg	<b>j</b>	'	
4-Chlorotoluene	< 10	ug/kg	į		
Dibromochloromethane	< 10	ug/kg			
1,2-Dibromoethane	< 10	ug/kg	ļ		
Dibromomethane	< 10	ug/kg	1		
1,2-Dibromo-3-chloropropane	< 10	ug/kg	[		
' <i>'</i>	1		[		
		1	l l		

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 13

Canton Analytical Laboratory, Inc.

QA/QC

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DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-5-29 REPORT NUMBER: 35848 SAMPLE ID: 4101687

DATE SAMPLED....: 10/19/94 TIME SAMPLED..... 15:30

DATE RECEIVED....: 10/20/94

TIME RECEIVED....: 16:50

,2-Dichlorobenzene ,3-Dichlorobenzene ,4-Dichlorobenzene ichlorodifluoromethane ,1-Dichloroethane	< 10 < 10 < 10	ug/kg ug/kg				
,3-Dichlorobenzene ,4-Dichlorobenzene ichlorodifluoromethane ,1-Dichloroethane	< 10 < 10	1 0. 0	1			
,4-Dichlorobenzene ichlorodifluoromethane ,1-Dichloroethane	< 10	lug/kg	1		1	
ichlorodifluoromethane ,1-Dichloroethane		U U	1		1	
,1-Dichloroethane		ug/kg	l i			
	< 10	ug/kg		i	I	
	< 10	ug/kg			1	
,2-Dichloroethane	< 10	ug/kg			1	
,1-Dichloroethene	< 10	ug/kg			1	
is-1,2 Dichloroethylene	<b>  &lt; 10</b>	ug/kg	}		}	
rans-1,2 Dichloroethylene	< 10	ug/kg				
,2-Dichloroethene (total)	< 10	ug/kg	{		1	
ichloromethane	< 30	ug/kg	}		1	
,2-Dichloropropane	< 10	ug/kg	}	l	1	
3-Dichloropropane	< 10	ug/kg	]		1	
2-Dichloropropane	< 10	ug/kg	į į		Į.	
1-Dichloropropene	< 10	ug/kg	1		1	
s-1,3-Dichloropropene	< 10	ug/kg			(	
ans-1,3-Dichloropropene	< 10	ug/kg	ļ		ł	
hylbenzene	< 10	ug/kg	ļ į		ļ	
exachlorobutadiene	< 10	ug/kg	[		İ	
Hexanone	< 100	ug/kg	i		[	
opropyltoluene	< 50 *	ug/kg				
Isopropyltoluene	< 50 *	ug/kg	}		J	
Methyl-2-Pentanone	< 100	ug/kg			1	
phthalene	< 10	ug/kg				
Propylbenzene	<b>) &lt; 10</b>	ug/kg	}			
yrene	< 10	ug/kg			ľ	
1,1,2-Tetrachloroethane -	< 10	ug/kg	İ			
1,2,2-Tetrachloroethane	< 10	ug/kg			ļ	
trachloroethylene	860	ug/kg	)		)	
luene	< 10	ug/kg			1	
2,3-Trichlorobenzene	< 10	ug/kg			]	
2,4-Trichlorobenzene	< 10	ug/kg			1	
1,1-Trichloroethane	< 10	ug/kg	j		ĺ	
1,2-Trichloroethane	< 10	ug/kg			1	
ichloroethene	47	ug/kg	}		i	
ichlorofluoromethane	< 10	ug/kg	]		]	
2,3-Trichloropropane	<b>√ 10</b>	ug/kg	1		}	
2,4-Trimethylbenzene	< 10	ug/kg	1		1	
3,5-Trimethlybenzene	< 10	ug/kg	İ		1	
nyl Acetate	< 5.0	ug/kg			ļ	
nyl Chloride	< 10	ug/kg	4		}	
-Xylene	< 20	ug/kg	i i			
Kylene	< 10	ug/kg	ļ			
lenes, Total	< 30	ug/kg				

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 14

Canton Analytical Laboratory, Inc.

GA/GC XV

REPORT NUMBER: 35848

SAMPLE ID: 4101687

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-5-29

DATE SAMPLED.....: 10/19/94 TIME SAMPLED.....: 15:30

DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	< 10	ug/kg	EPA 8260	10/29/94	vs
1						

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 15

Canton Analytical Laboratory, Inc.

DA/OC\_KI

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CLIENT: CLEAN TECH

REPORT NUMBER: 35905 SAMPLE ID: 4102042 DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB6-14

DATE SAMPLED.....: 10/20/94 TIME SAMPLED.....: 13:30 DATE RECEIVED ....: 10/25/94

TIME RECEIVED ....: 09:30

.}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
า				Ţ		\ <u></u>	
	TAL METALS		{				
	Aluminum, Total	1800	mg/kg		EPA 6010	10/31/94	SD
	Barium, Total	11	mg/kg		EPA 6010	10/28/94	SD
R	Beryllium, Total	< 0.5	mg/kg		EPA 6010	10/31/94	SD
	Cadmium, Total	0.72	mg/kg		EPA 6010	10/28/94	SD
	Calcium, Total	100000	mg/kg	ł	EPA 6010	10/31/94	SD
	Chromium, Total	5.0	mg/kg	1	EPA 6010	10/28/94	SD
	Cobalt, Total	< 1.0 3.2	mg/kg	1	EPA 6010	10/31/94	SD SD
4	Silver, Total	9.7	mg/kg	1	EPA 6010	10/28/94	SD SD
J	Copper, Total	210	mg/kg	i I	EPA 6010 EPA 6010	10/28/94 10/31/94	SD
_	Sodium, Total Vanadium, Total	< 1.0	mg/kg mg/kg	į į	EPA 6010	10/31/94	SD
	Zinc, Total	17	mg/kg	ì	EPA 6010	10/31/94	SD
1	Antimony Total, GFAA	< 1.5	mg/kg	į l	EPA 7041	11/03/94	NS
ı	Arsenic Total, GFAA	5.4	mg/kg		EPA 7041	11/01/94	NS
*	Lead, Total, GFAA	2.4	mg/kg	<u> </u>	EPA 7421	11/01/94	NS I
	Selenium Total, GFAA	< 1.0	mg/kg	[	EPA 7740	11/01/94	NS .
ור	Thallium Total, GFAA	1.8	mg/kg		EPA 7841	11/02/94	NS .
Ш	Mercury, Cold Vapor	< 0.04	mg/kg	ļ i	EPA 7471	10/27/94	SD
IJ	TCL VOLATILE ORGANICS, GC/MS		-	ľ	EPA 8260	10/29/94	vs
ļ	Acetone	< 100	ug/kg				ſ
٠,	Acrolein	< 50	ug/kg	i			1
H	Acrylonitrile	< 10	ug/kg		ļ		
J	Benzene	< 10	ug/kg		ł		ł
_	Bromobenzene	< 10	ug/kg	1	ł		ŀ
Į	Bromochloromethane	< 10	ug/kg				Į.
	Bromodichloromethane	< 10	ug/kg				
	Bromoform	< 10	ug/kg	1	Í		}
	Bromomethane	< 10	ug/kg	Ĭ	[		Į
	2-Butanone	< 100	ug/kg	j	i		į
	n-Butylbenzene	< 10 < 10	ug/kg		1		İ
И	Sec-Butylbenzene tert-Butylbenzene	< 10	ug/kg ug/kg				1
٥į	Carbon Disulfide	< 100	ug/kg ug/kg				į
1	Carbon Tetrachloride	< 10	ug/kg		)		1
-	Chlorobenzene	< 10	ug/kg	ţ			
201	Chloroform	< 10	ug/kg	İ	[		
J	Chioromethane	< 10	ug/kg		Ì		1
- [	2-Chlorotoluene	< 10	ug/kg	Ţ	[		į
-	4-Chlorotoluene	< 10	ug/kg		ľ		
	Dibromochloromethane		ug/kg	}			Ì
	1,2-Dibromoethane		ug/kg		1		1
	Dibromomethane		ug/kg	}	j		1
	1,2-Dibromo-3-chloropropane	< 10	ug/kg	}	}		1
H	<b>{</b>			\$	<b>{</b>		}

<sup>\*</sup> The method blank for 10/29/94 contained 20 ug/kg of Dichloromethane. (The method blank can range from < 10 to 30 ug/kg.)
\*\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2,EPA Method 8260

PAGE 1

CLIENT: CLEAN TECH

REPORT NUMBER: 35905 SAMPLE ID: 4102042 DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB6-14

DATE SAMPLED....: 10/20/94 TIME SAMPLED....: 13:30 DATE RECEIVED ....: 10/25/94

TIME RECEIVED....: 09:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
ANALISES PERFORMED	RESOLT	HEASONE		1 1100	T DATE	120114
1,2-Dichlorobenzene	< 10	ug/kg				
1.3-Dichlorobenzene	< 10	ug/kg		}		
1,4-Dichlorobenzene	< 10	ug/kg		ł	1	
Dichlorodifluoromethane	< 10	ug/kg		}		
1,1-Dichloroethane	< 10	ug/kg		<b>!</b>		
1,2-Dichloroethane	< 10	ug/kg		1	1	
1,1-Dichloroethene	< 10	ug/kg		}	1	
Cis-1,2 Dichloroethylene	< 10	ug/kg		1	1	
Trans-1,2 Dichloroethylene	< 10	ug/kg		1	1	
1,2-Dichloroethene (total)	< 10	ug/kg		ł	1	
Dichloromethane	37 *	ug/kg				
1,2-Dichloropropane	< 10	ug/kg		ł		
1,3-Dichloropropane	< 10	ug/kg		ļ		
2,2-Dichloropropane	< 10	ug/kg		ł	1	
1,1-Dichloropropene	< 10	ug/kg		ļ	1	
Cis-1,3-Dichloropropene	< 10	ug/kg		}		
Trans-1,3-Dichloropropene	< 10	ug/kg		ļ	j	
Ethylbenzene	< 10	ug/kg			1	
Hexachlorobutadiene	< 10	ug/kg			(	
2-Hexanone	< 100	ug/kg			i	
Isopropyltoluene	\ < 50 <b>**</b>	ug/kg			1	
4-Isopropyltoluene	< 50 **	ug/kg				
4-Methyl-2-Pentanone	< 100	ug/kg			1	
Naphthalene	< 10	ug/kg			1	
n-Propyl benzene	< 10	ug/kg		•	1	
Styrene	< 10	ug/kg			1	
1,1,1,2-Tetrachloroethane	< 10	ug/kg			1	
1,1,2,2-Tetrachloroethane	< 10	ug/kg			1	
Tetrachloroethylene	√ 38	ug/kg			J	
Toluene	< 10	ug/kg			}	
1,2,3-Trichlorobenzene	< 10	ug/kg			Į	
1,2,4-Trichlorobenzene	<b>) &lt; 10</b>	ug/kg			1	
1,1,1-Trichloroethane	< 10	ug/kg			ĺ	
1,1,2-Trichloroethane	< 10	ug/kg			ļ	
Trichloroethene	54	ug/kg			1	
Trichlorofluoromethane	< 10	ug/kg				
1,2,3-Trichloropropane	< 10	ug/kg	}		1	
1,2,4-Trimethylbenzene	< 10	ug/kg				
1,3,5-Trimethlybenzene	< 10	ug/kg	ì		1	
Vinyl Acetate	< 5.0	ug/kg	ł		Į.	
Vinyl Chloride	< 10	ug/kg			1	
MP-Xylene	< 20	ug/kg	ļ		1	
0-Xylene	< 10	lug/kg	1			
Kylenes, Total	< 30	ug/kg			1	
		i l			ŀ	

\* The method blank for 10/29/94 contained 20 ug/kg of Dichloromethane. (The method blank can range from < 10 to 30 ug/kg.)
\*\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2,EPA Method 8260

QA/QC KI

REPORT NUMBER: 35905 SAMPLE ID: 4102042 DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB6-14

DATE SAMPLED.....: 10/20/94 TIME SAMPLED.....: 13:30

DATE RECEIVED ....: 10/25/94

TIME RECEIVED....: 09:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	< 10	ug/kg				
			}			
					i i	

\* The method blank for 10/29/94 contained 20 ug/kg of Dichloromethane. (The method blank can range from < 10 to 30 ug/kg.)
\*\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 3

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 35905

SAMPLE ID: 4102043

DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB7-24

DATE SAMPLED.....: 10/20/94 TIME SAMPLED ..... 09:10

DATE RECEIVED....: 10/25/94 TIME RECEIVED....: 09:30

Ì		ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
J	ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
		<del>1</del>	,			de la seconda de la companya della companya della companya de la companya della c	
1					]	ŀ	ļ
l	TAL METALS			1		)	1
J	Aluminum, Total	1500	mg/kg	1	EPA 6010	10/31/94	SD
	Barium, Total	12	mg/kg	ì	EPA 6010	10/28/94	SD
_	Beryllium, Total	< 0.5	mg/kg	İ	EPA 6010	10/31/94	SD
B	Cadmium, Total	0.73	mg/kg	}	EPA 6010	10/28/94	SD
	Calcium, Total	140000	mg/kg		EPA 6010	10/31/94	SD
•	Chromium, Total	4.5	mg/kg	Î	EPA 6010	10/28/94	SD
	Cobalt, Total	< 1.0	mg/kg	1	EPA 6010	10/31/94	SD
	Silver, Total	2.8	mg/kg	}	EPA 6010	10/28/94	SD
7	Copper, Total	10	mg/kg	1	EPA 6010	10/28/94	SD
J	Sodium, Total	180	mg/kg		EPA 6010	10/31/94	SD
	Vanadium, Total	< 1.0	mg/kg	1	EPA 6010	10/31/94	SD
_	Zinc, Total	24	mg/kg	<b>\</b>	EPA 6010	10/28/94	SD
4	Antimony Total, GFAA	< 1.5	mg/kg		EPA 7041	11/03/94	NS
ı	Arsenic Total, GFAA	6.4	mg/kg	{	EPA 7060	11/01/94	NS'
7	Lead, Total, GFAA	2.5	mg/kg	ł	EPA 7421	11/01/94	NS
	Selenium Total, GFAA	< 1.0	mg/kg	1	EPA 7740	11/01/94	NS
7	Thallium Total, GFAA	1.2	mg/kg		EPA 7841	11/02/94	NS
ļ	Mercury, Cold Vapor	< 0.04	mg/kg		EPA 7471	10/27/94	SD
J	TCL VOLATILE ORGANICS, GC/MS				EPA 8260	10/29/94	vs [
	Acetone	< 100	ug/kg			1	[
	Acrolein	< 50	ug/kg			ļ	
٦.	Acrylonitrile	< 10	ug/kg	<b>!</b>		ļ	Į.
1	Benzene	<b>  &lt; 10</b>	ug/kg		•	1	į
	Bromobenzene	<b>  &lt; 10</b>	ug/kg				Ţ
1	Bromochloromethane	< 10	ug/kg	<u> </u>			i
,	Bromodichloromethane	< 10	ug/kg		į	į	Į.
	Bromoform	< 10	ug/kg	1			
J	Bromomethane	< 10	ug/kg	į į			ļ
1	2-Butanone	< 100	ug/kg				1
_	n-Butylbenzene	< 10	ug/kg	1			i
Į	Sec-Butylbenzene	< 10	ug/kg				ŀ
	tert-Butylbenzene	<b>  &lt; 10</b>	ug/kg				l
7	Carbon Disulfide	< 100	ug/kg				i
- (	Carbon Tetrachloride	< 10	ug/kg	[			ı
٦İ	Chlorobenzene	< 10	ug/kg	l i			
U	Chloroform	< 10	ug/kg	[			l
시	Chloromethane	< 10	ug/kg				İ
ı	2-Chlorotoluene	< 10	ug/kg	Į Į			į.
,	4-Chlorotoluene	< 10	ug/kg				- 1
	Dibromochloromethane	< 10	ug/kg		į		ł
31	1,2-Dibromoethane Dibromomethane	< 10	ug/kg				j
	1,2-Dibromo-3-chloropropane	< 10	ug/kg		Į		[
١	i, a violuno-3-chtoropropane	< 10	ug/kg		Ì		j
1							i
11				1	Ì		}

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 4

Canton Analytical Laboratory, Inc.

CLIENT: CLEAN TECH

REPORT NUMBER: 35905 SAMPLE ID: 4102043 DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB7-24

DATE SAMPLED....: 10/20/94 TIME SAMPLED....: 09:10 DATE RECEIVED....: 10/25/94

TIME RECEIVED ....: 09:30

	ANALYTICAL	UNITS OF	1	ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECH
,2-Dichlorobenzene	< 10	/k-				
,3-Dichlorobenzene	1	ug/kg		i		
	< 10 < 10	ug/kg	1	}	ì	
,4-Dichlorobenzene	1 '-	ug/kg		1	1	
richlorodifluoromethane	< 10	ug/kg		ľ		
,1-Dichloroethane	< 10	ug/kg		(	į.	
,2-Dichloroethane	< 10	ug/kg		}	1	
,1-Dichloroethene	< 10	ug/kg		}	1	
is-1,2 Dichloroethylene	< 10	ug/kg	1		1	
rans-1,2 Dichloroethylene	< 10	ug/kg	·	<b>{</b>	}	
,2-Dichloroethene (total)	< 10	ug/kg		f		
ichloromethane	< 30	ug/kg	j		1	
,2-Dichloropropane	< 10	ug/kg	I	l	J	
,3-Dichloropropane	< 10	ug/kg	1	}	}	
,2-Dichloropropane	< 10	ug/kg	1	[	1	
,1-Dichloropropene	< 10	ug/kg	1	i	1	
is-1,3-Dichloropropene	< 10	ug/kg	l .	Į	1	
rans-1,3-Dichloropropene	< 10	ug/kg	!		1	
thylbenzene	< 10	ug/kg	<b>,</b>			
exach lorobutadi ene	< 10	ug/kg			ŀ	
- Hexanone	< 100	ug/kg	1		j	
sopropyltoluene	< 50 *	ug/kg			1	
-Isopropyltoluene	< 50 *	ug/kg				
-Methyl-2-Pentanone	< 100	ug/kg	i i			
aphthalene	< 10	ug/kg	}		]	
-Propylbenzene	< 10	ug/kg	{			
tyrene	l < 10	ug/kg				
,1,1,2-Tetrachloroethane	< 10	ug/kg	ļ	1	į.	
,1,2,2-Tetrachloroethane	< 10	ug/kg				
etrachloroethylene	280	ug/kg	<b>\</b>		l	
oluene	< 10	ug/kg			ĺ	
,2,3-Trichlorobenzene	< 10	ug/kg			i	
,2,4-Trichlorobenzene	< 10	ug/kg			1	
1,1-Trichloroethane	< 10	ug/kg lug/kg	]		1	
1,2-Trichloroethane	< 10	ug/kg ug/kg	į į		l	
ichloroethene	20	ug/kg			ł	
richtoroftuoromethane	< 10	lug/kg	¦		{	
2,3-Trichloropropane	< 10	ug/kg ug/kg			l	
2,4-Trimethylbenzene	< 10	ug/kg ug/kg			}	
3,5-Trimethybenzene	< 10	lug/kg lug/kg				
nyl Acetate	< 5.0	1			1	
inyl Chloride	< 10	ug/kg			Ī	
rryk chtoride Y-Xylene	< 20	ug/kg			1	
O / Neure	_	ug/kg			ł	
Xylene Venes, Total	< 10 < 30	ug/kg ug/kg				

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 5

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DA/OC\_ KIL

		LABORATORY RESULTS CLIENT: CLEAN TECH	
REPORT NUMBER: 35905	SAMPLE ID: 4102043	DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB7-24	

DATE SAMPLED.....: 10/20/94 TIME SAMPLED.....: 09:10 DATE RECEIVED....: 10/25/94 TIME RECEIVED....: 09:30

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	< 10	ug/kg			
]	-					
	-					
						ì

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 6

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_KN

CLIENT: CLEAN TECH

REPORT NUMBER: 35848

SAMPLE ID: 4101686

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-8-24

DATE SAMPLED.....: 10/19/94 TIME SAMPLED....: 10:20

DATE RECEIVED....: 10/20/94

TIME RECEIVED....: 16:50

J	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
7				l			
İ	TAL METALS	Ì	j				
J	Aluminum, Total	1900000	ug/kg		EPA 6010	10/31/94	SD
	Barium, Total	11000	ug/kg	į	EPA 6010	10/21/94	SD
	Beryllium, Total	< 500	ug/kg		EPA 6010	10/26/94	SD
	Cadmium, Total	470	ug/kg	ĺ	EPA 6010	10/26/94	SD
	Calcium, Total	110000000	ug/kg		EPA 6010	10/31/94	SD
	Chromium, Total	6800	ug/kg	1	EPA 6010	10/26/94	SD
	Cobalt, Total	< 1000	ug/kg	i	EPA 6010	10/31/94	SD
	Silver, Total	2900	ug/kg		EPA 6010	10/26/94	SD
1	Copper, Total	7200	ug/kg	<u>'</u>	EPA 6010	10/21/94	SD
J	Sodium, Total	160000	ug/kg	<b>!</b>	EPA 6010	10/31/94	SD
	Vanadium, Total	< 1000	ug/kg		EPA 6010	10/31/94	SD
3	Zinc, Total	16000	ug/kg	1	EPA 6010	10/26/94	SD
	Antimony Total, GFAA	< 1500	ug/kg		EPA 7041	10/25/94	NS
•	Arsenic Total, GFAA	6500	ug/kg		EPA 7060	10/25/94	NS
	Lead, Total, GFAA	4100	ug/kg		EPA 7421	10/26/94	NS
_	Selenium Total, GFAA	< 1000	ug/kg		EPA 7740	10/25/94	NS
1	Thallium Total, GFAA Mercury, Cold Vapor	1100 < 40	ug/kg		EPA 7841 EPA 7470	10/27/94 10/25/94	NS SD
ı	TCL VOLATILE ORGANICS, GC/MS	<b>~ 4</b> 0	ug/kg		EPA 8260	10/29/94	VS
•	Acetone	< 100	ug/kg		EFA GEOU	10/27/74	V3
	Acrolein	< 50	ug/kg				
1	Acrylonitrile	< 10	ug/kg				
П	Benzene	< 10	ug/kg				
,	Bromobenzene	< 10	ug/kg				i
	Bromochloromethane	< 10	ug/kg				
1	Bromodichloromethane	< 10	ug/kg				
1	Bromoform	< 10	ug/kg				
	Bromomethane .	< 10	ug/kg				Į
	2-Butanone	< 100	ug/kg				
.	n-Butyl benzene	< 10	ug/kg				1
	Sec-Butylbenzene	< 10	ug/kg		1		
	tert-Butylbenzene	< 10	ug/kg				Ì
	Carbon Disulfide	< 100	ug/kg		ĺ		J
	Carbon Tetrachloride		ug/kg		J		]
	Chloroform Chloroform		ug/kg		j		}
1	Chloromethane	17	ug/kg	-			i
'	2-Chlorotoluene	1.7	ug/kg ug/kg		1		- 1
- }	4-Chlorotoluene	11	ug/kg	}	}		1
1	Dibromochloromethane	1.1	ug/kg	}	1		1
	1,2-Dibromoethane	11	ug/kg	1	İ		1
•	Dibromomethane		ug/kg	1	}		j
- 1	1,2-Dibromo-3-chloropropane		ug/kg	f	j		1
IJ			-3,3	İ	ĺ		}
H	}	}	}	-	ţ		}
ŀ	<del></del>			<u></u>			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 10

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Canton Analytical Laboratory, Inc.

QA/QC

CLIENT: CLEAN TECH

REPORT NUMBER: 35848 SAMPLE ID: 4101686 DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-8-24

DATE SAMPLED....: 10/19/94 TIME SAMPLED....: 10:20 DATE RECEIVED....: 10/20/94 TIME RECEIVED....: 16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECH
		TIENOOKE			1	
,2-Dichlorobenzene	< 10	ug/kg				
,3-Dichlorobenzene	< 10	ug/kg		·	ŀ	
,4-Dichlorobenzene	< 10	ug/kg				
ichlorodifluoromethane	< 10	ug/kg				
,1-Dichloroethane	< 10	ug/kg			}	
,2-Dichloroethane	< 10	ug/kg			1	
,1-Dichloroethene	< 10	ug/kg		1	1	
is-1,2 Dichloroethylene	< 10	ug/kg				
rans-1,2 Dichloroethylene	< 10	ug/kg			l	
,2-Dichloroethene (total)	< 10	ug/kg		l		
ichloromethane	< 30	ug/kg				
,2-Dichloropropane	< 10	ug/kg		1		
,3-Dichloropropane	< 10	ug/kg				
,2-Dichloropropane	< 10	ug/kg	Ţ			
,1-Dichloropropene	< 10	ug/kg				
is-1,3-Dichloropropene	< 10	ug/kg	J		1	
rans-1,3-Dichloropropene	< 10	ug/kg	1			
thylbenzene	< 10	ug/kg	}		1	
exachlorobutadiene	< 10	ug/kg			i	
Hexanone	< 100	ug/kg			1	
sopropyltoluene	< 50 *	ug/kg			1	
Isopropyltoluene	< 50 *	ug/kg	i			
Methyl-2-Pentanone	< 100	ug/kg	ļ			
phthalene	< 10	ug/kg	İ		1	
Propylbenzene	< 10	ug/kg	t		1	
Cyrene	< 10	ug/kg			ĺ	
1,1,2-Tetrachloroethane	< 10	ug/kg	1		1 .	
1,2,2-Tetrachloroethane	₹ 10	ug/kg				
trachloroethylene	480	ug/kg	{		}	
luene	< 10	ug/kg				
2,3-Trichlorobenzene	< 10	ug/kg	1			
2,4-Trichlorobenzene	< 10	ug/kg	ŧ			
1,1-Trichloroethane	< 10	ug/kg	j		1	
1,2-Trichloroethane	₹ 10	ug/kg				
ichloroethene	< 10	ug/kg	ļ			
ichlorofluoromethane	₹ 10	ug/kg	ĺ		1	
2,3-Trichloropropane	< 10	ug/kg				
2,4-Trimethylbenzene	\ \ \ 10	1 1	ļ		ŀ	
3,5-Trimethlybenzene	< 10	ug/kg				
nyl Acetate	< 5.0	ug/kg	1			
nyl Chloride	< 10	ug/kg			]	
-Xylene	< 20	ug/kg	j		ŀ	
xylene Xylene	< 10	ug/kg ug/kg	1		1	
	1 % IU	/ UU / K O	1		l	
lenes, Total	< 30	ug/kg	1		ļ	

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 11

Canton Analytical Laboratory, Inc.

ga/gc KN

CLIENT: CLEAN TECH

REPORT NUMBER: 35848

SAMPLE ID: 4101686

DESCRIPTION: PROJECT-DAYTON-SOIL STUDY SB-8-24

ATE	SAMPLED:	10/19/94	DATE	RECEIVED:	10/20/94
IME	SAMPLED:	10:20	TIME	RECEIVED:	16:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	< 10	ug/kg	EPA 8260	10/29/94	vs
			;		

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 12

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_

SAMPLE ID: 4102044 REPORT NUMBER: 35905 DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB9-19

DATE SAMPLED.....: 10/21/94

DATE RECEIVED ....: 10/25/94

TIME SAMPLED.....: 13:55 TIME RECEIVED....: 09:30

	· ANALYTICAL	UNITS OF	ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE	METHOD	DATE	TECHI
AL METALS luminum, Total	1600	mg/kg	EPA 6010	10/31/94	SD
		1	EPA 6010	1	SD
arium, Total	8.0	mg/kg	• **	10/28/94	
eryllium, Total	< 0.5	mg/kg	EPA 6010	10/31/94	SD
admium, Total	0.72	mg/kg	EPA 6010	10/28/94	SD
alcium, Total	77000	mg/kg	EPA 6010	10/31/94	SD
romium, Total	5.2	mg/kg	EPA 6010	10/28/94	SD
balt, Total	<u>&lt; 1.0</u>	mg/kg	EPA 6010	10/31/94	SD
ilver, Total	3.0	mg/kg	EPA 6010	10/28/94	SD
opper, Total	8.1	mg/kg	EPA 6010	10/28/94	SD
odium, Total	290	mg/kg	EPA 6010	10/31/94	SD
anadium, Total	< 1.0	mg/kg	EPA 6010	10/31/94	SD
inc, Total	19	mg/kg	EPA 6010	10/28/94	SD
ntimony Total, GFAA	< 1.5	mg/kg	EPA 7041	11/03/94	NS
senic Total, GFAA	5.2	mg/kg	EPA 7060	11/01/94	NS
ead, Total, GFAA	1.6	mg/kg	EPA 7421	11/01/94	NS
elenium Total, GFAA	] < 1.0	mg/kg	EPA 7740	11/01/94	NS
allium Total, GFAA	1.3	mg/kg	EPA 7841	11/02/94	NS
ercury, Cold Vapor	< 0.04	mg/kg	EPA 7471	10/27/94	SD
CL VOLATILE ORGANICS, GC/MS	ł		EPA 8260	10/29/94	VS
etone	< 100	ug/kg	ł		
rolein	< 50	ug/kg	1		
rylonitrile	< 10	ug/kg	ļ	i	
nzene	< 10	ug/kg			
omobenzene	< 10	ug/kg	l	l	
omochloromethane	< 10	ug/kg		1	
omodichloromethane	< 10	ug/kg			
omoform	< 10	ug/kg	ì	Î	
omomethane	< 10	ug/kg	1		
Butanone	< 100	1	}	j	
Butylbenzene	< 100	ug/kg	1	İ	
c-Butylbenzene	< 10	ug/kg	<b>I</b>	1	
rt-Butylbenzene		ug/kg		Ì	
rbon Disulfide	< 10 < 100	ug/kg			
rbon Tetrachloride	< 10	ug/kg		1	
lorobenzene	< 10	ug/kg		ł	
loroform	< 10	ug/kg	l e e e e e e e e e e e e e e e e e e e	i	
loromethane	1	ug/kg	i	i	
coromethane Chlorotoluene	< 10	ug/kg	)	1	
	< 10	ug/kg	J		
Chlorotoluene	< 10	ug/kg		}	
bromochloromethane	< 10	ug/kg		Į	
2-Dibromoethane	< 10	ug/kg		}	
bromomethane 2-Dibromo-3-chloropropane	< 10	ug/kg ug/kg		j	
/ =       DEOMO = { = ch   ononoonoono	< 10		•	1	

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 7

Canton Analytical Laboratory, Inc.

QA/QC Th

CLIENT: CLEAN TECH

REPORT NUMBER: 35905

SAMPLE ID: 4102044

DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB9-19

DATE SAMPLED....: 10/21/94 TIME SAMPLED....: 13:55 DATE RECEIVED....: 10/25/94

TIME RECEIVED....: 09:30

1		ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
. 3	ANALYSES PERFORMED	RESULT	MEASURE	1	METHOD	DATE	TECHN
		<u> </u>	1	<u></u>		20010 1 200 1 1 1 1	
1							
1	1,2-Dichlorobenzene	< 10	ug/kg	Į		1	
j	1,3-Dichlorobenzene	< 10	ug/kg	1	l		
	1,4-Dichlorobenzene	< 10	ug/kg				
	Dichlorodifluoromethane	< 10	ug/kg		<u> </u>		
l	1.1-Dichloroethane	< 10	ug/kg			Į	
ľ	1,2-Dichloroethane	< 10	ug/kg	i		İ	
	1,1-Dichloroethene	< 10	ug/kg			f	
	Cis-1,2 Dichloroethylene	15	ug/kg			i	
	Trans-1.2 Dichloroethylene	< 10	ug/kg	Į.		l	
1	1,2-Dichloroethene (total)	15	ug/kg			j	
J	Dichloromethane	33	ug/kg	}			
i	1,2-Dichloropropane	< 10	ug/kg				
ا_	1,3-Dichloropropane	< 10	ug/kg				
	2,2-Dichloropropane	< 10	ug/kg			,	
€	1,1-Dichloropropene	< 10	ug/kg				
4	Cis-1,3-Dichloropropene	< 10	ug/kg				
	Trans-1,3-Dichloropropene	< 10	ug/kg		į		
7	Ethylbenzene	< 10	ug/kg	1			
	Hexachlorobutadiene	< 10	ug/kg	]			
1	2-Hexanone	< 100	ug/kg	1			
ŀ	Isopropyltoluene	< 50 *	ug/kg				
ļ	4-Isopropyltoluene	< 50 *	ug/kg	]			
1	4-Methyl-2-Pentanone	< 100	ug/kg		İ		
¥I	Naphthalene	< 10	ug/kg				
•	n-Propylbenzene	< 10	ug/kg	(			
-	Styrene	< 10	ug/kg				
اه	1,1,1,2-Tetrachloroethane	< 10	ug/kg	[			
	1,1,2,2-Tetrachloroethane	< 10	ug/kg		Į.		İ
	Tetrachloroethylene	390	ug/kg	ļ ļ	Į		l
- ]	Toluene	< 10	ug/kg		l		
J	1,2,3-Trichlorobenzene	< 10	ug/kg				i
	1,2,4-Trichlorobenzene	< 10	ug/kg				ŀ
	1,1,1-Trichloroethane	< 10	ug/kg	l			
	1,1,2-Trichloroethane	< 10	ug/kg	Į			
	Trichloroethene	2600	ug/kg				
ŕ	Trichlorofluoromethane	< 10	ug/kg				ļ
	1,2,3-Trichtoropropane	< 10	ug/kg				
	1,2,4-Trimethylbenzene	< 10	ug/kg				f
Ţ	1,3,5-Trimethlybenzene	< 10	ug/kg		Ţ		ł
	Vinyl Acetate	< 5.0	ug/kg	1	ľ		1
	Vinyl Chloride	< 10	ug/kg	ſ	İ		1
	MP-Xylene	< 20	ug/kg		1		ſ
- 1	0-Xyl ene	< 10	ug/kg	1	1		ł
-[	Xylenes, Total	< 30	ug/kg	'			
1	}			1	j		1
				ļ			į.

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

The results above apply only to the sample gnalyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of CLEAN TECH PAGE 8

Canton Analytical Laboratory, Inc.

GA/GC KI

REPORT	NI IMBER:	35905	SAMPLE	10: 416	02044	DESCR

DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB9-19

DATE SAMPLED.....: 10/21/94 TIME SAMPLED.....: 13:55 DATE RECEIVED....: 10/25/94 TIME RECEIVED....: 09:30

]	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	< 10	ug/kg			
						i
\$2.00						
1						
1						

The results above apply only to the sample pshyzed. This report may only be reprotuced in full, and may only be submitted to a faird party with the written permission of CLEAN TECH PAGE 9

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

Canton Analytical Laboratory, Inc.

DA/OC KIL

REPORT NUMBER: 35905 SAMPLE ID: 4102045 DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB10-29

DATE SAMPLED....: 10/21/94 TIME SAMPLED....: 09:20 DATE RECEIVED....: 10/25/94
TIME RECEIVED....: 09:30

	ANALYTICAL	UNITS OF	ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE	METHOD	DATE	TECHN
TAL METALS	1400		EPA 6010	10/31/94	65
Aluminum, Total		mg/kg		1	SD
Barium, Total	9.6	mg/kg	EPA 6010	10/28/94	SD
Beryllium, Total	< 0.5	mg/kg	EPA 6010	10/31/94	SD
Cadmium, Total	0.46	mg/kg	EPA 6010	10/28/94	SD
Calcium, Total	70000	mg/kg	EPA 6010	11/04/94	SD
hromium, Total	4.8	mg/kg	EPA 6010	10/28/94	SD
obalt, Total	< 1.0	mg/kg	EPA 6010	10/31/94	SD
ilver, Total	2.4	mg/kg	EPA 6010	10/28/94	SD
opper, Total	8.1	mg/kg	EPA 6010	10/28/94	SD
odium, Total	140	mg/kg	EPA 6010	11/04/94	SD
anadium, Total	< 1.0	mg/kg	EPA 6010	10/31/94	SD
inc, Total	22	mg/kg	EPA 6010	10/28/94	SD
ntimony Total, GFAA	< 1.5	mg/kg	EPA 7041	11/03/94	NS
rsenic Total, GFAA	3.0	mg/kg	EPA 7060	11/01/94	NS
ead, Total, GFAA	2.8	mg/kg	EPA 7421	11/01/94	NS
elenium Total, GFAA	< 1.0	mg/kg	EPA 7740	11/01/94	NS
nallium Total, GFAA	0.94	mg/kg	EPA 7841	11/02/94	NS
ercury, Cold Vapor	< 0.04	mg/kg	EPA 7471	10/24/94	SD
CL VOLATILE ORGANICS, GC/MS	1	i 1	EPA 8260	10/29/94	٧S
cetone	< 100	ug/kg			
crolein	< 50	lug/kg			
crylonitrile	< 10	ug/kg			
enzene	< 10	ug/kg		į.	
romobenzene	< 10	ug/kg			
romochloromethane	< 10	ug/kg			
omodichloromethane	< 10	ug/kg		}	
romoform	< 10	ug/kg			
omomethane		ug/kg		1	
Butanone	< 100	ug/kg			
Butylbenzene	< 10	ug/kg	)	ì	
ec-Butylbenzene	l < 10	ug/kg			
ert-Butylbenzene	< 10	ug/kg			
rbon Disulfide	< 100	ug/kg			
arbon Tetrachloride	< 10	ug/kg			
lorobenzene	< 10	ug/kg		1	
loroform	< 10	ug/kg			
loromethane	< 10	ug/kg			
Chlorotoluene	< 10	ug/kg			
Chlorotoluene	< 10	ug/kg		1	
bromochloromethane	< 10	ug/kg		ļ	
2-Dibromoethane	< 10	ug/kg			
bromomethane	< 10	ug/kg			
2-Dibromo-3-chloropropane	< 10	ug/kg			
	1	1-9173	1	ľ	

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 10

DA/DC FU

CLIENT: CLEAN TECH

REPORT NUMBER: 35905

SAMPLE ID: 4102045

DESCRIPTION: PROJECT ACUSTAR-DAYTON/SB10-29

DATE SAMPLED....: 10/21/94 TIME SAMPLED....: 09:20 DATE RECEIVED ....: 10/25/94

TIME RECEIVED ....: 09:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
		Τ.,				
,2-Dichlorobenzene	< 10	ug/kg				
3-Dichlorobenzene	< 10	ug/kg				
4-Dichlorobenzene	< 10	ug/kg	İ	]		
chlorodifluoromethane	< 10	ug/kg		<b>!</b>		
1-Dichloroethane	< 10	ug/kg	1			
2-Dichloroethane	< 10	ug/kg		İ		
1-Dichloroethene	< 10	ug/kg				
s-1,2 Dichloroethylene	110	ug/kg	1	ļ	ł	
ans-1,2 Dichloroethylene	< 10	ug/kg				
2-Dichloroethene (total)	110	ug/kg	İ		į	
chloromethane	< 20	ug/kg				
2-Dichloropropane	< 10	ug/kg	1		1	
3-Dichloropropane	< 10	ug/kg			İ	
2-Dichloropropane	< 10	ug/kg			ļ	
1-Dichloropropene	< 10	ug/kg			1	
s-1,3-Dichloropropene	< 10	ug/kg	1		1	
ans-1,3-Dichloropropene	< 10	ug/kg				
hylbenzene	< 10	ug/kg			Ì	
xachlorobutadiene	< 10	ug/kg				
Hexanone	< 100	ug/kg			ĺ	
opropyltoluene	< 50 *	ug/kg			}	
Isopropyltoluene	< 50 *	ug/kg				
Methyl-2-Pentanone	< 100	ug/kg			i	
phthalene	< 10	ug/kg				
Propylbenzene	< 10	ug/kg				
yrene	< 10	ug/kg	1			
1,1,2-Tetrachloroethane	< 10	ug/kg	1			
1,2,2-Tetrachloroethane	<b>  &lt; 10</b>	ug/kg	]			
trachloroethylene	< 10	ug/kg	1			
luene	< 10	ug/kg	ļ			
2,3-Trichlorobenzene	<b>√ &lt; 10</b>	ug/kg	{			
2,4-Trichlorobenzene	< 10	ug/kg	1			
1,1-Trichloroethane	< 10	ug/kg	1			
1,2-Trichloroethane	< 10	ug/kg				
ichloroethene	3100	ug/kg				
ichlorofluoromethane	< 10	ug/kg				
2,3-Trichloropropane	< 10	ug/kg				
2,4-Trimethylbenzene	< 10	ug/kg	1			
5,5-Trimethlybenzene	< 10	ug/kg	}			
nyl Acetate	< 5.0	ug/kg	1			
nyl Chloride	< 10	ug/kg	}			
Xylene	< 20	ug/kg	1			
(ylene .enes, Total	< 10	ug/kg		ĺ		
	l < 30	ug/kg	1			

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 11

Canton Analytical Laboratory, Inc.

QA/QC XV

LABORATORY	RESULTS	_
CLIENT: CL	EAN TECH	
DESCRIPTION: PROJECT	ACUSTAR-DAYTON/SB10-29	

DATE SAMPLED.....: 10/21/94 TIME SAMPLED.....: 09:20

SAMPLE ID: 4102045

REPORT NUMBER: 35905

DATE RECEIVED....: 10/25/94

TIME RECEIVED....: 09:30

I		····		 		
)	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	< 10	ug/kg			<u> </u>
]						
]						
1						
1						
1						

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 12

Canton Analytical Laboratory, Inc.

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47801 Anchor Court, Plymouth, Michigan 48170 Canton Analytical Laboratory, Inc. (313) 459-8484 FAX (313) 459-8262

PHOJECT

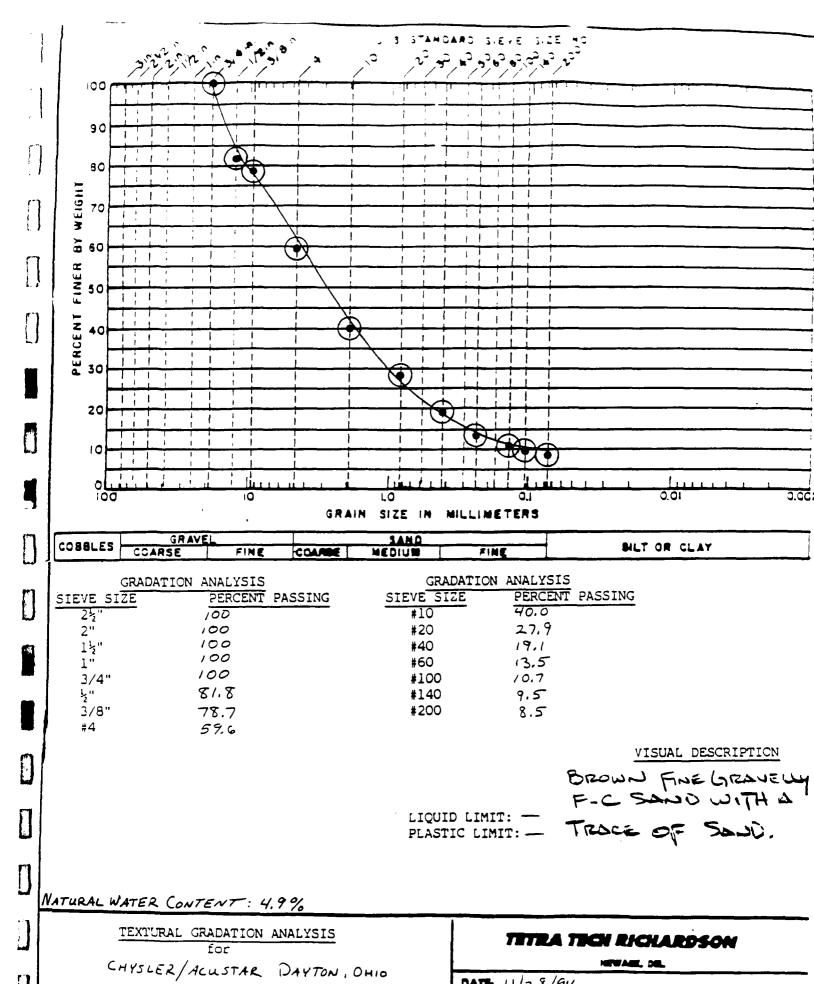
CHAIN OF CUSTODY SHEET

all.

REMARKS HECENED BY SAMPLE CONDITION UPON LAB RECEIPT DATE TIME COOLER TEMP UPON RECEIPT BY LAB RELINOUISHED BY REMARKS ANALYSIS DAYS (MAY RESULT IN ADDITIONAL CHARGE) NUMBER OF CONTAINERS RECEIVED BY ACOSTA( -COMP 0760 0350 12. S. D. 20 03/10 20 1330 9/50 PEGGINED TURNAROUND TIME (, i RUSH ADDRESS LEGA TECK SAMPLE LOCATION STANDARD SAMPLEH K AJAMAS ABAMUN

### ATTACHMENT H

Laboratory Results: Geotechnical
Soil Samples from Soil Borings
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



SAMPLE No. 53-5 (14'-16')

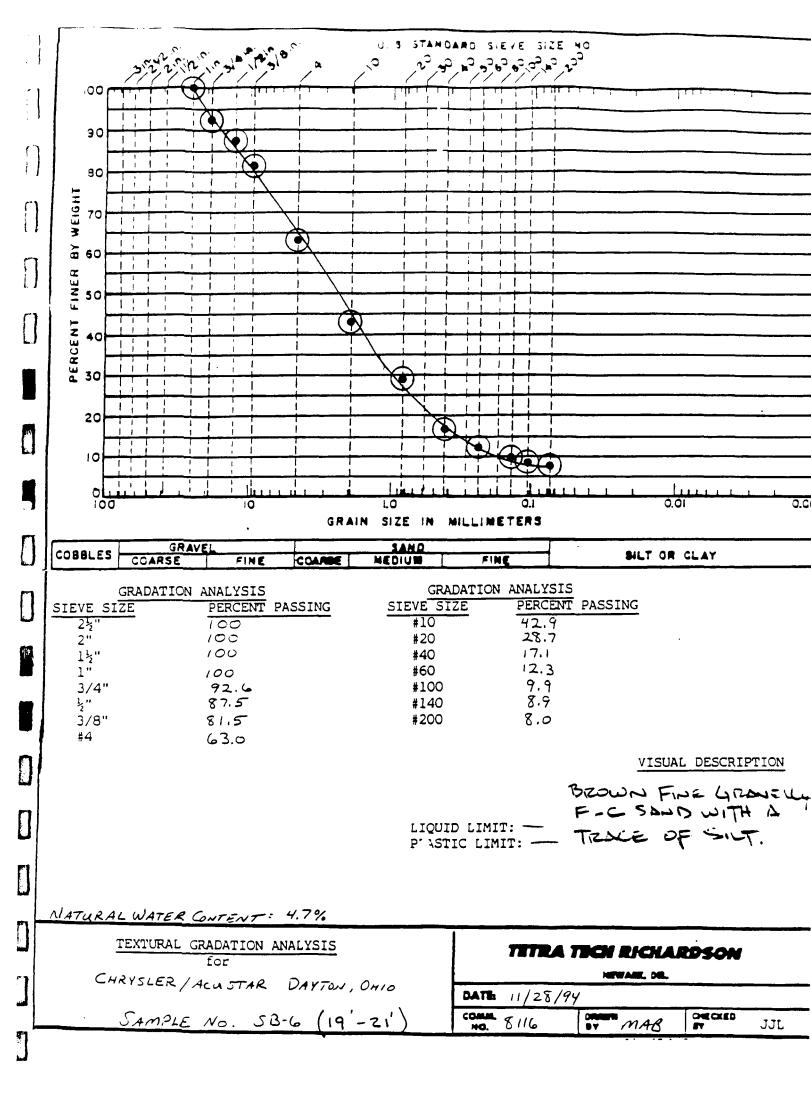
DATE 11/28/94

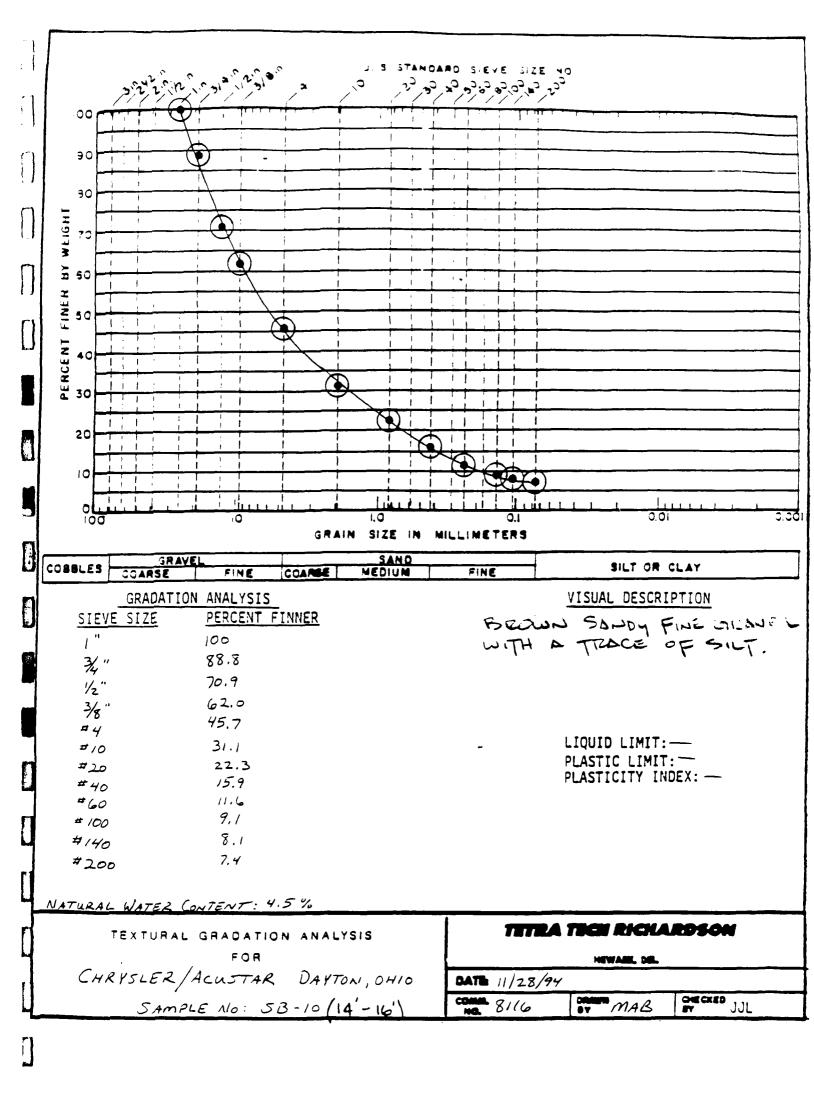
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TETRA TECH ENGINEERS ARCHITECTS	
SCIENTISTS	

#### **COMPUTATION SHEET**

SHEET NO	
otho	

TITLE OF PROJECT OR STUDY CHRYSLER ACUSTAR - DAY TON

SUBJECT LABORATORY MOISTURE CONTENT TEST

COMPUTER JJU DATE 12-5-94 CHECKED BY

DATE CHECKED\_\_\_\_\_19\_\_\_

# Sample No.

# MOISTURE CONTENT (ASTMD -2216)

#### **ATTACHMENT J**

Laboratory Results: Chemical
Soil Samples from Monitoring Well Borings
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404

CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE ID: 4112345 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA1-24

DATE SAMPLED.....: 11/14/94 TIME SAMPLED.....: 10:15 DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
CL VOLATILE ORGANICS. GC/MS	ì		EPA 8260	11/22/94	VS

-	[	4	1	ļ.	Į.	ł.	
	TCL VOLATILE ORGANICS, GC/MS		1		EPA 8260	11/22/94	/s
	Acetone	< 100	ug/kg				
	Acrolein	< 50	ug/kg	)	)		
	Acrylonitrile	< 10	ug/kg				
	Benzene	< 10	ug/kg	<b>\</b>	l .	ļ	
ĸ	Bromobenzene	< 10	ug/kg		ļ		
	Bromochloromethane	< 10	ug/kg		[	į	i
	Bromodichloromethane	< 10	ug/kg	i			
_	Bromoform	< 10	ug/kg				
ġ	Bromomethane	< 10	ug/kg		}	}	
	2-Butanone	< 100	ug/kg				
-	n-Butylbenzene	< 10	ug/kg				
	Sec-Butylbenzene	< 10	ug/kg				
2	tert-Butylbenzene	< 10	lug/kg				
	Carbon Disulfide	100	ug/kg				
7	Carbon Tetrachloride	< 10	ug/kg				
ŀ	Chlorobenzene	< 10	ug/kg				
ı	Chloroform	< 10	ug/kg				i
7	Chloromethane	< 10	ug/kg				
H	2-Chlorotoluene	< 10	ug/kg				ì
d	4-Chlorotoluene	< 10	ug/kg ug/kg				
	Dibromochloromethane	< 10	ug/kg				ĺ
, a	1.2-Dibromoethane	< 10	ug/kg				- 1
1	Dibromomethane	< 10	ug/kg				ļ
Ш	1,2-Dibromo-3-chloropropane	< 10	ug/kg ug/kg				i
_	1,2-Dichlorobenzene	< 10	ug/kg			l	1
	1,3-Dichlorobenzene	10	ug/kg				1
	1,4-Dichlorobenzene	< 10	ug/kg				I
	Dichlorodifluoromethane	< 10	ug/kg	,			
	1,1-Dichloroethane	₹ 10	ug/kg				ŀ
	1,2-Dichloroethane	< 10	ug/kg	į			ĺ
	1.1-Dichloroethene	< 10	ug/kg				1
	Cis-1,2 Dichloroethylene	< 10	ug/kg				
	Trans-1,2 Dichloroethylene	< 10	ug/kg				j
	1,2-Dichloroethene (total)	< 20	ug/kg				Ì
_	Dichloromethane	26 **	ug/kg	Ì			}
	1,2-Dichloropropane	< 10	ug/kg				i
	1,3-Dichloropropane	< 10	ug/kg				- {
	2,2-Dichloropropane	< 10	ug/kg	j			
- 1	1,1-Dichloropropene	< 10	ug/kg				ı
7	Cis-1,3-Dichloropropene	< 10	ug/kg				- 1
I	Trans-1,3-Dichloropropene	< 10	ug/kg				i
	Ethylbenzene	< 10	ug/kg		ļ		
	Hexachlorobutadiene	< 10	ug/kg				}
1	į	ĺ					- 1
<b>E</b> (							

<sup>\*\* 12</sup> ug/kg of Dichloromethane in Method Blank.

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PAGE 4

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE ID: 4112345 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA1-24

DATE SAMPLED.....: 11/14/94 TIME SAMPLED.....: 10:15

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

}	ANALYSES PERFORMED	ANALYTICAL	UNITS OF	}	ANALYTICAL METHOD	RUN	TECHN
	ANALISES PERFORMED	RESULT	MEASURE	<u> </u>	MCTHOU	DATE	ICUM
٠,		T				1	
1	2-Hexanone	< 100	ug/kg				
1	Isopropyltoluene	< 50 *	ug/kg		Ì	1	
-	4-Isopropyltoluene	< 50 *	ug/kg				
	4-Methyl-2-Pentanone	< 100	ug/kg		1	ì	
	Naphthalene	< 10	ug/kg			ļ	
	n-Propylbenzene	< 10	ug/kg		}	]	
	Styrene	< 10	ug/kg	ļ		ļ	
	1,1,1,2-Tetrachloroethane	< 10	ug/kg	]		ļ	
	1,1,2,2-Tetrachloroethane	< 10	ug/kg		<b>[</b>	(	
4	Tetrachloroethylene	5300	ug/kg			ļ	
1	Toluene	< 10	ug/kg			}	
_	1,2,3-Trichlorobenzene	< 10	ug/kg	1	1		
	1,2,4-Trichlorobenzene	< 10	ug/kg	1		1	
å,	1,1,1-Trichloroethane -	25	ug/kg				
	1,1,2-Trichloroethane	< 10	ug/kg	1		}	
7	Trichloroethene	91	ug/kg			ł	
	Trichlorofluoromethane	< 10	ug/kg	i	Ì	]	ì
-	1,2,3-Trichloropropane	< 10	ug/kg	İ			
П	1,2,4-Trimethylbenzene	< 10	ug/kg	İ		1	
J	1,3,5-Trimethlybenzene	< 10	ug/kg	İ	ł		İ
٦	Vinyl Acetate	< 100	ug/kg	i	1	1	
-	Vinyl Chloride	< 10	ug/kg		ł	1	
٦.	MP-Xylene	< 20	ug/kg	ļ		1	ŀ
11	0-Xylene	< 10	ug/kg	}	1		i
4	Xylenes, Total	< 30	ug/kg			i	ŀ
ĺ	Chloroethane	< 10	ug/kg	1		Ì	1
	TAL METALS					1	i
	Aluminum, Total -	2300000	ug/kg	]	EPA 6010	11/20/94	SD
Н	Barium, Total	12000	ug/kg	ļ	EPA 6010	11/21/94	SD (
- 1	Beryllium, Total	99	ug/kg		EPA 6010	12/01/94	SD
_	Cadmium, Total	300	ug/kg		EPA 6010	11/21/94	SD
	Calcium, Total	93000000	ug/kg		EPA 6010	12/01/94	SD
	Chromium, Total -	7400	ug/kg		EPA 6010	12/01/94	SD
	Cobalt, Total	5700	ug/kg		EPA 6010	12/01/94	SD
-1	Silver, Total	3700	ug/kg		EPA 6010	12/01/94	SD
ור	Copper, Total	9000	ug/kg		EPA 6010	11/21/94	SD
	Sodium, Total	140000	ug/kg		EPA 6010	12/01/94	SD
J	Vanadium, Total	9700	ug/kg		EPA 6010	12/01/94	SD
	Zinc, Total —	17000	ug/kg		EPA 6010	11/21/94	SD
	Antimony Total, GFAA Arsenic Total, GFAA	< 1500 6500	ug/kg		EPA 7041	11/23/94	NS
	Lead, Total, GFAA	4000	ug/kg		EPA 7060 EPA 7421	11/21/94	NS NS
	Selenium Total, GFAA	< 1000	ug/kg		EPA 7421	11/21/94 11/22/94	NS I
	Thallium Total, GFAA	2000	ug/kg		EPA 7740	11/22/94	NS I
- [	THECETORIE FOLGE, GEAR	2000	ug/kg		CPA 7041	11/22/74	49
7							J
Ш							

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 5



### LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE ID: 4112345 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA1-24

DATE SAMPLED....: 11/14/94
TIME SAMPLED....: 10:15

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Mercury, Cold Vapor	< 40	ug/kg	EPA 7471	11/22/94	SD
					<u> </u> 	
3						
7						
]						
	-					
		}				
1				;		
1				ļ		
1						

PAGE 6

Canton Analytical Laboratory, Inc.

QA/QC\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36123 SAMPLE ID: 4110328 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWA2-19

DATE SAMPLED....: 10/28/94 TIME SAMPLED....: 08:15 DATE RECEIVED....: 11/03/94

TIME RECEIVED....: 09:45

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHI
M5744.6					
TAL METALS	2700000	1	504 4040	44 (45 (0)	
lluminum, Total	2300000	ug/kg	EPA 6010	11/15/94	SD
arium, Total	13000	ug/kg	EPA 6010	11/08/94	SD
eryllium, Total	210	ug/kg	EPA 6010	11/15/94	SD
admium, Total	710	ug/kg	EPA 6010	11/15/94	SD
alcium, Total	22000000	ug/kg	EPA 6010	11/15/94	SD
hromium, Total	9000	ug/kg	EPA 6010	11/08/94	SD
obalt, Total	4900	ug/kg	EPA 6010	11/15/94	SD
ilver, Total	2700	ug/kg	EPA 6010	11/15/94	SD
opper, Total	9600	ug/kg	EPA 6010	11/08/94	SD
odium, Total	150000	ug/kg	EPA 6010	11/15/94	SD
anadium, Total	9600	ug/kg	EPA 6010	11/15/94	SD
inc, Total	19000	ug/kg	EPA 6010	11/08/94	SD
ntimony Total, GFAA	< 1500	ug/kg	EPA 7041	11/10/94	NS
rsenic Total, GFAA	4400	ug/kg	EPA 7060	11/09/94	NS
ead, Total, GFAA	3200	ug/kg	EPA 7421	11/09/94	NS
elenium Total, GFAA	< 1000	ug/kg	EPA 7740	11/10/94	NS
hallium Total, GFAA	1900	ug/kg	EPA 7841	11/09/94	NS
ercury, Cold Vapor	< 40	ug/kg	EPA 7471	11/08/94	SD
CL VOLATILE ORGANICS, GC/MS	1		EPA 8260	11/13/94	VS
cetone	i < 100	ug/kg	1	,,	
crolein	< 50	ug/kg			
crylonitrile	< 10	ug/kg	i	ł	
enzene	< 10	ug/kg			
romobenzene	< 10	ug/kg	1		
romochloromethane	< 10	ug/kg	]		
romodichloromethane	< 10	ug/kg	i		
comoform	₹ 10	ug/kg			
omomethane	< 10	1 - 1			
	1	ug/kg	1		
Butanone	< 100	ug/kg			
Butylbenzene	< 10	ug/kg			
c-Butylbenzene	< 10	ug/kg			
rt-Butylbenzene	< 10	ug/kg	1		
orbon Disulfide	< 100	ug/kg	ì		
rbon Tetrachloride	< 10	ug/kg	1		
lorobenzene	< 10	ug/kg			
loroform	< 10	ug/kg			
loromethane	< 10	ug/kg	[		
Chlorotoluene	< 10	ug/kg			
Chlorotoluene	< 10	ug/kg			
bromochloromethane	< 10	ug/kg			
2-Dibromoethane	< 10	ug/kg			
bromomethane	< 10	ug/kg (			
2-Dibromo-3-chloropropane	< 10	ug/kg			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 4

CLIENT: CLEAN TECH

SAMPLE ID: 4110328 REPORT NUMBER: 36123 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWA2-19

DATE SAMPLED....: 10/28/94 TIME SAMPLED....: 08:15

DATE RECEIVED....: 11/03/94 TIME RECEIVED....: 09:45

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
,2-Dichlorobenzene	29	ug/kg			1	
,3-Dichlorobenzene	<b>  &lt; 10</b>	ug/kg		1	1	
,4-Dichlorobenzene	< 10	ug/kg		}		
ichlorodifluoromethane	< 10	ug/kg			İ	
,1-Dichloroethane	< 10	ug/kg			1	
,2-Dichloroethane	< 10	ug/kg		}	1	
,1-Dichloroethene	< 10	ug/kg		1	!	
is-1,2 Dichloroethylene	< 10	ug/kg		i	1	
rans-1,2 Dichloroethylene	< 10	ug/kg			1	
,2-Dichloroethene (total)	< 20	ug/kg			ł	
ichloromethane	< 10	ug/kg		ĺ	1	
,2-Dichloropropane	< 10	ug/kg		1		
,3-Dichloropropane	< 10	ug/kg				
2-Dichloropropane	< 10	ug/kg			1	
,1-Dichloropropene	< 10	ug/kg			]	
is-1,3-Dichloropropene	< 10	ug/kg				
ans-1,3-Dichloropropene	< 10	ug/kg			1	
thylbenzene	< 10	ug/kg			1	
exachlorobutadiene	< 10	ug/kg			1	
·Hexanone	< 100	ug/kg			1	
sopropyltoluene	< 50 *	ug/kg			J	
Isopropyltoluene	< 50 *	ug/kg			ł	
Methyl-2-Pentanone	< 100	ug/kg		•	}	
phthalene	< 10	ug/kg	İ		1	
Propylbenzene	< 10	ug/kg				
yrene	< 10	ug/kg			l	
1,1,2-Tetrachloroethane	< 10	ug/kg				
1,2,2-Tetrachloroethane	< 10	ug/kg				
trachloroethylene	1800	ug/kg				
luene	< 10	ug/kg				
2,3-Trichlorobenzene	< 10	ug/kg		l		
2,4-Trichlorobenzene	< 10	ug/kg				
1,1-Trichloroethane	640	ug/kg				
1,2-Trichloroethane	< 10	ug/kg	[			
ichloroethene	200	ug/kg	Ì			
ichlorofluoromethane	< 10	ug/kg				
2,3-Trichloropropane	< 10	ug/kg				
2,4-Trimethylbenzene	< 10	ug/kg				
3,5-Trimethlybenzene	< 10	ug/kg	)			
nyl Acetate	< 10	ug/kg				
nyl Chloride	< 10	ug/kg				
-Xylene	< 20	ug/kg				
Xyĺ ene ¯	< 10	ug/kg				
lenes, Total	< 30	ug/kg	ļ			
• • • • • • • • • • • • • • • • • • • •	1 ***	1-5,				

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 5

QA/QC\_\_\_

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י [		LABO	RATORY R CLIENT: CLEAN T							
REPORT NUMBER: 36123 SAMPLE ID: 4110328 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWA2-19										
	DATE SAMPLED: 10/28/94 TIME SAMPLED: 08:15	DA T1	DATE RECEIVED: 11/03/94 TIME RECEIVED: 09:45							
	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN			
j	Chloroethane	< 10	ug/kg							
, 					i i					
T										
						į				
ı T										
7										
ij.										
r										
1					į					
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Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_KV

CLIENT: CLEAN TECH

SAMPLE ID: 4111959 DESCRIPTION: CHRYSLER ACUSTAR MWA3-24 REPORT NUMBER: 36415

DATE SAMPLED....: 11/11/94
TIME SAMPLED....: 10:05

DATE RECEIVED ....: 11/15/94

TIME RECEIVED....: 11:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
TCL VOLATILE ORGANICS, GC/MS :				EPA 8260	11/16/94	VS
Acetone	< 100	ug/kg			11,10,74	
Acrolein	< 50	ug/kg			<b>!</b>	
Acrylonitrile	< 10	ug/kg				
Benzene	< 10	ug/kg				
Bromobenzene	< 10	ug/kg			İ	
Bromochloromethane	< 10	ug/kg	•		j	
Bromodichloromethane	< 10	ug/kg			ļ	
Bromoform	< 10	ug/kg			j	
Bromomethane	< 10	ug/kg				
2-Butanone	< 100	ug/kg				
n-Butylbenzene	< 10	ug/kg	į			
Sec-Butylbenzene	< 10	ug/kg				
tert-Butylbenzene	< 10	ug/kg				
Carbon Disulfide	< 100	ug/kg	Į.			
Carbon Tetrachloride	< 10	ug/kg	Ĭ			
Chlorobenzene	< 10	ug/kg				
Chloroform	< 10	ug/kg	Į i			
Chloromethane	< 10	ug/kg				
2-Chlorotoluene	< 10	ug/kg				
4-Chlorotoluene	< 10	ug/kg				
Dibromochloromethane	< 10	ug/kg				
1,2-Dibromoethane	< 10	ug/kg	]			
Dibromomethane	< 10	ug/kg				
1,2-Dibromo-3-chloropropane	< 10	ug/kg				
1,2-Dichlorobenzene	< 10	ug/kg				
1,3-Dichlorobenzene	< 10	ug/kg				
1,4-Dichlorobenzene	< 10	ug/kg	l i			
Dichlorodifluoromethane	< 10	ug/kg	i l			
1,1-Dichloroethane	< 10	ug/kg				
1,2-Dichloroethane	< 10	ug/kg				
1,1-Dichloroethene	< 10	ug/kg				
Cis-1,2 Dichloroethylene	< 10	ug/kg				
Trans-1,2 Dichloroethylene	< 10	ug/kg				
1,2-Dichloroethene (total)	< 20	ug/kg				
Dichloromethane	< 10	ug/kg				
1,2-Dichloropropane	< 10	ug/kg				
1,3-Dichtoropropane	< 10	ug/kg	}			
2,2-Dichloropropane	< 10	ug/kg				
1,1-Dichloropropene	< 10	ug/kg				
Cis-1,3-Dichloropropene	< 10	ug/kg				
Trans-1,3-Dichloropropene	< 10	ug/kg				
Ethylbenzene	< 10	ug/kg				
Hexachlorobutadiene	< 10	ug/kg				
	1	1	i i			

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 1

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Canton Analytical Laboratory, Inc.

### LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36415 SAMPLE ID: 4111959 DESCRIPTION: CHRYSLER ACUSTAR MWA3-24

DATE SAMPLED....: 11/11/94 TIME SAMPLED....: 10:05 DATE RECEIVED....: 11/15/94
TIME RECEIVED....: 11:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECH
- Hexanone	< 100	ug/kg				
sopropyltoluene	< 50 *	ug/kg		ł	1	
-Isopropyltoluene	< 50 *	ug/kg			]	
- raopropy ( to ( derie - Methyl - 2 - Pentanone	< 100	ug/kg	1	J	Ì	
aphthalene	< 10	ug/kg			Ì	
-Propylbenzene	< 10	ug/kg	}			
tyrene	< 10				i	
,1,1,2-Tetrachloroethane	< 10	ug/kg				
	< 10	ug/kg	ļ			
,1,2,2-Tetrachloroethane	,	ug/kg				
etrachloroethylene	260	ug/kg				
oluene	< 10	ug/kg				
2,3-Trichlorobenzene	< 10	ug/kg				
2,4-Trichlorobenzene	< 10	ug/kg				
1,1-Trichloroethane	160	ug/kg			1	
1,2-Trichloroethane	< 10	ug/kg				
ichloroethene	52	ug/kg				
ichlorofluoromethane	< 10	ug/kg			1	
2,3-Trichloropropane	< 10	ug/kg			j	
2,4-Trimethylbenzene	< 10	ug/kg	1	}	1	
3,5-Trimethlybenzene	< 10	ug/kg			ļ	
nyl Acetate	< 10	ug/kg	ļ		!	
nyl Chloride	< 10	ug/kg				
-Xylene	< 20	ug/kg		}	İ	
Xylene	< 10	ug/kg				
lenes, Total	< 30	ug/kg			}	
loroethane	< 10	ug/kg	ļ	İ		
L METALS		29/ 29				
uminum, Total	3300000	ug/kg		EPA 6010	11/29/94	SD
rium, Total	11000	ug/kg		EPA 6010	11/29/94	SD
ryllium, Total	260	ug/kg		EPA 6010	11/29/94	SD
dmium, Total	780			EPA 6010		
lcium, Total	■ · · · · · · · · · · · · · · · · · · ·	ug/kg	İ	1	11/29/94	SD
	90000000	ug/kg		EPA 6010	11/29/94	SD
romium, Total Dalt, Total	7100	ug/kg	1	EPA 6010	11/29/94	SD
	4100	ug/kg		EPA 6010	11/29/94	SD
lver, Total	3300	ug/kg	]	EPA 6010	11/16/94	SD
oper, Total	7300	ug/kg		EPA 6010	11/29/94	SD
dium, Total	120000	ug/kg		EPA 6010	11/29/94	SD
nadium, Total	8700	ug/kg		EPA 6010	11/29/94	SD
nc, Total	20000	ug/kg		EPA 6010	11/29/94	SD
timony Total, GFAA	< 1500	ug/kg		EPA 7041	11/17/94	NS
senic Total, GFAA	5600	ug/kg		EPA 7060	11/17/94	NS
ad, Total, GFAA	3300	ug/kg	1	EPA 7421	11/17/94	NS
lenium Total, GFAA	< 1000	ug/kg	ì	EPA 7740	11/18/94	NS
allium Total, GFAA	900	ug/kg	l	EPA 7841	11/17/94	NS

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 2

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Canton Analytical Laboratory, Inc.

QA/QC SV

CLIENT: CLEAN TECH

REPORT NUMBER: 36415

SAMPLE ID: 4111959

DESCRIPTION: CHRYSLER ACUSTAR MWA3-24

DATE SAMPLED.....: 11/11/94
TIME SAMPLED.....: 10:05

DATE RECEIVED....: 11/15/94 TIME RECEIVED....: 11:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Mercury, Cold Vapor	< 100	ug/kg	EPA 7471	11/17/94	SD
			:		

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 3

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CLIENT: CLEAN TECH

REPORT NUMBER: 36036 SAMPLE ID: 4102741 DESCRIPTION: MWA4-24

DATE SAMPLED.....: 10/24/94 TIME SAMPLED.....: 15:45 DATE RECEIVED....: 10/31/94

TIME RECEIVED ....: 11:30

	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
TO 100 ATT 5 000 ATT 5				504 6040	44 (07 (0)	
TCL VOLATILE ORGANICS, GC/MS	4.00	0	Ì	EPA 8260	11/03/94	VS
Acetone	< 120	ug/kg				
Acrolein	< 50	ug/kg				
Acrylonitrile	< 5.0	ug/kg	1		ļ	
Benzene	< 10	ug/kg			į	
Bromobenzene	< 10	ug/kg				
Bromochloromethane	< 10	ug/kg				
Bromodichloromethane	< 10	ug/kg	1	Į	[	
Bromoform	< 10	ug/kg			ł	
Bromomethane	< 10	ug/kg		1		
2-Butanone	< 100	ug/kg	Ì	1	1	
n-Butylbenzene	52	ug/kg	1	1	]	
Sec-Butylbenzene	50 < 10	ug/kg	İ			
tert-Butylbenzene Carbon Disulfide	1	ug/kg	1	1	]	
Carbon Disulfide   Carbon Tetrachloride	< 100	ug/kg	ì	Ì	1	
Chlorobenzene	< 10 < 10	ug/kg		1	i	
Chloroform	< 10	ug/kg	1	1	1	
Chloromethane	< 10	ug/kg		j	1	
2-Chlorotoluene	< 10	ug/kg	1	ļ	ì	
14-Chlorotoluene	< 10	ug/kg ug/kg				
Dibromochloromethane	< 10	ug/kg ug/kg		ĺ		
1.2-Dibromoethane	< 10	ug/kg ug/kg			ĺ	
Dibromomethane	< 10	ug/kg		1	ļ	
1,2-Dibromo-3-chloropropane	< 10	ug/kg		1	ļ	
1,2-Dichlorobenzene	< 10	ug/kg	1		1	
1,3-Dichlorobenzene	< 10	ug/kg		,		
1,4-Dichlorobenzene	< 10	ug/kg	1		}	
Dichlorodifluoromethane	< 10	ug/kg	1			
1,1-Dichloroethane	< 10	ug/kg				
1,2-Dichloroethane	< 10	ug/kg	Į .			
1,1-Dichloroethene	< 10	ug/kg				
Cis-1,2 Dichloroethylene	140	ug/kg				
Trans-1,2 Dichloroethylene	< 10	ug/kg				
1,2-Dichloroethene (total)	140	ug/kg				
Dichloromethane	< 30	ug/kg	1		1	
1,2-Dichloropropane	< 10	ug/kg				
1,3-Dichloropropane	< 10	ug/kg				
2,2-Dichtoropropane	< 10	ug/kg		ļ		
1,1-Dichloropropene	< 10	ug/kg	]			
Cis-1,3-Dichloropropene	< 10	ug/kg				
Trans-1,3-Dichloropropene	< 10	ug/kg				
Ethylbenzene	< 10	ug/kg		'		
Hexachlorobutadiene	< 10	ug/kg	1			
		1				
L	L		<u> </u>			
		<del></del>				

PAGE 1

CLIENT: CLEAN TECH

REPORT NUMBER: 36036 SAMPLE ID: 4102741 DESCRIPTION: MWA4-24

DATE SAMPLED....: 10/24/94 TIME SAMPLED....: 15:45 DATE RECEIVED....: 10/31/94 TIME RECEIVED....: 11:30

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
_							
1	2.11-11-1-1	100				}	
l	2-Hexanone	< 100 < 50	ug/kg	1			
-	Isopropyltoluene	< 50   < 50	ug/kg	}			
	4-Isopropyltoluene   4-Methyl-2-Pentanone	< 30   < 100	ug/kg	1		ļ	
	4-Methyt-2-rentanone   Naphthalene	< 10	ug/kg			i	
ı	n-Propylbenzene	28	ug/kg	1		]	
	Styrene	20   < 10	ug/kg	1			
	1,1,1,2-Tetrachloroethane	< 10	ug/kg ug/kg	İ			
	1,1,2,2-Tetrachloroethane	< 10	ug/kg				
l.	Tetrachloroethylene	150	ug/kg				
	Toluene	< 10	ug/kg			İ	
	1,2,3-Trichlorobenzene	< 10	ug/kg			i	
	1,2,4-Trichlorobenzene	< 10	ug/kg			i	
3	1.1.1-Trichloroethane	< 10	ug/kg				
	1,1,2-Trichloroethane	< 10	ug/kg	1	•		
•	Trichloroethene	1300	ug/kg			ŀ	
	Trichlorofluoromethane	< 10	ug/kg				
7	1,2,3-Trichloropropane	< 10	ug/kg			}	1
1	1,2,4-Trimethylbenzene	< 10	ug/kg	ľ		]	i
J	1,3,5-Trimethlybenzene	15	ug/kg	1			i
	Vinyl Acetate	< 10	ug/kg				
	Vinyl Chloride	< 10	ug/kg				
П	MP-Xylene	< 20	ug/kg			ĺ	1
Ш	0-Xylene	< 10	ug/kg	1	,	1	
_	Xylenes, Total	< 30	ug/kg	i i			
-	Chloroethane	< 10	ug/kg	1			ł
۹ ا	TAL METALS	1	ł .	<b>\</b>			į
*	Aluminum, Total	1900000	ug/kg	1	EPA 6010	11/01/94	SD
	Barium, Total	15000	ug/kg		EPA 6010	11/01/94	SD
ļ	Beryllium, Total	< 500	ug/kg	1	EPA 6010	11/10/94	SD
_	Cadmium, Total	660	ug/kg	1	EPA 6010	11/01/94	SD
	Calcium, Total	10000000	ug/kg	İ	EPA 6010	11/10/94	SD
	Chromium, Total	6400	ug/kg	{	EPA 6010	11/01/94	SD
	Cobalt, Total	2600	ug/kg		EPA 6010	11/10/94	SD
	Silver, Total	2100	ug/kg		EPA 6010	11/01/94	SD
4	Copper, Total Sodium, Total	11000 140000	ug/kg		EPA 6010	11/01/94	SD
4	Vanadium, Total	5600	ug/kg		EPA 6010	11/10/94	SD
	Zinc, Total	23000	ug/kg	j	EPA 6010 EPA 6010	11/01/94	SD C
	Antimony Total, GFAA	< 1500 < 1500	ug/kg ug/kg	}	EPA 7041	11/01/94	SD NS
	Arsenic Total, GFAA	2600	ug/kg ug/kg		EPA 7060	11/03/94 11/01/94	NS I
	Lead, Total, GFAA	3200	ug/kg		EPA 7421	11/01/94	NS
	Selenium Total, GFAA	< 1000	ug/kg ug/kg		EPA 7740	11/01/94	NS
-	Thallium Total, GFAA	820	ug/kg		EPA 7841	11/02/94	NS NS
			~ <b>₽/ ^8</b>		C-7 10-1	11/06/74	73
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PAGE 2

Canton Analytical Laboratory, Inc.

QA/QC ()(2)

## LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36036 SAMPLE ID: 4102741 DESCRIPTION: MWA4-24

DATE SAMPLED.....: 10/24/94 TIME SAMPLED .....: 15:45

DATE RECEIVED ....: 10/31/94

TIME RECEIVED ....: 11:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	 ANALYTICAL METHOD	RUN DATE	TECHN
Mercury, Cold Vapor	< 40	ug/kg	EPA 7471	11/02/94	SO
		1			
	<u></u>		 		<del></del>

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PAGE 3

Canton Analytical Laboratory, Inc.

QA/QC\_

### LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36514 SAMPLE ID: 4112343 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA5-24

DATE SAMPLED....: 11/15/94 TIME SAMPLED....: 09:00

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
1						
l	TCL VOLATILE ORGANICS, GC/MS		1	EPA 8260	12/02/94	VS
1	Acetone	< 100	lug/kg	-		
	Acrolein	< 50	ug/kg	į		
	Acrylonitrile	< 10	ug/kg			
i	Benzene	< 10	ug/kg			
B	Bromobenzene	< 10	ug/kg			
•	Bromochloromethane	< 10	ug/kg			
	Bromodichloromethane	< 10	ug/kg	}		
	Bromoform	< 10	lug/kg	1		
4	Bromomethane	< 10	lug/kg	Ì		
J	2-Butanone	< 100	ug/kg			
	n-Butylbenzene	< 10	ug/kg	İ	•	
_	Sec-Butylbenzene	< 10	ug/kg			
1	tert-Butylbenzene	< 10	lug/kg	1		
	Carbon Disulfide	< 100	ug/kg			
•	Carbon Tetrachloride	< 10	ug/kg			1
	Chlorobenzene	< 10	ug/kg			
'ד	Chloroform	< 10	ug/kg	1		
Ł	Chloromethane	< 10	ug/kg	]		
J	2-Chlorotoluene	< 10	ug/kg			
	4-Chlorotoluene	< 10	ug/kg			í
_[	Dibromochloromethane	< 10	ug/kg			
П	1,2-Dibromoethane	< 10	ug/kg			
	Dibromomethane	< 10	ug/kg			
-	1,2-Dibromo-3-chloropropane	< 10	ug/kg			
١	1,2-Dichlorobenzene	< 10	ug/kg			
4	1,3-Dichlorobenzene	< 10	ug/kg	İ		
4	1,4-Dichlorobenzene Dichlorodifluoromethane	< 10 < 10	ug/kg	}		- {
	1,1-Dichloroethane	< 10	ug/kg ug/kg	ļ		
١	1,2-Dichloroethane	< 10	ug/kg	}		- 1
	1,1-Dichloroethene	< 10	ug/kg			i
	Cis-1,2 Dichloroethylene	< 10	ug/kg	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \		
	Trans-1,2 Dichloroethylene	< 10	ug/kg			
j	1,2-Dichloroethene (total)	< 20	ug/kg			}
_	Dichloromethane	< 10	ug/kg			j
П	1,2-Dichloropropane	< 10	ug/kg			- 1
	1,3-Dichloropropane	i i	ug/kg			- 1
7	2,2-Dichloropropane	< 10	ug/kg			- 1
	1,1-Dichloropropene	< 10	ug/kg			1
٦!	Cis-1,3-Dichloropropene		ug/kg			İ
	Trans-1,3-Dichloropropene		ug/kg			i
-	Ethylbenzene	1	ug/kg			
-	Hexachlorobutadiene	< 10	ug/kg			ļ
7	}					}
J٨						
- 1						ŀ

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_

CLIENT: CLEAN TECH

SAMPLE ID: 4112343 REPORT NUMBER: 36514 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA5-24

DATE SAMPLED.....: 11/15/94 TIME SAMPLED.....: 09:00

DATE RECEIVED....: 11/18/94

TIME RECEIVED....: 09:30

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}		ANALYTICAL	UNITS OF	1	ANALYTICAL	RUN	
-	ANALYSES PERFORMED	RESULT	MEASURE	<u></u>	METHOD	DATE	TECHN
					T	iwe	
1	1		i				
ŀ	2-Hexanone	< 100	ug/kg				
3	Isopropyltoluene	< 50 *	ug/kg			1	
	4-Isopropyltoluene	< 50 *	ug/kg	l .			
	4-Methyl-2-Pentanone	< 100	ug/kg				
ı	Naphthalene	< 10	ug/kg		į.		
ı	n-Propylbenzene	< 10	ug/kg				
•	Styrene	< 10	ug/kg		ł		
	1,1,1,2-Tetrachloroethane	< 10	ug/kg		]	}	
	1,1,2,2-Tetrachloroethane	< 10	ug/kg				
ł	Tetrachloroethylene	300	ug/kg		ĺ		
L	Toluene	< 10	ug/kg		}		
-	1,2,3-Trichlorobenzene	< 10	ug/kg		}	}	
	1,2,4-Trichlorobenzene	< 10	ug/kg				
1	1,1,1-Trichloroethane	39	ug/kg	}	}		
ł	1,1,2-Trichloroethane	< 10	ug/kg				
Į	Trichloroethene	64	ug/kg ug/kg		1	Į.	
	Trichlorofluoromethane	< 10	1 7 7			j	
_	1,2,3-Trichloropropane	< 10	ug/kg		1	1	
1	1,2,4-Trimethylbenzene	< 10	ug/kg		J	1	
i		< 10	ug/kg	•			
4	1,3,5-Trimethlybenzene	< 100	ug/kg	1	İ	}	
	Vinyl Acetate		ug/kg			}	i
	Vinyl Chloride	< 10	ug/kg				
ł	MP-Xylene	< 20	ug/kg			ļ	]
ı	0-Xylene	< 10	ug/kg	·	ļ	1	1
-	Xylenes, Total	< 30	ug/kg				l
1	Chloroethane	< 10	ug/kg				
4	TAL METALS						
1	Aluminum, Total —	1800000	ug/kg	ĺ	EPA 6010	12/02/94	SD (
B	Barium, Total —	9000	ug/kg		EPA 6010	11/21/94	SD
į	Beryllium, Total —	100	⊔g/kg		EPA 6010	12/01/94	SD [
	Cadmium, Total 🖚	680	ug/kg		EPA 6010	11/21/94	SD
	Calcium, Total -	150000000	ug/kg		EPA 6010	12/01/94	SD
	Chromium, Total —	7900	ug/kg		EPA 6010	11/21/94	SD
	Cobalt, Total	5400	ug/kg		EPA 6010	12/01/94	SD
١	Silver, Total	4200	ug/kg		EPA 6010	11/21/94	SD I
J	Copper, Total	8700	ug/kg		EPA 6010	11/21/94	SD
ı	Sodium, Total -	190000	ug/kg		EPA 6010	12/01/94	SD
J	Vanadium, Total	8700	ug/kg	į	EPA 6010	12/01/94	SD
٦	Zinc, Total	15000	ug/kg		EPA 6010	11/21/94	SD
١	Antimony Total, GFAA	< 1500	ug/kg		EPA 7041	11/23/94	NS
7	Arsenic Total, GFAA -	2500	ug/kg	,	EPA 7060	11/21/94	NS
ı	Lead, Total, GFAA -	3500	ug/kg		EPA 7421	11/21/94	NS
4	Selenium Total, GFAA	< 1000	ug/kg		EPA 7740	11/22/94	NS
ı	Thallium Total, GFAA	2600	ug/kg	j	EPA 7841	11/22/94	NS
Į,	]		-ur ng	ļ		,, , .	··
I				1			
1				1			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 5

he results above apply only to the sample unsiyzed. This report may only be repro-tuded in full, and may only be submitted to a tird party with the written permission of CLEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36514

SAMPLE ID: 4112343

DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA5-24

DATE SAMPLED.....: 11/15/94 TIME SAMPLED.....: 09:00

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

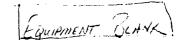
		<del>~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~</del>	 	<del></del>	
ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Mercury, Cold Vapor	< 40	ug/kg	EPA 7471	11/22/94	SD
					i
	<del>- L</del>	<del>1</del>		<u></u>	

PAGE 6

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QA/QC\_\_\_

F The results above apply only to the sample unityzed. This report may only be repro-tituded in full, and may only be submitted to a third party with the written permission of CLEAN TECH



CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE ID: 4112346 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA5-24

DATE SAMPLED.....: 11/15/94 TIME SAMPLED.....: 09:05

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

]	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
1	TCL VOLATILE ORGANICS, GC/MS			EPA 8260	11/20/94	vs
J	Acetone	< 50	ug/l			
ı	Acrolein	< 50	ug/l			
_	Acrylonitrile	< 1.0	ug/l			
H	Benzene	< 1.0	lug/l			
П	Bromobenzene	< 1.0	ug/l			
•	Bromochloromethane	< 1.0	ug/l			
١	Bromodichloromethane	< 1.0	ug/(			
	Bromoform	< 1.0	ug/l			
1	Bromomethane	< 1.0	ug/l			
3	2-Butanone	< 50	ug/l			
- 1	n-Butylbenzene	< 1.0	ug/l			
.	Sec-Butylbenzene	< 1.0	ug/L			
H	tert-Butylbenzene	< 1.0	ug/l			
	Carbon Disulfide	< 50	ug/l		•	
,	Carbon Tetrachloride	< 1.0	lug/l			
	Chlorobenzene	< 1.0	lug/l			
T .	Chloroform	< 1.0	ug/l			
	Chloromethane	< 1.0	lug/l			
	2-Chlorotoluene	< 1.0	ug/l			
	4-Chlorotoluene	< 1.0	ug/L			
	Dibromochloromethane	< 1.0	ug/l			į.
	1,2-Dibromoethane	< 1.0	ug/l			ļ
	Dibromomethane	< 1.0 < 1.0	ug/L			J
	1,2-Dibromo-3-chloropropane 1,2-Dichlorobenzene	< 1.0	ug/l ug/l			
	1.3-Dichlorobenzene	< 1.0	ug/t ug/t			f
	1.4-Dichlorobenzene	< 1.0	ug/l			
	Dichlorodifluoromethane	< 1.0	ug/l			1
	1.1-Dichloroethane	< 1.0	ug/l			ì
	1,2-Dichloroethane	< 1.0	ug/l			i i
- 1	1,1-Dichloroethene	< 1.0	ug/l			
	Cis-1,2 Dichloroethylene	< 1.0	ug/l			}
	Trans-1,2 Dichloroethylene	< 1.0	ug/l			1
	1,2-Dichloroethene (total)	< 2.0	ug/l	į		ļ
	Dichloromethane	< 1.0	ug/l			
ı	1,2-Dichloropropane	< 1.0	ug/l			
	1,3-Dichtoropropane	< 1.0	ug/l			
1	2,2-Dichloropropane	< 1.0	ug/l	Į		Į.
	1,1-Dichloropropene	< 1.0	ug/l			
	Cis-1,3-Dichloropropene	< 1.0	ug/l	1		ŀ
	[rans-1,3-Dichloropropene	< 1.0	ug/l	Ţ		1
	thylbenzene	< 1.0	ug/l			- 1
1	lexachlorobutadiene	< 1.0	ug/t	}		1
ı	1			Ì		)
	<del></del>			1		
						Ì

PAGE 7

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE ID: 4112346 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA5-24

DATE SAMPLED.....: 11/15/94 TIME SAMPLED..... 09:05

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Trichlorofluoromethane	< 50 < 20 * < 20 * < 20 * < 50 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0	ug/l ug/l ug/l ug/l ug/l ug/l ug/l ug/l			
祷						
5						}

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 8

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Canton Analytical Laboratory, Inc.

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## LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36036 SAMPLE ID: 4102742 DESCRIPTION: MWA6-24

DATE SAMPLED....: 10/25/94 TIME SAMPLED .....: 15:45

DATE RECEIVED....: 10/31/94 TIME RECEIVED....: 11:30

		ANALYTICAL	UNITS OF	ANALYTICAL	RUN	
,	ANALYSES PERFORMED	RESULT	MEASURE	METHOD	DATE	TECHN
. !					l	
١	TCL VOLATILE ORGANICS, GC/MS		<b>,</b>	EPA 8260	11/03/94	vs
	Acetone	< 110	ug/kg	LFA OLOO	11/03/74	•••
	Acrolein	< 50	ug/kg	ł	[	
	Acrylonitrile	< 5.0	ug/kg			
ı	Benzene	< 10	lug/kg			
	Bromobenzene	< 10	ug/kg		}	
'	Bromochloromethane	< 10	ug/kg	ł	}	
	Bromodichloromethane	< 10	ug/kg		{	
	Bromoform	< 10	ug/kg	1		
1	Bromomethane	< 10	ug/kg		}	
4	2-Butanone	< 100	ug/kg	1	ì	
	n-Butylbenzene	< 10	ug/kg		1	
اء	Sec-Butylbenzene	< 10	ug/kg	1	Į.	
	tert-Butylbenzene	< 10	ug/kg	Į.	ļ	
١,	Carbon Disulfide	< 100	ug/kg	Į.		
	Carbon Tetrachloride	< 10	ug/kg	1		
_	Chlorobenzene	< 10	ug/kg	1		
11	Chloroform	< 10	lug/kg	1		- 1
Ш	Chloromethane 2-Chlorotoluene	< 10 < 10	lug/kg	4		- 1
"	4-Chlorotoluene	< 10	lug/kg			
- 1	Dibromochloromethane	< 10	ug/kg ug/kg	1		
	1,2-Dibromoethane	< 10	ug/kg	}	ı	
	Dibromomethane	< 10	ug/kg	}		
	1,2-Dibromo-3-chloropropane	< 10	ug/kg		1	
1	1,2-Dichlorobenzene	< 10	ug/kg	{		l
3	1,3-Dichlorobenzene	< 10	ug/kg			l l
Ш	1,4-Dichlorobenzene	< 10	ug/kg	]		]
-1	Dichlorodifluoromethane	< 10	ug/kg	Ì		ì
	1,1-Dichloroethane	< 10	ug/kg	1		i
	1,2-Dichloroethane	< 10	ug/kg			
	1,1-Dichloroethene	< 10	ug/kg			Į.
	Cis-1,2 Dichloroethylene	< 10	lug/kg	1		1
	Trans-1,2 Dichloroethylene 1,2-Dichloroethene (total)	< 10 < 20	ug/kg			1
	Dichloromethane	< 30	ug/kg ug/kg			j
	1,2-Dichloropropane	< 10	ug/kg	!		1
	1,3-Dichloropropane	< 10	ug/kg			ŀ
	2.2-Dichloropropane	< 10	ug/kg	[		
	1,1-Dichloropropene	< 10	ug/kg			j
	Cis-1,3-Dichloropropene	< 10	ug/kg			j
I	Trans-1,3-Dichtoropropene	< 10	ug/kg	j		j
	Ethylbenzene	< 10	ug/kg	[		i
	Hexach Lorobutadiene	< 10	ug/kg			
1		Į		{		ł
L		l		<u>                                      </u>		1
₹						

PAGE 4

Canton Analytical Laboratory, Inc.

じ QA/QC\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36036 SAMPLE ID: 4102742 DESCRIPTION: MWA6-24

DATE SAMPLED....: 10/25/94 TIME SAMPLED....: 15:45

DATE RECEIVED....: 10/31/94 TIME RECEIVED....: 11:30

J	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
-3							
1	2-Hexanone	< 100	ug/kg			Į	
. 3	Isopropyltoluene	< 50	ug/kg				
	4-Isopropyltoluene	< 50	ug/kg			}	
_	4-Methyl-2-Pentanone	< 100	ug/kg			}	
H	Naphthalene	< 10	ug/kg			ļ	
	n-Propylbenzene	< 10	ug/kg				
	Styrene	< 10	ug/kg			ļ	
	1,1,1,2-Tetrachloroethane	< 10	ug/kg			İ	
	1,1,2,2-Tetrachloroethane	< 10	ug/kg			[	
	Tetrachloroethylene	< 10	ug/kg			[	
	Toluene	< 10	ug/kg			ĺ	
	1,2,3-Trichlorobenzene	< 10	ug/kg				
	1,2,4-Trichlorobenzene	< 10	ug/kg	1	Į		
	1,1,1-Trichloroethane	< 10	ug/kg (	į			
7	1,1,2-Trichloroethane	< 10	ug/kg				
1	Trichloroethene	90	ug/kg	j		)	
_	Trichlorofluoromethane	< 10	ug/kg		•		
П	1,2,3-Trichloropropane	< 10	ug/kg [				
	1,2,4-Trimethylbenzene	< 10	ug/kg				
_	1,3,5-Trimethlybenzene	< 10	ug/kg				
- 1	Vinyl Acetate	< 10	ug/kg	1	1		
ו ד	Vinyl Chloride	< 10	ug/kg		į		
Ш	MP-Xylene	< 20	ug/kg				
J	0-Xylene	< 10	ug/kg				
- 1	Xylenes, Total Chloroethane	< 30 < 10	ug/kg	- 1	,	ı	
-a	TAL METALS	* 10	ug/kg				
1	Aluminum, Total	2300000	ua /ka		EPA 6010	11/01/94	SD
	Barium, Total	14000	ug/kg ug/kg	1	EPA 6010	11/01/94	SD
- {	Beryllium, Total	< 500	ug/kg	1	EPA 6010	11/10/94	SD
_ [	Cadmium, Total	710	ug/kg	1	EPA 6010	11/01/94	SD
Fj	Calcium, Total	100000000	ug/kg	j	EPA 6010	11/10/94	SD
	Chromium, Total	5000	ug/kg	1	EPA 6010	11/01/94	SD
-	Cobalt, Total	2800	ug/kg	}	EPA 6010	11/10/94	SD
- {	Silver, Total	2000	ug/kg	,	EPA 6010	11/01/94	SD
	Copper, Total	11000	ug/kg		EPA 6010	11/01/94	SD
	Sodium, Total		ug/kg	ì	EPA 6010	11/10/94	SD
	Vanadium, Total		ug/kg	1	EPA 6010	11/01/94	SD
	Zinc, Total		ug/kg	1	EPA 6010	11/01/94	SD
	Antimony Total, GFAA		ug/kg	ł	EPA 7041	11/03/94	NS
11	Arsenic Total, GFAA		ug/kg	1	EPA 7060	11/01/94	NS
	Lead, Total, GFAA		ug/kg	}	EPA 7421	11/01/94	NS
1	Selenium Total, GFAA	1	ug/kg		EPA 7740	11/02/94	NS (
	Thallium Total, GFAA	610	ug/kg		EPA 7841	11/02/94	NS
11	1	ŀ	1		[		
							1
- [							

PAGE 5

The results above apply only to the sample analyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of CLEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_

## LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36036

SAMPLE ID: 4102742

DESCRIPTION: MWA6-24

DATE SAMPLED.....: 10/25/94 TIME SAMPLED....: 15:45

DATE RECEIVED....: 10/31/94 TIME RECEIVED....: 11:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
Hercury, Cold Vapor	< 40	ug/kg		EPA 7471	11/02/94	SD
-						
		·		<u> </u>		

PAGE 6

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_\_

The results above apply only to the sample hashyzed. This report may only be represented in full, and may only be submitted to a third party with the written permission of CLEAN TECH

// E.

Canton Analytical Laboratory, Inc. 47801 Anchor Court, Plymouth, Michigan 48170 (313) 459-8484 FAX (313) 459-8262

ž REMARKS Analy RECEIVED BY ) ad CUSTODY # COOLER TEMP. UPON SAMPLE CONDITION UPON LAB RECEIPT RECEIPT BY LAB: DATE/TIME RELINQUISHED BY ANALYSIS DAYS (MAY RESULT IN ADDITIONAL CHARGE) ત Q NUMBER OF CONTAINERS BARD RECEIVED BY COMP 1545 1915 1915 1545 िया 1545 PROJECT PHONE **BMIT** 10/24 **DATE** CHAIN OF CUSTODY SHEET CLIENT NAME LICAN TECH ☐ RUSH ; TURNAROUND TIME MWAL - 24 M WA 6 - 20 MWA4-24 SAMPLE LOCATION SAMPLER (SIGNATURE) MINDARD JISHED B. LINOUISHED BY ADDP SAMPLE REBMUN

18597

WHITE: ACCOMPANIES SHIPMENT

08597 ż REMARKS RECEIVED BY CUSTOOY \* COOLER TEMP. UPON SAMPLE CONDITION UPON LAB RECEIPT RECEIPT BY LAB: DATEITIME 47801 Anchor Court, Ptimouth, Michigan 48170 (313) 459-8484 FAX (313) 459-8262 Canton Analytical Laboratory, Inc. RELINCUISHED BY: AMAL YSIS (MAY RESULT IN ADDITIONAL CHARGE) NUMBER OF CONTAINERS BARD RECEIVED BY COMP. 0 25 1545 1545 1925 1945 0 M 1545 PROJECT PHONE TIME म्युटा DAYS 3TAG CHAIN OF CUSTODY SHEET Ū RUSH: , TUPNAROUND TIME SAMPLE LOCATION MWAL SAMPLER (SIGNATURE) NDARD JISHED BY ADOA SAMPLE

K Blazk ž COOLER TEMP UPON SAMPLE CONDITION UPON LAB RECEIPT RECEIPT BY LAB DATE/TIME 3 RELINQUISHED BY ANALYSIS N N DAYS (MAY RESULT IN ADDITIONAL CHARGE) NUMBER OF CONTAINERS M **GRAB** RECEIVED BY COMP TIME 1891 18ch RUSH REGULAÇED TURNAROUND TIME SAMPLE LOCATION MWBD SAMPLER (SIGNATUHE) SAMPLE HBBMUN

YELLOW FILE COPY

WHITE ACCOMPANIES SHIPMENT

1

47801 Anchor Court, Plymouth, Michigan 48170

(313) 459-8484 FAX (313) 459-8262

CHAIN OF CUSTODY SHEET

Canton Analytica, Laboratory, Inc.

Ĭ

47801 Anchor Court, Plymouth, Michigan 48170 (313) 459-8484 FAX (313) 459-8262 Canton Analytical Laboratory, Inc.

CLIENT NAME OF CUSTONY SHEET

ر آگ REMARKS RECEIVED BY CUSTODY # COOLER TEMP. UPON SAMPLE CONDITION UPON LAB RECEIPT RECEIPT BY LAB DATE/TIME RELINQUISHED BY ANALYSIS WHITE ACCOMPANIES SHIPMENT d 7 DAYS (MAY RESULT IN ADDITIONAL CHARGE) RECEIVED FOR LABORATOPY BY NUMBER OF CONTAINERS (V) Chaysler Heuste GRAB RECEIVED ". сомь 11/15 0900 TIME **3TAO** Clesa Tech MWA7-24 MWA5-24 REQUIRED LURINAROUND TIME L ] RUSH SAMPLE LOCATION SAMPLE PTS GNATURE) STANDARD RELIMOUISHED BY RELINOUISHED BY NUMBER SAMPLE

YELLOW, FILE COPY

MW3-2 1 24-26 573 GTLOGORAN

LABORATORY RESULTS

CLIENT: CLEAN TECH

REPORT NUMBER: 36514 SAMPLE ID: 4112342 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA7-24

DATE SAMPLED....: 11/15/94 TIME SAMPLED....: 14:05 DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

4-Chlorotoluene       < 10       ug/kg         Dibromochloromethane       < 10       ug/kg         1,2-Dibromoethane       < 10       ug/kg         Dibromomethane       < 10       ug/kg         1,2-Dibromo-3-chloropropane       < 10       ug/kg	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
Acetone	TCL VOLATILE ODCANICE CC (NC				EDA 8260	12/02/0/	\/C
Acrolein		100			EPA 0200	12/02/74	٧3
Acrylonitrile		1 '**					
Benzene		1		}	Ì	}	
Bromoberzene		1		}		}	
Bromodichioromethane	f			1		1	
Bromodich for methane						Į.	
Bromoform			1	Į.	1		
Bromomethane						ł	
2-Butanone	1 - 1 - 1 - 1		,			1	
n-Butyl benzene         < 10         ug/kg           Sec-Butyl benzene         < 10         ug/kg           tert-Butyl benzene         < 10         ug/kg           Carbon Tetrachloride         < 100         ug/kg           Carbon Tetrachloride         < 10         ug/kg           Chlorobenzene         < 10         ug/kg           Chloroform         < 10         ug/kg           Chloroform         < 10         ug/kg           Chlorofoluene         < 10         ug/kg           2-Chlorotoluene         < 10         ug/kg           4-Chlorotoluene         < 10         ug/kg           9 bibromochloromethane         < 10         ug/kg           1 j.2-Dibromochlane         < 10         ug/kg           1 j.2-Dichlorobenzene         < 10         ug/kg           1 j.2-Dichlorobenzene         < 10         ug/kg           1 j.3-Dichlorobenzene         < 10         ug/kg           1 j.4-Dichlorobenzene         < 10         ug/kg           1 j.4-Dichlorobenzene         < 10         ug/kg           1 j.1-Dichloroethane         < 10         ug/kg           1 j.2-Dichloroethylene         < 10         ug/kg           1 j.2-Dichloroethylene	T						
Sec-Butylbenzene		1	1	}	1	ì	
tert-Butylbenzene				}		1	
Carbon Disulfide       < 100       ug/kg         Carbon Tetrachloride       < 10       ug/kg         Chlorobenzene       < 10       ug/kg         Chloroform       < 10       ug/kg         Chloromethane       < 10       ug/kg         2-Chlorotoluene       < 10       ug/kg         4-Chlorotoluene       < 10       ug/kg         9-Chlorotoluene       < 10       ug/kg         9-Chlorotoluene       < 10       ug/kg         9-Chlorotoluene       < 10       ug/kg         9-Chlorotoluene       < 10       ug/kg         1,2-Dibromoethane       < 10       ug/kg         1,2-Dibromoethane       < 10       ug/kg         1,2-Dichlorobenzene       < 10       ug/kg         1,2-Dichlorobenzene       < 10       ug/kg         1,4-Dichlorobenzene       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,2-Dichloroethylene       < 10       ug/kg         1-a-Dichloroethylene       < 10       ug/kg         1,2-Dichloroethylene       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg		1 '-		1		}	
Carbon Tetrachloride       < 10       ug/kg         Chloroform       < 10       ug/kg         Chloroform       < 10       ug/kg         Chlorotoluene       < 10       ug/kg         2-Chlorotoluene       < 10       ug/kg         4-Chlorotoluene       < 10       ug/kg         0bbromochloromethane       < 10       ug/kg         1,2-Dibnomothane       < 10       ug/kg         1,2-Dibnomo-3-chloropropane       < 10       ug/kg         1,2-Dibnomo-3-chloropropane       < 10       ug/kg         1,2-Dichlorobenzene       < 10       ug/kg         1,3-Dichlorobenzene       < 10       ug/kg         1,4-Dichlorobenzene       < 10       ug/kg         1,4-Dichlorobenzene       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,2-Dichloroethylene       < 10       ug/kg         1;2-Dichloroethylene       < 10       ug/kg         1;2-Dichloroethylene       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,2-Dichloropropane       <		,		1	į	}	
Chloroform			( • •	İ		1	
Chloroform						1	
Chloromethane	Chloroform	< 10	1 -			1	
4-Chlorotoluene       < 10	Chloromethane	< 10					
Dibromochloromethane         < 10	2-Chlorotoluene	< 10	ug/kg	1		1	
1,2-Dibromoethane       < 10	4-Chlorotoluene	< 10	ug/kg	1		1	
Dibromomethane       < 10	Dibromochloromethane	< 10	ug/kg			j	
1,2-Dibromo-3-chloropropane       < 10       ug/kg         1,2-Dichlorobenzene       < 10       ug/kg         1,3-Dichlorobenzene       < 10       ug/kg         1,4-Dichlorobenzene       < 10       ug/kg         Dichlorodifluoromethane       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,1-Dichloroethylene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         2,2-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg	1,2-Dibromoethane	< 10	ug/kg	<b>,</b>	ļ		
1,2-Dichlorobenzene       < 10       ug/kg         1,3-Dichlorobenzene       < 10       ug/kg         1,4-Dichlorobenzene       < 10       ug/kg         Dichlorodifluoromethane       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,1-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         (is-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg	Dibromomethane	< 10	ug/kg		Į.		
1,3-Dichlorobenzene       < 10       ug/kg         1,4-Dichlorobenzene       < 10       ug/kg         Dichlorodifluoromethane       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         0ichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg	1,2-Dibromo-3-chloropropane		ug/kg	1	1	1	
1,4-Dichlorobenzene       < 10       ug/kg         Dichlorodifluoromethane       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,1-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         0ichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         2,2-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg	1,2-Dichlorobenzene				ľ	1	
Dichlorodifluoromethane       < 10       ug/kg         1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,1-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         2,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg		1	( * *	Ì	[	l	
1,1-Dichloroethane       < 10       ug/kg         1,2-Dichloroethane       < 10       ug/kg         1,1-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg				j	j	j	
1,2-Dichloroethane       < 10       ug/kg         1,1-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg			1 .	}		{	
1,1-Dichloroethene       < 10       ug/kg         Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Trans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg					ł		
Cis-1,2 Dichloroethylene       < 10       ug/kg         Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Trans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg		1 1	1	į.		Į.	
Trans-1,2 Dichloroethylene       < 10       ug/kg         1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Trans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg					Į.	ł	
1,2-Dichloroethene (total)       < 20       ug/kg         Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Irans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg		<b> </b>			ĺ	-	
Dichloromethane       < 10       ug/kg         1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Irans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg	Trans-1,2 Dichloroethylene				1	ľ	
1,2-Dichloropropane       < 10       ug/kg         1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Irans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg		II.		)	1		
1,3-Dichloropropane       < 10       ug/kg         2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Trans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg				}		}	
2,2-Dichloropropane       < 10       ug/kg         1,1-Dichloropropene       < 10       ug/kg         Cis-1,3-Dichloropropene       < 10       ug/kg         Trans-1,3-Dichloropropene       < 10       ug/kg         Ethylbenzene       < 10       ug/kg		1	1		1	1	
1,1-Dichloropropene < 10					<b>f</b>		
Cis-1,3-Dichloropropene < 10 ug/kg Trans-1,3-Dichloropropene < 10 ug/kg Ethylbenzene < 10 ug/kg	•			1	<b>(</b>	\	
Trans-1,3-Dichloropropene < 10 ug/kg Ethylbenzene < 10 ug/kg							
Ethylbenzene < 10 ug/kg					1		
					1		
	•	I : -	1 -				
		1 ''	~3/ ~B				
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			<del></del>	<u> </u>	L	<u> </u>	

PAGE 1

he results above apply only to the sample individed. This report may only be reprotuded in full, and may only be submitted to ad third party with the written permission of CLEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_()[)

CLIENT: CLEAN TECH

SAMPLE ID: 4112342 REPORT NUMBER: 36514 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MWA7-24

DATE SAMPLED.....: 11/15/94 TIME SAMPLED..... 14:05

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	2-Hexanone Isopropyltoluene 4-Isopropyltoluene	< 100 < 50 * < 50 *	ug/kg ug/kg ug/kg			
	4-Methyl-2-Pentanone Naphthalene n-Propylbenzene Styrene 1,1,1,2-Tetrachloroethane	< 100 < 10 < 10 < 10 < 10	lug/kg lug/kg lug/kg lug/kg lug/kg			
	1,1,2,2-Tetrachloroethane Tetrachloroethylene Toluene 1,2,3-Trichlorobenzene 1,2,4-Trichlorobenzene	< 10 2600 < 10 < 10 < 10	lug/kg lug/kg lug/kg lug/kg lug/kg			
	1,1,1-Trichloroethane 1,1,2-Trichloroethane Trichloroethene Trichlorofluoromethane	21 < 10 38 < 10 < 10	ug/kg ug/kg ug/kg ug/kg ug/kg			
	1,2,3-Trichloropropane 1,2,4-Trimethylbenzene 1,3,5-Trimethlybenzene Vinyl Acetate Vinyl Chloride MP-Xylene	< 10 < 10 < 100 < 10 < 20	ug/kg ug/kg ug/kg ug/kg ug/kg			
	O-Xylene Xylenes, Total Chloroethane TAI METAIS	< 10 < 30 < 10	ug/kg ug/kg ug/kg	504 (010	12,02,07	<b>CD</b>
7	Aluminum, Total Barium, Total Beryllium, Total Cadmium, Total Calcium, Total	2200000 13000 95 750 12000000	ug/kg ug/kg ug/kg ug/kg ug/kg	EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010	12/02/94 11/21/94 12/01/94 11/21/94 12/01/94	SD SD SD SD SD
	Chromium, Total Cobalt, Total Silver, Total Copper, Total Sodium, Total	8400 5800 3400 11000 150000	ug/kg ug/kg ug/kg ug/kg ug/kg	EPA 6010 EPA 6010 EPA 6010 EPA 6010 EPA 6010	11/21/94 12/01/94 11/21/94 11/21/94 12/01/94	SD SD SD SD SD
-	Vanadium, Total Zinc, Total Antimony Total, GFAA Arsenic Total, GFAA	9200 25000 < 1500 4000	ug/kg ug/kg ug/kg ug/kg	EPA 6010 EPA 6010 EPA 7041 EPA 7060	12/01/94 11/21/94 11/23/94 11/21/94	SD SD NS NS
افت	lead, Total, GFAA selenium Total, GFAA Thallium Total, GFAA	3600 < 1000 1800	ug/kg ug/kg ug/kg	EPA 7421 EPA 7740 EPA 7841	11/21/94 11/22/94 11/22/94	NS NS NS

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

results above apply only to the sample elyzed. This report may only be repro-ted in full, and may only be submitted to which party with the written permission of CLEAN TECH

PAGE 2

Canton Analytical Laboratory, Inc.

71		• • •	RATORY RE	SULTS	<del></del>				
<b>                                     </b>		CLIENT: CLEAN TECH							
REPOR	T NUMBER: 36415 SAMPLE ID: 41119	761 DESCRIPTION	H: CHRYSLER ACUST	'AR MUB1-49					
DATE TIME	SAMPLED: 10/27/94 SAMPLED: 12:00	: 10/27/94 DATE RECEIVED: 11/15/94: 12:00 TIME RECEIVED: 11:50							
<u>i.</u> ]	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE TECHN			
ai	Organic Carbon	-17000	mg/kg		EPA 9060H	11/25/94. AF			
\$				:					
		•							
			·.						
	<del></del>	·							

a shows apply only to the compo-This report may only be repreit full, and may only be estimated to party with the system permanion of N TECH PAGE 7

Canton Analytical Laboratory, Inc.

DA/OC XIV

CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE 10: 4112344 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MU82-24

DATE SAMPLED ....: 11/16/94

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
t VOLATILE ORGANICS, GC/MS				EPA 8260	11/22/94	VS
cetone	< 100	ug/kg	ł			
captein	< 50	ug/kg	ì		'	
vionitrile ·	< 10	ug/kg	<b>j</b> .			
tene ·	< 10	ug/kg	į			
romobenzene	< 10	ug/kg	,			
romoch (orome thane	< 10	ug/kg	, ·	٠.		
modichloromethane	< 10	ug/kg			ļ	
Anoform	< 10	ug/kg		•		
chome thane	< 10	ug/kg	<b>!</b>			
Butanone	< 100	ug/kg				
-Puty (benzene	< 10	ug/kg	1			
Butylbenzene	< 10	ug/kg	•			
Butylbenzene	< 10.	ug/kg				
arbon Disulfide	< 100	ug/kg	1			
rbon Tetrachloride	< 10	ug/kg				
Probenzene	'< 10	Ug/kg		•		
roform	< 10	ug/kg				
romethane	<.10.	ug/kg	1			
Chlorotoluene	< 10"	ug/kg				
Chlorotoluene	< 10	ug/kg	<u>.</u> .	. 5		
bmoch Lorome thane	< 10	ug/kg	1	· .	•	
Dibromoethane	< 10	ug/kg	j			
bromomethane	< 10	ug/kg				
2-Dibromo-3-chloropropane	< 10	ug/kg		i		
pich Lorobenzene	< 10	ug/kg		j		
ichlorobenzene	< 10	ug/kg	]			
ich Lorobenzene	·< 10	ug/kg	[			
chlorodifluoromethane	< 10	ug/kg	Į į			
1-Dichtoroethane	< 10	ug/kg		•		
ichloroethane	< 10	ug/kg				
ichloroethene	< 10	ug/kg		. }		
2 Dichloroethylene	< 10	ug/kg		}		
ans-1,2 Dichtoroethylene	< 10	ug/kg		. 1		
(ichloroethene (total)		ug/kg	-	i		
oromethane	Ψ- (	Ug/kg	'	ł	•	
ichtoropropane		ug/kg	1	j·	•	
i-Dichloropropene		ug/kg	,	1		
-Dichtoropropane		ug/kg		Į.		
ichlaropropene	· · · · · · · · · · · · · · · · · · ·	ug/kg	1	ŀ		1
P* 1 7 A (= 6.1 = = = = =		ug/kg	1	•		
13-1,3-01chtoropropene		ug/kg		ł		
/		ug/kg	į	1	•	
pahlorobutadiene		ug/kg	j	}		
	}		. {	İ		
[ ]	i		1	i		

ug/kg of Dichloromethane in Method Blank.

PAGE 1

185 QA/QC\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36515 SAMPLE ID: 4112344 DESCRIPTION: PROJECT CHRYSLER ACUSTAR/MUB2-24

DATE SAMPLED ....: 11/16/94

DATE RECEIVED....: 11/18/94 TIME RECEIVED....: 09:30

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
-Hexanone	< 100	ug/kg				
sopropyitoluene	< 50 *	ug/kg				
-Isopropyl toluene	< 50 *	ug/kg		1	1	
14-Methyl-2-Pentanone	< 100	ug/kg		· •	1	
Naph that ene	< 10	ug/kg	<i>'</i>		1	
Propyl begrege	< 10	ug/kg			(	
tyrene 1 2 Terrachi ornerhane	< 10	ug/kg		<u> </u>	l	
1,1,1,2-Tetrachloroethane	< 10	ug/kg		ì	1	
1,1,2,2-Tetrachloroethane	< 10	ug/kg				. [
Letrachi oroethylene	4000	ug/kg			1	
pluene	< 10 ·	ug/kg	1.		ļ	
2,3-Trichlarobenzene	< 10	ug/kg				
1,2,4-Trichlorobenzene	< 10	ug/kg			1	j
1,1,1-Trichloroethane	< 10	ug/kg	1	1.	İ	. [
[7,1,2-Trichloroethane	< 10:	ug/kg		· ·	j·.	ł
richloroethene	< 10	ug/kg	· <b>!</b> .			
richtoroftuoromethane .	. < 10	ug/kg	1 .		·	· .
1,2,3-Trichtoropropane	< 10	ug/kg	1.			
1,2,4-Trimethylbenzene	< 10 €	ug/kg	1	· ·		
3,5-Trimethlybenzene	< 10	ug/kg	1	1.		
inyl Acetate	< 100	ug/kg				
Vinyt Chloride	< 10	ug/kg				j
MP-Xylene .	< 20	ug/kg	<b>[</b> .			}
Xylene	< 10	ug/kg	ļ'			i
vienes, Total	< 30	ug/kg	ľ			
loroethane	< 10	ug/kg	į i			- !
TAL METALS		1				
Aluminum, Total	2100000	ug/kg		EPA 6010	11/20/94	SD
Arium, Total	12000	ug/kg		EPA 6010	11/21/94	SO
eryllium, Total	88	ug/kg	] . ]	EPA 6010	12/01/94	SD
Coloium, Total	620	ug/kg		EPA 6010	11/21/94	SD
Calcium, Total	83000000	ug/kg	ļ ·	EPA 6010	12/01/94	<b>SD</b>
posit, Total	7200	ug/kg		EPA 6010	12/01/94	20
iver, Total	5100	ug/kg	1	EPA 6010	12/01/94	SD
Copper, Total	3700	ug/kg	1	EPA 6010	12/01/94	20
Sodium, Total	10000	ug/kg	{ · {	EPA 6010	11/21/94	SD SD
anadium, Total	140000	ug/kg	i i	EPA 6010	12/01/94	SO .
inc, Total	9500 22000	ug/kg		EPA 6010	12/01/94	20
Antimony Total, GFAA	< 1500	ug/kg	<b> </b>	EPA 6010 EPA 7041	11/21/94	NS
( Tenic Total GEAA	7100	ug/kg	[· · ]	EPA 7060	11/23/94	NS NS
Alien, lotal cras	7100 3500	ug/kg	l .		11/21/94	NS
[ CONTINUE TO SEA		ug/kg	i i	EPA 7421	11/21/94	NS
Tallium Total, GFAA	< 1000	ug/kg	· · ·	EPA 7740	11/22/94	NS NS
THE STATE OF THE S	2000	ug/kg		EPA 7841	11/22/94	49 E
						}

entative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 2

Canton Analytical Laboratory, Inc.

QA/QC\_

REPORT NUMBER: 36270 SAMPLE ID: 4111214 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MJB3-24

DATE SAMPLED....: 11/03/94 TIME SAMPLED....: 11:45 DATE RECEIVED....: 11/09/94 TIME RECEIVED....: 09:45

	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECH
				1		
	· · · · · · · · · · · · · · · · · · ·		1.			
TAL METALS						
Aluminum, Total	1500000	ug/kg	· ·	EPA 6010	11/10/94	<b>SD</b>
Barium, Total	12000	ug/kg		EPA 6010	11/10/94	SD
Beryllium, Total	300	ug/kg	l	EPA 6010	11/21/94	20
Cadmium, Total	680	ug/kg		EPA 6010	11/10/94	20
Calcium, Total	140000	mg/kg	· ·	EPA 6010	11/17/94	20
Chromium, Total	6800	ug/kg		EPA 6010	11/10/94	SD
Cobalt, Total .	<b>8200</b> .	ug/kg		EPA 6010	11/18/94	SD
Silver, Total	3100	ug/kg		EPA 6010	11/10/94	SD
Copper, Total	12000	ug/kg		EPA 6010	11/10/94	SD
Sodium, Total	190000	ug/kg	·	EPA 6010	11/17/94	SD
Vanadium, Total	9500	ug/kg		EPA 6010	11/18/94	<b>SD</b>
Zinc, Total	19000	ug/kg		EPA 6010	11/10/94	20
Antimony Total, GFAA	< 1500	ug/kg	1	EPA 7041	11/11/94	NS
Arsenic Total, GFAA	6000	ug/kg	1	EPA 7060	11/11/94	NS
Lead, Total, GFAA	.3000	ug/kg		EPA 7421	11/11/94	NS
Selenium Total, GFAA	< .1000	ug/kg		EPA 7740	11/11/94	NS
Thallium Total, GFAA	4600	ug/kg		EPA 7841	11/11/94	NS
Mercury, Cold Vapor	< 100	ug/kg	1:	EPA 7471	11/11/94	SD
TCL VOLATILE ORGANICS, GC/MS	-		- 1	EPA 8260	11/16/94	٧S
Acetone	∫ < 100	ug/kg	1			
Acrolein	< 50	ug/kg		į		
Acrylonitrile	< 10	ug/kg	<b>.</b>	Ĭ	· ·	
Benzene	< 10	ug/kg	1	1	}	
Bromobenzene	< 10	ug/kg	1	į.	1	
Bromoch Loromethane	< 10	ug/kg	ŀ	i	i	
Bromodichloromethane	< 10	ug/kg	-	İ		
Bromoform	< 10	ug/kg			1	
Bromomethane	< 10	ug/kg	1	İ	1	
2-Butanone	< 100	ug/kg	ł			
n-Butylbenzene	< 10	ug/kg			İ	
Sec-Butyl benzene	< 10	ug/kg		J		
tert-Butylbenzene	< 10	ug/kg		]	Ì	
Carbon Disulfide	< 100	ug/kg.		1	ţ	
Carbon Tetrachloride	< 10	ug/kg		}	1	
Chiorobenzene	< 10	ug/kg		}	1	
Chlaroform	< 10	ug/kg				•
Chloromethane	< 10				Ì	
2-Chlorotoluene	< 10	ug/kg				
4-Chlorotoluene	< 10	ug/kg				
Dibromochloromethane	< 10	ug/kg	1		1	
1,2-Dibromoethane	< 10	ug/kg			1	
0 fbromomethane	< 10	ug/kg				
1,2-0fbromo-3-chloropropane		ug/kg			į	
. te a rei ein. 3. cut o Labrabaus	< 10	ug/kg			j	
		1		i	1	

<sup>\*</sup> Tentative identification and quantitation based on a computer generated Library search per section 7.6.1.2, EPA Method 8260

PAGE 1

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 36270

SAMPLE ID: 4111214

DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB3-24

DATE SAMPLED....: 11/03/94 TIME SAMPLED....: 11:45

DATE RECEIVED....: 11/09/94
TIME RECEIVED....: 09:45

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL . METHOD	RUN DATE	TECHN
]	1,2-Dichlorobenzene 1,3-Dichlorobenzene 1,4-Dichlorobenzene	< 10 < 10 < 10	ug/kg ug/kg ug/kg				
1	Dichlorodifluoromethane	< 10	ug/kg	•			

ug/kg 1,1-Dichloroethane < 10 ug/kg 1,2-Dichloroethane < 10 ug/kg < 10 1.1-Dichloroethene ug/ka Cis-1,2 Dichloroethylene 81 ug/kg < 10 Trans-1,2 Dichloroethylene ug/kg 81 1,2-Dichloroethene (total) ug/kg 15 Dichloromethane ug/kg 1,2-Dichloropropane < 10 ug/kg 1,3-Dichloropropane < 10 ug/kg 2,2-Dichloropropane < 10 ug/kg 1,1-Dichloropropene < 10 ug/kg Cis-1,3-Dichloropropene < 10 ug/kg Trans-1,3-Dichloropropene < 10. ug/kg Ethylbenzene < 10 ug/kg Hexach Lorobutadiene < 10. ug/kg 2-Hexanone < 100 ug/kg Isopropyltoluene < 50 \* ug/kg < 50 \* 4-Isopropyltoluene ug/kg 4-Methyl-2-Pentanone < 100. ug/kg Naphthalene < 10 ug/kg

n-Propylbenzene < 10 ug/kg Styrene < 10 ug/kg 1,1,1,2-Tetrachloroethane < 10 ua/ka 1,1,2,2-Tetrachloroethane < 10 ug/kg Tetrachioroethylene < 10 ug/kg Totuene < 10 ua/ka 1,2,3-Trichlorobenzene < 10 ug/kg 1,2,4-Trichtorobenzene < 10 ug/kg 1,1,1-Trichloroethane < 10 ua/ka 1,1,2-Trichloroethane < 10 ug/kg Trichloroethene 1200 ug/kg Trichlorofluoromethane < 10 ug/kg 1,2,3-Trichloropropane < 10 ug/kg 1,2,4-Trimethylbenzene < 10 ug/kg

ug/kg

ug/kg

< 10

< 10

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 2

1,3,5-Trimethlybenzene

Vinyl Acetate

GAZOC KW

CLIENT: CLEAN TECH

SAMPLE 1D: 4110331 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR HWB4-19 REPORT NUMBER: 36123

DATE SAMPLED.....: 11/01/94 TIME SAMPLED.....: 08:45

DATE RECEIVED ....: 11/03/94

TIME RECEIVED ....: 09:45

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
1						
TAL METALS	1	[	t	(		
Aluminum, Total	1900000	ug/kg	1	EPA 6010	11/15/94	SD
Barium, Total	13000	ug/kg	1	EPA 6010	11/08/94	SD
Beryllium, Total	210	ug/kg	Ì	EPA 6010	11/15/94	SD
Cadmium, Total	670	ug/kg	1 .	EPA 6010	11/15/94	SD
Calcium, Total	260000000	ug/kg	} '	EPA 6010	11/15/94	SD
Chromium, Total	6200	ug/kg	1	EPA 6010	11/15/94	SD
Cobalt, Total	4800	ug/kg	ł	EPA 6010	11/15/94	SD
Silver, Total	2500	ug/kg	ì	EPA 6010	11/08/94	SD
Copper, Total	9100	ug/kg	1	EPA 6010	11/08/94	SD
Sodium, Total	130000	ug/kg	}	EPA 6010	11/15/94	SD
Vanadium, Total	8600	ug/kg	ł	EPA 6010	11/15/94	SD
Zinc, Total	18000	ug/kg	}	EPA 6010	11/08/94	SD
Antimony Total, GFAA	< 1500	ug/kg		EPA 7041	11/10/94	NS
Arsenic Total, GFAA	4100	ug/kg		EPA 7060	11/09/94	NS
Lead, Total, GFAA	4500	ug/kg	1	EPA 7421	11/09/94	NS
Selenium Total, GFAA	< 1000	ug/kg		EPA 7740	11/10/94	NS
النس Total, GFAA	1600	ug/kg		EPA 7841	11/09/94	NS
ury, Cold Vapor	< 40	ug/kg	İ	EPA 7471	11/08/94	SD
VOLATILE ORGANICS, GC/MS			İ	EPA 8260	11/13/94	VS .
Acetone	< 100	ug/kg	{	ļ	<b>\</b>	
Acrolein	< 50	ug/kg			ł	
Acrylonitrile	i < 10	ug/kg	İ		}	
8enzene	<b>  &lt; 10</b>	ug/kg	1		}	
Bromobenzene	< 10	ug/kg	ţ		ţ	
Bromochioromethane	< 10	ug/kg				
Bromodichloromethane .	< 10	ug/kg	j			
Bromoform	< 10	ug/kg	}			
Bromomethane	< 10	ug/kg	l .			
2-Butanone	< 100	ug/kg	i			
n-Butylbenzene	< 10	ug/kg	1			
Sec-Butylbenzene	∫ < 10	ug/kg	Į į			,
tert-Butylbenzene	< 10	ug/kg	1			
Carbon Disulfide	<b>  &lt; 100</b>	ug/kg	]			
Carbon Tetrachloride	< 10	ug/kg	}			
Chlorobenzene	< 10	ug/kg	Į l			
Chloroform	< 10	ug/kg	l i			
Chloromethane	< 10	ug/kg				j
2-Chlorotoluene	< 10	ug/kg	]	1		1
4-Chlorotoluene	< 10	ug/kg			1	ļ
Dibromochloromethane	< 10	ug/kg	]		1	j
1,2-Dibromoethane	< 10	ug/kg				1
Dibromomethane	< 10	ug/kg		į		į
1,2-Dibromo-3-chloropropane	< 10	ug/kg	]	Í		i
	1			ì		1
	ı	1	ı			ļ

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 13

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Canton Analytical Laboratory, Inc.

ga/gc KN

CLIENT: CLEAN TECH

REPORT NUMBER: 36123 SAMPLE 1D: 4110331 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB4-19

DATE SAMPLED.....: 11/01/94 TIME SAMPLED.....: 08:45

DATE RECEIVED....: 11/03/94

TIME RECEIVED ....: 09:45

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
						*************
1,2-Dichlorobenzene	< 10	ug/kg		1	1	
1,3-Dichlorobenzene	< 10	ug/kg		İ		
1,4-Dichlorobenzene	) < 10	ug/kg		1	1	
)ichlorodifluoromethane	< 10	ug/kg		į	· ·	
1,1-Dichloroethane	) < 10	ug/kg			]	
1,2-Dichloroethane	< 10	ug/kg		1		
1,1-Dichloroethene	< 10	ug/kg		ļ		
Cis-1,2 Dichloroethylene	< 10	ug/kg		)	1	
rans-1,2 Dichloroethylene	< 10	ug/kg			1	
1,2-Dichloroethene (total)	< 20	ug/kg		Ì	1	
ichloromethane	<b>  &lt; 10</b>	ug/kg		İ	1	
,2-Dichloropropane	< 10	ug/kg		i	1	
,3-Dichloropropane	< 10	ug/kg		1	{	
2,2-Dichloropropane	< 10	ug/kg		i	İ	
,1-Dichloropropene	< 10	ug/kg		}	}	
is-1,3-Dichloropropene	ļ < 10	ug/kg [		ļ	ł	
rans-1,3-Dichloropropene	<b>√ &lt; 10</b>	ug/kg		}	ł	
"   Yl benzene	< 10	ug/kg		<b>{</b>	1	
,chi orobutadiene	<b>  &lt; 10</b>	ug/kg			1	
rriexanone	< 100	ug/kg			}	
sopropyltoluene	< 50 *	ug/kg			ļ	
-lsopropyltoluene	< 50 *	ug/kg		Ì	1	
-Methyl-2-Pentanone	< 100	ug/kg		1	Į	
aph tha lene	j < 10	ug/kg			ļ	
-Propylbenzene	< 10	ug/kg			1	
tyrene	< 10	ug/kg				
,1,1,2-Tetrachloroethane	<b>  &lt; 10</b>	ug/kg		)	1	
,1,2,2-Tetrachloroethane	< 10	ug/kg			l	
etrachloroethylene	< 10	ug/kg (			ł	
oluene	< 10	ug/kg			ļ	
,2,3-Trichlorobenzene	< 10	lug/kg				
,2,4-Trichlorobenzene	< 10	ug/kg	;		ſ	
,1,1-Trichloroethane	< 10	ug/kg			Ī	
,1,2-Trichloroethane	< 10	ug/kg			İ	
richloroethene	< 10	ug/kg			(	
richlorofluoromethane	< 10	ug/kg			ĺ	
,2,3-Trichloropropane	< 10	ug/kg		•	1	
,2,4-Trimethylbenzene	< 10	ug/kg			l	
,3,5-Trimethlybenzene	< 10	ug/kg			ĺ	
inyl Acetate	< 10	ug/kg	į		į	
inyl Chloride	< 10	ug/kg			ł	
P-Xylene	< 20	ug/kg			}	
-Xylene	< 10	ug/kg			!	
ylenes, Total	( < 30	lug/kg			1	

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 14

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LABORATORY RESULTS CLIENT: CLEAN TECH	
DESCRIPTION: PROJECT-CHRYSLER ACUSTAR	
	DATE RECEIVED: 11/03/94 TIME RECEIVED: 09:45

ANALYTICAL ANALYTICAL UNITS OF ANALYSES PERFORMED RESULT MEASURE METHOD DATE TECHN Chloroethane < 10 ug/kg

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 15

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REPORT NUMBER: 36123

DATE SAMPLED.....: 11/01/94

TIME SAMPLED.....: 08:45

SAMPLE ID: 4110331

REPORT NUMBER: 36270 SAMPLE ID: 4111215 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB5-24

DATE SAMPLED....: 11/07/94 TIME SAMPLED....: 11:25

DATE RECEIVED....: 11/09/94

TIME RECEIVED ....: 09:45

		1		T ANALYZION	T SUN	
ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
TAL METALS		1	1		]	
Aluminum, Total	2100000	ug/kg	}	EPA 6010	11/10/94	SD
Barium, Total	10000	ug/kg	i	EPA 6010	11/10/94	SD
Beryllium, Total	270	ug/kg	ļ	EPA 6010	11/21/94	SD
Cadmium, Total	550	ug/kg		EPA 6010	11/10/94	SD
Calcium, Total	73000	mg/kg		EPA 6010	01/17/95	SD
Chromium, Total	6300	ug/kg	į.	EPA 6010	11/10/94	SD
Cobalt, Total	8300	ug/kg		EPA 6010	11/18/94	SD
Silver, Total	2700	ug/kg		EPA 6010	11/10/94	SD
Copper, Total	9100	ug/kg		EPA 6010	11/10/94	SD
Sodium, Total	170000	ug/kg		EPA 6010	11/17/94	SD
Vanadium, Total	11000	ug/kg	1	EPA 6010	11/18/94	SD
Zinc, Total	22000	ug/kg		EPA 6010	11/10/94	SD
Antimony Total, GFAA	< 1500	ug/kg		EPA 7041	11/11/94	NS
Arsenic Total, GFAA	8400	ug/kg		EPA 7060	11/11/94	NS
Lead, Total, GFAA	3300	ug/kg	]	EPA 7421	11/11/94	NS
Selenium Total, GFAA	< 1000	ug/kg	1	EPA 7740	11/11/94	NS
allium Total, GFAA	2100	ug/kg		EPA 7841	11/11/94	NS
cury, Cold Vapor	< 100	ug/kg	}	EPA 7471	11/11/94	SD
ಳುವರ VOLATILE ORGANICS, GC/MS		1	I	EPA 8260	11/16/94	VS
Acetone	< 100	ug/kg	1	1	]	
Acrolein	< 50	ug/kg	Ţ		1	
Acrylonitrile	< 10	ug/kg		1		
Benzene	< 10	ug/kg	<b>{</b>	1	1	
Bromobenzene	< 10	ug/kg	ľ	1	1	
Bromochloromethane	· < 10	ug/kg	1	1	1	
Bromodichloromethane	< 10	ug/kg	ł	Į.	Į.	
Bromoform	< 10	ug/kg	1	i		
Bromomethane	< 10	ug/kg	}	}	1	
2-Butanone	< 100	ug/kg		}		
n-Butylbenzene	< 10	ug/kg	}	}	1	
Sec-Butylbenzene	< 10	ug/kg	ł	Į.		
tert-Butylbenzene	< 10	ug/kg	ł	1	]	
Carbon Disulfide	< 100	ug/kg	1	<b>\</b>		
Carbon Tetrachloride	< 10	ug/kg		ł		
Chlorobenzene Chloroform	< 10	ug/kg	1	}		
	< 10	ug/kg	l		(	
Chloromethane 2-Chlorotoluene	< 10	ug/kg				
4-Chlorotoluene	< 10	ug/kg		1	}	
Dibromochloromethane	< 10 < 10	ug/kg			1	
1,2-Dibromoethane	< 10	ug/kg			l .	
Dibromomethane	< 10	ug/kg			l	
1,2-Dibromo-3-chloropropane	< 10	ug/kg			ł	
The bibliomorp circuit opi opane	1 10	ug/kg			}	
	1	1			ł	

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 4

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Canton Analytical Laboratory, Inc.

GA/GC KN

CLIENT: CLEAN TECH

REPORT NUMBER: 36270 SAMPLE ID: 4111215 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB5-24

DATE SAMPLED.....: 11/07/94 TIME SAMPLED.....: 11:25 DATE RECEIVED....: 11/09/94

TIME RECEIVED ....: 09:45

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
,2-Dichlorobenzene	< 10	ug/kg		[	ļ	
,3-Dichlorobenzene	< 10	ug/kg		i		
,4-Dichlorobenzene	< 10	ug/kg		1	1	
ichlorodifluoromethane	< 10	ug/kg		ĺ		
,1-Dichloroethane	< 10	ug/kg	•	ł	ł	
,2-Dichloroethane	∫ < 10	ug/kg	1	}	}	
,1-Dichloroethene	< 10	ug/kg		1	1	
is-1,2 Dichloroethylene	Ì < 10	ug/kg		1	ł	
rans-1,2 Dichloroethylene	< 10	ug/kg		1	1	
,2-Dichloroethene (total)	< 10	ug/kg		1	ļ	
ichloromethane	14	ug/kg		i	1	
,2-Dichloropropane	<b>⟨ &lt; 10</b>	ug/kg	1	{	Į.	
,3-Dichloropropane	< 10	ug/kg				
,2-Dichloropropane	i < 10	ug/kg		1	]	
,1-Dichloropropene	< 10	ug/kg	1	!		
is-1,3-Dichloropropene	< 10	ug/kg			1	
rans-1,3-Dichloropropene	< 10	ug/kg	i	)	1	
thylbenzene	< 10	ug/kg	į	ļ	ļ	
wachlorobutadiene	< 10	ug/kg	1	<u> </u>		
lexanone	< 100	ug/kg	j		1	
sopropyltoluene	< 50 *	ug/kg		]	l	
-Isopropyltoluene	< 50 *	ug/kg			1	
Methyl-2-Pentanone	< 100	ug/kg	}		<u> </u>	
aphthalen <b>e</b>	< 10	ug/kg			l	
Propylbenzene	< 10	ug/kg			1	
tyrene	< 10	ug/kg			}	
1,1,2-Tetrachloroethane	l < 10	ug/kg			ĺ	
1,2,2-Tetrachloroethane	< 10	ug/kg			i	
etrachloroethylene	< 10	ug/kg	}		}	
luene	< 10	ug/kg			ł	
2.3-Trichlorobenzene	< 10	ug/kg			)	
2,4-Trichlorobenzene	< 10	ug/kg	<b>(</b>		ļ	
1,1-Trichloroethane	14	ug/kg			}	
1,2-Trichloroethane	< 10	ug/kg				
ichloroethene	470	ug/kg			ļ	
ichlorofluoromethane	< 10	ug/kg			ļ	
2,3-Trichloropropane	< 10	ug/kg	1		1	
2,4-Trimethylbenzene	< 10	ug/kg			ļ	
3,5-Trimethlybenzene	< 10	ug/kg	1			
nyl Acetate	< 10	ug/kg	}		)	
nyl Chloride	< 10	ug/kg	1			
-Xylene	< 20	ug/kg				
Xylene	< 10	ug/kg				
lenes, Total	< 30	1	1 1			
tenes. totat		lug/kg	1			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 5

Canton Analytical Laboratory, Inc.

GAZGC KN

]	···						
( 		LABO	RATORY RECLIENT: CLEAN TO				
.	REPORT NUMBER: 36270 SAMPLE ID: 4111	215 DESCRIPTI	ON: PROJECT-CHRYS	LER ACUSTAR MWB5-	24		
7	DATE SAMPLED: 11/07/94 TIME SAMPLED: 11:25			DA Tii	TE RECEIVED	: 11/09/94 : 09:45	
_}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	< 10	ug/kg				
ì							
d)							
2							
~							
*							
	_						
***	-						
			·				
7							
_							

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 6

Canton Analytical Laboratory, Inc.

GA/GC KIN

REPORT NUMBER: 36270 SAMPLE ID: 4111218 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB5-49

DATE SAMPLED.....: 11/07/94 TIME SAMPLED.....: 13:35

DATE RECEIVED....: 11/09/94

TIME RECEIVED....: 09:45

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
	Total Organic Carbon	21000	mg/kg		EPA 9060M	11/16/94	AF
•							
1	···						
	_)						
		<u>1</u>	<u>_</u>	<u> </u>			

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PAGE 9

Canton Analytical Laboratory, Inc.

REPORT NUMBER: 36415 SAMPLE ID: 4111960

DESCRIPTION: CHRYSLER ACUSTAR MWB6-24

DATE SAMPLED....: 11/09/94 TIME SAMPLED....: 08:30

DATE RECEIVED....: 11/15/94

TIME RECEIVED....: 11:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
TCL VOLATILE ORGANICS, GC/MS				EPA 8260	11/16/94	vs
Acetone	< 100	ug/kg		L. A. OLOG	11,10,74	••
Acrolein	< 50	ug/kg				
Acrylonitrile	< 10	ug/kg	1			
Benzene	< 10	ug/kg	]	Ĭ		
Bromobenzene	< 10	ug/kg			i	
Bromochloromethane	< 10	1 4. 4				
	< 10	ug/kg	]	1	}	
Bromodichloromethane	< 10	ug/kg	İ			
Bromoform	< 10	ug/kg				
Bromomethane		ug/kg	· ·	l	ļ	
2-Butanone	< 100	ug/kg		]		
n-Butylbenzene	< 10	ug/kg		i		
Sec-Butylbenzene	< 10	ug/kg		1	ı	
ert-Butylbenzene	< 10	ug/kg		ţ		
Carbon Disulfide	< 100	ug/kg				
Carbon Tetrachloride	< 10	ug/kg		1	İ	
thlorobenzene	< 10	ug/kg			ļ	
hloroform	< 10	ug/kg			}	
hloromethane	< 10	ug/kg				
!-Chlorotoluene	< 10	ug/kg				
-Chlorotoluene	< 10	ug/kg		}	ļ	
ibromochloromethane	< 10	ug/kg		İ	i	
.2-Dibromoethane	< 10	ug/kg			ļ	
ibromomethan <b>e</b>	< 10	ug/kg				
,2-Dibromo-3-chloropropane	< 10	ug/kg		ł		
,2-Dichlorobenzene	< 10	ug/kg		Ï	)	
,3-Dichlorobenzene	< 10	ug/kg				
,4-Dichlorobenzene	< 10	ug/kg				
ichlorodifluoromethane	< 10	ug/kg				
,1-Dichloroethane	17	ug/kg				
,2-Dichloroethane	< 10	ug/kg				
,1-Dichloroethene	< 10	ug/kg				
is-1,2 Dichloroethylene	53	ug/kg ug/kg	1			
rans-1,2 Dichloroethylene	< 10					
	53	ug/kg			1	
,2-Dichloroethene (total)	< 10	ug/kg				
ichloromethane	< 10	ug/kg	1			
,2-Dichloropropane	1 7 7	ug/kg				
,3-Dichloropropane	< 10 < 10	ug/kg				
,2-Dichloropropane		ug/kg	1 .			
,1-Dichloropropene	< 10	ug/kg				
is-1,3-Dichloropropene	< 10	ug/kg				
rans-1,3-Dichloropropene	< 10	ug/kg				
thylbenzene	< 10	ug/kg	ĺ			
exachlorobutadiene	< 10	ug/kg				

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 4

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Canton Analytical Laboratory, Inc.

GA/GC KV

CLIENT: CLEAN TECH

REPORT NUMBER: 36415 SAMPLE ID: 4111960 DESCRIPTION: CHRYSLER ACUSTAR MWB6-24

DATE SAMPLED....: 11/09/94 TIME SAMPLED.....: 08:30

DATE RECEIVED....: 11/15/94 TIME RECEIVED....: 11:50

ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
Mariana	< 100					· · · · · · · · · · · · · · · · · · ·
2-Hexanone	< 50 *	ug/kg				
Isopropyltoluene	< 50 *	ug/kg	i			
-Isopropyltoluene	1	ug/kg				
-Methyl-2-Pentanone	< 100 < 10	ug/kg				
laphthalene		ug/kg	l	•	ļ	
n-Propylbenzene	< 10	ug/kg	<b>!</b>		ţ	
Styrene	< 10	ug/kg				
,1,1,2-Tetrachloroethane	< 10	ug/kg	1		ļ	
,1,2,2-Tetrachloroethane	< 10	ug/kg				
etrachloroethylene	< 10	ug/kg	<u> </u>		1	
oluene	< 10	ug/kg			İ	
,2,3-Trichlorobenzene	< 10	ug/kg				
,2,4-Trichlorobenzene	< 10	ug/kg	İ			
,1,1-Trichloroethane	420	ug/kg				
,1,2-Trichloroethane	< 10	ug/kg				
richloroethene	400	ug/kg	ĺ			
richlorofluoromethane	< 10	ug/kg				
,2,3-Trichloropropane	< 10	ug/kg				
,2,4-Trimethylbenzene	< 10	ug/kg				
,3,5-Trimethlybenzene	< 10	ug/kg				
inyl Acetate	<b>&lt;</b> 10	ug/kg				
inyl Chloride	< 10	ug/kg				
P-Xyl ene	< 20	ug/kg				
-Xylene	< 10	ug/kg				
ylenes, Total	< 30	ug/kg				
hloroethane	< 10	ug/kg				
AL METALS						
luminum, Total	2900000	ug/kg		EPA 6010	11/29/94	SD
arium, Total	10000	ug/kg		EPA 6010	11/29/94	SD
eryllium, Total	280	ug/kg		EPA 6010	11/29/94	SD
admium, Total	770	ug/kg	i	EPA 6010	11/29/94	SD
alcium, Total	100000000	ug/kg		EPA 6010	11/29/94	SD
hromium, Total	7000	ug/kg		EPA 6010	11/29/94	SD
obalt, Total	4400	ug/kg		EPA 6010	11/29/94	SD
ilver, Total	3100	ug/kg		EPA 6010	11/16/94	SD
opper, Total	8100	ug/kg		EPA 6010	11/29/94	SD
odium, Total	130000	ug/kg		EPA 6010	11/29/94	SD
anadium, Total	8000	ug/kg		EPA 6010	11/29/94	SD
inc, Total	18000	ug/kg		EPA 6010	11/29/94	SD
ntimony Total, GFAA	< 1500	ug/kg		EPA 7041	11/17/94	NS
rsenic Total, GFAA	3500	ug/kg	•	EPA 7060	11/17/94	NS
ead, Total, GFAA	2600	ug/kg		EPA 7421	11/17/94	NS
elenium Total, GFAA	< 1000	ug/kg		EPA 7740	11/18/94	NS
nallium Total, GFAA		ug/kg	ì	EPA 7841	11/17/94	NS

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 5

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REPORT NUMBER: 36415

SAMPLE ID: 4111960

DESCRIPTION: CHRYSLER ACUSTAR MWB6-24

DATE SAMPLED....: 11/09/94 TIME SAMPLED....: 08:30 DATE RECEIVED....: 11/15/94 TIME RECEIVED....: 11:50

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Mercury, Cold Vapor	< 100	ug/kg	EPA 7471	11/17/94	SD
<u>!</u>					
				·	

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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Canton Analytical Laboratory, Inc.

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Descripte MDR-4 19-21 Sig p 47 GT Lightert.

#### LABORATORY RESULTS CLIENT: CLEAN TECH

REPORT NUMBER: 36123 SAMPLE ID: 4110330

DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB7-19

DATE SAMPLED....: 10/31/94 TIME SAMPLED....: 10:15 DATE RECEIVED....: 11/03/94 TIME RECEIVED....: 09:45

ANALYSES SECTIONS	ANALYTICAL	UNITS OF	ANALYTICAL	RUN	****
ANALYSES PERFORMED	RESULT	MEASURE	METHOD	DATE	TECHN
AL METALS	1800000		504 (040	44.415.404	00
Numinum, Total		ug/kg	EPA 6010	11/15/94	SD
Barium, Total	11000	ug/kg	EPA 6010	11/08/94	SD
eryllium, Total	190	ug/kg	EPA 6010	11/15/94	SD
Cadmium, Total	630	ug/kg	EPA 6010	11/15/94	SD
alcium, Total	230000000	ug/kg	EPA 6010	11/15/94	SD
hromium, Total	5900	ug/kg	EPA 6010	11/15/94	SD
obalt, Total	4200	ug/kg	EPA 6010	11/15/94	SD
ilver, Total	3600	ug/kg	EPA 6010	11/08/94	SD
opper, Total	8400	ug/kg	EPA 6010	11/08/94	SD
odium, Total	130000	ug/kg	EPA 6010	11/15/94	SD
anadium, Total	7800	ug/kg	EPA 6010	11/15/94	SD
inc, Total	16000	ug/kg	EPA 6010	11/08/94	SD
ntimony Total, GFAA	< 1500	ug/kg	EPA 7041	11/10/94	NS
rsenic Total, GFAA	2500	ug/kg	EPA 7060	11/09/94	NS
ead, Total, GFAA	3000	ug/kg	EPA 7421	11/09/94	NS
elenium Total, GFAA	< 1000	ug/kg	EPA 7740	11/10/94	NS
∖ium Total, GFAA	1900	ug/kg	EPA 7841	11/09/94	NS
Jury, Cold Vapor	< 40	ug/kg	EPA 7471	11/08/94	SD
TE VOLATILE ORGANICS, GC/MS	- 1		EPA 8260	11/13/94	٧S
cetone	< 100	ug/kg i	ļ		
crolein	< 50	ug/kg	l l		
rylonitrile	< 10	ug/kg		i	
enzene	< 10	ug/kg	ļ		
romobenzene	< 10	ug/kg			
romochloromethane	< 10	ug/kg			
omodichloromethane -	< 10	ug/kg	Ì	1	
omoform	i < 10	ug/kg			
omomethane	< 10	ug/kg	<b>\</b>	(	
Butanone	< 100	ug/kg			
Butylbenzene	< 10	ug/kg			
c-Butylbenzene	< 10	ug/kg			
rt-Butylbenzene	< 10	ug/kg			
rbon Disulfide	< 100	ug/kg			
rbon Tetrachloride	< 10	ug/kg			
lorobenzene	< 10	ug/kg		ļ	
loroform	< 10	ug/kg			
loromethane	< 10	ug/kg	4		
Chlorotoluene	< 10	ug/kg			
Chlorotoluene	< 10	ug/kg	[ ]		
bromochloromethane	< 10	ug/kg			
2-Dibromoethane	< 10	ug/kg			
bromomethane	< 10	ug/kg	}		
2-Dibromo-3-chloropropane	< 10		]		
E DIDIGIO-3-CITOTODI ODANE	1 10	lug/kg			

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 10

le results above apply only to the sample balayzed. This report may only be reproduced in full, and may only be submitted to a third parry with the written permission of , LEAN TECH

Canton Analytical Laboratory, Inc.

GA/GC KV

CLIENT: CLEAN TECH

REPORT NUMBER: 36123 SAMPLE ID: 4110330 DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB7-19

DATE SAMPLED.....: 10/31/94 TIME SAMPLED.....: 10:15

DATE RECEIVED....: 11/03/94 TIME RECEIVED....: 09:45

ŧ.	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
_		1					
	1 3-Dishlanshans	   < 10	115 /4-				
¥.	1,2-Dichlorobenzene 1,3-Dichlorobenzene	< 10	ug/kg ug/kg	1			
	1,4-Dichtorobenzene	< 10	ug/kg				
	Dichlorodifluoromethane	< 10	ug/kg	ŀ		İ	
Н	1,1-Dichloroethane	< 10	ug/kg	1		}	
	1,2-Dichloroethane	< 10	ug/kg	[		ł	
	1,1-Dichloroethene	< 10	ug/kg	1		[	
	Cis-1,2 Dichloroethylene	< 10	ug/kg			ĺ	
	Trans-1,2 Dichloroethylene	< 10	ug/kg	1		Ì	
E.	1,2-Dichloroethene (total)	< 20	ug/kg			ĺ	
ĭ	Dichloromethane	< 10	ug/kg			!	
ļ	1,2-Dichloropropane	< 10	ug/kg				
	1,3-Dichloropropane	< 10	ug/kg		İ	1	
	2,2-Dichloropropane	< 10	ug/kg	}			
	1,1-Dichloropropene	< 10	ug/kg	!			
J	Cis-1,3-Dichloropropene	< 10	ug/kg	[			
-	Trans-1,3-Dichloropropene	< 10	ug/kg	[			
	\\lbenzene	< 10	ug/kg	1			
į.	chlorobutadiene	< 10	ug/kg	!			j
Т	Z-Hexanone	< 100	ug/kg	ŀ			
ı	Isopropyltoluene	< 50 *	ug/kg				
1	4-Isopropyltoluene	< 50 *	ug/kg				ļ
	4-Metnyl-2-Pentanone	< 100	ug/kg				
	Naph that ene	< 10	ug/kg	ļ .	1		į
	n-Propylbenzene	< 10	ug/kg				
	Styrene	< 10	ug/kg	]			
235	1,1,1,2-Tetrachloroethane	< 10 < 10	ug/kg		ļ		J
	1,1,2,2-Tetrachloroethane Tetrachloroethylene	< 10	ug/kg		•		<b>1</b>
	Toluene	< 10	ug/kg				ł
Ė		-	ug/kg				Į.
	1,2,3-Trichlorobenzene 1,2,4-Trichlorobenzene	< 10 < 10	ug/kg				1
	1,1,1-Trichtoroethane	< 10	ug/kg	ľ	ì		)
Τ	1,1,2-Trichtoroethane	< 10	ug/kg ug/kg				1
	Trichloroethene	< 10	ug/kg				ŀ
	Trichlorofluoromethane	< 10	ug/kg		ĺ		i
	1.2. Z. Trichloronno	< 10	ug/kg	]	l l		ľ
7	1,2,4-Trimethylbenzene	< 10	ug/kg		l.		- 1
1	1,3,5-Trimethlybenzene	< 10	ug/kg		ļ		1
	Vinyl Acetate	< 10	ug/kg	Į.	ţ		l.
	Vinyl Chloride	< 10	ug/kg		1		ļ
	MP-Xylene	< 20	ug/kg	İ	j		i
	O-Xylene	< 10	ug/kg		1		I
- 2	Kylenes, Total	< 30	ug/kg	1	Į.		ļ
ı					[		j
	1			t e	1		

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 11

he results above apply only to the sample salyzed. This report may only be repro-duced in full, and may only be submitted to a third party with the written permission of "LEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_KN

# LABORATORY RESULTS CLIENT: CLEAN TECH DESCRIPTION: PROJECT-CHRYSLER ACUSTAR MWB7-19

DATE SAMPLED.....: 10/31/94 TIME SAMPLED.....: 10:15

SAMPLE ID: 4110330

REPORT NUMBER: 36123

DATE RECEIVED....: 11/03/94 TIME RECEIVED....: 09:45

( ) (	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	< 10	ug/kg			
1						
Ţ						
"						
1	<u> </u>					
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<u>ل</u> ـــ						

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 12

Canton Analytical Laboratory, Inc.

GA/GC TN

results above apply only to the sample yzed. This report may only be represeded in full, and may only be submitted to third party with the written permission of "LEAN TECH

AWBY-24 Sample Anyles REMARKS COOLER TEMP. UPON SAMPLE CONDITION UPON LAB RECEIPT RECEIPT BY LAB (313) 459-8484 FAX (313) 459-8262 HELINOUIS-ED BY ANALYSIS MAY RESULT IN ADDITIONAL CHARGE)  $\beta$ NUMBER OF CONTAINERS N Chrysler Acusta RECEIVED BY COMP 1330 2780 1330 0200 TIME 16/31 18/0 **DATE** CHAIN OF CUSTODY SHEET Clean Tech [] RUSH REQUIRED TURNAROUND TIME SAMPLE LOCATION MWAZ--CHMH MWB7 MWRY IN SAMPLER (SIGNATURE) STANDARD
RELINGUISHED BY HEO BY CLIENT NAME ADDRESS 3J9MAS R38MUN

47801 Anchor Court, Plymouth, Michigan 48170

Canton Anaig.

ž

CUSTODY #

YELLOW FILE COPY

WHITE. ACCOMPANIES SHIPMENT

Canton Analytical Laboratory, Inc. 47801 Anchor Court. Plymouth. Michigan 48170 (313) 459-8484 FAX (313) 459-8262

CHAIN OF CUSTODY SHEET

STREET FROM PORTION OF CONTRINERS OF CONTRINERS OF CONTRINERS OF CONTRINERS OF CONTRINERS OF CONTRINING OF CONTRINING OF CONTRINING OF CONTRION UPON LAB RECEIPT BY LAB.    Interior   Days   MAY RESULT IN ADDITIONAL CHARGE   RECEIPT BY LAB.   Interior   Days   MAY RESULT IN ADDITIONAL CHARGE   RECEIPT BY LAB.   Interior   Days   D	REMARKS	Aralyses per	EIPT Ng 78531
COMP  DAYS (MAY RESULT IN ADDITIONAL CHARGE)  Offering  PROJECT  P			AMPLE CONDITION UPON LAB RECE
DAYS  BATEL  DAYS  BATEL  DAYS  BATEL  DAYS  DATE  DAT	3	( N	COOLER TEMP. UPON RECEIPT BY LAB:  C REUMQUISHED BY THEMMRIS:
A DATE DATE	TIME GRAB	/ / / / / / / / / / / / / / / / / / / /	(MAY RESULT IN ADDITIONAL CHAINE FOR LABORATORY
	Clean Tech LOO Wabster Strees SENATURED OF SAMPLE LOCATION		WAROUND TIME:  BUSH: DAYS  APPEND

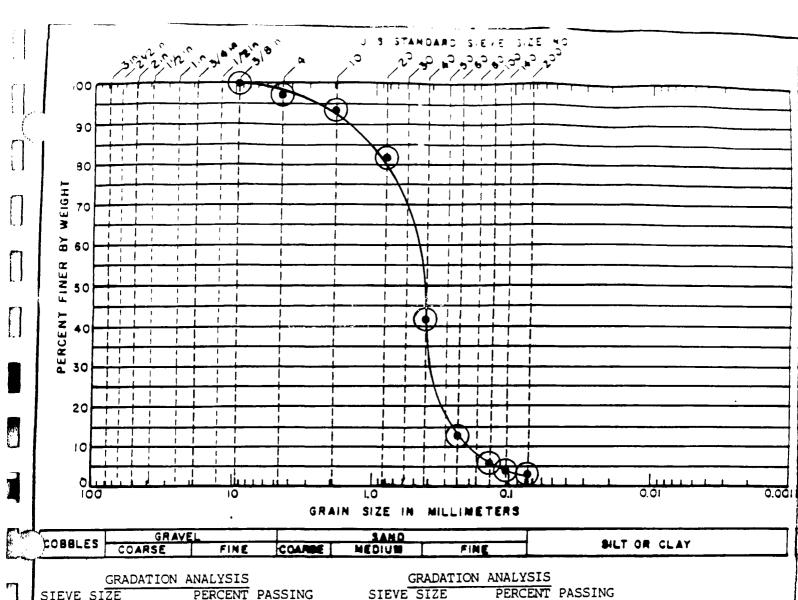
WHITE ACCOMPANIES SHIPMENT YELLOW, FILE COPY

Samples 2500 2500 9 A PRIVES ace HS/MSD MWB4-24 Ž VECSING) 27 CUSTODY COOLER TEMP. UPON SAMPLE CONDITION LPON LAB RECEIPT RECEIPT BY LAB: 47801 Anchor Court. Plymouth. Michigan 48170 (313) 459-8484 FAX (313) 459-8262 YELLOW FILE COPY NELAGUIS-ED BY: ANALYSIS L'AY RESULT IN ADDITIONAL CHARGEL Canton Alway. 7 NUMBER OF CONTAINERS GRAB PECEIVED BY COMP /ze 08/5 0815 1330 020 10/31; 1330 **JMIT** CHAIN OF CUSTODY SHEET Tech FUS+: RECUIRED TUPNAROUND TIME SAMPLE LOCATION MW.A2 - 19 HWB7-1 71-785 HWBd-589-14 - CHUM AN BR HW84 SAMPLER (SIGNATURE) ☐ STANDAPC REUNON SMED 3 CLIENT NAME ADORESS SJAMAS REBMUN

		Frayses Der	PECEIPT PECEIPT Nº (45)
			SAMPLE CONDITION UPON LAB FECEIPT DATETIME FEC
, Inc. Kum 48170	2,20		<del>┥╸</del> ┝╾┼╼┩╺ <del>┞</del> ╼╬╼╼═┤ │ │
tialytical Laboratory, Inc. Court, Plymouth, Michigun 48170 -8484 FAX (31.3) 459-8262	ANALYSIS	18 18	COOLER RECEIPT
47801 Anchor Court. (213) 459-8484	OMP.		FIN ADCITIONAL CO ECE.VED BY WHITE: ADOM
	HON HAIT	N 2 2 2 0 1 1 1	DAYS (MAY RESUL)  DAYETIME
CHAIN OF CUSTODY SHEET	ran Tech	-24 -30 104 -49 5-49 3-49	
CHAIN OFICE	SAMPLER SIGNATURE C'ENT ASORESS SAMPLER SIGNATURE SAMPLER SAMPLELOCATION	HW33 MWB4 NWB3	REQUIRED TURNARD JND TIME:  STANDARD  C. RUSH  PELMOUSHED BY  RELNOUSHED BY:

#### ATTACHMENT K

Laboratory Results: Geotechnical
Soil Samples from Monitoring Well Borings
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404



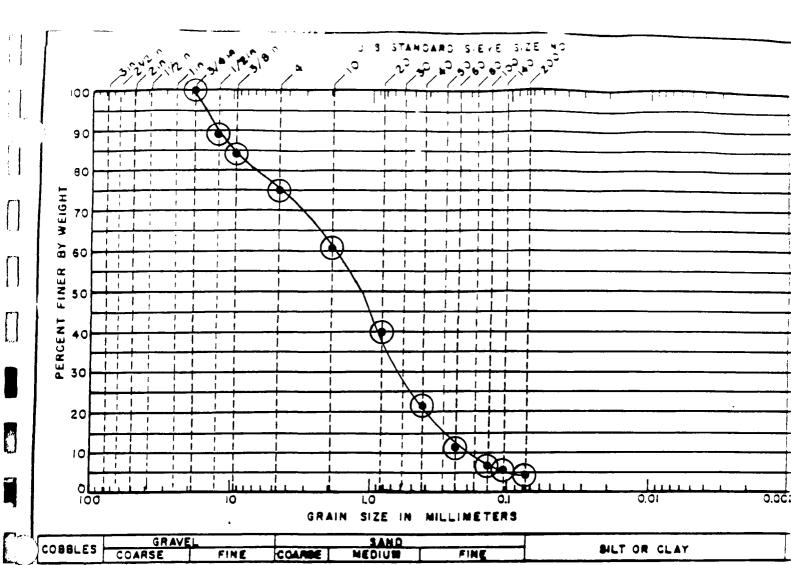
GRADAT	TION ANALYSIS	GRADATIC	N ANALYSIS
SIEVE SIZE	PERCENT PASSING	SIEVE SIZE	PERCENT PASSING
215"	100	#10	93.5
2"	100	#20	82.0
1½"	100	#40	41.9
1"	100	#60	/3./
3/4"	100	#100	5.6
ا <sub>ک</sub> ے "	100	#140	4,2
3/8"	100	#200	3.3
#4	97.3		

DARK BROWN F-LA SAUD WITH ATRACE OF SILT

LIQUID LIMIT: PLASTIC LIMIT:

4 FINE GRAVEL.

TEXTURAL GRADATION ANALYSIS  for  CHRYSLER / Acust 4.2	TETRA TECH RICHARDSON				
CHRYSLER / ACUSTAR DAYTON, OHIO	DATE 11/30/94				
SAMPLE NO. MWAY	NO. 8/16 OF MAS OF JJL				



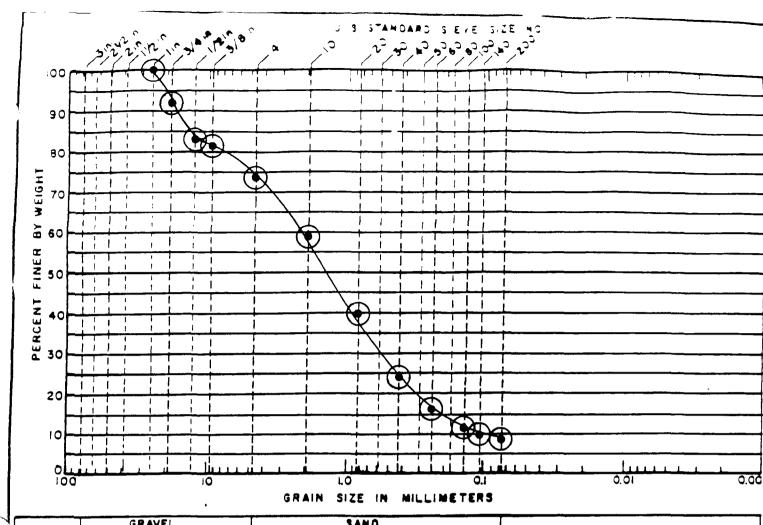
GRAD	ATION ANALYSIS	GRADATIO	ON ANALYSIS
SIEVE SIZE	PERCENT PASSING	SIEVE SIZE	PERCENT PASSING
21/3"	100	#10	60.6
2"	100	#20	40.0
15"	100	<b>#4</b> 0	21.8
	100	#60	11.5
3/4"	100	#100	6.7
1/2"	89.0	#140	5.4
3/8"	84.3	#200	4.6
#4	75.3		

DARK BROWN F-M SAND WITH A LITTLE FINE GRAVEL.

LIQUID LIMIT: PLASTIC LIMIT:

& TRACE OF SILT.

TEXTURAL GRADATION ANALYSIS  for  CHRYSIER / Ann. 57.42	TETRA TECH RICHARDSON				
CHRYSLER / ACUSTAR DAYTON, OHIO	DATE 11/30/94				
SAMPLE NO. MWAS	NO. 8116 DOMEN MAB OMECKED JJL				



``		GRA	VEL		SANO		AU - AB AI AV
	COBBLES	COARSE	FINE	COARSE	MEDIUM	FINE	SILT OR CLAY
,,,,,					<u> </u>		
		GRADATION	ANALYSIS		GRAI	ATION ANALYSIS	
	SIEVE S	IZE	PERCENT I	PASSING	SIEVE SIZ	E PERCENT	PASSING

Ol GIDIII	1011 111111111111				
SIEVE SIZE	PERCENT PASSING	SIEVE SIZE	PERCENT PASSING		
	100	#10	59.0		
2"	100	#20	<i>39.</i> 7		
1½"	100	#40	24.1		
1"	100	#60	16.4		
3/4"	92.0	#100	11.6		
1/2"	83.2	#140	9.8		
3/8"	81.8	#200	8.5		
#4	74.0				

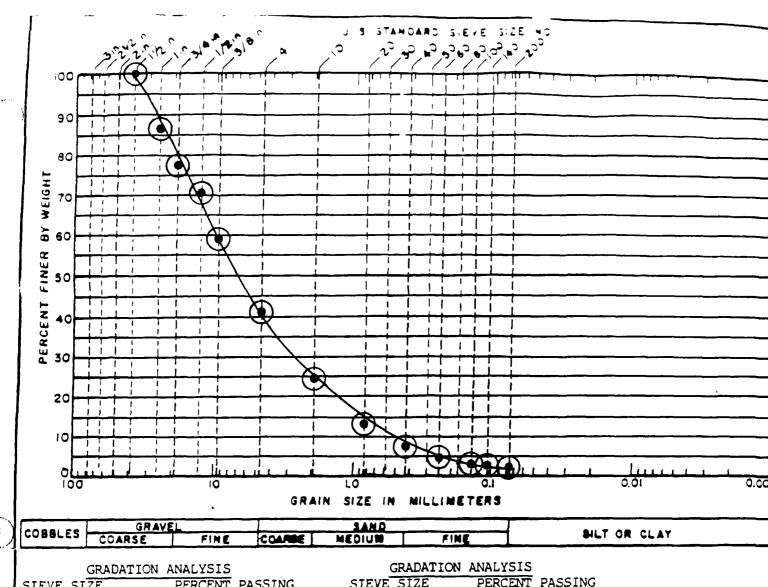
VISUAL DESCRIPTION

GRAY GRAVELLY F-C SOND WITH A TRACE

LIQUID LIMIT: PLASTIC LIMIT:

مة عالم.

TEXTURAL GRADATION ANALYSIS  for  CHRYSLER   ACUSTAR DAYTON, OHIO	TETRA TECH RICHARDSON MENARL DEL			
75'	DATE 11/30/94	<u> </u>		
SAMPLE No. MWB2	NO. 8/16	OF MAB	ST CHECKED	JJL

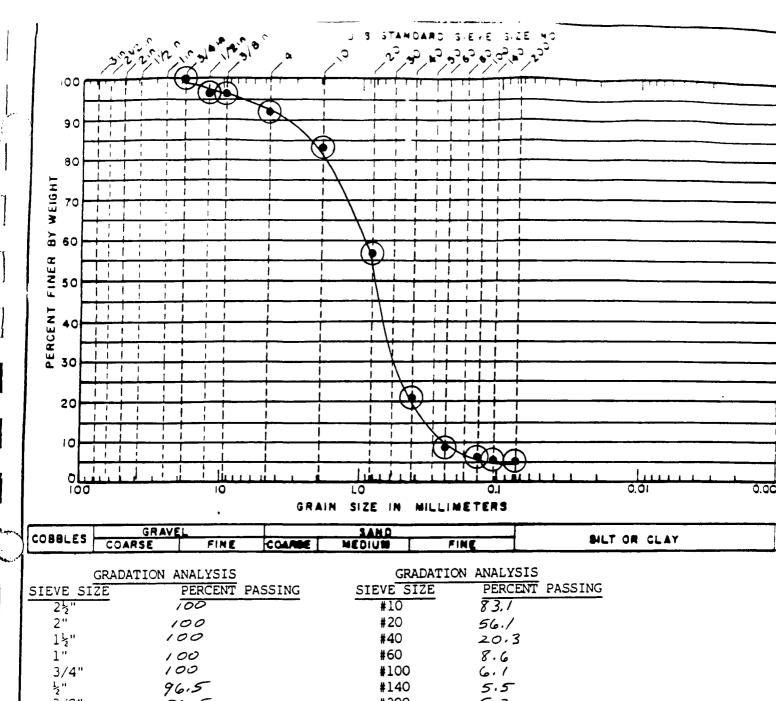


GRADAT:	ION ANALYSIS	GRADATIC	N ANALYSIS
SIEVE SIZE  2½" 2" 1½" 1" 3/4" ½" 3/8" #4	PERCENT PASSING 100 100 100 86.6 77.3 70.6 58.9 41.1	#10 #20 #40 #60 #100 #140 #200	PERCENT PASSING 24.2 13.7 7.6 4.7 3.5 2.8 2.3

GRAY SANDY FLC GRAVEL WITH ATTACE OF SILT.

LIQUID LIMIT: PLASTIC LIMIT:

TEXTURAL GRADATION ANALYSIS  for  CHRYSLER   ACUSTAR DAYTON, OHIO	TETRA TECH RICHARDSON					
104 - 106	DATE 11/30/94					
SAMPLE NO. MWCI	NO. 8/16 OF MAR OF JJL					



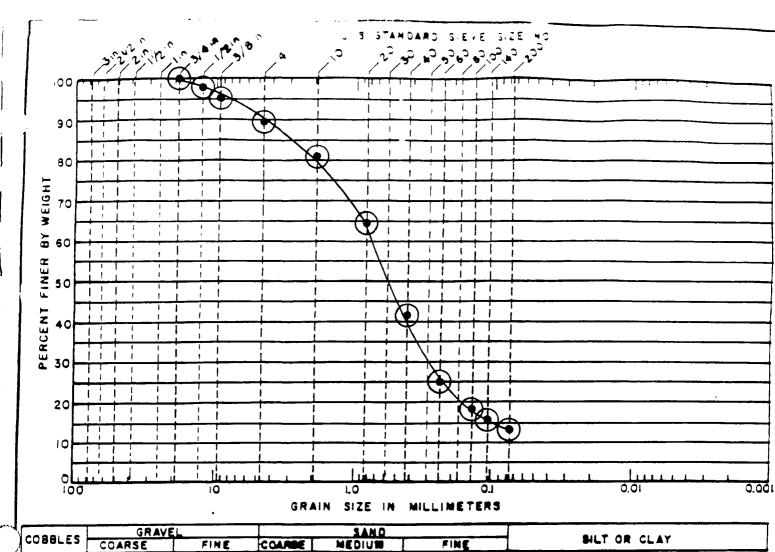
GRAD	DATION ANALYSIS	GRADATIC	N ANALISIS
SIEVE SIZE	PERCENT PASSING	SIEVE SIZE	PERCENT PASSING
2½"	100	#10	83.1
2"	100	#20	56./
1½"	100	#40	20.3
1"	100	#60	8.6
3/4"	100	#100	6.1
1/2"	96.5	#140	5. <b>5</b>
3/8"	96.5	#200	5.2
#4	92.4		

GRAY F-M SOND

WITH A TRACE OF

LIQUID LIMIT: PLASTIC LIMIT: SILT & FINE GRAVEL.

TEXTURAL GRADATION ANALYSIS  for  CHRYSCER / ACUSTAR DAVIDOR	TETRA TECH RICHARDSON				
CHRYSLER/ACUSTAR DAYTON, OHIO	DATE 11/30/9	4			
SAMPLE NO. MWC2	NO. 8116	ON MAB	OCCUED BY	JJL	



	/	100			
•	GRADA	TION ANALYSIS	GRADATI	ON ANALYSIS	
	SIEVE SIZE	PERCENT PASSING	SIEVE SIZE	PERCENT PASSING	
	21/5"	100	#10	80.6	
	2"	100	#20	64.8	
ı	1 ½"	100	#40	41.1	
ı	1"	100	#60	25.1	
	3/4"	100	#100	18.4	
Ì	15"	97.9	#140	15.8	
1	3/8"	95.3	#200	/3.5	
1	#4	89.7			

DARK BROWN F-M SAND

WITH & LITTLE SILT

LIQUID LIMIT: PLASTIC LIMIT: & FINE GRAVEL.

TEXTURAL GRADATION ANALYSIS  for  CHRYSLER (A. C. T. C. C. C. C. C. C. C. C. C. C. C. C. C.	TETRA TECH RICHARDSON HEWARL DE					
CHRYSLER / ACUSTAR DAYTON, J410 76-78'	DATE 11/30/94					
SAMPLE NO. MWC3	NO. 8/16 OF MB OF JUL					

#### ATTACHMENT M

Laboratory Results: Chemical
First Round Groundwater Samples
Chrysler Corporation
Dayton Thermal Products Plant
1600 Webster Street
Dayton, Ohio 45404

REPORT NUMBER: 37188 SAMPLE ID: 4121985 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA1-12

DATE SAMPLED.....: 12/13/94 TIME SAMPLED.....: 19:30 DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

L	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION	ANALYTICAL METHOD	RUN DATE	TECHN
<b>,</b> , ,	AIRLIGES FERTORIES	REJUEI	HEASURE	LIMIT	HETHOU	JAIL	LUIN
	TAL METALS						
i	Aluminum, Dissolved	< 140	ug/l		EPA 200.7	12/29/94	SD
ئے	Barium, Dissolved	210	ug/L	}	EPA 200.7	12/22/94	SD
	Beryllium, Dissolved	< 10	ug/l	1	EPA 200.7	12/29/94	SD
	Cadmium, Dissolved	< 10	ug/l	1	EPA 200.7	12/22/94	SD
	Calcium, Dissolved	130	mg/l		EPA 200.7	12/29/94	SD
	Chromium, Dissolved	< 20	lug/l		EPA 200.7	12/22/94	SD
,,	Cobalt, Dissolveed	< 20	ug/l		EPA 200.7	12/29/94	SD
	Silver, Dissolved	45	ug/L		EPA 200.7	12/22/94	SD
7	Copper, Dissolved	< 20 73	ug/l	[	EPA 200.7	12/22/94	SD
}	Sodium, Dissolved	73 < 20	mg/l	1	EPA 200.7 EPA 200.7	12/29/94	SD SD
ات	Vanadium, Dissolved Zinc, Dissolved	< 20 < 20	lug/l		EPA 200.7	12/29/94 12/22/94	SD
	Antimony Diss., GFAA	6.6	ug/l		EPA 200.7	12/23/94	NS NS
	Arsenic Diss., GFAA	< 2.0	ug/l  ug/l		EPA 204.2	12/23/94	NS NS
J	Lead, Diss., GFAA	< 1.0	lug/l	1	EPA 239.2	12/22/94	NS
	Selenium Diss., GFAA	< 2.0	ug/l	!	EPA 270.2	12/27/94	NS
	" `\u · · ·	13	ug/l	)	EPA 279.2	12/29/94	NS
ŧ.	cury, Cold Vapor	< 0.2	ug/l	1	EPA 245.2	12/22/94	SD
	TCL VOLATILE ORGANICS, GC/MS	- • <del>-</del>			EPA 8260	12/21/94	vs
- 1	Acetone	ND	ug/l	50			· · · · · · · · · · · · · · · · · · ·
r	Acrolein	ND	ug/l	50			į
ŀ	Acrolein Acrylonitrile Benzene	ND	ug/l	5.0	!		
۳	Benzene	ND	ug/l	1.0	j		ì
-	Bromobenzene	ND	ug/l	1.0	1		ł
	Bromochloromethane	ND	ug/l	1.0	1		j
Ŋ	Bromochloromethane Bromodichloromethane Bromoform	ND	ug/l	1.0			ł
	Bromoform	ND	ug/l	] 1.0	}		}
	Bromomethane	ND	ug/l	1.0			ł
_	2-Butanone	ND	ug/l	50	j		]
	n-Butyl benzene	ND	ug/l	1.0			ł
	Sec-Butylbenzene	ND	ug/l	1.0	j		ì
	tert-Butylbenzene	ND	ug/l	1.0			ĺ
	Carbon Disulfide	ND	ug/l	50	]		1
	Carbon Tetrachloride	ND	ug/l	1.0			1
	Chlorobenzene	ND	ug/l	1.0	}		]
	Chloroform	ND	ug/l	1.0			i
	Chloromethane	ND	ug/l	1.0			- 1
	2-Chlorotoluene	ND [	ug/l	1.0			[
	4-Chlorotoluene	ND )	ug/l	1.0	İ		- 1
	Dibromochloromethane 1,2-Dibromoethane	ND .	ug/l	1.0			1
			ug/l	1.0			
	Dibromomethane 1,2-Dibromo-3-chloropropane	ND	ug/l	1.0			į
''1	1,2 o foi olio-3-ch conopropane	ND	ug/l	1.0			l l
- [			{		1		

ND= None Detected at or above the limit reported

Prentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 13

c results above apply only to the sample had lyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of TEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121985 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA1-12

DATE SAMPLED....: 12/13/94 TIME SAMPLED.....: 19:30

DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
1,2-Dichlorobenzene	ND	ug/l	1.0	1	1	
1,3-Dichlorobenzene	ND	ug/l	1.0	<u> </u>		
1,4-Dichlorobenzene	ND	ug/l	1.0		1	
)ichlorodifluoromethane	470	ug/(	1.0		1	
1,1-Dichloroethane	65	ug/l	1.0	}	[	
1,2-Dichloroethane	ND	ug/l	1.0	ļ	1	
1,1-Dichloroethene	81	ug/l	1.0		í	
Cis-1,2 Dichloroethylene	110	ug/l	1.0		Į.	
rans-1,2 Dichloroethylene	ND	ug/l	1.0			
,2-Dichloroethene (total)	110	ug/l	2.0		1	
ichloromethane	ND	lug/l	1.0		1	
,2-Dichloropropane	ND	lug/l	1.0		ľ	
,3-Dichloropropane	ND	ug/l	1.0	ļ		
,2-Dichloropropane	ND	ug/l	1.0		ļ	
,1-Dichlaropropene	ND	ug/l	1.0		ļ	
is-1,3-Dichloropropene	ND	ug/l	1.0		[	
mons-1,3-Dichloropropene	ND	ug/l	1.0			
benzene	ND	ug/l	1.0		]	
chlorobutadiene مراكم	ND	ug/l	1.0			
-Hexanone	ND	ug/l	50		Ĭ	
sopropyltoluene	ND	ug/l	1.0 *		i	
-Isopropyltoluene	ND	ug/l	1.0 *		1	
-Methyl-2-Pentanone	ND	ug/l	50			
aphthalene	ND .	ug/l	11.0		1	
-Propylbenzene	ND	ug/l	1.0		1	
tyrene	ND	ug/l	1.0		ì	
,1,1,2-Tetrachloroethane	ND	ug/l	l 1.0		l	
,1,2,2-Tetrachloroethane	ND	ug/l	11.0		1	
etrachloroethylene	2500	ug/l	1.0		j	
oluene	ND	ug/l	1.0			
,2,3-Trichlorobenzene	ND	ug/l	1.0		<b>{</b>	
,2,4-Trichlorobenzene	ND	ug/l	11.0			
,1,1-Trichloroethane	3300	ug/l	1.0		ļ	
,1,2-Trichloroethane	ND	ug/l	1.0			
richloroethene	350	ug/t	1.0			
richlorofluoromethane	ND	ug/l	1.0			
,2,3-Trichloropropane	ND	ug/l	1.0			
,2,4-Trimethylbenzene	ND	ug/l	1.0	l		
3,5-Trimethlybenzene	ND	ug/l	1.0			
inyl Acetate inyl Chloride P-Xylene	ND	ug/l	50			
inyl Chloride	ND	ug/t	1.0			
)-Xyl ene	ND	ug/l	2.0			
V I	ND	ug/l	1.0			
·Xylene						
xylene /lenes, Total	ND	ug/l	3.0			

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 14

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QA/QC\_\_\_\_

LABORATORY RESULTS CLIENT: CLEAN TECH							
REPORT NUMBER: 37188 SAMPLE ID: 4121985 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA1-12							
DATE SAMPLED: 12/13/94 TIME SAMPLED: 19:30		_	DAT TIM	TE RECEIVED	: 12/16/94 : 14:20		
ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE TE		
hloroethane	ND	ug/l	1.0				
·.							
.)							
None Detected at or above the liferative identification and quant	mit reported itation based on a co	omputer generated	l library search per	section 7.6.1.	2, EPA Method 826		
			<del></del>				

REPORT NUMBER: 37188 SAMPLE ID: 4121983 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA2-12

DATE SAMPLED.....: 12/13/94 TIME SAMPLED.....: 17:45 DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

		ANALYTICAL	UNITS OF	DETECTION	ANALYTICAL	RUN	
1	ANALYSES PERFORMED	RESULT	MEASURE	LIMIT	METHOD	DATE	TECHN
-							THE SUID TO
1		ĺ		j		1	
	TAL METALS	<b>{</b>				{	
	Aluminum, Dissolved	< 140	ug/l		EPA 200.7	12/29/94	SD
	Barium, Dissolved	160	ug/l		EPA 200.7	12/22/94	SD
	Beryllium, Dissolved	< 10	ug/l	ļ	EPA 200.7	12/29/94	SD
	Cadmium, Dissolved	<b>&lt; 10</b> .	ug/l	Ì	EPA 200.7	12/22/94	SD
7	Calcium, Dissolved	100	mg/l	į i	EPA 200.7	12/29/94	SD
- 1	Chromium, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
4	Cobalt, Dissolved	< 20	ug/l	]	EPA 200.7	12/29/94	SD
	Silver, Dissolved	34	ug/l	]	EPA 200.7	12/22/94	SD
	Copper, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
	Sodium, Dissolved	72	mg/l		EPA 200.7	12/29/94	SD
	Vanadium, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
	Zinc, Dissolved	< 20	ug/l	[ .	EPA 200.7	12/22/94	SD
	Antimony Diss., GFAA	5.0	ug/l	}	EPA 204.2	12/23/94	NS
	Arsenic Diss., GFAA	< 2.0	ug/l	]	EPA 206.2	12/29/94	NS
	Lead, Diss., GFAA	< 1.0	ug/l	}	EPA 239.2	12/22/94	NS
بلور	Selenium Diss., GFAA	< 2.0	ug/l	}	EPA 270.2	12/27/94	NS
1	lium Diss., GFAA	15	ug/l	1	EPA 279.2	12/29/94	NS
ښا	Jury, Cold Vapor	< 0.2	lug/(	ļ	EPA 245.2	12/22/94	SD
	TCL VOLATILE ORGANICS, GC/MS		1		EPA 8260	12/21/94	VS
	Acetone	ND	ug/l	50			
	Acrolein	ND	ug/l	50			
	Acrylonitrile Benzene	ND	ug/l	5.0			
7	Benzene	ND	ug/l	1.0			
	Bromobenzene Bromochloromethane	ND	ug/l	1.0		; 	
	Bromochtoromethane	ND	ug/l	1.0			
100	Bromoform	ND ND	ug/l	1.0			
	Bromomethane	ND ND	ug/l	1.0			j
	2-Butanone	ND ND	ug/l	50			ì
Ľ		ND .	ug/l				j
	n-Butylbenzene Sec-Butylbenzene	ND	ug/l ug/l	1.0 1.0			1
	tert-Butylbenzene	ND ND	ug/l	1.0	İ		
	Carbon Disulfide	ND	ug/l	50	l		]
	Carbon Tetrachloride	ND	ug/l	1.0	ļ		i
44.53	thtorobenzene	ND	ug/l	1.0			ŀ
1. 1	thloroform	ND	ug/l	1.0			ļ
4	Chloromethane	ND	ug/l	1.0			İ
13	2-Chlorotoluene	ND	ug/l	1.0			i
	-Chlorotoluene	ND .	ug/l	1.0			ı
	ibromochloromethane	ND	ug/l	1.0	}		ì
W	,2-Dibromoethane	ND	ug/l	1.0			
10	ibromomethane	ND	ug/l	1.0	i		1
11	,2-Dibromo-3-chloropropane	ND	ug/l	1.0	İ		ł
- 1	1			1	Ì		Ì
4	į		ļ		į		į

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 7

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Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121983 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA2-12

DATE SAMPLED....: 12/13/94 TIME SAMPLED....: 17:45 DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

. 1	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION	ANALYTICAL METHOD	RUN DATE	TECHN
	ANALISES FERTORIED	KEJOET	ALAJUKL	LIMIT	I FICTION	1 2015	, CCIIN
1			1				
J	1,2-Dichlorobenzene	ND	ug/l	1.0			
	1,3-Dichlorobenzene	ND	ug/l	1.0		}	
	1,4-Dichlorobenzene	ND	ug/l	1.0	}		
	Dichlorodifluoromethane	94	ug/l	1.0	}	<b>}</b>	
	1,1-Dichloroethane	3.6	ug/l	1.0	]		
-	1,2-Dichloroethane	ND	ug/l	1.0	Î	1	
	1,1-Dichloroethene	84	ug/l	1.0		1	
-	Cis-1,2 Dichloroethylene	8.3	ug/l	1.0	ŀ	1	
Δi	Trans-1,2 Dichloroethylene	ND	ug/l	1.0	ł	i	
	1,2-Dichloroethene (total)	8.3	ug/l	2.0	1	{	
_	Dichloromethane	ND	ug/l	1.0		l	
	1,2-Dichloropropane	ND	ug/l	1.0	<b>!</b>	1	
2	1,3-Dichloropropane	ND	ug/l	1.0	1	1	
	2,2-Dichloropropane	ND	ug/l	1.0	ł	Į.	
7	1,1-Dichloropropene	ND	ug/l	1.0		1	
- (	Cis-1,3-Dichloropropene	ND	ug/l	1.0		1	
_	Trans-1,3-Dichloropropene	ND	ug/l	1.0	Ì	i	
•	hylbenzene	ND	ug/l	11.0	ļ	İ	
de.	kachlorobutadiene	ND	ug/l	1.0		i	
۳,	rz-Hexanone	ND	ug/l	50		ł	
ı	Isopropyltoluene	ND	ug/l	1.0 *		1	
ni	4-Isopropyltoluene	ND	ug/l	1.0 *		t	
	4-Methyl-2-Pentanone	ND	ug/l	50		ļ	
J	Naphthalene	ND	ug/l	1.0		l	
J	n-Propylbenzene	ND	ug/l	1.0		[	
	Styrene	ND	ug/l	1.0		ſ	
	1,1,1,2-Tetrachloroethane	ND	ug/l	1.0		1	
	1,1,2,2-Tetrachloroethane	ND	ug/l	1.0			
	Tetrachloroethylene	2400	ug/l	1.0		ĺ	
	Toluene	ND	ug/l	1.0		1	
	1,2,3-Trichlorobenzene	ND	ug/l	11.0			
	1,2,4-Trichlorobenzene	ND	ug/l	1.0		1	
	1,1,1-Trichloroethane	8600	ug/l	1.0		1	
	1,1,2-Trichloroethane	ND	ug/l	1.0		1	
	Trichloroethene	110	ug/l	1.0		1	
	Trichlorofluoromethane	ND	ug/l	1.0		i	
	1,2,3-Trichloropropane	ND	ug/l	1.0		}	
	1,2,4-Trimethylbenzene	ND	ug/l	1.0		1	
	1,3,5-Trimethlybenzene	ND	ug/l	1.0		1	
7	Vinyl Acetate	ND	ug/l	50		1	
ŀ	Vinyl Chloride	ND	ug/l	1.0			
	MP-Xylene	ND	ug/l	2.0			
1	0-Xylene	ND	ug/l	1.0		ł	
1	(ylenes, Total	ND	ug/l	3.0		!	
1	•			}			
1		1		}			

ND= None Detected at or above the limit reported

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

CLIENT: CLEAN TECH

REPORT NUMBER: 37188

SAMPLE ID: 4121983

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA2-12

DATE SAMPLED....: 12/13/94 TIME SAMPLED....: 17:45

DATE RECEIVED ....: 12/16/94

TIME RECEIVED....: 14:20

١.,	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	ND	ug/l	1.0			
	·						
F							
- -			ī	i i			
	-						
-							
-							
	-						
_	<del> </del>	<del></del>					

ND= None Detected at or above the limit reported

PAGE 9

results above apply only to the sample red. This report may only be repround in full, and may only be submitted to third party with the written permission of LEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

REPORT NUMBER: 37188 SAMPLE ID: 4121984 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA3-12

DATE SAMPLED....: 12/13/94 TIME SAMPLED....: 18:30 DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

		ANALYTICAL	UNITS OF	DETECTION	ANALYTICAL	RUN	
	ANALYSES PERFORMED	RESULT	MEASURE	LIMIT	METHOD	DATE	TECHN
П						1	
	TAL METALS	1					
	Aluminum, Dissolved	< 140	ug/l		EPA 200.7	12/29/94	SD
ļ	Barium, Dissolved	250	ug/l		EPA 200.7	12/22/94	SD
	Beryllium, Dissolved	< 10	ug/l		EPA 200.7	12/29/94	SD
	Cadmium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
4	Calcium, Dissolved	150	mg/l		EPA 200.7	12/29/94	SD
	Chromium, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
	Cobalt, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
. 1	Silver, Dissolved	38	ug/l		EPA 200.7	12/22/94	SD
	Copper, Dissolved	< 20	ug/l	1	EPA 200.7	12/22/94	SD
	Sodium, Dissolved	96	mg/l	]	EPA 200.7	12/29/94	SD
	Vanadium, Dissolved	< 20	ug/l	(	EPA 200.7	12/29/94	SD
	Zinc, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
	Antimony Diss., GFAA	4.5	lug/l	1	EPA 204.2	12/23/94	NS
- }	Arsenic Diss., GFAA	< 2.0	ug/l		EPA 206.2	12/29/94	NS
- [	Lead, Diss., GFAA	< 1.0	ug/l		EPA 239.2	12/22/94	NS
4	Selenium Diss., GFAA	< 2.0	lug/l		EPA 270.2	12/27/94	NS
۲	Clium Diss., GFAA	21	ug/l		EPA 279.2	12/29/94	NS
ω,	cury, Cold Vapor	< 0.2	ug/l		EPA 245.2	12/22/94	SD
	TCL VOLATILE ORGANICS, GC/MS		1	1 1	EPA 8260	12/21/94	VS
	Acetone	ND	ug/l	50			
	Acrolein	ND	ug/l	50			
2	Acrolein Acrylonitrile Benzene	ND	ug/l	5.0			
	Bromobenzene	ND ND	ug/l	1.0			
	Bromochloromethane	ND	lug/l lug/l	1.0			
10073	Bromodichloromethane	ND	ug/t ug/t	1.0			
	Bromoform	ND	ug/l ug/l	1.0			
	Bromomethane	ND	ug/l	1.0			
- 1	2-Butanone	ND	ug/l	50			
- 1	n-Butylbenzene	ND	ug/l	1.0			
	Sec-Butylbenzene	ND	ug/l	1.0			
	tert-Butylbenzene	ND	ug/l	1.0			
	Carbon Disulfide	ND	ug/l	50	Į		
	Carbon Tetrachloride	ND	ug/(	1.0	ļ		
5	Chlorobenzene	ND	ug/l	1.0			
27.	Chloroform	ND	ug/l	1.0			
7	Chloromethane	ND	ug/l	1.0	ŀ		ļ
- }	2-Chlorotoluene	ND	ug/l	1.0			j
	4-Chlorotoluene	ND	ug/l	1.0			
	Dibromochloromethane	ND	ug/l	1.0	]		}
	1,2-Dibromoethane	ND	ug/l	1.0			
	Dibromomethane	ND	ug/l	1.0			1
1	1,2-Dibromo-3-chloropropane	ND (	ug/i	1.0	ļ		į
- 1					]		ı
				. 1			

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 10

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Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121984 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA3-12

DATE SAMPLED.....: 12/13/94
TIME SAMPLED.....: 18:30

DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

ANALYCE DEPENDED	ANALYTICAL	UNITS OF	DETECTION	ANALYTICAL	RUN	TECHN
ANALYSES PERFORMED	RESULT	MEASURE	LIMIT	METHOD	DATE	TECHN
1,2-Dichlorobenzene	ND	ug/l	1.0	(		
1,3-Dichlorobenzene	ND	ug/l	1.0		1	
1,4-Dichlorobenzene	ND	ug/l	1.0	ļ.	1	
Dichlorodifluoromethane	79	ug/l	1.0		1	
1,1-Dichloroethane	590	ug/l	1.0	1		
1,2-Dichloroethane	5.0	ug/l	1.0	[		
1,1-Dichloroethene	240	ug/l	1.0	1	1	
Cis-1,2 Dichloroethylene	160	ug/l	1.0	İ	j	
Trans-1,2 Dichloroethylene 1.2-Dichloroethene (total)	4.3	ug/l	1.0	)	1	
	160	ug/l	2.0		ĺ	
Dichloromethane	ND ND	ug/l	1.0	Ì	1	
1,2-Dichloropropane	ND ND	ug/l	1.0	1	1	
1,3-Dichloropropane	ND ND	ug/l	1.0		1	
2,2-Dichloropropane 1,1-Dichloropropene	ND ND	ug/l ug/l	1.0		ļ	
Cis-1,3-Dichloropropene	ND ND	ug/t ug/t	1.0	[	1	
Fans-1,3-Dichloropropene	ND	ug/l	1.0	(	1	
\	ND	ug/l	1.0		1	
achlorobutadiene	ND	ug/l	1.0	1	<b>\</b>	
2-Hexanone	ND	ug/l	50		1	
Isopropyltoluene	ND	ug/l	1.0 *			
4-Isopropyltoluene	ND	ug/l	1.0 *			
4-Methyl-2-Pentanone	ND	ug/l	50		İ	
Naphthalene	ND	ug/l	1.0		ĺ	
n-Propylbenzene	ND	ug/l	1.0			
-Styrene	ND	ug/l	1.0		[	
1,1,1,2-Tetrachloroethane	ND	ug/l	1.0		1	
1,1,2,2-Tetrachloroethane	ND	ug/l	1.0		ļ	
Tetrachloroethylene	2200	ug/l	1.0			
Toluene	ND	ug/l	1.0	•	ł	
1,2,3-Trichlorobenzene	ND	ug/l	1.0		1	
1,2,4-Trichlorobenzene	ND	ug/l	1.0		1	
1,1,1-Trichloroethane	5500	ug/l	1.0		1	
1,1,2-Trichloroethane	ND	ug/l	1.0		1	
Irichloroethene	240	ug/l	1.0			
richlorofluoromethane	ND	ug/l	1.0		1	
,2,3-Trichloropropane	ND	ug/l	1.0		į	
1,2,4-Trimethylbenzene	ND	ug/l	1.0		!	
1,3,5-Trimethlybenzene	ND	ug/l	1.0		!	
Vinyl Acetate Vinyl Chloride	ND	ug/l	50			
,,	1.7	ug/l	1.0		<b> </b>	
MP-Xylene	ND	ug/l	2.0		1	
O-Xylene	ND	ug/t	1.0			
Xylenes, Total	ND	ug/l	3.0			

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated Library search per section 7.6.1.2, EPA Method 8260

PAGE 11

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_

	LAB	ORATORY CLIENT: CLEAN						
REPORT NUMBER: 37188 SAMPLE ID: 4121984 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA3-12								
DATE SAMPLED: 12/13/94 TIME SAMPLED: 18:30					DATE RECEIVED: 12/16/94 TIME RECEIVED: 14:20			
ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN		
Chloroethane	1.3	ug/l	1.0					
ND= None Detected at or above the ** Tentative identification and qua	limit reported antitation based on a co	omputer generated	d library search pe	r section 7.6.1.	2, EPA Metho	d 8260		

the results above apply only to the sample balyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of LEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121998 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA4-12

DATE SAMPLED.....: 12/15/94 TIME SAMPLED.....: 13:00

DATE RECEIVED ....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION	ANALYTICAL METHOD	RUN DATE	TECHN
			T			
'AL METALS		1				
lluminum, Dissolved	< 140	ug/l		EPA 200.7	12/28/94	SD
Barium, Dissolved	320	ug/l		EPA 200.7	12/22/94	SD
eryllium, Dissolved	< 10	ug/l		EPA 200.7	12/28/94	SD
admium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
alcium, Dissolved	150	mg/l		EPA 200.7	12/29/94	SD
thromium, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
obalt, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
ilver, Dissolved	< 30	ug/l		EPA 200.7	12/22/94	SD
opper, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
odium, Dissolved	85	lmg/l	Į.	EPA 200.7	12/29/94	SD
anadium, Dissolved	29	ug/l	1	EPA 200.7	12/29/94	SD
inc, Dissolved	< 20	ug/l	}	EPA 200.7	12/22/94	SD
ntimony Diss., GFAA	< 3.0	ug/l	,	EPA 204.2	12/23/94	NS
rsenic Diss., GFAA	< 2.0	lug/l		EPA 206.2	12/29/94	NS
ead, Diss., GFAA	< 1.0	ug/t		EPA 239.2	12/22/94	NS
elenium Diss., GFAA	< 2.0	ug/l	1	EPA 270.2	12/27/94	NS
\lium Diss., GFAA	17	ug/t	1	EPA 279.2	12/29/94	NS
ury, Cold Vapor	< 0.2	ug/t	1	EPA 245.2	12/22/94	SD
CL VOLATILE ORGANICS, GC/MS	"-	-3, ·		EPA 8260	12/21/94	VS
cetone	ND	ug/l	50	LIX OLOG	1 12/21//4	
crolein	ND	ug/l	50			
crylonitrile	ND ND	ug/l	5.0		t.	
enzene	ND	ug/l	1.0		i	
romobenzene	ND	ug/l	1.0			
romochloromethane	ND	) -	1		1	
romodichloromethane	ND ND	ug/l	1.0		ŧ	
		ug/l	1.0		1	
romoform romomethane	ND	ug/l	1.0		1	
	ND	ug/l	1.0	į.		
Butanone	ND	ug/l	50	i	1	
Butylbenzene	ND NO	ug/l	1.0	ĺ	ì	
ec-Butylbenzene	ND ND	ug/l	1.0	1	1	
ert-Butylbenzene	ND ND	ug/t	1.0	1		
arbon Disulfide	ND ND	lug/(	50	1	]	
arbon Tetrachloride	ND	lug/l	1.0	}	}	
llorobenzene	ND ND	ug/l	1.0	1	[	
loroform	ND ND	ug/l	1.0			
Chlosethane	ND	lug/l	1.0	}		
Chlorotoluene	ND	ug/l	1.0	(	1	
Chlorotoluene	ND	ug/l	1.0		ĺ	
bromochloromethane	ND	ug/l	1.0	1	1	
2-Dibromoethane	ND	lug/l	1.0		,	
bromomethane 2-Dibromo-3-chloropropane	ND	ug/l	1.0			
	ND ND	ug/l	1.0			

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 48

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121998 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA4-12

DATE SAMPLED.....: 12/15/94 TIME SAMPLED.....: 13:00

DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL	UNITS OF MEASURE	DETECTION	ANALYTICAL METHOD	RUN	TECHA
				l		
,2-Dichlorobenzene	ND	ug/l	1.0			
,3-Dichlorobenzene	סא (	ug/l	1.0	)	1	
,4-Dichlorobenzene	ND	ug/l	1.0			
ichlorodifluoromethane	ND	ug/l	1.0			
,1-Dichloroethane	ND	ug/l	1.0	ļ		
,2-Dichloroethane	ND	ug/l	1.0			
,1-Dichloroethene	42	ug/l	1.0		1	
is-1,2 Dichloroethylene	30000	ug/l	1.0		1	
rans-1,2 Dichloroethylene	j 110	ug/l	1.0			
,2-Dichloroethene (total)	30000	ug/l	2.0			
ichloromethane	ND	ug/l	1.0			
,2-Dichloropropane	ND ND	ug/l	1.0		1	
,3-Dichloropropane	ND	ug/l	1.0		1	
,2-Dichloropropane	ND	ug/l	1.0		1	
,1-Dichloropropene	ND	ug/l	1.0		1	
is-1,3-Dichloropropene	ND	ug/t	1.0		1	
rans-1,3-Dichloropropene	ND	ug/l	1.0		1	
, `\lbenzene	ND	ug/l	1.0			
chlorobutadiene	ND	ug/l	1.0		1	
Hexanone	ND	ug/l	50		1	
sopropyltoluene	ND	ug/l	1.0 *			
-Isopropyltoluene	ND	ug/l	1.0 *		Į	
-Methyl-2-Pentanone	ND	ug/l	50		!	
aphthalene	ND	ug/l	1.0		}	
-Propylbenzene	ND ND	ug/l	1.0			
tyrene	ND	ug/l	1.0		1	
1,1,2-Tetrachloroethane	ND	ug/l	1.0		l	
1,2,2-Tetrachloroethane	ND	ug/l	1.0			
etrachloroethylene	15	ug/l	1.0		1	
luene	ND	ug/l	1.0			
2,3-Trichlorobenzene	ND	ug/l	1.0		1	
2,4-Trichlorobenzene	ND	ug/l	1.0		ł .	
1,1-Trichloroethane	28	ug/l	1.0		I	
1,2-Trichloroethane	ND	ug/l	1.0		1	
ichloroethene	76000	ug/l	1.0			
ichlorofluoromethane	ND	ug/l	1.0			
2,3-Trichloropropane	ND	ug/l	1.0		Į.	
2,4-Trimethylbenzene	ND	ug/l	1.0		<b>!</b>	
3,5-Trimethlybenzene	ND	ug/l	1.0		Ţ.	
nyl Acetate	ND	ug/l	50		[	
nyl Chloride	1100	ug/l	1.0		j	
-Xylene	ND	ug/l	2.0		[	
Xylene	ND	ug/l	1.0			
lenes, Total	ND	ug/l	3.0		ļ	
	1	1	i l		i	

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 49

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_

	L A B	ORATORY   CLIENT: CLEAN				
REPORT NUMBER: 37188 SAMPLE		ION: ACUSTAR-DAY				
DATE SAMPLED: 12/15/94 TIME SAMPLED: 13:00				TE RECEIVED		
ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	ND	ug/l	1.0			
<u>T</u>						
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PAGE 50

results above apply only to the sample 2nd. This report may only be reproued in full, and may only be submitted to third party with the written permission of \*\*EAN TECH\*\*

Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121987 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA5-12

DATE SAMPLED.....: 12/14/94

TIME SAMPLED.....: 10:30

DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION	ANALYTICAL METHOD	RUN DATE	TECHN
						. 1911 8 44 91
AL METALS	(		1	-	Į.	
luminum, Dissolved	1 < 140	lug/l		EPA 200.7	12/29/94	SD
arium, Dissolved	290	ug/l	}	EPA 200.7	12/22/94	SD
eryllium, Dissolved	< 10	ug/l	1	EPA 200.7	12/29/94	SD
admium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
alcium, Dissolved	140	mg/l	1	EPA 200.7	12/29/94	SD
hromium, Dissolved	< 20	ug/l	ľ	EPA 200.7	12/22/94	SD
obalt, Dissolved	< 20	ug/l	ł	EPA 200.7	12/29/94	SD
ilver, Dissolved	50	1 -		EPA 200.7	12/22/94	SD
		lug/l	1	EPA 200.7		SD
opper, Dissolved	< 20	lug/l			12/22/94	
odium, Dissolved	100	mg/l	[	EPA 200.7	12/29/94	SD
anadium, Dissolved	< 20	lug/l	1	EPA 200.7	12/29/94	SD
inc, Dissolved	< 20	ug/l	ļ	EPA 200.7	12/22/94	\$D
ntimony Diss., GFAA	4.9	ug/l		EPA 204.2	12/23/94	NS
senic Diss., GFAA	< 2.0	ug/l	į	EPA 206.2	12/29/94	NS
ead, Diss., GFAA	< 1.0	lug/l		EPA 239.2	12/22/94	NS
eLenium Diss., GFAA	< 2.0	ug/l		EPA 270.2	12/27/94	NS
∖\ium Diss., GFAA	<b>11</b>	ug/i	Ì	EPA 279.2	12/29/94	NS
ury, Cold Vapor	< 0.2	ug/l		EPA 245.2	12/22/94	SD
L'VOLATILE ORGANICS, GC/MS	<b>i</b>	1	ļ.	EPA 8260	12/21/94	٧S
cetone	ND	ug/l	50			
crolein	ND ND	ug/l	50	1		
crylonitrile	ND	ug/l	5.0	ĺ		
enzene	ND	ug/l	1.0	<u> </u>	(	
romobenzene	ND	ug/l	1.0			
omochloromethane	ND	ug/l	1.0		Į.	
omodichloromethane	ND	ug/l	1.0		l	
omoform	ND	ug/l	1.0			
omomethane	ND	ug/l	1.0		]	
Butanone	ND ND	ug/l	50	J	J	
	ND		1.0	}	]	
Butylbenzene	1	ug/l		1	1	
c-Butylbenzene	ND	ug/l	1.0	1	Ì	
rt-Butylbenzene	ND	ug/l	1.0	1		
rbon Disulfide	ND	ug/l	50	1	)	
rbon Tetrachloride	ND	ug/l	1.0	1		
lorobenzene	ND	ug/l	1.0	}	}	
loroform	ND	ug/l	1.0	ŀ		
loromethane	ND	ug/l	1.0	}		
Chlorotoluene	ND	ug/l	1.0	}		
Chlorotoluene	ND	ug/l	1.0	}		
bromochloromethane	ND	ug/l	1.0			
2-Dibromoethane	ND	ug/l	1.0			
bromomethane	l ND	ug/l	1.0	I	l	
2-Dibromo-3-chloropropane	1	- <i>0</i> / `	,,,,		•	

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 19

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CLIENT: CLEAN TECH

REPORT NUMBER: 37188

SAMPLE ID: 4121987

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA5-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED....: 10:30 DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
			1		1	
1,2-Dichlorobenzene	ND	ug/i	1.0			
1,3-Dichlorobenzene	) ND	ug/l	1.0	]		
1,4-Dichlorobenzene	ND	ug/l	1.0			
Dichlorodifluoromethane	1.5	ug/l	1.0	İ	j	
1,1-Dichloroethane	74	ug/l	1.0	[	!	
1,2-Dichloroethane	5.0	ug/l	1.0	}	1	
1,1-Dichloroethene	260	ug/l	1.0	İ		
Cis-1,2 Dichloroethylene	59	ug/l	1.0	<b>S</b>	ł	
Trans-1,2 Dichloroethylene 1.2-Dichloroethene (total)	2.1	ug/l	1.0	i		
1,2-Dichloroethene (total)	61	ug/l	2.0	ļ	ľ	
Dichloromethane	ND	ug/l	1.0			
1,2-Dichloropropane	ND	ug/t	1.0	1	1	
1,3-Dichloropropane	ND	ug/l	1.0			
2,2-Dichloropropane	ND	ug/l	1.0	İ		
1,1-Dichloropropene	ND	ug/l	1.0		1	
Cis-1,3-Dichloropropene	ND	ug/t	1.0		!	
ans-1,3-Dichtoropropene	ND	ug/l	1.0		l	
Vlbenzene	ND	ug/l	1.0		1	
kachlorobutadiene	ND	ug/l	1.0		J	
2-Hexanone	ND	lug/l	50		]	
Isopropyltoluene	ND ND	ug/l	1.0 *		ì	
	ND	lug/l	1.0 *			
4-Isopropyltoluene 4-Methyl-2-Pentanone	ND ND	ug/l	50		Ì	
Naphthalene	ND	ug/t	1.0			
n-Propylbenzene	ND	ug/l	1.0		1	
Styrene	ND	lug/l	1.0			
1,1,1,2-Tetrachloroethane	ND	1 -	1.0			
1,1,2,2-Tetrachloroethane	ND	lug/l	11.0			
Tetrachioroethylene	240	lug/l	1.0			
Toluene	ND	ug/l	I I	'		
	ND ND	ug/l	1.0			
1,2,3-Trichlorobenzene	1	lug/l	1.0			
1,2,4-Trichlorobenzene	ND 2600	ug/l	1.0	Í		
11,1,1-Trichloroethane		ug/l	1.0	İ		
1,1,2-Trichloroethane	2.1	ug/l	1.0	1		
Trichloroethene	1100	ug/l	1.0	i		
. The state of the state	2.1	ug/l	1.0	{		
11,2,3-Trichloropropane	ND	ug/l	1.0			
1,2,4-Trimethylbenzene	ND	ug/l	1.0			
1,3,5-Trimethlybenzene	ND	ug/l	1.0	ſ		
Tvinyl Acetate Vvinyl Chloride TMP-XVIene	ND ND	ug/l	50			
TVINYL UNIONIGE	ND	ug/l	1.0			
1,	ND	ug/l	2.0			
0-Xylene	ND	ug/l	1.0	1		
Xylenes, Total	ND	ug/l	3.0			
i		]	}	1		j
J	1	1	l I			

ND= None Detected at or above the limit reported

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 20

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Canton Analytical Laboratory, Inc.

QA/QC 3

IJ	er.						
		LABO	ORATORY R CLIENT: CLEAN				
	REPORT NUMBER: 37188 SAMPLE ID: 412	1987 DESCRIPTI	ON: ACUSTAR-DAYTO	ON 94-1-MWA5-12			
	DATE SAMPLED: 12/14/94 TIME SAMPLED: 10:30			DA'	TE RECEIVED:	: 12/16/94 : 14:20	
	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	ND	ug/l	1.0			
1							
					ľ		
1							
1							
1						 	
Ĭ	ND= None Detected at or above the limit Tentative identification and quantitati	reported ion based on a co	mputer generated	library search per	r section 7.6.1.2	PA Meth	od 8260
auc	e results above apply only to the sample styred. This report may only be repro- ted in full, and may only be submitted to		PAGE 21	ahanata - T		Ţ.	? ```
	hird party with the written permission of EAN TECH	Canton .	anaiyiicai La	aboratory, Inc	. QA/G	IC	<u>.'</u>

REPORT NUMBER: 37188 SAMPLE ID: 4121988 DESCRIPTION: ACUSTAR-DAYTON 94-2-MWA5-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED....: 10:35 DATE RECEIVED ....: 12/16/94

TIME RECEIVED....: 14:20

			00 <b>0</b> 000000000000000000000000000000000			an an Tara Darin Ar	V
L. I		ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
	ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
şî.						64 CC	
	TAL METALS		1	(		1	
i.	TAL METALS	/0	1	į	ED 200 7	12/20/0/	CD
į	Aluminum, Dissolved	< 140   310	ug/l	į l	EPA 200.7	12/29/94	SD
	Barium, Dissolved	Į.	ug/l	1	EPA 200.7 EPA 200.7	12/22/94	SD
	Beryllium, Dissolved Cadmium, Dissolved	< 10   < 10	lug/l lug/l	1	EPA 200.7	12/27/94	SD SD
	Calcium, Dissolved	140	ug/ t   mg/ t	l i	EPA 200.7	12/22/94	SD
	Chromium, Dissolved Chromium, Dissolved	< 20	ug/t	[	EPA 200.7	12/22/94	SD
j	Cobalt, Dissolved	< 20	lug/l	]	EPA 200.7	12/29/94	SD
	Silver, Dissolved	34	ug/l	1	EPA 200.7	12/22/94	SD
	Copper, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
~{	Sodium, Dissolved	110	mg/l	<b>,</b>	EPA 200.7	12/29/94	SD
ı	Vanadium, Dissolved	49	ug/l	(	EPA 200.7	12/29/94	SD
ز	Zinc, Dissolved	< 20	ug/l	[ ]	EPA 200.7	12/22/94	SD
	Antimony Diss., GFAA	3.3	ug/l	j	EPA 204.2	12/23/94	NS
٦	Arsenic Diss., GFAA	< 2.0	ug/l	}	EPA 206.2	12/29/94	NS
Į	Lead, Diss., GFAA	< 1.0	ug/l	[	EPA 239.2	12/22/94	NS
H	Selenium Diss., GFAA	< 2.0	ug/l		EPA 270.2	12/27/94	NS
ľ	\lium Diss., GFAA	13	ug/l	]	EPA 279.2	12/29/94	NS
L.	Lury, Cold Vapor	< 0.2	ug/l		EPA 245.2	12/22/94	SD
ľ	74/			1		<b>}</b>	
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П	}					[	
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PAGE 22

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Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121993 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA6-12

DATE SAMPLED.....: 12/14/94 TIME SAMPLED.....: 17:30

DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL	UNITS OF	DETECTION	ANALYTICAL	RUN	TECUN
ANALYSES PERFORMED	RESULT	MEASURE	LIMIT	METHOD	DATE	TECHN
					1	
TAL METALS	<b>\</b>	}	}	}		
Aluminum, Dissolved	< 140	ug/l		EPA 200.7	12/29/94	SD
Barium, Dissolved	150	ug/l		EPA 200.7	12/22/94	SD
Beryllium, Dissolved	< 10	ug/l	1	EPA 200.7	12/29/94	SD
Cadmium, Dissolved	< 10	ug/l	}	EPA 200.7	12/22/94	SD
Calcium, Dissolved	140	mg/t		EPA 200.7	12/29/94	SD
Chromium, Dissolved	< 20	ug/l	1	EPA 200.7	12/22/94	SD
☑ Cobalt, Dissolved	j 61	ug/l		EPA 200.7	12/29/94	SD
Silver, Dissolved	<b>  &lt; 30</b>	ug/l	ì	EPA 200.7	12/22/94	SD
Copper, Dissolved	< 20	ug/l	}	EPA 200.7	12/22/94	SD
Sodium, Dissolved	81	mg/l	t	EPA 200.7	12/27/94	SD
	< 20	lug/l	1	EPA 200.7	12/29/94	SD
Zinc, Dissolved	< 20	ug/l	1	EPA 200.7	12/22/94	SD
Antimony Diss., GFAA	4.3	ug/l		EPA 204.2	12/23/94	NS
Arsenic Diss., GFAA	< 2.0	ug/l	ľ	EPA 206.2	12/29/94	NS
Lead, Diss., GFAA	< 1.0	ug/l		EPA 239.2	12/22/94	NS
Selenium Diss., GFAA	< 2.0	ug/l	1	EPA 270.2	12/27/94	NS
\lium Diss., GFAA	8.4	ug/l		EPA 279.2	12/29/94	NS
Jury, Cold Vapor	< 0.2	ug/l	]	EPA 245.2	12/22/94	\$D
TICE VOLATILE ORGANICS, GC/MS		1	150	EPA 8260	12/21/94	VS
Acetone	ND	ug/l	50	1	]	
Acrolein	ND	ug/l	50	<b>{</b>		
Acrylonitrile	ND	ug/l	5.0	[		
Benzene	ND	ug/l	1.0			
Bromobenzene	ND	ug/l	1.0	i		
Bromochloromethane	ND	ug/l	1.0			
Bromodichloromethane	ND	ug/l	1.0			
3romoform	ND	ug/l	1.0			
Bromomethane	ND	ug/l	1.0			
12-Butanone	ND	ug/t	50		İ	
h-Butylbenzene	ND	ug/l	1.0			
Sec-Butylbenzene	ND	ug/l	1.0			
tert-Butylbenzene Carbon Disulfide	ND ND	ug/l	1.0		li	
Carbon Disulfide		ug/l	50			
Larbon retrachtoride Ehlorobenzene	ND ND	lug/l	1.0			
chloroform	ND	ug/l ug/l	1.0			
T Chloromethane	ND ND	ug/l	1.0			
2-Chlorotoluene	ND ND	lug/t	1.0			
- Chlorotoluene	ND ND	ug/l	1.0			
Dibromochloromethane	ND ND	ug/t	1.0			
.2-Dibromoethane	ND ND	ug/l	1.0		l	
Dibromomethane	ND ND	ug/l	1.0			
	ND ND	ug/l	1.0			
12 2 151 Ollo 3 Circor opi opane	עמ ו	Lag/ C	1.0	Į.		
1		1				
	1	1	1	i		

ND= None Detected at or above the limit reported

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 35

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Canton Analytical Laboratory, Inc.

QA/QC

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121993 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA6-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED....: 17:30

DATE RECEIVED....: 12/16/94

TIME RECEIVED ....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
	T T					
1,2-Dichlorobenzene	ND	ug/t	1.0	[	1	
1,3-Dichlorobenzene	ND ND	ug/l	1.0	l	1	
1,4-Dichlorobenzene	ND ND	ug/l	1.0	ļ	ŀ	
Dichlorodifluoromethane	ND ND	ug/t	1.0	}	1	
1,1-Dichloroethane	∫ ND	ug/l	1.0	ļ	}	
1,2-Dichloroethane	l ND	ug/l	1.0			
1,1-Dichloroethene	19	ug/l	1.0	ļ		
Cis-1,2 Dichloroethylene	73	ug/l	1.0	[	1	
Trans-1,2 Dichloroethylene	1.3	ug/l	1.0		1	
1,2-Dichloroethene (total)	74	ug/l	2.0	ĺ	Į.	
Dichloromethane	ND	ug/l	1.0		1	
1,2-Dichloropropane	ND ND	ug/l	1.0	ĺ	1	
1,3-Dichloropropane	ND	ug/l	1.0		1	
2,2-Dichloropropane	ND	ug/l	1.0			
,1-Dichloropropene	ND	ug/l	1.0		1	
is-1,3-Dichloropropene	) ND	ug/l	1.0		1	
cans-1,3-Dichloropropene	ND	ug/l	1.0			
Ylbenzene	ND	ug/l	1.0			
achlorobutadiene	ND	ug/l	1.0		1	
-Hexanone	ND	ug/l	50		1	
sopropyltoluene	ND.	ug/l	1.0 *		1	
- Isopropyltoluene	ND	ug/l	1.0 *		1	
-Methyl-2-Pentanone	ND ON	ug/l	50		1	
aphthalene	ND	ug/l	1.0		]	
-Propylbenzene	ND	ug/t	1.0		i	
tyrene	ND	·	1.0		1	
,1,1,2-Tetrachioroethane	ND ND	ug/l	1.0			
,1,2,2-Tetrachtoroethane	ND ND	lug/l	1		}	
	1	lug/l	1.0		İ	
etrachloroethylene oluene	1.9	ug/l	1.0			
	ND	ug/l	1.0			
,2,3-Trichlorobenzene ,2,4-Trichlorobenzene	ND	ug/l	1.0		ľ	
,2,4-Trichtorobenzene ,1,1-Trichloroethane	ND 640	ug/l	1.0		}	
,1,2-Trichloroethane		ug/l			1	
richloroethene	ND 2600	ug/L	1.0		l	
		ug/l	11.0		1	
cichlorofluoromethane	ND	ug/l	1.0		1	
,2,3-Trichloropropane	ND	ug/l	1.0		İ	
2,4-Trimethylbenzene	ND ND	ug/l	1.0		1	
,3,5-Trimethlybenzene inyl Acetate	ND	lug/l	1.0		1	
inyl Acetate inyl Chloride	ND	ug/l	50		1	
•	ND ND	lug/l	1.0		}	
2-Xylene	ND ND	ug/l	2.0		{	
-Xylene	ND ND	ug/l	1.0		f	
ylenes, Total	ND	ug/l	3.0		1	

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 36

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Canton Analytical Laboratory, Inc.

QA/QC\_

REPORT NUMBER: 37188

SAMPLE ID: 4121993

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWA6-12

DATE SAMPLED.....: 12/14/94 TIME SAMPLED.....: 17:30 DATE RECEIVED....: 12/16/94

TIME	RECEIVED:	14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	ND ND	ug/l	1.0			
·						
( <u></u> ( <del>)</del> )						
na			·			
j						
l						

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 37

Canton Analytical Laboratory, Inc.

QA/QC 3

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CLIENT: CLEAN TECH

REPORT NUMBER: 37188

SAMPLE ID: 4121981

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB1-12

DATE SAMPLED.....: 12/13/94 TIME SAMPLED..... 12:30

DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
AL METALS	}	1				
lluminum, Dissolved	< 140	ug/l	ì	EPA 200.7	12/29/94	SD
Sarium, Dissolved	110	ug/l	1	EPA 200.7	12/22/94	SD
Beryllium, Dissolved	< 10	ug/l	[	EPA 200.7	12/29/94	SD
admium, Dissolved	< 10	ug/l	į.	EPA 200.7	12/22/94	SD
alcium, Dissolved	150	mg/l	j i	EPA 200.7	12/29/94	SD
hromium, Dissolved	< 20	ug/l	ſ	EPA 200.7	12/22/94	SD
obalt, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
ilver, Dissolved	< 30	ug/l		EPA 200.7	12/22/94	SD
opper, Dissolved	<b>  &lt; 20</b>	ug/l		EPA 200.7	12/22/94	SD
odium, Dissolved	56	mg/l	}	EPA 200.7	12/29/94	SD
anadium, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
inc, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
ntimony Diss., GFAA	< 3.0	ug/l		EPA 204.2	12/23/94	NS
rsenic Diss., GFAA	2.4	ug/l	}	EPA 206.2	12/29/94	NS
ead, Diss., GFAA	< 1.0	ug/l		EPA 239.2	12/22/94	NS
elenium Diss., GFAA	< 2.0	ug/l		EPA 270.2	12/27/94	NS
"Nlium Diss., GFAA	13	ug/l		EPA 279.2	12/29/94	NS
ury, Cold Vapor	0.28	ug/l	1	EPA 245.2	12/22/94	SD
VOLATILE ORGANICS, GC/MS	1 5.25	ug/ t		EPA 8260	12/21/94	VS
cetone	ND	ug/l	50	EFA 0200	12/21/34	•3
crolein	ND	ug/t	50	1	}	
crylonitrile	ND ND	1	5.0	}		
enzene	ND ND	ug/l	1.0	Į.	į	
	1 ''	lug/l	1 * * *	1		
romobenzene	ND ND	ug/l	1.0	İ		
romochloromethane	ND	ug/l	1.0	}		
romodichloromethane	ND	lug/l	1.0	1		
romoform	ND	ug/l	1.0			
romomethane	ND	ug/l	1.0	)		
Butanone	ND	ug/l	50	}		
Butylbenzene	ND	ug/l	1.0			
ec-Butylbenzene	ND	ug/l	1.0			
ert-Butylbenzene	ND	ug/l	1.0	]		
rbon Disulfide	ND	ug/l	50	}		
rbon Tetrachloride	ND	ug/l	1.0			
Lorobenzene	ND	ug/l	1.0			
loroform	ND	ug/l	1.0			
Coromethane	ND	ug/l	1.0	1		
Chlorotoluene	ND	ug/l	1.0	į		
Chlorotoluene	ND	ug/l	1.0			
bromochloromethane	ND	ug/l	1.0	Ì		
2-Dibromoethane	ND	ug/l	1.0			
bromomethane	ND	ug/l	1.0			
2-Dibromo-3-chloropropane	l ND	ug/l	1.0			

results above apply only to the sample allyzed. This report may only be repro-ticed in full, and may only be submitted to third party with the written permission of "LEAN TECH

ND= None Detected at or above the limit reported

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 1

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121981 DESCRIPTI

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB1-12

DATE SAMPLED....: 12/13/94 TIME SAMPLED....: 12:30 DATE RECEIVED....: 12/16/94

TIME RECEIVED ....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
					1	
1,2-Dichlorobenzene 1,3-Dichlorobenzene 1,4-Dichlorobenzene Dichlorodifluoromethane 1,1-Dichloroethane 1,2-Dichloroethane 1,1-Dichloroethene Cis-1,2 Dichloroethylene Trans-1,2 Dichloroethylene 1,2-Dichloroethene (total) Dichloromethane 1,2-Dichloropropane	ND r>1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 2.0 1.0					
1,3-Dichloropropane 2,2-Dichloropropane 1,1-Dichloropropene Cis-1,3-Dichloropropene Trans-1,3-Dichloropropene	ND ND ND ND ND	ug/l ug/l ug/l ug/l ug/l	1.0   1.0   1.0   1.0   1.0			
ylbenzene achlorobutadiene mexanone	ND ND ND	ug/l ug/l ug/l	1.0 1.0 50			
Isopropyltoluene 4-Isopropyltoluene 4-Methyl-2-Pentanone Naphthalene n-Propylbenzene Styrene 1,1,1,2-Tetrachloroethane 1,1,2,2-Tetrachloroethane Tetrachloroethylene Toluene 1,2,3-Trichlorobenzene 1,2,4-Trichlorobenzene 1,1,1-Trichloroethane Trichloroethane Trichloroethene Trichlorofluoromethane 1,2,3-Trichloropropane 1,2,4-Trimethylbenzene 1,3,5-Trimethylbenzene 1,3,5-Trimethlybenzene Vinyl Acetate Vinyl Chloride MP-Xylene 0-Xylene Xylenes, Total	ND ND ND ND ND NO	ug/l ug/l ug/l ug/l ug/l ug/l ug/l ug/l	1.0 * 1.0 * 50 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.			

ND= None Detected at or above the limit reported

the results above apply only to the sample alyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of LEAN TECH.

PAGE 2

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<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

		ORATORY R CLIENT: CLEAN				
REPORT NUMBER: 37188 SAMPLE ID	D: 4121981 DESCRIPT	TION: ACUSTAR-DAYT	TON 94-1-MWB1-12			
DATE SAMPLED: 12/13/94 TIME SAMPLED: 12:30			DA TI	ATE RECEIVED	.: 12/16/94 .: 14:20	
ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TEC
hloroethane	ND	ug/l	1.0			
` \						
<i>I</i>						
			!			
			1			
			!			
			1			

PAGE 3

Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121992 DESCRIPTION: ACUSTAR-DAYTON 94-1-MW82-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED....: 16:45 DATE RECEIVED....: 12/16/94
TIME RECEIVED....: 14:20

<b>5</b>	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
٦				I		Ĭ	
1	TAL METALS				1	İ	
	Aluminum, Dissolved	< 140	ug/l		EPA 200.7	12/29/94	SD
	Barium, Dissolved	190	ug/l	ŀ	EPA 200.7	12/22/94	SD
	Beryllium, Dissolved	< 10	ug/l		EPA 200.7	12/29/94	SD
	Cadmium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
	Calcium, Dissolved	160	mg/l	1	EPA 200.7	12/29/94	SD
1	Chromium, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
	Cobalt, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
	Silver, Dissolved	< 30	ug/l	}	EPA 200.7	12/22/94	SD )
J	Copper, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD (
ı	Sodium, Dissolved	180	mg/l	]	EPA 200.7	12/27/94	SD
ا_	Vanadium, Dissolved	36	ug/l	}	EPA 200.7	12/29/94	SD
	Zinc, Dissolved	20	ug/l	1	EPA 200.7	12/22/94	SD
4	Antimony Diss., GFAA	5.2	ug/l	1	EPA 204.2	12/23/94	NS
1	Arsenic Diss., GFAA	< 2.0	ug/l	1	EPA 206.2	12/29/94	NS
1	Lead, Diss., GFAA	1.1   < 2.0	ug/l  ug/l	1	EPA 239.2 EPA 270.2	12/22/94 12/27/94	NS I
	"Selenium Diss., GFAA Xlium Diss., GFAA	42	lug/l	İ	EPA 270.2	12/29/94	NS I
	cury, Cold Vapor	< 0.2	ug/l	j	EPA 245.2	12/22/94	SD
ì	TCL VOLATILE ORGANICS, GC/MS	. 0.2	ug/ t		EPA 8260	12/21/94	vs
ł	Acetone	ND	lug/l	50	LFA GEOD	12/21//4	''
4	Acrolein	ND	ug/l	50		}	1
ł	Acrylonitrile	ND	ug/l	5.0			
4	Benzene	ND	ug/l	1.0			
١	Bromobenzene	ND	ug/l	1.0			ľ
	Bromochloromethane	ND	ug/l	1.0			ł
Ĭ	Bromodichloromethane	ND	ug/l	1.0			]
Ņ	Bromoform	ND	ug/l	1.0			T.
1	Bromomethane	ND	ug/l	1.0			1
_	2-Butanone	ND	ug/l	50			1
	n-Butylbenzene	ND	ug/l	1.0		1	l
	Sec-Butylbenzene	ND	ug/l	1.0			1
	tert-Butylbenzene	ND	ug/l	1.0			- [
	Carbon Disulfide	ND ND	ug/l	50			i
	Carbon Tetrachloride Chlorobenzene	ND ND	ug/l	1.0			1
	Chloroform	ND	ug/l	1.0			ŀ
ж.	Chloromethane	ND	ug/l ug/l	1.0			ł
•	2-Chlorotoluene	ND	ug/l	1.0	ł		į.
	4-Chlorotoluene	ND	ug/l	1.0	j		1
	Dibromochloromethane	ND	ug/l	1.0			Į
	1,2-Dibromoethane	ND	ug/l	1.0			ł
h	Dibromomethane	ND	ug/l	1.0			f
	1,2-Dibromo-3-chloropropane	ND	ug/l	1.0			1
l	· · ·		-	-			j
1		[					

ND= None Detected at or above the limit reported

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 32

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Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121992 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB2-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED....: 16:45

DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
					GDCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCC	
1,2-Dichlorobenzene	ND	ug/l	1.0	ł	ļ	
1,3-Dichlorobenzene	ND	ug/l	1.0		ŀ	
1,4-Dichlorobenzene	ND	ug/l	1.0		1	
Dichlorodifluoromethane	ND	ug/l	1.0		ĺ	
1,1-Dichloroethane	ND	ug/l	1.0		Ì	
1,2-Dichloroethane	( ND	ug/l	1.0		į.	
1,1-Dichloroethene	ND	ug/l	1.0			
Cis-1,2 Dichloroethylene	ND	ug/l	1.0		}	
Trans-1,2 Dichloroethylene	ND	ug/l	1.0		1	
1,2-Dichloroethene (total)	ND	∫ug/l	2.0		1	
Dichloromethane	( ND	ug/l	1.0		ļ	
1,2-Dichloropropane	ND	ug/l	1:0		Į	
1,3-Dichloropropane	ND	ug/i	1.0		}	
2,2-Dichloropropane	ND	ug/l	1.0		1	
1,1-Dichloropropene	ND	ug/l	] 1.0		]	
Cis-1,3-Dichloropropene	ND	ug/l	1.0			
Irans-1,3-Dichloropropene	ND	ug/l	1.0		!	
)( benzene	ND	ug/l	1.0		}	
dchlorobutadiene	ND	ug/l	1.0		}	
2-Hexanone	ND	ug/l	50		}	
Isopropyltoluene	ND	ug/l	1.0 *			
4-Isopropyltoluene	ND	ug/l	1.0 *			
4-Methyl-2-Pentanone	ND	ug/l	50			
Naph tha lene	ND	ug/l	1.0			
n-Propylbenzene	ND	ug/t	1.0			
Styrene	ND	ug/l	1.0			
1,1,1,2-Tetrachloroethane	ND	ug/l	1.0			
1,1,2,2-Tetrachloroethane	ND	ug/l	1.0			
Tetrachloroethylene	1.6	ug/l	1.0			
Toluene	ND	ug/t	1.0			
1,2,3-Trichlorobenzene	ND	ug/l	11.0			
1,2,4-Trichlorobenzene	ND	ug/l	1.0			
1,1,1-Trichloroethane	ND	ug/l	1.0		•	
1,1,2-Trichloroethane Trichloroethene	ND ND	ug/l	1.0			
rrichtoroethene Trichtorofluoromethane	ND ND	ug/t	1.0			
	ND ND	ug/l	1.0			
1,2,3-Trichloropropane 1,2,4-Trimethylbenzene		ug/l	1.0			
,2,4-Trimethytbenzene ,3,5-Trimethlybenzene	ND ND	ug/l	1.0	}		
inyl Acetate	ND	lug/l	1.0	Į.		
inyl Acetale inyl Chloride	ND ND	ug/L				
P-Xylene	ND	lug/l	1.0	\$		
)-Xylene	ND	lug/l	2.0	Ì		
ylenes, Total	ND ND	ug/l	1.0			
y concey rocat	ND	ug/l	3.0			
	•					

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 33

results above apply only to the sample wallyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of LEAN TECH

Canton Analytical Laboratory, Inc.

	er j														
( []															
[ ]	REPORT NUMBER: 37188 SAMPLE ID: 41219	992 DESCRIPTIO	N: ACUSTAR-DAYTON	94-1-MWB2-12											
	DATE SAMPLED: 12/14/94 TIME SAMPLED: 16:45			TE RECEIVED:											
	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN								
	Chloroethane	ND	ug/l	1.0											
H															
JF	Aero.														
			:												
٦															
١.															
` ]															
7	ND= None Detected at or above the limit re * Tentative identification and quantitation	ported on based on a com	puter generated l	library search pe	r section 7.6.1.2	, EPA Method	1 8260								

PAGE 34

Canton Analytical Laboratory, Inc.

REPORT NUMBER: 37188 SAMPLE ID: 4121996 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB3-12

DATE SAMPLED.....: 12/15/94 TIME SAMPLED.....: 11:40 DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
				1		
TAL METALS					•	
Aluminum, Dissolved	< 140	ug/l		EPA 200.7	12/27/94	SD
Barium, Dissolved	110	ug/l		EPA 200.7	12/22/94	SD
Beryllium, Dissolved	< 10	lug/l	}	EPA 200.7	12/27/94	SD
admium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
Calcium, Dissolved	170	mg/l	ì	EPA 200.7	12/29/94	SD
hromium, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
obalt. Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
ilver, Dissolved	< 30	ug/l		EPA 200.7	12/22/94	SD
opper, Dissolved	< 20	ug/l	1	EPA 200.7	12/22/94	SD
odium, Dissolved	120	mg/l		EPA 200.7	12/29/94	SD
anadium, Dissolved	42	ug/l		EPA 200.7	12/29/94	SD
inc, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
ntimony Diss., GFAA	5.1	ug/l		EPA 204.2	12/23/94	NS
rsenic Diss., GFAA	< 2.0	ug/l		EPA 206.2	12/29/94	NS
ead, Diss., GFAA	1.4	ug/l		EPA 239.2	12/22/94	NS
Lenium Diss., GFAA	< 2.0	ug/l		EPA 270.2	12/27/94	NS
\ium Diss., GFAA	23	ug/l	1	EPA 279.2	12/29/94	NS
ary, Cold Vapor	< 0.2	*	}	EPA 245.2	12/22/94	SD
L'VOLATILE ORGANICS, GC/MS	\ 0.2	ug/l		EPA 8260	12/21/94	VS
cetone	ND ND	1	50	EPA 0200	12/21/94	۸2
	1 "	lug/l			i	
crolein	ND	ug/l	50		I	
rylonitrile	ND	ug/l	5.0	1		
nzene	2.4	ug/l	1.0		į	
omobenzene	ND	ug/l	1.0	İ	i	
omochloromethane	ND	ug/l	1.0	l	Į	
omodichloromethane	ND	ug/l	1.0			
omoform	ND	ug/l	[ 1.0		1	
omomethane	ND	ug/l	1.0		1	
Butanone	ND	ug/l	50	į .		
Butylbenzene	ND ND	ug/l	1.0	i	1	
c-Butylbenzene	l ND	lug/t	1.0		ļ	
rt-Butylbenzene	ND	ug/l	1.0		ł	
rbon Disulfide	ND	ug/l	50	ĺ	ļ	
rbon Tetrachloride	ND	ug/l	1.0			
lorobenzene	ND	ug/t	1.0		<b> </b>	
loroform	1.9	ug/l	1.0			
loromethane	ND	ug/l	1.0			
Chlorotoluene	ND	ug/l	1.0			
Chlorotoluene	ND	ug/l	1.0			
bromochloromethane	ND	ug/l	1.0			
2-Dibromoethane	ND	ug/l	1.0			
bromomethane	ND	ug/l	1.0			
2-Dibromo-3-chloropropane	ND	ug/l	1.0	,		
·· -· - <b>r</b> · - <b>r</b>	{					

ND= None Detected at or above the limit reported

results above apply only to the sample red. This report may only be reproduced in full, and may only be submitted to third party with the written permission of IEAN TECH

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 43

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121996 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB3-12

DATE SAMPLED.....: 12/15/94 TIME SAMPLED.....: 11:40

DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
2-Night anghanga-	ND		1.0			·····
,2-Dichlorobenzene	ND ND	ug/l	1.0		1	
,3-Dichlorobenzene	ND	ug/l	1.0	1	1	
,4-Dichlorobenzene	ND	ug/l	1.0			
ichlorodifluoromethane	ND .	ug/l	1.0		1	
,1-Dichloroethane	14	ug/l	1.0		i	
,2-Dichloroethane	11	ug/l	1.0			
,1-Dichloroethene	27	ug/l	1.0	1		
is-1,2 Dichloroethylene	1700	ug/l	1.0			
rans-1,2 Dichloroethylene	24	ug/l	1.0	ļ	]	
,2-Dichloroethene (total)	J 1700	ug/l	2.0		j	
ichloromethane	ND ND	ug/l	1.0			
,2-Dichloropropane	ND	ug/l	1.0			
,3-Dichloropropane	ND	ug/l	1.0		i	
,2-Dichloropropane	ND	ug/l	1.0			
,1-Dichloropropene	ND	ug/l	1.0			
is-1,3-Dichloropropene	ND	ug/l	1.0			
ans-1,3-Dichloropropene	ND	ug/l	1.0		İ	
<sub>te</sub> ∖benzene	Į ND	ug/l	1.0		1	
🚈 ,chlorobutadiene	ÍND	ug/l	1.0			
Hexanone	ND	ug/l	50		1	
sopropyltoluene	ND	ug/l	1.0 *			
Isopropyltoluene	ND ND	ug/l	1.0 *		1	
Methyl-2-Pentanone	ND ND	ug/l	50		ļ	
phthalene .	l ND	ug/l	1.0			
Propylbenzene	ND	ug/l	1.0			
yrene	ND	ug/l	1.0			
1,1,2-Tetrachloroethane	ND	ug/l	i.o			
1,2,2-Tetrachloroethane	ND	ug/l	1.0			
trachloroethylene	ND.	ug/l	1.0		j	
luene	ND	ug/l	1.0		[	
2,3-Trichlorobenzene	ND ND	ug/t	1.0		1	
2,4-Trichlorobenzene	ND	ug/t	11.0		I	
1,1-Trichloroethane	320	ug/l	11.0			
1,2-Trichloroethane	2.5	ug/t	1.0		1	
ichloroethene	9900	ug/l	11.0		1	
ichlorofluoromethane	3.8	ug/l	1.0		1	
2,3-Trichloropropane	ND ND	ug/l	1.0		1	
2,4-Trimethylbenzene	ND ND	lug/l	1.0		1	
3,5-Trimethlybenzene	ND ND		1.0			
nyl Acetate	ND ND	lug/l	50			
nyl Chloride	21	ug/l	1 1		!	
-Xylene	= ·	ug/l	1.0			
Kylene	ND ND	ug/l	2.0		}	
lenes, Total	ND NO	ug/l	1.0			
cenes, local	ND	lug/l	3.0		I	

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 44

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QA/QC\_

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121996 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB3-12

DATE SAMPLED....: 12/15/94 TIME SAMPLED....: 11:40 DATE RECEIVED....: 12/16/94

TIME RECEIVED ....: 14:20

			Ramores				
3	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
	Chloroethane	ND	ug/l	1.0			
			]				
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				;			

ND= None Detected at or above the Limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 45

ic results above apply only to the sample sublyzed. This report may only be reproduced in full, and may only be submitted to a third party with the written permission of LEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC ()

REPORT NUMBER: 37188 SAMPLE ID: 4121990 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB4-12

DATE SAMPLED.....: 12/14/94 TIME SAMPLED.....: 13:40

DATE RECEIVED....: 12/16/94 TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
					T	
AL METALS_		1				
turninum, Dissolved	< 140	ug/l	1	EPA 200.7	12/29/94	SD
arium, Dissolved	180	ug/l	1	EPA 200.7	12/22/94	SD
eryllium, Dissolved	< 10	ug/l		EPA 200.7	12/29/94	SD
admium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
alcium, Dissolved	100	mg/l		EPA 200.7	12/29/94	SD
hromium, Dissolved	< 20	ug/l	l	EPA 200.7	12/22/94	SD
obalt, Dissolved	48	ug/l		EPA 200.7	12/29/94	SD
ílver, Dissolved	33	ug/l		EPA 200.7	12/22/94	SD
opper, Dissolved	< 20	ug/l	ĺ	EPA 200.7	12/22/94	SD
odium, Dissolved	78	mg/l		EPA 200.7	12/29/94	SD
enadium, Dissolved	30	ug/l	1	EPA 200.7	12/29/94	SD
inc, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
ntimony Diss., GFAA	3.0	ug/l		EPA 204.2	12/23/94	NS
rsenic Diss., GFAA	< 2.0	ug/t		EPA 206.2	12/29/94	NS
ead, Diss., GFAA	< 1.0	ug/l	1	EPA 239.2	12/22/94	NS
્યું enium Diss., GFAA	< 2.0	ug/l		EPA 270.2	12/27/94	NS
lium Diss., GFAA	15	_ ·		EPA 279.2	12/29/94	NS
∠ury, Cold Vapor	₹ 0.2	ug/t		EPA 245.2	,, ,	
	1 0.2	ug/l			12/22/94	SD
CL VOLATILE ORGANICS, GC/MS	1 110			EPA 8260	12/21/94	VS
cetone	ND	ug/l	50			
crolein	ND	ug/l	50		}	
crylonitrile	ND	ug/l	5.0			
enzene	ND	ug/l	1.0			
omobenzene	ND	ug/l	1.0			
omochloromethane	ND	ug/l	1.0			
omodichloromethane	ND	lug/l	1.0			
omoform	ND	lug/l	1.0			
omomethane	ND	ug/l	1.0			
Butanone	ND	ug/l	50			
Butylbenzene	ND	ug/l	1.0			
c-Butylbenzene	ND	ug/l	1.0			
rt-Butylbenzene	ND	ug/l	1.0			
rbon Disulfide	ND	ug/l	50			
rbon Tetrachloride	ND	ug/l	1.0			
Lorobenzene	ND ND	ug/t lug/t	1.0			
loroform	ND ND	ug/l ug/l				
loromethane	ND		1.0	1		
Chlorotoluene	}	ug/L	1.0			
Chlorotoluene Chlorotoluene	ND ND	ug/l	1.0	i		
- ··	ND ND	ug/l	1.0	1		
bromochloromethane	ND	ug/l	1.0			
2-Dibromoethane	ND	ug/l	1.0			
promomethane	ND	ug/l	1.0			
2-Dibromo-3-chloropropane	ND	ug/l	1.0			

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 26

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Canton Analytical Laboratory, Inc.

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CLIENT: CLEAN TECH

REPORT NUMBER: 37188

SAMPLE ID: 4121990

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB4-12

DATE SAMPLED.....: 12/14/94 TIME SAMPLED.....: 13:40

DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

L		ANALYTICAL	UNITS OF	DETECTION	ANALYTICAL	RUN	
į	ANALYSES PERFORMED	RESULT	MEASURE	LIMIT	METHOD	DATE	TECHN
-							****
			Ì.	1		[	
Ш	1,2-Dichlorobenzene	ND	ug/l	1.0		ł	
1	1,3-Dichlorobenzene	ND	ug/l	1.0			
┙	1,4-Dichlorobenzene	ND	ug/l	1.0	ł	i	
	Dichlorodifluoromethane	ND	lug/l	1.0		ļ	
	1,1-Dichloroethane	ND	ug/l	1.0		i	
7	1,2-Dichloroethane	ND	ug/l	1.0	İ		
- [	1,1-Dichloroethene	ND	ug/l	1.0			
	Cis-1,2 Dichloroethylene	ND	ug/l	1.0		ļ	
16	Trans-1,2 Dichloroethylene	ND	ug/l	1.0			
U	1,2-Dichloroethene (total)	ND	ug/l	2.0		l	
Ì	Dichloromethane	ND	ug/l	1.0		<b>[</b>	
1	1,2-Dichloropropane	ND	ug/l	1.0		1	
	1,3-Dichloropropane	ND	ug/l	1.0			
4	2,2-Dichloropropane	ND	ug/l	1.0			
- [	1,1-Dichloropropene	ND	ug/l	1.0			ļ
- 1	Cis-1,3-Dichloropropene	ND	ug/l	1.0	1		i
:-4	Trans-1,3-Dichloropropene	ND	ug/l	1.0	}		1
į.	\\lbenzene	ND	ug/l	1.0			
1	/chlorobutadiene	ND	ug/l	1.0			ł
	2 nexanone	ND	ug/L	50			
	Isopropyltoluene	ND	ug/l	1.0 *			
	4-Isopropyltoluene	ND	lug/l	1.0 *			ľ
£ [	4-Methyl-2-Pentanone	ND	ug/l	50	i .		1
الهنقة	Naphthalene	ND	ug/l	1.0	]		
	n-Propylbenz <i>e</i> ne	ND	ug/l	1.0			İ
	Styrene	ND	ug/l	1.0			
	1,1,1,2-Tetrachloroethane	ND	ug/l	1.0			
	1,1,2,2-Tetrachloroethane	ND	ug/l	1.0			ļ
	Tetrachloroethylene	ND	ug/l	1.0			1
	Toluene	ND	ug/l	1.0			
	1,2,3-Trichlorobenzene	ND	ug/l	1.0	1		ŀ
	1,2,4-Trichlorobenzene	ND	ug/l	1.0			İ
	1,1,1-Trichloroethane	ND	ug/l	1.0	]		
- 1	1,1,2-Trichtoroethane	ND_	ug/l	1.0	i		i
	Trichloroethene	8.7	ug/l	1.0	)		
	richlorofluoromethane	ND	ug/l	1.0			ľ
	1,2,3-Trichloropropane	ND	ug/l	1.0			
	1,2,4-Trimethylbenzene	ND	ug/l	1.0			1
	,3,5-Trimethlybenzene		ug/l	1.0			ł
	/inyl Acetate	ND	ug/l	50			J
	rinyl Chloride	ND I	ug/l	1.0			j
	IP-Xylene	ND )	ug/l	2.0	Ì		1
	0-Xylene		ug/l	1.0			l
···1	ylenes, Total	ND	ug/l	3.0	]		1
- [				j			ļ
1					}		1

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 27

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L	A	В	0	R	A	T	0	R	Y	R	E	S	U	L	T	S	
				•	1 1	<i> </i> = 1	ut.	. ,	1 6	AM 1	c	٦u					

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121990

DESCRIPTION: ACUSTAR-DAYTON 94-1-MW84-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED....: 13:40 DATE RECEIVED ....: 12/16/94

TIME RECEIVED....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
Chloroethane	ND	ug/l	1.0			
]						
-						
<b>.</b> 1						
]					·	

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 28

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QA/QC 3

CLIENT: CLEAN TECH

REPORT NUMBER: 37188

SAMPLE ID: 4121989

DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB5-12

DATE SAMPLED.....: 12/14/94 TIME SAMPLED..... 12:45

DATE RECEIVED....: 12/16/94

TIME RECEIVED....: 14:20

	ANALYTICAL	UNITS OF	DETECTION	ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE	LIMIT	METHOD	DATE	TECHN
						<u> </u>
TAL METALS		1				
Aluminum, Dissolved	< 140	ug/l	i	EPA 200.7	12/29/94	SD
Barium, Dissolved	150	ug/l		EPA 200.7	12/22/94	SD
Beryllium, Dissolved	< 10	ug/l	Į.	EPA 200.7	12/29/94	SD
Cadmium, Dissolved	< 10	ug/l		EPA 200.7	12/22/94	SD
Calcium, Dissolved	150	mg/l		EPA 200.7	12/29/94	SD
Chromium, Dissolved	23	ug/l		EPA 200.7	12/22/94	SD
Cobalt, Dissolved	< 20	ug/l		EPA 200.7	12/29/94	SD
Silver, Dissolved	36	ug/l		EPA 200.7	12/22/94	SD
Copper, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD
Sodium, Dissolved	150	mg/l		EPA 200.7	12/29/94	SD
Vanadium, Dissolved	41	ug/l		EPA 200.7	12/29/94	SD
Zinc, Dissolved	< 20	ug/l		EPA 200.7	12/22/94	SD .
Antimony Diss., GFAA	4.7	ug/l		EPA 204.2	12/23/94	NS
Arsenic Diss., GFAA	< 2.0	ug/l		EPA 206.2	12/29/94	NS
Lead, Diss., GFAA	1.1	ug/l	1	EPA 239.2	12/22/94	NS
A Selenium Diss., GFAA	< 2.0	ug/l	1	EPA 270.2	12/27/94	NS
t \lium Diss., GFAA	26	ug/l		EPA 279.2	12/29/94	NS
Łury, Cold Vapor	< 0.2	ug/l		EPA 245.2	12/22/94	SD
TCL VOLATILE ORGANICS, GC/MS	ł			EPA 8260	12/21/94	vs
Acetone	ND	ug/l	50	l	1	
Acrolein	ND	ug/l	50		•	
Acrylonitrile	ND	ug/l	5.0			
Benzene	ND	ug/l	1.0			
Bromobenzene	ND	ug/l	1.0			
Bromochloromethane	ND	ug/l	1.0			
M Bromodichloromethane	ND	ug/l	1.0			
Bromoform	ND ND	ug/l	1.0			
Bromomethane	ND	lug/l	1.0			
2-Butanone	ND	ug/l	50			
n-Butylbenzene	ND	ug/l	1.0			
Sec-Butylbenzene	ND	ug/l	1.0			
tert-Butylbenzene	ND	ug/l	1.0			
Carbon Disulfide	ND	ug/l	50			
Carbon Tetrachloride	ND	ug/l	1.0			
Chloroform	ND	ug/l	1.0			
Chloroform	ND	ug/l	1.0			
Chloromethane	ND	ug/l	1.0	İ		
2-Chlorotoluene	ND	ug/l	1.0			
4-Chlorotoluene	ND	ug/l	1.0			
Dibromochloromethane	ND	ug/l	1.0			
1,2-Dibromoethane	ND	ug/t	1.0	]		
Dibromomethane	ND	ug/l	1.0	ļ		
1,2-Dibromo-3-chloropropane	ND	ug/l	1.0			
	1	]				
.]	1	1	1 1			

ND= None Detected at or above the limit reported

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 23

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QA/QC\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 37188 SAMPLE ID: 4121989 DESCRIPTION: ACUSTAR-DAYTON 94-1-MWB5-12

DATE SAMPLED....: 12/14/94 TIME SAMPLED.....: 12:45

DATE RECEIVED ....: 12/16/94

TIME RECEIVED ....: 14:20

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	DETECTION LIMIT	ANALYTICAL METHOD	RUN DATE	TECHN
	I	Ī				
1,2-Dichlorobenzene	ND	ug/l	1.0	ļ		
1,3-Dichlorobenzene	ND	ug/l	1.0		1	
1.4-Dichlorobenzene	ND	ug/l	1.0	Į		
Dichlorodifluoromethane	ÍND	ug/l	1.0		1	
1,1-Dichloroethane	ND	lug/l	1.0		Į.	
1.2-Dichloroethane	ND	ug/l	1.0		1	
1,1-Dichloroethene	ND	ug/l	1.0		1	
Cis-1,2 Dichloroethylene	ND	ug/l	1.0		1	
Trans-1 2 Bichloroethylane	ND	ug/l	1.0		[	
1,2-Dichloroethene (total)	ND	ug/l	2.0		1	
Dichloromethane	ND	ug/l	1.0		1	
1,2-Dichloropropane	) ND	ug/l	1.0		1	
1,3-Dichloropropane	ND	ug/l	1.0		1	
2,2-Dichloropropane	ND	ug/i	1.0		1	
1,1-Dichtoropropene	ND	ug/l	1.0		1	
Cis-1,3-Dichtoropropene	) ND	ug/l	1.0		Ì	
Trans-1,3-Dichloropropene	ND	ug/l	1.0		Ì	
ylbenzene	ND	ug/l	1.0		1	
ach lorobutadiene	ND	ug/l	1.0		ļ	
Ž-Hexanone	ND	ug/l	50		ľ	
Isopropyltoluene	ND	ug/l	1.0 *		1	
4-Isopropyltoluene	ND	ug/l	1.0 *		ľ	
4-Methyl-2-Pentanone	ND	ug/l	50			
Naph thalene	ND	ug/l	1.0		}	
n-Propylbenzene	ND	ug/l	1.0		!	
Styrene	ND	ug/l	1.0		)	
1,1,1,2-Tetrachloroethane	ND	ug/l	1.0		1	
1,1,2,2-Tetrachloroethane	ND	ug/l	1.0		i	
Tetrachloroethylene	ND	ug/l	1.0		ļ	
Toluene	ND	ug/l	1.0		1	
1,2,3-Trichlorobenzene	ND	ug/l	1.0		j	
1,2,4-Trichlorobenzene	ND	ug/l	1.0		1	
1,1,1-Trichloroethane	ND	ug/l	1.0		1	
1,1,2-Trichloroethane	ND	ug/l	1.0		1	
Trichloroethene	1.0	ug/l	1.0		1	
richlorofluoromethane	ND	ug/l	1.0		ì	
,2,3-Trichloropropane	ND	ug/l	1.0			
1,2,4-Trimethylbenzene	ND	ug/l	1.0		1	
1,3,5-Trimethlybenzene	ND	ug/l	1.0		1	
Vinyl Acetate	ND	ug/l	50			
Vinyl Chloride	ND	ug/l	1.0			
#P-Xylene	ND	ug/l	2.0			
0-Xylene	ND	ug/l	1.0			
Xylenes, Total	ND	ug/l	3.0			
		1	1			
i		1	1			

ND= None Detected at or above the limit reported

Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 24

results above apply only to the sample subject. This report may only be reprotuced in full, and may only be submitted to a third party with the written permission of LEAN TECH

Canton Analytical Laboratory, Inc.

QA/QC\_\_\_\_\_B

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022522 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWA6-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED.....: 14:10 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
					1	
1,2-Dibromo-3-chloropropane	< 1.0	ug/l			1	
1,2-Dichlorobenzene	< 1.0	ug/l	1			
1,3-Dichlorobenzene	₹ 1.0	ug/l				
1,4-Dichtorobenzene	< 1.0	ug/l		1	l	
Dichlorodifluoromethane	< 1.0	ug/l		}	1	
1,1-Dichloroethane	4.3	ug/l	1		l	
1,2-Dichloroethane	< 1.0	ug/l	1			
1,1-Dichloroethene	34	ug/l				
Cis-1,2 Dichloroethylene	110	ug/l			1	
Trans-1,2 Dichloroethylene	3.1	ug/l	1	}		
1,2-Dichloroethene (total)	110	ug/t			ļ	
Dichloromethane	< 1.0	ug/l	1	l	1	
1,2-Dichloropropane	< 1.0	ug/l				
1,3-Dichloropropane	< 1.0	ug/l				
2,2-Dichloropropane	< 1.0	ug/l			•	
1,1-Dichloropropene	< 1.0	ug/l			ļ	
Cis-1,3-Dichloropropene	< 1.0	ug/l		i		
Trans-1,3-Dichloropropene	< 1.0	ug/l				
Ethylbenzene	< 1.0	ug/l	1			
Hexachlorobutadiene	< 2.0	ug/l	1			
2-Hexanone	< 50	ug/l				
Isopropyltoluene	< 1.0	ug/l				
4-Isopropyltoluene *	< 1.0	ug/l	İ			
4-Methyl-2-Pentanone	< 50	ug/l	ļ			
Naphthalene	< 5.0	ug/l	ŀ			
n-Propylbenzene	< 1.0	ug/l				
Styrene	< 1.0	ug/l				
1,1,1,2-Tetrachloroethane	< 1.0	ug/l	1 1			
1,1,2,2-Tetrachloroethane	< 1.0	ug/l	l I			
Tetrachloroethylene	7.3	ug/L	1			
Toluene	< 1.0	ug/l		1		
1,2,3-Trichlorobenzene	< 1.0	ug/L	i	ľ		J
1,2,4-Trichlorobenzene	< 1.0	ug/L	1			
1,1,1-Trichloroethane	420 **	ug/t				
1,1,2-Trichloroethane Trichloroeth e	< 1.0	ug/l				
Trichloroflucromethane	1400 **	ug/l		į.		
1,2,3-Trichloropropane	< 1.0 < 1.0	lug/L	1			
1,2,4-Trimethylbenzene	< 1.0	ug/l ug/l	j i			
1,3,5-Trimethlybenzene	< 1.0	lug/l				
Vinyl Acetate	< 100	ug/l				
Vinyl Chloride	< 1.0	ug/t	ĺ			
HP-Xylene	₹ 2.0	ug/l		1		i
0-Xylene	< 1.0	ug/l	[	[		
•	` ''•	·		l		ĺ
	l l	1	l l	ŀ		ľ

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/8/95.

PAGE 51

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022522 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWA6-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED....: 14:10

DATE RECEIVED ....: 02/27/95

TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
ylenes, Total hloroethane	< 3.0 < 1.0	ug/l ug/l			
				į.	

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/8/95.

PAGE 52

KN QA/QC\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022505 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MW81-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 10:40 DATE RECEIVED ....: 02/27/95

TIME RECEIVED....: 10:40

i i	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	,	ANALYTICAL METHOD	RUN DATE	TECHN
71			T				***************************************
	DISSOLVED METALS						
1	TAL METALS		Í			_	
	Aluminum	< 0.14	mg/l		EPA 200.7	02/28/95	SD
	Barium	0.10	mg/l		EPA 200.7	02/28/95	SD
	Beryllium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
	Cadmium	< 0.01	mg/l	Į i	EPA 200.7	02/28/95	SD
	Calcium	110	mg/l		EPA 200.7	03/03/95	SD
_	Chromium	< 0.02	mg/l	1	EPA 200.7	02/28/95	SD
紬	Cobalt	0.034	mg/l	İ	EPA 200.7	02/28/95	SD
	Silver	< 0.03	mg/l		EPA 200.7	02/28/95	SD
	Copper	< 0.02	mg/l	1	EPA 200.7	02/28/95	SD
	Sodium	40	mg/l	]	EPA 200.7	03032/95	SD
	Vanadium	< 0.02	mg/l		EPA 200.7	02/28/95	SD
	Zinc	< 0.02	mg/t		EPA 200.7	02/28/95	SD
7	Antimony, GFAA	0.0044	mg/L		EPA 204.2	03/02/95	NS
ı	Arsenic, GFAA	0.0026	mg/l	j i	EPA 206.2	03/02/95	NS
-	Lead, GFAA	< 0.001	mg/L		EPA 239.2	03/01/95	NS
	lenium, GFAA	< 0.002	mg/l	1	EPA 270.2	03/01/95	NS
2	illium, GFAA	0.011	mg/l	Į į	EPA 279.2	03/02/95	NS
ື່ ໃ	raercury, Cold Vapor	< 0.0002	mg/l	[	EPA 245.2	02/28/95	SD
	TCL VOLATILE ORGANICS, GC/MS	_		}	EPA 8260	03/06/95	JB
7	Acetone	< 50	ug/l				
1	Acrolein	< 50	ug/l				
J	Acrylonitrile	< 5.0	ug/l				
ł	Benzene	< 1.0	ug/l				
	Bromobenzene	< 1.0	ug/l				
	Bromochloromethane	< 1.0	ug/l				
	Bromodichloromethane	< 1.0	ug/l				
_	Bromoform	< 1.0	ug/L	İ			
Į	Bromomethane	< 1.0	ug/l		i		
	2-Butanone	< 50	ug/l	İ			- 1
	n-Butylbenzene	< 1.0	ug/l	l	Į		Į.
	Sec-Butylbenzene	< 1.0	ug/l	[	i		ĺ
ŀ	tert-Butylbenzene	< 1.0	ug/l				I
20	Carbon Disulfide		ug/l				Į
	Carbon Tetrachloride Chlorobenzene		ug/l	ſ			1
81	Chloroform		ug/l				J
	Chloromethane		ug/l				j
	2-Chlorotoluene		ug/l				1
- 1			ug/l				l
			ug/l ug/l				ĺ
7				ļ	ļ		ļ
			ug/l				l
7		` '."	ug/l	1			ļ
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<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 4

Canton Analytical Laboratory, Inc.

ga/gc\_\_\_KW\_\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022505 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB1-2

DATE SAMPLED....: 02/20/95 TIME SAMPLED....: 10:40 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
						******************
,2-Dibromo-3-chloropropane	< 1.0	ug/l		}	1	
,2-Dichlorobenzene	< 1.0	ug/l				
,3-Dichlorobenzene	< 1.0	ug/l				
,4-Dichlorobenzene	< 1.0	ug/l			į.	
ichlorodifluoromethane	< 1.0	ug/l		Ì		
,1-Dichloroethane	< 1.0	ug/l			i	
,2-Dichloroethane	< 1.0	ug/l		1	ŀ	
,1-Dichloroethene	< 1.0	ug/l			1	
is-1,2 Dichloroethylene	< 1.0	ug/l				
rans-1,2 Dichloroethylene	< 1.0	ug/l				
,2-Dichloroethene (total)	< 2.0	ug/l		İ	i	
ichloromethane	< 1.0	ug/l			1	
,2-Dichloropropane	< 1.0	ug/l			i	
,3-Dichloropropane	< 1.0	ug/l				
,2-Dichloropropane	< 1.0	ug/l		1	ļ	
,1-Dichloropropene	< 1.0	ug/l				
is-1,3-Dichloropropene	< 1.0	ug/l		i	1	
cans-1,3-Dichloropropene	< 1.0	ug/l			1	
hylbenzene	i < 1.0	ug/l			J	
xachlorobutadiene	< 2.0	ug/l			1	
-Hexanone	< 50	ug/l		1		
sopropyltoluen <del>e</del>	< 1.0	ug/l			1	
-Isopropyltoluene *	< 1.0	ug/t				
-Methyl-2-Pentanone	< 50	ug/t			i	
aphthalene	< 5.0	ug/l		1	1	
Propylbenzene	< 1.0	ug/l				
tyrene	< 1.0	ug/t			ļ	
,1,1,2-Tetrachloroethane	< 1.0	ug/l		ì	i	
1,2,2-Tetrachloroethane	< 1.0	ug/l			1	
etrachloroethylene	< 1.0	ug/l				
oluene	< 1.0	ug/l			ł	
2,3-Trichlorobenzene	< 1.0	ug/l		1		
2,4-Trichlorobenzene	< 1.0	ug/t				
1,1-Trichloroethane	< 1.0	ug/l		ĺ		
1,2-Trichloroethane	< 1.0	ug/l		!		
ichloroethene	< 1.0	ug/l				
ichlorofluoromethane	< 1.0	ug/t				
2,3-Trichloropropane	< 1.0	ug/l				
2,4-Trimethylbenzene	< 1.0	ug/l				
3,5-Trimethlybenzene	< 1.0	ug/l	ı	l		
nyl Acetate	< 100	ug/l				
nyl Chloride	< 1.0	ug/t				
-Xylene	₹ 2.0	ug/l				
Xylene	< 1.0	ug/l				
· • · · · · ·	1	""				
	1	1				

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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Canton Analytical Laboratory, Inc.

GA/GC KN

DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB1-2

SAMPLE ID: 5022505

DATE SAMPLED.....: 02/20/95 TIME SAMPLED .....: 10:40

REPORT NUMBER: 38436

DATE RECEIVED ....: 02/27/95

TIME RECEIVED ....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE TECHN
Xylenes, Total Chloroethane	< 3.0 < 1.0	ug/l ug/l		

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

he sesults above apply only to the sen callyand. This report may only be rep above in full, and may only be submitted a third party with the written permission CLEAN TECH

PAGE 6

Canton Analytical Laboratory, Inc.

QA/QC\_\_

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022507 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB2-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 12:55

DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

	0.0000000000000000000000000000000000000	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	***************************************
	ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
<b>6</b> 7							***************************************
Н	DICCOLVED METALE	1	1	1		1	
L	DISSOLVED METALS TAL METALS	1	1				
	Aluminum	0.15	mg/l		EPA 200.7	02/28/95	SD
_	Barium	0.16	mg/l		EPA 200.7	02/28/95	SO
	Beryllium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
	Cadmium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
	Calcium	130	mg/l	1	EPA 200.7	03/03/95	SD
_	Chromium	< 0.02	mg/l		EPA 200.7	02/28/95	SD
Ŏ	Cobalt	0.032	mg/l		EPA 200.7	02/28/95	SD
	Silver	0.037	mg/t	ļ	EPA 200.7	02/28/95	SD
_	Copper	< 0.02	mg/l	]	EPA 200.7	02/28/95	SD
	Sodium	140	mg/l	]	EPA 200.7	03/03/95	SD
	Vanadium	< 0.02	mg/L	1	EPA 200.7	02/28/95	SD
	ZIIC	< 0.02	mg/L		EPA 200.7	02/28/95	SD
. 1	Antimony, GFAA	< 0.003	mg/L	[	EPA 204.2	03/02/95	NS
ı	Arsenic, GFAA Lead, GFAA	< 0.002 < 0.001	mg/l		EPA 206.2 EPA 239.2	03/02/95	NS NS
F-1	- Venium, GFAA	< 0.001	mg/t mg/t	}	EPA 270.2	03/01/95 03/01/95	NS I
	Illium, GFAA	0.029	mg/L	<u> </u>	EPA 270.2	03/01/95	NS NS
١.,	crcury, Cold Vapor	< 0.0002	mg/l	l	EPA 245.2	02/28/95	SD
ı	TCL VOLATILE ORGANICS, GC/MS			]	EPA 8260	03/06/95	JB
Ы	Acetone	< 50	ug/l	į į			l
	Acrolein	< 50	ug/l				ı
Ц	Acrylonitrile	< 5.0	ug/l	1			ŀ
- 1	Benzene	< 1.0	ug/l	1			ŀ
	Bromobenzene	< 1.0	ug/l		i		ſ
	Bromochloromethane Bromodichloromethane	< 1.0	ug/l				[
	Bromodichtoromethane Bromoform	< 1.0	ug/l	]	Į.		ļ
	Bromomethane	< 1.0 < 1.0	ug/l ug/l		ł		ļ
	2-Butanone	< 50	ug/t ug/t				
	n-Butylbenzene	< 1.0	ug/l				
	Sec-Butylbenzene	< 1.0	ug/l	]			1
	tert-Butylbenzene	< 1.0	ug/l				i
$oldsymbol{ol}}}}}}}}}}}}}}}}$	Carbon Disulfide	< 50	ug/l		ł		į
F	Carbon Tetrachloride	< 1.0	ug/l	<b> </b>	}		1
ŧ)	Chlorobenzene	< 1.0	ug/l	l	ļ		- 1
7	Chlonomoth	< 1.0	ug/l		1		ł
	Carbon Tetrachloride Chlorobenzene Chloroform Chloromethane 2-Chlorotoluene	< 1.0	ug/l		[		Į
	4-Chlorotoluene	< 1.0 < 1.0	ug/l		}		Į
	Dibromochloromethane		ug/l ug/l	Ì			ĺ
7	1,2-Dibromoethane	< 1.0	ug/t ug/t		}		İ
	Dibromomethane	1	ug/l	1	}		j
7	1				[		i
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\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 10

Canton Analytical Laboratory, Inc.

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022507 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB2-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 12:55

DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
1,2-Dibromo-3-chloropropane	< 1.0	ug/l				
1,2-Dichlorobenzene	< 1.0	ug/l				
1,3-Dichlorobenzene	< 1.0	ug/l			1	
1,4-Dichlorobenzene	< 1.0	ug/l		ľ		
Dichlorodifluoromethane	< 1.0	ug/l			1	
1,1-Dichloroethane	< 1.0	ug/l				
1,2-Dichloroethane	< 1.0	ug/l			ļ	
1,1-Dichloroethene	< 1.0	ug/l				
Cis-1,2 Dichloroethylene	< 1.0	ug/l			ł	
Trans-1,2 Dichloroethylene	< 1.0	ug/l	1		1	
1,2-Dichloroethene (total)	< 2.0	ug/l				
Dichloromethane	< 1.0	ug/l			ĺ	
1,2-Dichloropropane	< 1.0	ug/l		•		
1,3-Dichloropropane	< 1.0	ug/l	ł		1	
2,2-Dichloropropane	< 1.0	ug/l				
1,1-Dichloropropene	< 1.0	ug/l	!			
Cis-1,3-Dichloropropene	∤ < 1.0	ug/l				
rans-1,3-Dichloropropene	< 1.0	ug/l				
hylbenzene	< 1.0	ug/l				
nexachlorobutadiene	< 2.0	ug/l				
2-Hexanone	< 50	ug/l				
Isopropyltoluene	< 1.0	ug/l				
-Isopropyltoluene *	< 1.0	ug/l				
-Methyl-2-Pentanone	< 50	ug/l				
laph tha Lene	< 5.0	ug/l				
n-Propylbenzene	< 1.0	ug/l				
Styrene	< 1.0	ug/l				
,1,1,2-Tetrachloroethane	< 1.0	ug/l				
,1,2,2-Tetrachloroethane	< 1.0	ug/l				
etrachloroethylene	< 1.0	ug/t				
oluene	< 1.0	ug/l		1		
,2,3-Trichlorobenzene	< 1.0	ug/l	!			
,2,4-Trichlorobenzene	< 1.0	ug/l	j			
,1,1-Trichloroethane	< 1.0	ug/l				
,1,2-Trichloroethane	< 1.0	ug/t				
richloroethene	< 1.0	ug/t	i	1		
richlorofluoromethane	< 1.0	ug/l	ĺ	ļ		
,2,3-Trichloropropane	<b>  &lt; 1.0</b>	ug/l	į	1		
,2,4-Trimethylbenzene	< 1.0	ug/l		}		
,3,5-Trimethlybenzene	< 1.0	ug/l		1		
inyl Acetate	< 100	ug/l	}	i		
inyl Chloride	< 1.0	ug/t		)		
P-Xylene	< 2.0	ug/l		Ì		
-Xylene	< 1.0	ug/l				
	1	j	}			
		j j				

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 11

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Canton Analytical Laboratory, Inc.

QA/QC KV

CLIENT: CLEAN TECH

REPORT NUMBER: 38436

SAMPLE ID: 5022507

DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB2-2

DATE SAMPLED..... 02/20/95 TIME SAMPLED.....: 12:55

DATE RECEIVED....: 02/27/95

TIME RECEIVED....: 10:40

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Xylenes, Total Chloroethan <del>e</del>	< 3.0 < 1.0	ug/l ug/l			
]						
4			i			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 12

Canton Analytical Laboratory, Inc.

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REPORT NUMBER: 38436 SAMPLE ID: 5022508 DESCRIPTION: CHRYSLER ACUSTAR 95-2-MWB2-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 13:00

DATE RECEIVED....: 02/27/95

TIME RECEIVED....: 10:40

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
A A B B B C C C C C C S C S V Z A A A L L L	ISSOLVED METALS AL METALS Luminum arium eryllium admium alcium hromium obalt ilver opper odium anadium inc ntimony, GFAA rsenic, GFAA recir, GFAA illium, GFAA illium, GFAA ircury, Cold Vapor	0.21 0.16 < 0.01 < 0.02 0.029 0.036 < 0.02 < 0.02 < 0.003 < 0.002 < 0.003 < 0.002 < 0.003 < 0.002 < 0.0002 < 0.001 < 0.002 < 0.002 < 0.002	mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7 EPA 204.2 EPA 204.2 EPA 205.2 EPA 279.2 EPA 279.2 EPA 245.2	02/28/95 02/28/95 02/28/95 02/28/95 02/28/95 02/28/95 02/28/95 02/28/95 02/28/95 03/03/95 03/02/95 03/01/95 03/01/95 03/02/95 02/28/95	3 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

PAGE 13

Canton Analytical Laboratory, Inc.

DA/OC\_KN

he results above apply only to the sample subyand. This report may only be repro-duced in full, and may only be submitted to a third party with the written permission of CLEAN TECH

REPORT NUMBER: 38436

SAMPLE ID: 5022511

DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB3-2

DATE SAMPLED..... 02/20/95 TIME SAMPLED .....: 15:50

DATE RECEIVED ....: 02/27/95

TIME RECEIVED....: 10:40

AVALVEC DEDECTOR	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	TECUM
ANALYSES PERFORMED	RESULT	MEASURE	l	METHOD	DATE	TECHN
DISSOLVED METALS						
TAL METALS		l	İ	F04 300 7	03/30/05	
Aluminum	< 0.14	mg/l		EPA 200.7	02/28/95	SD
Barium	0.11	mg/l		EPA 200.7	02/28/95	SD
Beryllium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
Cadmium	< 0.01	mg/l	:	EPA 200.7	02/28/95	SD
Calcium	150	mg/L	ĺ	EPA 200.7	03/03/95	SD
Chromium	< 0.02	mg/t	1	EPA 200.7	02/28/95	SD
Cobalt	< 0.02	mg/l		EPA 200.7	02/28/95	SD
Silver	< 0.03	mg/l	İ	EPA 200.7	02/28/95	SD
Copper	< 0.02	mg/l		EPA 200.7	02/28/95	SD
Sodium	100	mg/L		EPA 200.7	03/03/95	SD
/anadium	< 0.02	mg/l	ŀ	EPA 200.7	02/28/95	SD
Zinc	< 0.02	mg/L		EPA 200.7	02/28/95	SD
Antimony, GFAA	< 0.003	mg/L	1	EPA 204.2	03/02/95	NS
Arsenic, GFAA	< 0.002	mg/l		EPA 206.2	03/02/95	NS
ead, GFAA	< 0.001	mg/l		EPA 239.2	03/01/95	NS
lenium, GFAA	< 0.002	mg/l		EPA 270.2	03/01/95	NS
allium, GFAA	0.015	mg/t		EPA 279.2	03/02/95	NS
ercury, Cold Vapor	< 0.0002	mg/l		EPA 245.2	02/28/95	SD
CL VOLATILE ORGANICS, GC/MS		1		EPA 8260	03/06/95	JB
cetone	< 50	ug/l		1		
crolein	< 50	ug/l				
crylonitrile	< 5.0	ug/L		ļ		
enzene	2.1	ug/l			ı	
romobenzene	< 1.0	ug/L				
romoch Loromethane -	< 1.0	ug/l		]		
romodichloromethane	< 1.0	ug/l		ĺ		
romoform	< 1.0	ug/l				
omomethane	< 1.0	ug/l				
Butanone	< 50	ug/l				
Butylbenzene	< 1.0	ug/l				
ec-Butylbenzene	< 1.0	ug/t				
ert-Butylbenzene	< 1.0	ug/l				
rbon Disulfide	< 50	ug/L	į	ļ		
rbon Tetrachloride	< 1.0	ug/l	i	j		
lorobenzene	< 1.0	ug/l				
loroform	4.8	ug/L				
loromethane	< 1.0	ug/l				
Chlorotoluene	< 1.0	ug/l		]		
Chlorotoluene	< 1.0	ug/l		1		
bromochloromethane	< 1.0	ug/t	į	ļ		
2-Dibromoethane	< 1.0	ug/l				
bromomethane	< 1.0	ug/l		İ		
	1	]	į	ļ		
	1	1 1		i		

<sup>\*</sup> Tentative identification and quantitation based on a computer generated Library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/7/95.

PAGE 20

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CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022511 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB3-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED .....: 15:50

DATE RECEIVED ....: 02/27/95

TIME RECEIVED ....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECH
						500000 001.00000000000000000000000000000
,2-Dibromo-3-chloropropane	< 1.0	ug/l		ł	ĺ	
,2-Dichlorobenzene	< 1.0	ug/l	Ì	Ì	Ì	
,3-Dichlorobenzene	< 1.0	ug/l			1	
,4-Dichlorobenzene	< 1.0	ug/L				
ichlorodifluoromethane	< 1.0	ug/l	ľ	į	ł	
,1-Dichloroethane	28	ug/l				
,2-Dichloroethane	42	ug/l			[	
1-Dichloroethene	28	ug/l		1	j	
is-1,2 Dichloroethylene	3500 **	ug/l			1	
ans-1,2 Dichloroethylene	28	ug/l				
2-Dichloroethene (total)	3500	ug/l			1	
chloromethane	< 1.0	ug/t	1		İ	
2-Dichloropropane	< 1.0	ug/l	1	1		
3-Dichloropropane	< 1.0	ug/l	1		1	
2-Dichloropropane	< 1.0	ug/l				
1-Dichloropropene	< 1.0	ug/l		1	1	
s-1,3-Dichloropropene	< 1.0	ug/l		1		
ans-1,3-Dichloropropene	< 1.0	ug/l	}			
ylbenzene xachlorobutadiene	< 1.0	ug/l				
	< 2.0	ug/l	Į.		l	
Hexanone	< 50	ug/l				
opropyltoluene Isopropyltoluene *	< 1.0	ug/l			•	
Methyl-2-Pentanone	< 1.0	ug/t	1	1	1	
phthalene	< 50	ug/l		!		
Propylbenzene	< 5.0	ug/l				
yrene	< 1.0 < 1.0	ug/l	İ			
1,1,2-Tetrachloroethane	< 1.0	ug/l				
1,2,2-Tetrachtoroethane	< 1.0	ug/L				
trachloroethylene	< 1.0	ug/l				
luene	< 1.0	ug/L				
2,3-Trichlorobenzene	< 1.0	ug/l ug/l			•	
2,4-Trichtorobenzene 2,4-Trichtorobenzene	< 1.0	-				
1,1-Trichloroethane	350 **	ug/l ug/l				
1,2-Trichtoroethane	3.9	ug/t ug/t	1			
ichloroethene	1100 **	ug/t				
ichlorofluoromethane	< 1.0	ug/l	1			
2,3-Trichloropropane	< 1.0	ug/l	]			
2,4-Trimethylbenzene	< 1.0	ug/t				
5,5-Trimethlybenzene	1.0	ug/l				
Nyl Acetate	< 100	ug/l				
nyl Chloride	26	ug/t	1			
Xylene	< 2.0	ug/l				
lylene	< 1.0	ug/l				
•	1	[ -4/ •	1			

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/7/95.

PAGE 21

Canton Analytical Laboratory, Inc.

QA/QC KV

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CLIENT: CLEAN TECH

REPORT NUMBER: 38436

SAMPLE ID: 5022511

DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB3-2

DATE SAMPLED....: 02/20/95

TIME SAMPLED.....: 15:50

DATE RECEIVED....: 02/27/95

TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Xylenes, Total Chloroethane	< 3.0 < 1.0	ug/l ug/l			
		·			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/7/95.

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Canton Analytical Laboratory, Inc.

GA/GC KV

CLIENT: CLEAN TECH

REPORT NUMBER: 38436

SAMPLE 10: 5022512

DESCRIPTION: CHRYSLER ACUSTAR 95-2-MWB3-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 15:55 DATE RECEIVED....: 02/27/95
TIME RECEIVED....: 10:40

)		ANALYTICAL	UNITS OF		ANALYTICAL	RUN	·····
	ANALYSES PERFORMED	RESULT	MEASURE	<u> </u>	METHOD	DATE	TECHN
1		I				**************************************	***************************************
l	TCL VOLATILE ORGANICS, GC/HS		1	i	EPA 8260	03/06/95	СВ
)	Acetone	< 50	ug/l		LFA 0200	03,00,73	00
	Acrolein	< 50	ug/l			1	
_	Acrylonitrile	₹ 5.0	นฐ/โ		Ι.	Į.	
ı	Benzene	2.4	ug/l	1		ì	
	Bromobenzene	< 1.0	ug/l				
_	Bromochloromethane	< 1.0	ug/l	1		ŀ	
	Bromodichloromethane	< 1.0	ug/l		1		
ď	Bromoform	₹ 1.0	ug/l			ĺ	
1	Bromomethane	< 1.0	ug/l			Į	
j	1 2-Butanone	< 50	ug/l				
	n-Butylbenzene	< 1.0	ug/l				
	Sec-Butylbenzene	< 1.0	ug/l		i		
	tert-Butylbenzene	< 1.0	ug/l	i		i	
ł	Carbon Disulfide	< 50	ug/l		1	ļ	
•	Carbon Tetrachloride	< 1.0	ug/l				
	Chlorobenzene	< 1.0	ug/l	1			ĺ
	Long to the state of the state	8.1	ug/l	1			1
	r)loromethane	< 1.0	ug/l	<b>\</b>	}	1	j
	,∡4Chlorotoluene	< 1.0	ug/l			}	i
	4-Chlorotoluene	< 1.0	ug/l				Į.
4	Dibromochloromethane	< 1.0	ug/l	i			ſ
1	1,2-Dibromoethane	< 1.0	ug/l	<b>f</b>			1
)	Dibromomethane	< 1.0	ug/l			ľ	- 1
	1,2-Dibromo-3-chloropropane	< 1.0	ug/l	1		1	i
	1,2-Dichlorobenzene	< 1.0	ug/l				i
1	1,3-Dichlorobenzene	< 1.0	ug/l				ł
	1,4-Dichlorobenzene	< 1.0	ug/l				ļ
•	Dichlorodifluoromethane	< 1.0	ug/l				I
	1,1-Dichloroethane	49	ug/l				- 1
Bĺ	1,2-Dichloroethane 1,1-Dichloroethene	39	ug/l				1
ı	Cis-1,2 Dichloroethylene	61 1500 **	ug/L	i			- 1
,	Trans-1,2 Dichloroethylene	48	ug/l	1			- 1
]	1,2-Dichloroethene (total)	1500 **	ug/l ug/l	]			}
٠l	Dichloromethane	< 1.0	ug/l	l			Ĭ
H	1,2-Dichloropropane	< 1.0	ug/l				
"	1,3-Dichloropropane	< 1.0	ug/l				ļ
-	2,2-Dichloropropane	< 1.0	ug/l		[		i
IJ	1,1-Dichloropropene	< 1.0	ug/l	İ	i		i
H	Cis-1,3-Dichloropropene	< 1.0	ug/l	i i			
H	Trans-1,3-Dichloropropene	< 1.0	ug/l				i
1	Ethylbenzene	< 1.0	ug/l				l
-	Hexachlorobutadiene	< 2.0	ug/l				İ
Н			-				į
١	į –			l l	i		

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/7/95.

PAGE 23

Canton Analytical Laboratory, Inc.

ga/gc KN

REPORT NUMBER: 38436 SAMPLE ID: 5022512 DESCRIPTION: CHRYSLER ACUSTAR 95-2-MWB3-2

DATE SAMPLED....: 02/20/95 TIME SAMPLED....: 15:55 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN	
	2-Hexanone Isopropyltoluene 4-Isopropyltoluene 4-Isopropyltoluene 84-Methyl-2-Pentanone Naphthalene n-Propylbenzene Styrene 1,1,2-Tetrachloroethane 1,1,2-Tetrachloroethane Tetrachloroethylene Toluene 1,2,3-Trichlorobenzene 1,1,1-Trichloroethane 1,1,2-Trichloroethane 1,1,2-Trichloroethane Trichloroethene Trichlorofluoromethane 1,2,3-Trichloropropane 1,2,4-Trimethylbenzene 1,3,5-Trimethylbenzene Vinyl Acetate Vinyl Chloride MP-Xylene 0-Xylene Xylene Xylenes, Total Chloroethane	< 50 < 1.0 < 1.0 < 50 < 5.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0 < 1.0	ug/l ug/l ug/l ug/l ug/l ug/l ug/l ug/l				2

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/7/95.

PAGE 24

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_KV

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022516 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB4-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED.....: 10:25

DATE RECEIVED ....: 02/27/95 TIME RECEIVED....: 10:40

	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
DISSOLVED METALS						
TAL METALS	l	ļ	1		ŀ	
Aluminum	< 0.14	mg/l	<b>,</b>	EPA 200.7	02/28/95	SD
Barium	0.18	mg/l		EPA 200.7	02/28/95	SD
Beryllium	< 0.01	mg/l	Í	EPA 200.7	02/28/95	SD
Cadmium	< 0.01	mg/t	1	EPA 200.7	02/28/95	SD
Calcium	96	mg/t	İ	EPA 200.7	03/03/95	SD
Chromium	< 0.02	mg/l		EPA 200.7	02/28/95	SD
Cobalt	< 0.02	mg/l		EPA 200.7	02/28/95	SD
Silver	< 0.03	mg/l	}	EPA 200.7	02/28/95	SD
Copper	< 0.02	mg/l		EPA 200.7	02/28/95	SD
Sodium	59	mg/t		EPA 200.7	03/03/95	SD
Vanadium	< 0.02	mg/t	1	EPA 200.7	02/28/95	SD
Zinc	< 0.02	mg/t		EPA 200.7	02/28/95	SD
Antimony, GFAA	< 0.003	mg/l		EPA 204.2	03/02/95	NS
Arsenic, GFAA	< 0.002	mg/t		EPA 206.2	03/02/95	NS
Lead, GFAA	< 0.001	mg/L		EPA 239.2	03/02/95	NS
Selenium, GFAA	< 0.002	mg/L	1	EPA 270.2	03/01/95	NS
allium, GFAA	0.013	mg/t	Į.	EPA 279.2	03/02/95	NS
rercury, Cold Vapor	< 0.002	mg/t		EPA 245.2	02/28/95	SD
TCL VOLATILE ORGANICS, GC/MS	1 0.0002	ING/ C		EPA 8260	03/07/95	JB
Acetone	< 50	ug/l		EPA 0200	03/07/93	75
Acrolein	< 50	ug/l				
Acrylonitrile	< 5.0	ug/l			1	
Benzene	< 1.0	ug/l				
Bromobenzene	< 1.0		1		1	
Bromochloromethane	< 1.0	ug/l ug/l				
Bromodichloromethane	< 1.0	1 -				
Bromoform	< 1.0	lug/l lug/l	<b>j</b>			
Bromomethane	< 1.0	ug/l				
2-Butanone	< 50	ug/t				
n-Butylbenzene	< 1.0	ug/l				
Sec-Butylbenzene	< 1.0	ug/l	• 1	1		
cert-Butylbenzene	< 1.0	ug/l				
Carbon Disulfide	< 50	ug/t		ı	ı	
Carbon Tetrachloride	1 < 1.0	ug/l				
hlorobenzene	< 1.0	ug/t				
hloroform	< 1.0	ug/t				
Chloromethane	< 1.0	ug/i	!			
-Chlorotoluene	₹ 1.0	ug/l				
-Chlorotoluene	< 1.0	ug/t	ľ	į		
ibromochloromethane	₹ 1.0	ug/t				
,2-Dibromoethane	< 1.0	ug/t		j		
ibromomethane	< 1.0					
	1	ug/ t		}		
. P. CHEMIS CHAIS		ug/l				

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 34

QA/QC\_

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REPORT NUMBER: 38436 SAMPLE ID: 5022516 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB4-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED.....: 10:25 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
						***************************************
,2-Dibromo-3-chloropropane	< 1.0	ug/l				
,2-Dichlorobenzene	< 1.0	ug/l				
,3-Dichlorobenzene	< 1.0	ug/l	i	]		
,4-Dichlorobenzene	< 1.0	ug/l			1	
ichlorodifluoromethan <del>e</del>	< 1.0	ug/l			]	
,1-Dichloroethane	( < 1.0	ug/l	[	Į	ł	
,2-Dichloroethane	< 1.0	ug/l			ļ	
,1-Dichloroethene	< 1.0	ug/l			Í	
is-1,2 Dichloroethylene	1.7	ug/l			ļ	
rans-1,2 Dichloroethylene	< 1.0	[ug∕l	ļ		ŀ	
2-Dichloroethene (total)	1.7	ug/l			ļ	
chloromethane	< 1.0	ug/l			ľ	
2-Dichloropropane	< 1.0	ug/l			i	
3-Dichloropropane	< 1.0	ug/l			ļ	
2-Dichloropropane	< 1.0	ug/l				
1-Dichloropropene	√ 1.0	ug/l		i		
s-1,3-Dichloropropene	< 1.0	ug/l				
ans-1,3-Dichloropropene	< 1.0	ug/l				
hylbenzene	< 1.0	ug/l				
xachlorobutadiene	< 2.0	ug/l				
Hexanone	< 50	ug/t				
opropyltoluene	< 1.0	ug/l				
Isopropyltoluene *	< 1.0	ug/l				
Methyl-2-Pentanone	< 50	ug/l	ì	i		
phthalene	< 5.0	ug/l	j			
Propylbenzene	< 1.0	ug/t				
yrene	l < 1.0	ug/l				
1,1,2-Tetrachloroethane	< 1.0	ug/l	ļ			
1,2,2-Tetrachloroethane	< 1.0	ug/l				
trachloroethylene	5.6	ug/l	1			
luene	< 1.0	ug/l	i			
2,3-Trichlorobenzene	< 1.0	ug/l	i			
2,4-Trichlorobenzene	< 1.0	ug/i	ļ	i		
1,1-Trichloroethane	1.3	ug/l				
1,2-Trichloroethane	< 1.0	ug/l				
ichloroethene	6.8	ug/l	ľ			
ichlorofluoromethane	< 1.0	ug/l	}	I		
2,3-Trichloropropane	< 1.0	ug/l	ļ	į		
2,4-Trimethylbenzene	< 1.0	ug/i	<b>F</b>	1		
5,5-Trimethlybenzene	< 1.0	ug/l	1			
nyl Acetate	< 100	ug/l		1		
nyl Chloride	< 1.0	ug/l	j			
	< 2.0	ug/l				
Xylene	1 , 5.0	109/1	1			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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Canton Analytical Laboratory, Inc.

QA/QC\_\_\_KV

REPORT NUMBER: 38436 SAMPLE ID: 5022516 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB4-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED....: 10:25

DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
Xylenes, Total Chloroethane	< 3.0 < 1.0	ug/l ug/l			

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PAGE 36

Canton Analytical Laboratory, Inc.

QA/QC KW

<sup>\*</sup> Tentative identification and quantitation based on a computer generated Library search per section 7.6.1.2, EPA Method 8260

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022509 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB5-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 14:05 DATE RECEIVED....: 02/27/95
TIME RECEIVED....: 10:40

	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	DISSOLVED METALS TAL METALS					
j.	Aluminum Barium Beryllium Cadmium	0.18 0.14 < 0.01 < 0.01	mg/l mg/l mg/l mg/l	EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7	02/28/95 02/28/95 02/28/95 02/28/95	SD SD SD SD
4	Calcium Chromium Cobalt Silver	140 < 0.02 0.029 0.036	mg/l mg/l mg/l mg/l	EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7	03/03/95 02/28/95 02/28/95 02/28/95	SD SD SD
	Copper Sodium Vanadium Zinc	< 0.02 130 < 0.02 < 0.02	mg/l mg/l mg/l mg/l	EPA 200.7 EPA 200.7 EPA 200.7 EPA 200.7	02/28/95 03/03/95 02/28/95 02/28/95	SD SD SD
7	Antimony, GFAA Arsenic, GFAA Lead, GFAA Lenium, GFAA	< 0.003 < 0.002 < 0.001 < 0.002	mg/l mg/l mg/l mg/l	EPA 204.2 EPA 206.2 EPA 239.2 EPA 270.2	03/02/95 03/02/95 03/01/95 03/01/95	NS NS NS NS
7	Allium, GFAA	0.025 < 0.0002 < 50 < 50	mg/l mg/l ug/l	EPA 279.2 EPA 245.2 EPA 8260	03/02/95 02/28/95 03/06/95	JB SD NS
	Acrylonitrile Benzene Bromobenzene Bromochloromethane	< 5.0 < 1.0 < 1.0 < 1.0	ug/l ug/l ug/l ug/l ug/l			
	Bromodichloromethane Bromomethane Bromomethane 2-Butanone	< 1.0 < 1.0 < 1.0 < 50	ug/l ug/l ug/l ug/l ug/l			
	Sec-Butylbenzene Sec-Butylbenzene tert-Butylbenzene Carbon Disulfide	< 1.0 < 1.0 < 1.0	ug/l ug/l ug/l ug/l ug/l			
	Carbon Tetrachloride Chlorobenzene Chloroform Chloromethane	< 1.0 < 1.0 < 1.0	ug/l ug/l ug/l ug/l			
1	2-Chlorotoluene 4-Chlorotoluene Dibromochloromethane ,2-Dibromoethane	< 1.0 < 1.0 < 1.0 < 1.0	ug/l ug/l ug/l ug/l			
1	)ibromomethane	< 1.0	ug/l			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 14

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QA/QC KN

REPORT NUMBER: 38436 SAMPLE ID: 5022509 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB5-2

DATE SAMPLED....: 02/20/95 TIME SAMPLED....: 14:05 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

		ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
	ANALYSES PERFORMED	RESULT	MEASURE	· ·	METHOD	DATE	TECHN
T						000000000000000000000000000000000000000	****************
į	Í	1	1	<b>f</b>	[	[	
)	1,2-Dibromo-3-chloropropane	< 1.0	ug/l				
	1,2-Dichlorobenzene	< 1.0	ug/l	1		[	
_	1,3-Dichlorobenzene	< 1.0	ug/l			Ì	
ľ	1,4-Dichlorobenzene	< 1.0	ug/l			1	
	Dichlorodifluoromethane	< 1.0	ug/l			ł	
•	1,1-Dichloroethane 1,2-Dichloroethane	< 1.0 < 1.0	ug/L	İ		l	
	1.1-Dichloroethene	< 1.0	ug/l	1	·	i	
	Cis-1,2 Dichloroethylene	1 < 1.0	ug/l ug/l				
1	Trans-1,2 Dichloroethylene	< 1.0	ug/t	İ		•	
ا د	1.2-Dichloroethene (total)	< 2.0	ug/t				
	Dichloromethane	< 1.0	ug/l				
•	1,2-Dichloropropane	< 1.0	ug/l	)			
ß	1,3-Dichloropropane	< 1.0	ug/l				
	2,2-Dichloropropane	< 1.0	ug/l	<u> </u>			
- 1	1,1-Dichloropropene	< 1.0	ug/l				
_ [	Cis-1,3-Dichloropropene	< 1.0	ug/l				
	~ans-1,3-Dichloropropene	< 1.0	ug/t				
18,	hylbenzene	< 1.0	ug/l	1			
*	méxach lorobutadiene	< 2.0	ug/l				
ı	2-Hexanone	< 50	ug/l	[ ]	i		
	Isopropyltaluene	< 1.0	ug/l	ł l			
	4-Isopropyltoluene *	< 1.0	ug/l	1			
	4-Methyl-2-Pentanone Naphthalene	< 50 < 5.0	ug/l	i i			
	n-Propylbenzene	< 1.0	ug/l	1			
	Styrene	< 1.0	ug/l ug/l	[			
	1,1,1,2-Tetrachloroethane	< 1.0	ug/t ug/t				
	1,1,2,2-Tetrachloroethane	< 1.0	ug/l	ļ			
	Tetrachloroethylene	< 1.0	ug/l	1			
	Toluene	< 1.0	ug/l				
H	1,2,3-Trichlorobenzene	< 1.0	ug/l	1	1		
И	1,2,4-Trichlorobenzene	< 1.0	ug/l		ł		
-1	1,1,1-Trichloroethane	< 1.0	ug/l				
_ [	1,1,2-Trichloroethane	< 1.0	ug/l		1		
	Trichloroethene	< 1.0	ug/t				
	Trichlorofluoromethane	< 1.0	ug/l	Ì	]		
•	1,2,3-Trichloropropane	< 1.0	ug/l		1		
1.	1,2,4-Trimethylbenzene	< 1.0	ug/l		ł		
1 .	1,3,5-Trimethlybenzene Vinyl Acetate	< 1.0	ug/l	I			
١,	Vinyl Chloride		ug/l		Ţ		1
	4P-Xylene		ug/l	İ	1		Į
	D-Xylene		ug/l		l		
ıl`	- ··•,··•	` '	ug/l		İ		

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 15

Canton Analytical Laboratory, Inc.

OA/OC KN

SAMPLE ID: 5022509 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB5-2 REPORT NUMBER: 38436

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 14:05

DATE RECEIVED ....: 02/27/95

TIME RECEIVED....: 10:40

1				 		
}	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE	ANALYTICAL METHOD	RUN DATE	TECHN
	Xylenes, Total Chloroethane	< 3.0 < 1.0	ug/l ug/l			
_  *+						
	-					

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

PAGE 16

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REPORT NUMBER: 38436 SAMPLE ID: 5022520 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB6-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED .....: 13:00

DATE RECEIVED....: 02/27/95

TIME RECEIVED ....: 10:40

QA/QC KV

}		ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
	ANALYSES PERFORMED	RESULT	MEASURE		METHOD	DATE	TECHN
				T .			************
	DISSOLVED METALS			1			
}	TAL METALS						
	Aluminum	< 0.14	mg/l		EPA 200.7	02/28/95	SD
į	Barium	0.16	mg/l		EPA 200.7	02/28/95	SD
	Beryilium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
)	Cadmium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
	Calcium	130	mg/l		EPA 200.7	03/03/95	SD
	Chromium	< 0.02	mg/l		EPA 200.7	02/28/95	SD
ŀ	Cobalt	0.02	mg/l		EPA 200.7	02/28/95	SD
	Silver	< 0.03	mg/t		EPA 200.7	02/28/95	SD
ʻi	Copper	< 0.02	mg/L		EPA 200.7	02/28/95	SD
į	Sodium	80	mg/l		EPA 200.7	03/03/95	SD
,	Vanadium	< 0.02	mg/l		EPA 200.7	02/28/95	SD
	Zinc	< 0.02	mg/l		EPA 200.7	02/28/95	SD
IJ	Antimony, GFAA	< 0.003	mg/l		EPA 204.2	03/02/95	NS
ı	Arsenic, GFAA	< 0.002	mg/l	İ	EPA 206.2	03/02/95	NS
ل	Lead, GFAA	< 0.001	mg/t		EPA 239.2	03/01/95	NS
	i≒elenium, GFAA	< 0.002	mg/l		EPA 270.2	03/01/95	NS
	∌allium, GFAA	0.01	mg/l		EPA 279.2	03/02/95	NS
۳,	⊣mercury, Cold Vapor	< 0.0002	mg/l		EPA 245.2	02/28/95	SD
- 1	TCL VOLATILE ORGANICS, GC/MS			i	EPA 8260	03/07/95	CB
ıİ	Acetone	< 50	ug/l		į		
П	Acrolein	< 50	ug/l				
IJ	Acrylonitrile	< 5.0	ug/l		İ		
	Benzene	< 1.0	ug/l				
Л	Bromobenzene	< 1.0	ug/l	ļ			
[ ]	Bromochloromethane	< 1.0	ug/l				
	Bromodichloromethane	< 1.0	ug/l	ĺ			
	Bromoform	< 1.0	ug/l	ļ			
	Bromomethane		ug/l	ł	1		
	2-Butanone		ug/l	j	1		
	n-Butylbenzene Sec-Butylbenzene	< 1.0	ug/l	İ	ļ		
	tert-Butylbenzene	< 1.0	ug/l		ĺ		
	Carbon Disulfide	< 1.0 < 50	ug/l	j	Ì		
	Carbon Jisutfide		ug/l	ŀ			
	Chlorobenzene		ug/l ug/l	ļ	ļ		
	Chloroform		ug/l	ļ	1		
•			ug/l	ľ	ļ		
	2-Chlorotoluene		ug/l				
	4-Chlorotoluene		ug/l	ļ			
	Dibromochloromethane		ug/l		1		
1	1,2-Dibromoethane		ug/l				
			ug/l				
1				}			i
1	i			1			

<sup>\*</sup> Tentative identification and quantitation based on a computer generated Library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/8/95.

PAGE 44

REPORT NUMBER: 38436 SAMPLE ID: 5022520 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB6-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED.....: 13:00 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

ANALYSES PERFORMED	ANALYTICAL RESULT	. UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
1 2 0/h 7 -h		4				
1,2-Dibromo-3-chloropropane	< 1.0	ug/L		1		
1,3-Dichlorobenzene	< 1.0	ug/t				
1,4-Dichlorobenzene	< 1.0 < 1.0	ug/L			1	
Dichlorodifluoromethane	4.2	ug/t		}	Ì	
1.1-Dichloroethane	130	lug/l		İ	l	
1,2-Dichloroethane	1.0	ug/l	1			
1,1-Dichloroethene	27	lug/l			ł	
Cis-1,2 Dichloroethylene	210	ug/l		ļ	Ì	
Trans-1,2 Dichloroethylene	14	ug/l	1	ĺ	1	
1,2-Dichloroethene (total)	230	ug/l ug/l	ľ			
Dichloromethane	< 1.0	ug/l				
1,2-Dichloropropane	< 1.0	1	1			
1,3-Dichloropropane	< 1.0	lug/l lug/l				
2,2-Dichloropropane	1 < 1.0	ug/t ug/t				
1,1-Dichtoropropene	< 1.0	ug/l				
Cis-1,3-Dichloropropene	< 1.0	ug/t				
Trans-1,3-Dichloropropene	< 1.0	ug/l				
thylbenzene	< 1.0	lug/l				
	< 2.0	ug/l		1		
2-Hexanone	< 50	ug/l				
Isopropyltoluene	< 1.0	ug/t				
4-Isopropyltoluene *	< 1.0	ug/l				
4-Methyl-2-Pentanone	< 50	ug/l				
Naphthalene	< 5.0	ug/t				
n-Propylbenzene	< 1.0	ug/l				
Styrene	< 1.0	ug/l	!			
1,1,1,2-Tetrachloroethane	< 1.0	ug/l	i !			
1,1,2,2-Tetrachloroethane	< 1.0	ug/l				
Tetrachloroethylene	1.2	ug/l				
Toluene	< 1.0	ug/l				
1,2,3-Trichlorobenzene	< 1.0	ug/i		İ		
1,2,4-Trichlorobenzene	₹ 1.0	ug/l	i			
1,1,1-Trichloroethane	84	ug/l		İ		
1,1,2-Trichloroethane	< 1.0	ug/l	]	Ì		
Trichloroethene	760 **	ug/l	]			
Trichlorofluoromethane	< 1.0	ug/t	1	ĺ		
1,2,3-Trichloropropane	< 1.0	ug/l		į		
1,2,4-Trimethylbenzene	< 1.0	ug/l		İ		
1,3,5-Trimethlybenzene	< 1.0	ug/l	]			
Vinyl Acetate	< 100	ug/l	1	ļ		
Vinyl Chloride	130	ug/t	į	ľ		
MP-Xylene	< 2.0	ug/l				
0-Xylene	< 1.0	ug/l				
•	' ''	~8/ ·				
	1	i				

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/8/95.

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PAGE 45

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REPORT NUMBER: 38436 SAMPLE ID: 5022520 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWB6-2

DATE SAMPLED.....: 02/21/95 TIME SAMPLED....: 13:00 DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

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	ANALYTICAL	UNITS OF		ANALYTICAL	RUN	
EDECIDMEN	DECHIT	MEACHIDE		METHOD	DATE	TECHN

	ANALYSES PERFORMED	RESULT	MEASURE	METHOD	DATE	TECHN
ſ	Xylenes, Total Chloroethane	< 3.0 < 1.0	ug/l ug/l			
·I	)					
			3			
L						

<sup>\*</sup> Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260 \*\* Re-analysis and quantitation performed 3/8/95.

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PAGE 46

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DA/OC\_KW

CLIENT: CLEAN TECH

REPORT NUMBER: 38436 SAMPLE ID: 5022504 DESCRIPTION: CHRYSLER ACUSTAR 95-1-MWC1-2

DATE SAMPLED.....: 02/20/95 TIME SAMPLED.....: 09:15

DATE RECEIVED....: 02/27/95 TIME RECEIVED....: 10:40

]. ]	ANALYSES PERFORMED	ANALYTICAL RESULT	UNITS OF MEASURE		ANALYTICAL METHOD	RUN DATE	TECHN
	DISSOLVED METALS TAL METALS						
	Aluminum	0.14	mg/l		EPA 200.7	02/28/95	SD
_	Barium	0.21	mg/l		EPA 200.7	02/28/95	SD
	Beryllium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
٥	Cadmium	< 0.01	mg/l		EPA 200.7	02/28/95	SD
_	Calcium	44	mg/l		EPA 200.7	03/03/95	SD
	Chromium	< 0.02	mg/l		EPA 200.7	02/28/95	SD
	Cobalt	0.033	mg/l		EPA 200.7	02/28/95	SD
	Silver	0.046	mg/L		EPA 200.7	02/28/95	SD
	Copper	< 0.02	mg/L		EPA 200.7	02/28/95	SO
	Sodium	7.9	mg/l		EPA 200.7	03/03/95	SD .
2	Vanadium	< 0.02	mg/L	İ	EPA 200.7	02/28/95	SO
	Zinc	< 0.02	mg/L		EPA 200.7	02/28/95	SO
	Antimony, GFAA	0.0056	mg/L	1	EPA 204.2	03/02/95	NS NS
	Arsenic, GFAA Lead, GFAA	0.01	ing/L	1	EPA 206.2 EPA 239.2	03/02/95	NS NS
<del></del>	Ledu, uran	< 0.002	mg/l mg/l	1	EPA 270.2	03/01/95 03/01/95	NS
	Relenium, GFAA )llium, GFAA	0.0045	mg/L	1	EPA 270.2	03/02/95	NS
kir.	rcury, Cold Vapor	< 0.0002	mg/l	1	EPA 245.2	02/28/95	SD
	TCL VOLATILE ORGANICS, GC/MS	1 0.000	11197	1	EPA 8260	03/06/95	JB
	Acetone	< 50	ug/l	1	LIA OLOO	03,00,75	
	Acrolein	< 50	ug/l			1	
	Acrylonitrile	< 5.0	ug/t			ļ	I
	Benzene	< 1.0	ug/l				
	Bromobenzene	< 1.0	ug/l	1			
4	Bromochloromethane	< 1.0	ug/l	]		•	1
1	Bromodichloromethane	< 1.0	ug/l				1
	Bromoform	< 1.0	ug/l				
	Bromomethane	< 1.0	ug/l				I
_	2-Butanone	< 50	ug/l	1			ľ
	n-Butylbenzene	< 1.0	ug/l	1			l l
	Sec-Butylbenzene	< 1.0	ug/l	]			1
	tert-Butylbenzene	< 1.0	ug/l	<b>\</b>			Į.
_	Carbon Disulfide Carbon Tetrachloride	< 50	ug/t	1			İ
M	Chlorobenzene	< 1.0 < 1.0	ug/l	1			į
	Chloroform	< 1.0	ug/l ug/l	1 1			
$\neg$	Chloromethane	< 1.0	ug/l	1			1
	2-Chlorotoluene	< 1.0	ug/i	1			- 1
A.	4-Chlorotoluene	< 1.0	ug/l	Į Į			Į
	Dibromochloromethane	< 1.0	ug/l	1			į.
۳	1,2-Dibromoethane	< 1.0	ug/l				- 1
-	Dibromomethane	< 1.0	ug/l	1			ŀ
		'''			Į		į
		ļ					}
	<del></del>	L	L	1 .l.			

\* Tentative identification and quantitation based on a computer generated library search per section 7.6.1.2, EPA Method 8260

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PAGE 1

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